Electronic Supplementary Information

Ag/TiO₂ and Ag/SiO₂ Composite Spheres: Synthesis, Characterization and Antibacterial Properties

Qiushuang Zhang^a, Junwei Ye^{*, a}, Peng Tian^a, Xinyi Lu^a, Yuan Lin^a, Qi Zhao^{*, b}, Guiling Ning^{*, a}

^aState Key Laboratory of Fine Chemicals and School of Chemical Engineering, Dalian University of Technology, 2 Linggong Road, Dalian 116012, P. R. China. E-mail: junweiye@dlut.edu.cn; ninggl@dlut.edu.cn; Fax: +86-411-84986067.

^bDepartment of Mechanical Engineering, University of Dundee, DD1 4HN, UK. E-mail: Q.Zhao@dundee.ac.uk

CAPTIONS:

- Fig. S1 SEM image of Ag/SiO₂ composite.
- **Fig. S2** XRD pattern of Ag/SiO₂ composite (the enlarged view of 2θ range from 5° to 35° in inset).
- **Fig. S3** XRD pattern of TiO₂ hollow spheres after calcined at 500 $^{\circ}$ C.
- **Fig. S4** EDX image of Ag/TiO₂ composite.

Fig. S5 TG curve of $Ag_2CO_3/TiO_2 \cdot xH_2O$.

- **Fig. S6** FT-IR spectra of (a) CaMg(CO₃)₂, (b) CaMg(CO₃)₂/TiO₂·xH₂O, (c) Ag₂CO₃/TiO₂·xH₂O and (d) Ag/TiO₂ composite.
- Fig. S7 XRD patterns of samples obtained with different concentration of AgNO₃ solution.
- Fig. S8 SEM images of samples obtained with different concentration of AgNO₃ solution: (a) 0.1 mol/L, (b) 0.3 mol/L and (c) 0.5 mol/L.

Fig. S9 EDX of Ag/TiO₂ composite obtained at different reaction time: (a) 4h and (b) 8 h.

- Fig. S10 XRD patterns of Ag/TiO₂ composite obtained at different calcined temperature: (a) 400 °C,
 (b) 500 °C and (c) 600 °C.
- Fig. S11 SEM images of Ag/TiO₂ composite obtained at different calcined temperature: (a) 400 °C,
 (b) 500 °C and (c) 600 °C.

Fig. S12 The relationship between antibacterial activity and concentration of Ag/TiO₂ composite.Fig. S13 SEM images of Ag-NPs, SiO₂ and TiO₂ hollow spheres.



Fig. S1 SEM image of Ag/SiO₂ composite



Fig. S2 XRD pattern of Ag/SiO₂ composite (the enlarged view of 2θ range from 5° to 35° in inset).



Fig. S3 XRD pattern of TiO₂ hollow spheres after calcined at 500 $^{\circ}$ C.



Fig. S4 EDX image of Ag/TiO₂ composite.



Fig. S5 TG curve of $Ag_2CO_3/TiO_2 \cdot xH_2O$.



Fig. S6 FT-IR spectra of (a) $CaMg(CO_3)_2$, (b) $CaMg(CO_3)_2/TiO_2 \cdot xH_2O$, (c) $Ag_2CO_3/TiO_2 \cdot xH_2O$ and (d) Ag/TiO_2 composite.



Fig. S7 XRD patterns of samples obtained with different concentration of AgNO₃ solution.



Fig. S8 SEM images of samples obtained with different concentration of AgNO₃ solution: (a) 0.1 mol/L, (b) 0.3 mol/L and (c) 0.5 mol/L.



Fig. S9 EDX of Ag/TiO₂ composite obtained at different reaction time: (a) 4h and (b) 8 h.



Fig. S10 XRD patterns of Ag/TiO₂ composite obtained at different calcined temperature: (a) 400 $^{\circ}$ C, (b) 500 $^{\circ}$ C and (c) 600 $^{\circ}$ C



Fig. S11 SEM images of Ag/TiO₂ composite obtained at different calcined temperature: (a) 400 $^{\circ}$ C, (b) 500 $^{\circ}$ C and (c) 600 $^{\circ}$ C.



Fig. S12 The relationship between antibacterial activity and concentration of Ag/TiO₂ composite.



Fig. S13 SEM images of Ag-NPs, SiO_2 and TiO_2 hollow spheres.