

Electronic Supplementary Material

**Synthesis of non-peripheral amine substituted nickel(II) phthalocyanine capped gold nanoparticles and their immobilization on electrode for the electrocatalytic oxidation of hydrazine**

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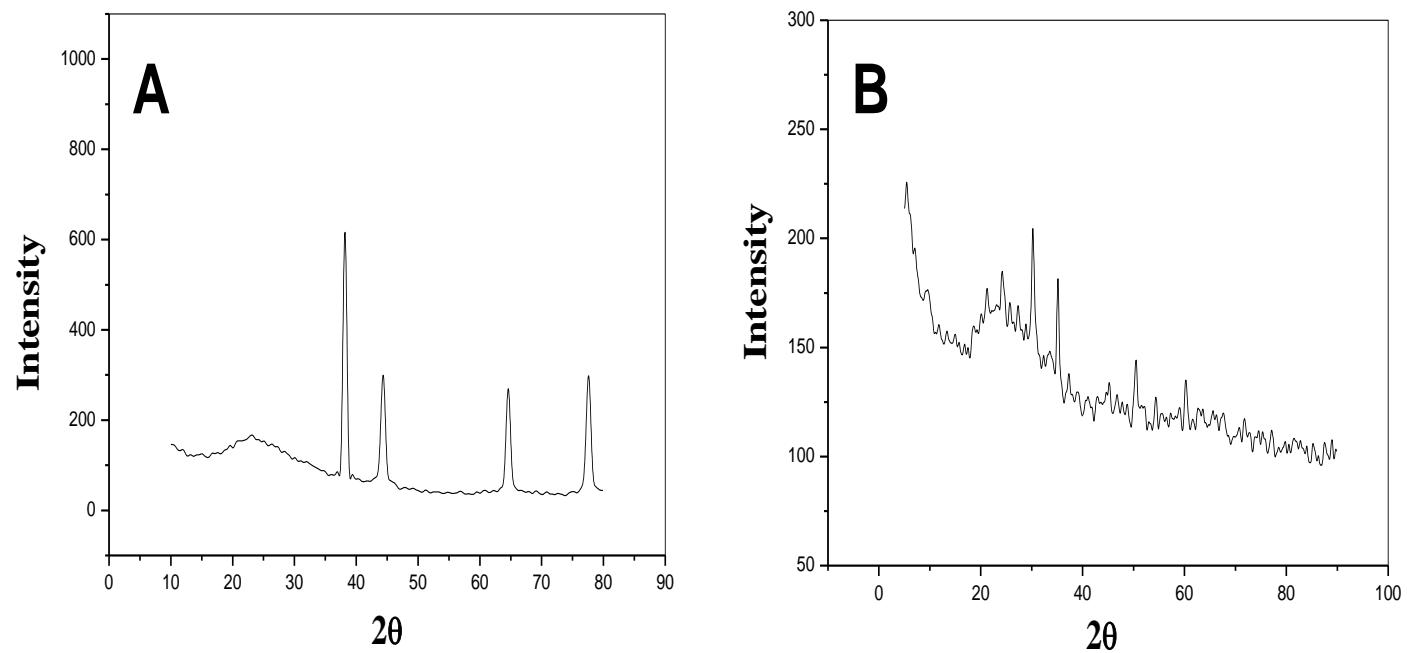


Fig. S1. X-ray diffraction pattern obtained for (A)  $4\alpha$ -Ni<sup>II</sup>TAPc-AuNPs and (B) ITO/MPTS/ $4\alpha$ -Ni<sup>II</sup>TAPc-AuNPs.

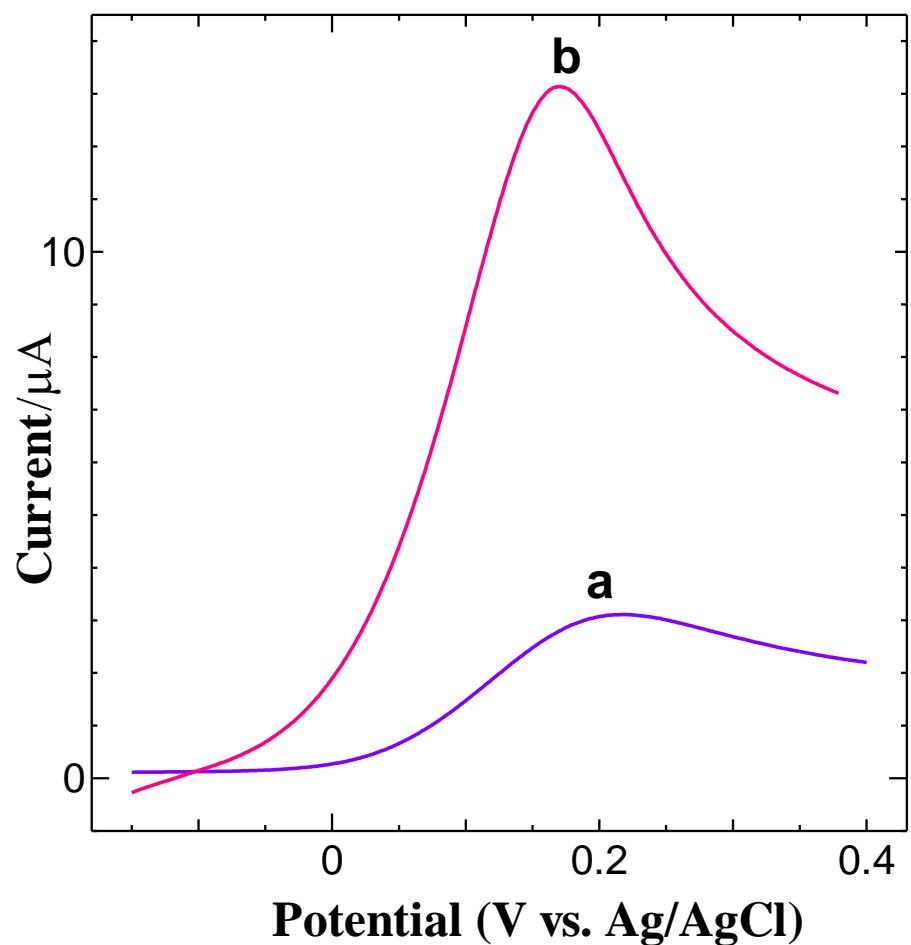
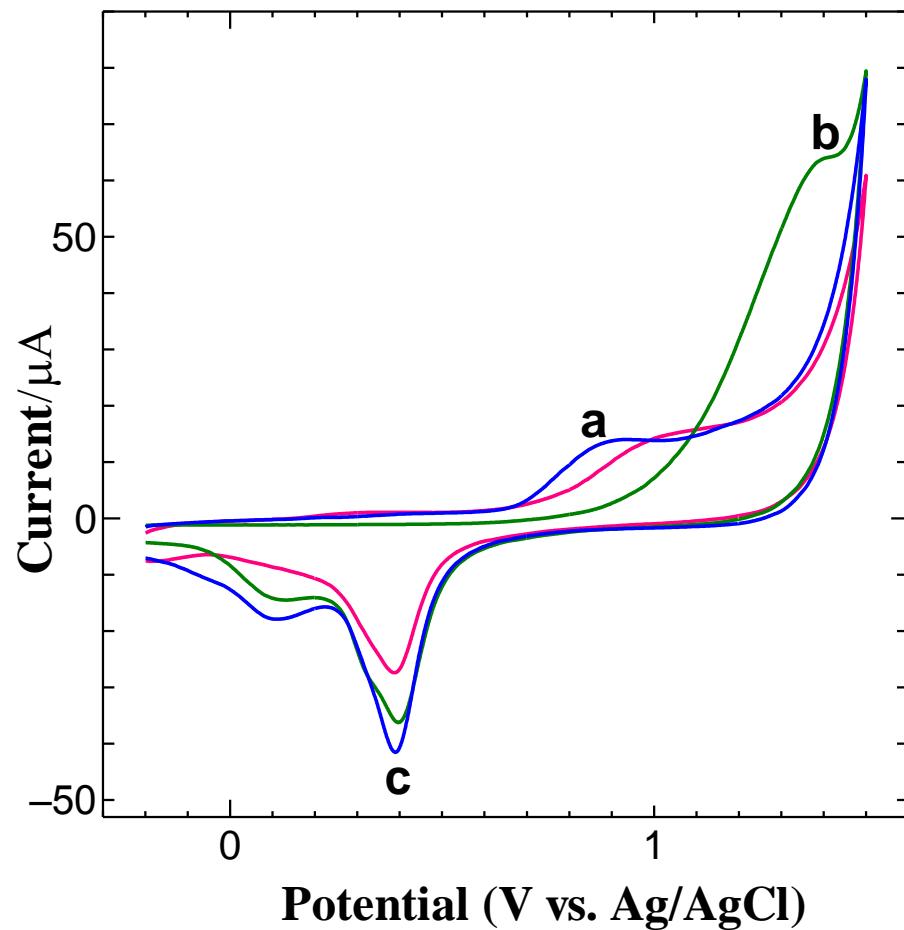
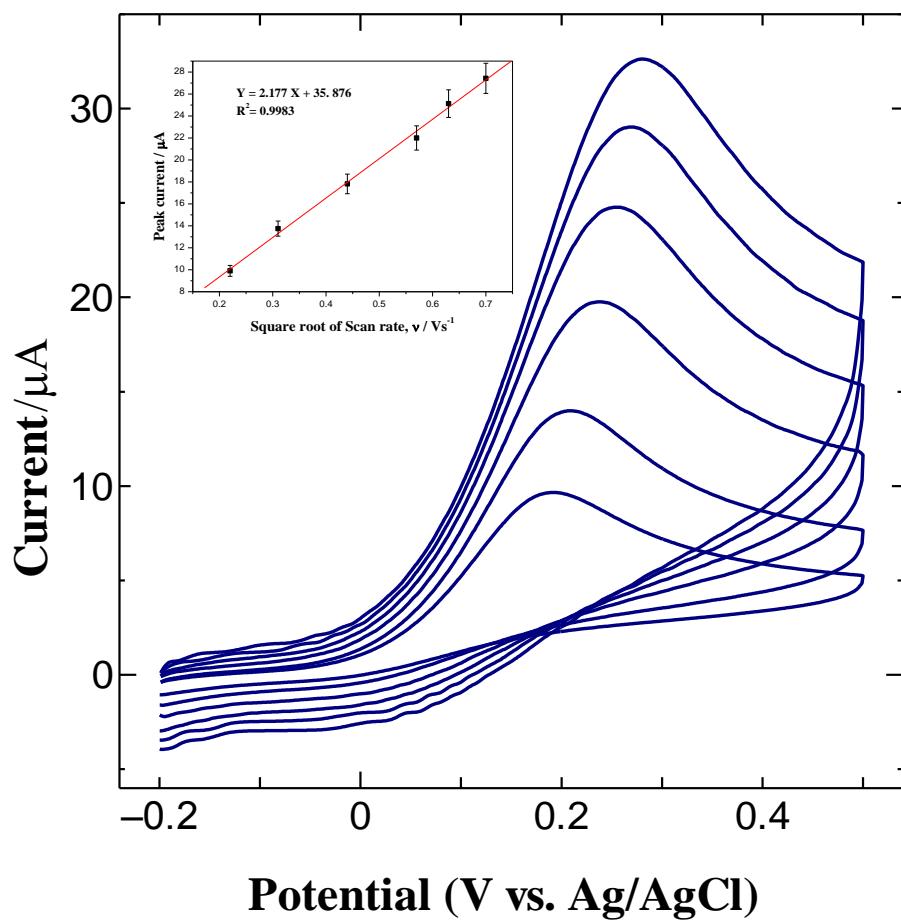


Fig. S2. LSVs obtained for 0.5 mM hydrazine at (a) Au/HDT/C-AuNPs and (b) Au/HDT/Ni<sup>II</sup>TAPc-AuNPs modified electrodes in 0.2 M PB solution (pH 7.2) at a scan rate of 0.05 V s<sup>-1</sup>.



**Fig. S3.** CVs obtained for (a) bare Au, (b) Au/HDT/C-AuNPs and (c) Au/HDT/4 $\alpha$ -Ni<sup>III</sup>TAPc-AuNPs modified electrodes in 0.2 M PB solution (pH 7.2) at a scan rate of 0.05 V s<sup>-1</sup>.



**Fig. S4.** CVs obtained for 0.5 mM hydrazine at Au/HDT/4 $\alpha$ -Ni<sup>II</sup>TAPc-AuNPs modified Au electrode in 0.2 M PB solution (pH 7.2) at scan rates of 0.05, 0.1, 0.2, 0.3, 0.4 and 0.5  $\text{V s}^{-1}$ .

Table S1

Table for impedance data

Parameter	Bare Au	Au/HDT	Au/HDT /4α-Ni <sup>II</sup> TAPc-AuNPs
R <sub>s</sub> (kΩ)	0.211	0.120	0.110
CPE (F)	$4.288 \times 10^{-6}$	$7.092 \times 10^{-7}$	$5.868 \times 10^{-7}$
R <sub>CT</sub> (kΩ)	12.93	103.67	17.55
k <sub>et</sub> (cm s <sup>-1</sup> )	$6.628 \times 10^{-4}$	$8.278 \times 10^{-5}$	$4.889 \times 10^{-4}$