ELECTRONIC SUPPLEMENTARY INFORMATION FOR

## New nanocomposite proton conducting membranes based on a core-shell nanofiller for low relative humidity fuel cells

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**Table S1.** Phases contained within the ZrTa filler nanoparticles determined by analyzing the electron diffraction pattern shown in Figure 1(c).

Interplanar Distance (Å)					
Measured Values	Literature Values <sup>1</sup>	Component	Lattice	Space Group	Miller Index
3.79	3.77	Ta <sub>2</sub> O <sub>5</sub>	Face-centered Monoclinic	F	111
3.15	3.16	m-ZrO <sub>2</sub>	Primitive Monoclinic	$P2_{1}/c$ (14)	-1 1 1
3.04	3.04	$Ta_2O_5$	Face-centered Monoclinic	F	117
3.00	3.00	$Ta_2O_5$	Face-centered Monoclinic	F	0 0 12
2.95	2.95	t-ZrO <sub>2</sub>	Primitive Tetragonal	P4 <sub>2</sub> /nmc (137)	101
2.64	2.62	m-ZrO <sub>2</sub>	Primitive Monoclinic	$P2_{1}/c$ (14)	002
2.24	2.25	$Ta_2O_5$	Face-centered Monoclinic	F	0 0 16
1.85	1.81	t-ZrO <sub>2</sub>	Primitive Tetragonal	P4 <sub>2</sub> /nmc (137)	112
1.57	1.55, 1.54	t-ZrO <sub>2</sub>	Primitive Tetragonal	P4 <sub>2</sub> /nmc (137)	211,103







Figure S1. EDX spectra of different ZrTa filler nanoparticles.



**Figure S2.** Polarization and power curves to illustration the single fuel cell performance of the MEAs. The oxidant is air, and the back pressure is 1 bar. The membrane thickness is ca. 110 and 140  $\mu$ m for Nafion and [Nafion/(ZrTa) $_{\Psi}$ ], respectively. The data are not corrected for IR losses.



**Figure S3.** Polarization and power curves to illustration the single fuel cell performance of the MEAs. The oxidant is pure oxygen, and the back pressure is 4 bar. The membrane thickness is ca. 110 and 140  $\mu$ m for Nafion and [Nafion/(ZrTa) $_{\Psi}$ ], respectively. The data are not corrected for IR losses.



**Figure S4.** Polarization and power curves to illustration the single fuel cell performance of the MEAs. The oxidant is air, and the back pressure is 4 bar. The membrane thickness is ca. 110 and 140  $\mu$ m for Nafion and [Nafion/(ZrTa) $_{\Psi}$ ], respectively. The data are not corrected for IR losses.



**Figure S5.** Dependence of the MEA conductivity on  $a_{H_2O}$  for Nafion and [Nafion/(ZrTa)<sub> $\Psi$ </sub>]. The conductivities were determined from the polarization curves: a) at the cell potential difference corresponding to the maximum of the power curve; and b) at lower current densities where the contribution of self-humidification is reduced. The back pressure was 1 bar and the oxidant was oxygen. The lines are a guide for the eye. The membrane thickness is ca. 110 and 140 µm for Nafion and [Nafion/(ZrTa)<sub> $\Psi$ </sub>], respectively. The data are not corrected for IR losses.



**Figure S6**. Dependence on  $a_{H_2O}$  of the current density values at a cell potential difference of 0.6 V. The back pressure was either 4 or 1 bar. The oxidant was either air or oxygen. The lines are meant as a guide to the eye. The membrane thickness is ca. 110 and 140 µm for Nafion and [Nafion/(ZrTa)<sub>Ψ</sub>], respectively. The data are not corrected for IR losses.

## REFERENCES

1. Powder Diffraction File, JCPDS International Centre for Diffraction Data, Swarthmore PA, PDF no. 33-1391 (Ta<sub>2</sub>O<sub>5</sub>), 86-1450 (*m*-ZrO<sub>2</sub>), 88-1007 (*t*-ZrO<sub>2</sub>).