

Photodecomposition of *N*-hydroxyurea in argon matrices. FTIR and theoretical studies

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Supplementary information: additional computational details

Figure S1 The optimized structures of the HNCO-NH₂OH complexes at the MP2/6-311++G(2d,2p) level of theory. The selected bond distances (in Å) and interaction energies ΔE^{CP} (in kJ mol⁻¹) are also presented; in parentheses the $\Delta E_{\text{ZPE}}^{\text{CP}}$ values (in kJ mol⁻¹) are given.

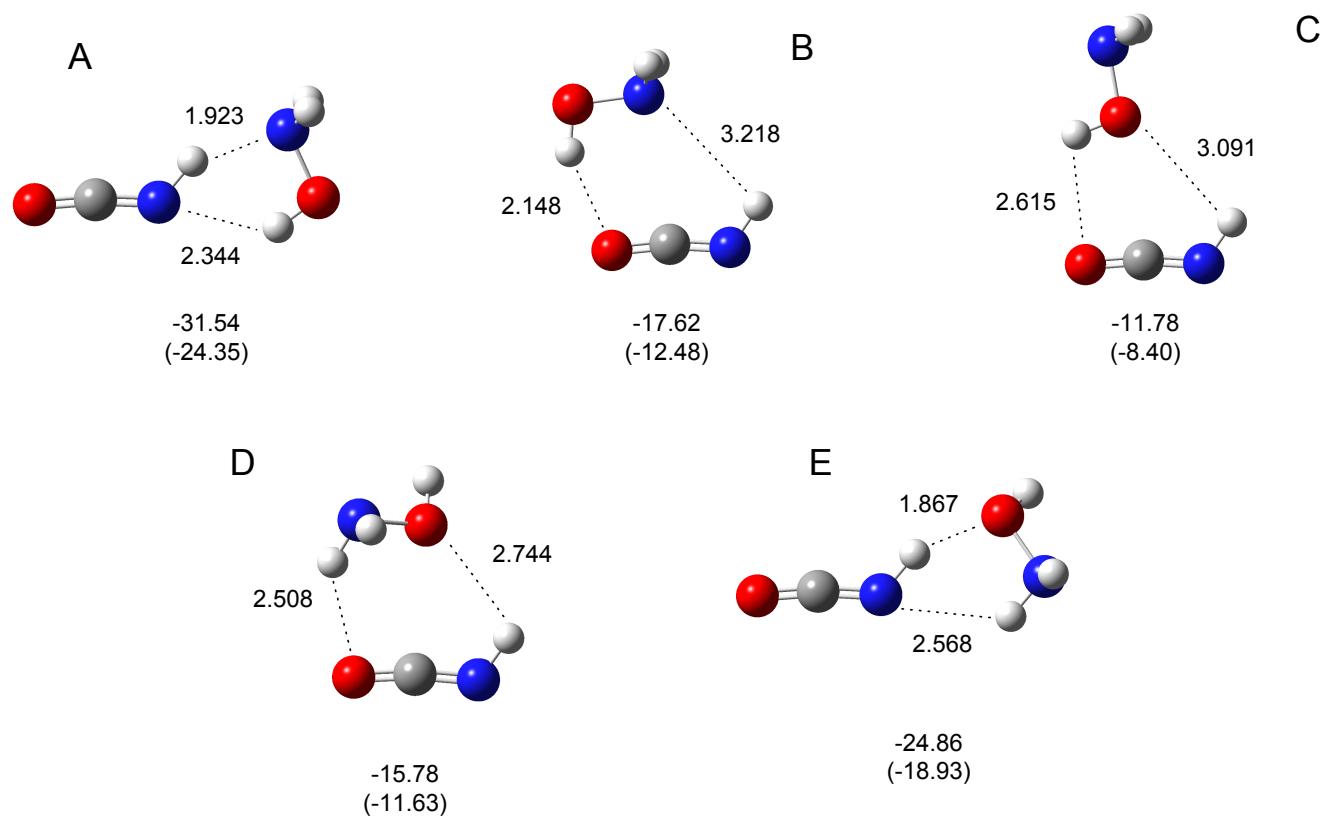


Table S1 Selected structural parameters^a calculated for the HNCO-NH₂OH complexes at the MP2/6-311++G(2d,2p) level of theory

Property	A	B	C	D	E
r(NH(C))	1.0241	1.0045	1.0050	1.0061	1.0150 (1.0192) ^b
r(NC)	1.2166	1.2185	1.2198	1.2198	1.2172 (1.2166)
r(CO)	1.1757	1.1760	1.1745	1.1751	1.1754 (1.1761)
r(OH)	0.9641	0.9621	0.9604	0.9594	0.9600 (0.9632)
r(NO)	1.4407	1.4454	1.4478	1.4513	1.4518 (1.4467)
r(NH _a)	1.0123	1.0128	1.0122	1.0132	1.0143 (1.0177)
r(NH _b)	1.0123	1.0128	1.0122	1.0122	1.0123 (1.0153)
r((N)H···N), r((N)H···O)	1.9232	3.2185	3.0909	2.7444	1.9242 (1.8667)
r((O)H···N), r((O)H···O),	2.3446	2.1479	2.6151		
r((N)H _a ···O), r((N)H _a ···N)				2.5078	2.6362 (2.5684)
ϕ(HNC)	127.17	121.71	121.52	120.39	126.47 (126.90)
ϕ(OCN)	173.09	171.14	172.15	172.27	172.57 (172.58)
ϕ(NOH)	102.64	102.00	102.15	101.62	102.02 (102.14)
ϕ(H _a NO)	104.33	103.64	103.81	103.68	103.83 (103.75)
ϕ(NH···N), ϕ(NH···O)	147.88	99.60	98.45	109.51	152.93 (153.08)
ϕ(OH···N), ϕ(OH···O)	131.08	152.62	107.92		
ϕ(NH _a ···O), ϕ(NH _a ···N)				140.43	121.11 (121.09)
θ(NH···NH _a), θ(NH···OH)	-110.57	123.58	0.01	-124.78	117.18 (116.27)
θ(OH···NH), θ(OH···OC)	0.00	0.02	0.00		
θ(NH _a ···OC), θ(NH _a ···NH)				7.48	9.08 (8.11)

^a the interacting atoms refer to the Figure S1; the bond distances are given in Å, the angles in degrees.

^b the calculation performed at the aug-cc-pVTZ basis set.

Table S2 Wavenumbers and wavenumbers shifts (in cm^{-1}) calculated for the HNCO-NH₂OH complexes at the MP2/6-311++G(2d,2p) level of theory

	Monomer	A	B	C	D	E	Assignment
	v	v	Δv	v	Δv	v	Δv
HNCO	3722 (172) ^a	3324 (1218)	-398	3711 (152)	-11	3706 (160)	-16
2308 (668)	2311 (713)	3	2302 (636)	-6	2305 (633)	-2	2301 (613) -7
1289 (0,1)	1292 (9)	3	1289 (1)	0	1287 (0,3)	-2	1287 (0) -2
813 (238)	972 (398)	159	820 (291)	7	831 (270)	18	842 (270) 29
630 (0,3)	742 (86)	112	629 (2)	-1	628 (2)	-2	631 (0) 1
575 (75)	656 (10)	81	559 (76)	-16	564 (77)	-11	560 (80) -15
NH ₂ OH	3872 (53)	3795 (134)	-77	3828 (139)	-56	3858 (63)	-14
3606 (6)	3606 (15)	0	3595 (8)	-11	3607 (7)	1	3604 (14) 2
3517 (0,6)	3518 (0)	1	3509 (0,5)	-8	3516 (0)	-1	3514 (3) -3
1683 (17)	1682 (22)	-1	1684 (20)	1	1683 (22)	0	1688 (15) 5
1415 (27)	1478 (33)	63	1455 (46)	40	1417 (42)	2	1411 (25) -4
1336 (0)	1331 (0)	-5	1335 (0)	-1	1338 (0)	2	1349 (3) 13
1164 (127)	1188 (154)	24	1184 (127)	20	1165 (149)	1	1175 (145) 11
929 (11)	937 (172)	8	937 (8)	8	933 (11)	4	927 (10) -2
434 (178)	404 (29)	-30	507 (165)	73	466 (170)	32	447 (179) 13
	575 (113)	-	-	-	-	-	-
-	245 (1)	-	245 (72)	-	207 (104)	-	280 (103) -
-	234 (33)	-	170 (19)	-	98 (8)	-	127 (1) -
-	124 (0)	-	156 (3)	-	94 (0)	-	109 (9) -
-	68 (2)	-	110 (2)	-	80 (0)	-	98 (0) -
-	14 (0)	-	83 (5)	-	40 (5)	-	58 (1) -
-	-	-	40 (4)	-	32 (0)	-	18 (1) 16 (0)

^a The IR calculated intensities expressed in km mol^{-1} .

Table S3 Selected structural parameters^a calculated for the N₂-H₂O-CO complexes at the MP2/aug-cc-pVTZ level of theory

Property	A	B	C
r(NN)	1.1141	1.1138	1.1138
r(OH(···N))	0.9612	0.9617	0.9629
r(OH(···C))	0.9641	0.9630	0.9613
r(CO)	1.1375	1.1376	1.1384
r((O)H···N)		2.3937	2.3502
r((O)H···C)	2.3645	2.3956	
r((H ₂)O···C)			3.0898
r((H ₂)O···N)	3.1365		
ϕ(HOH)	104.37	104.56	104.38
ϕ(OH···N)		160.34	165.96
ϕ(OH···C)	163.89	166.32	
ϕ(NN···H)		168.45	166.01
ϕ(OC···H)	167.26	171.73	
ϕ(NN···O(H ₂))	97.79		
ϕ(OC···O(H ₂))			101.82
θ(NN···HO)		-0.03	-0.01
θ(OH···CO)	-0.01	0.00	
θ(OC···OH)			0.30
θ(NN···OH)	-0.10		

^a the interacting atoms refer to the Figure 6; the bond distances are given in Å, the angles in degrees.

Table S4 Wavenumbers (in cm^{-1}) of the $\text{N}_2\text{-H}_2\text{O}\text{-CO}$ complexes calculated by the MP2/aug-cc-pVTZ method

A	B	C	Assignmt
3926 (141) ^a	3925 (254)	3935 (134)	$\nu_{\text{as}}\text{H}_2\text{O}$
3794 (69)	3804 (39)	3810 (34)	$\nu_{\text{s}}\text{H}_2\text{O}$
1636 (58)	1638 (36)	1632 (71)	$\delta\text{H}_2\text{O}$
2123 (31)	2121 (33)	2115 (35)	νCO
2186 (0)	2189 (0.2)	2189 (0.3)	νN_2
352 (60)	340 (40)	286 (64)	intermolecular
226 (83)	259 (102)	188 (82)	Intermolecular
108 (5)	244 (114)	102 (3)	Intermolecular
103 (114)	103 (0.7)	96 (119)	Intermolecular
86 (7)	88 (0.8)	82 (1)	Intermolecular
70 (0)	59 (1)	68 (2)	Intermolecular
57 (16)	54 (2)	51 (10)	Intermolecular
49 (1)	44 (6)	50 (16)	Intermolecular
32 (0)	32 (13)	38 (0.7)	Intermolecular
30 (0)	10 (0)	31 (0.4)	intermolecular

^a The IR calculated intensities expressed in km mol^{-1} .