

Supporting Information

Poly(ionic liquid)-derived nitrogen-doped hollow carbon spheres: synthesis and loading with Fe₂O₃ for high-performance lithium ion batteries

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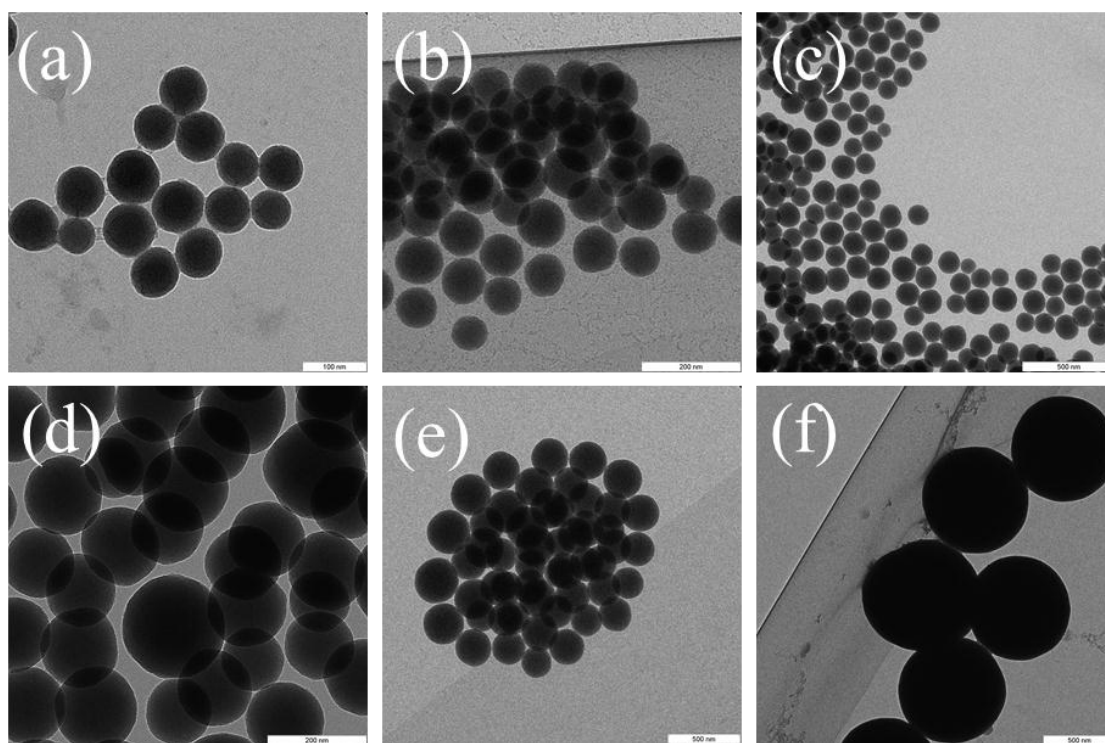


Fig. S1. TEM images of SiO₂ with different size of 70, 90, 130, 180, 250 and 790 nm in (a) to (f) respectively. The scale bars in (a) – (f) are 100, 200, 500, 200, 500 and 500 nm, respectively.

Table S1. Textural properties and nitrogen content of NHCSs.

Sample	AN (mol%)	CL (mol%)	SiO ₂ size (nm)		NHCS size (nm)		S _{BET} (m ² g ⁻¹)	N content wt%
			DLS	TEM	particle	wall		
			NHCS70	-	-	72		
NHCS90	-	-	89	93	82	2.0	512	7.4
NHCS130	-	-	129	127	121	3.7	571	8.1
NHCS180	-	-	180	174	172	2.1	254	8.0
NHCS250	-	-	256	249	237	2.7	206	8.2
NHCS790	-	-	792	770	780	9.2	19	9.3

AN: acrylonitrile; CL: crosslinking agent; DLS: dynamic light scattering; Particle sizes and wall-thickness of NHCS were estimated by TEM; S_{BET}, BET specific surface area; Nitrogen content was determined by elemental analysis.

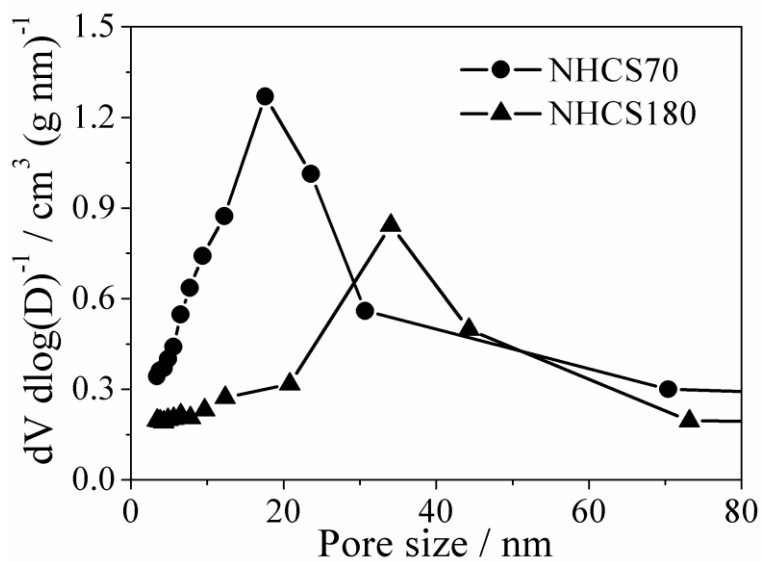


Fig. S2. Pore size distribution plots obtained from the adsorption branch of the isotherms for samples NHCS70 and NHCS180.

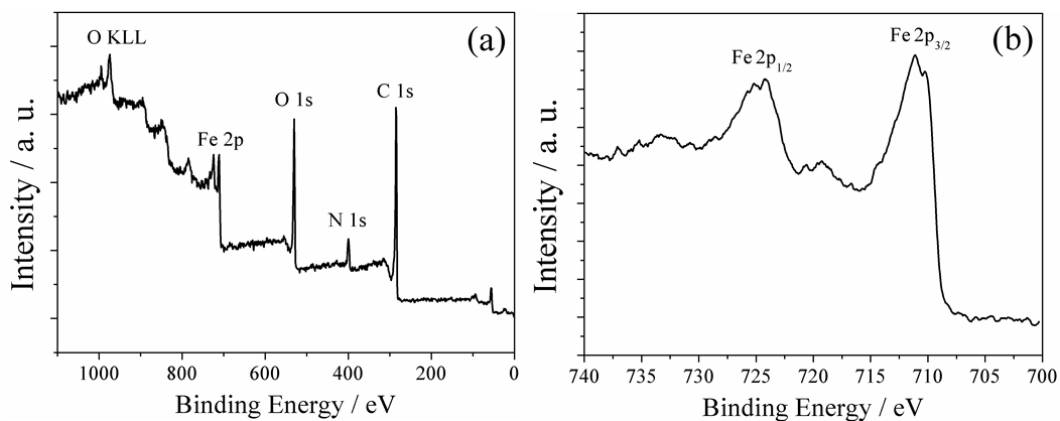


Fig. S3. (a) XPS spectra of Fe₂O₃-NHCS180 composite and (b) XPS spectrum of Fe 2p. The level of Fe2p_{3/2} and Fe2p_{1/2} are 711.1 and 724.2, respectively. In addition, Fig. S3b presents of a satellite peak at 719.0 eV. These results confirm that the iron oxide nanoparticles in the composite are maghemite -Fe₂O₃ rather than Fe₃O₄.

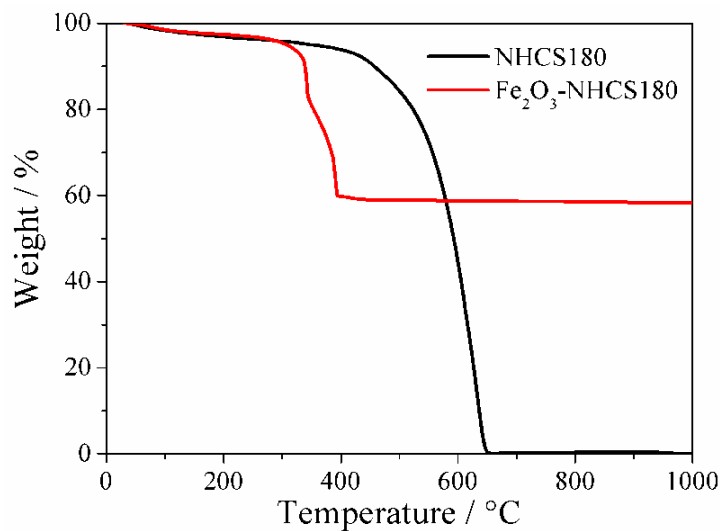


Fig. S4. TG analysis (in air-flow) of NHCS180 (black line) and Fe₂O₃-NHCS180 composite (red line). The Fe₂O₃ content in the composite material was determined to be 57.8 wt%.

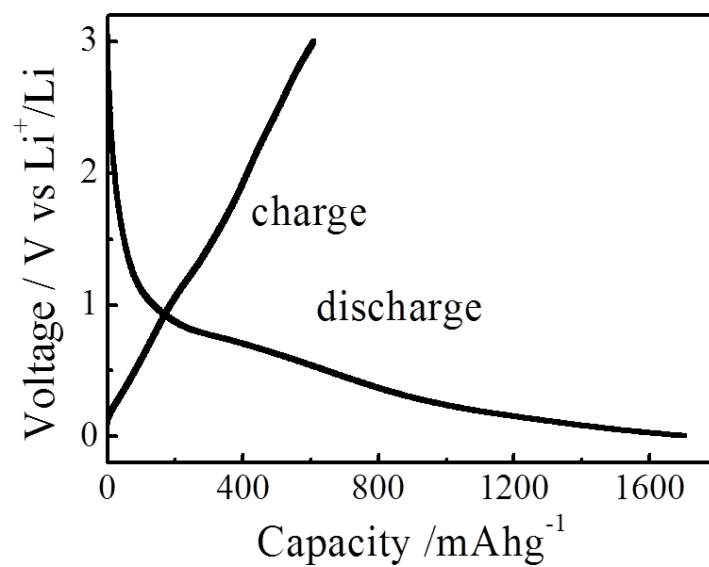


Fig. S5. Discharge and charge curves of NHCS180 at the current density of 100 mA g⁻¹ in a voltage window of 0.005-3.00 V.