Supplementary Data

Synthesis, Chemo-selective Properties of Substituted 9-Aryl-9H-fluorenes from Triarylcarbinols and Enantiomerical Kinetics of Chiral 9-Methoxy-11-(naphthalen-1-yl)-11H-benzo[a]fluorene

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I. General Procedure for Preparation of 1c, 1d, 1e, 1g, 1h, 1k, 1l) (except 1i).^[S1]



A mixture of bromoarene (32.0 mmol) and 20 mL of anhydrous THF is added to magnesium (2.30 g, 96.0 mmol), and the flask warmed gently until the reaction start. A mixture of bromoarene (64.0 mmol) and 40 mL of anhydrous THF is run in at such a rate as to cause vigorous refluxing, the mixture is warmed to refluxing for two hours, in order to dissolve all the magnesium. The reaction was cooled to room temperature, aromatic ester (40.0 mmol) is added into the Grinard solution, and the mixture is warmed to reflux over night. The reaction mixture is cooled in an ice-salt bath and then added slowly, with constant stirring, into 60 mL saturated NH_4Cl aqueous solution. The mixture is distilled to remove THF and stream-distilled to remove biphenyl and unchanged bromobenzene. The product is filtered, washed with water, and recrystallized from ethanol and water.

(4-Bromophenyl)bis(3-methoxyphenyl)methanol (1c)



Colorless solid; m.p. 127-128 °C; $R_f = 0.43$ (petroleum ether/ethyl acetate = 4/1); product 1.41 g (yield 35.3%).

¹H NMR (300M Hz, CDCl₃, TMS): δ = 7.42 (d, 2H, *J* = 8.7 Hz; H-3, 5), 7.23 (t, 2H, *J* = 8.7 Hz; H-5'), 7.18 (d, 2H, *J* = 8.7 Hz; H-2, 6), 6.84 (s, 2H; H-2'), 6.84-6.78 (m, 4H; H-4', 6'), 3.75 (s, 6H; H-3').

¹³C NMR (75M Hz): δ = 159.2, 147.8, 145.5, 130.1, 129.6, 129.0, 121.4, 120.3, 113.8, 112.5, 81.5, 55.1.

FT-IR (cm⁻¹): v_{max}(KBr/cm⁻¹), 3554, 2969, 1597, 1484, 1421, 1321, 1286, 1251, 1166, 1137, 1072, 1042, 1022, 1010, 934, 865, 833, 775, 704.

Anal. Calc. for C₂₁H₁₉BrO₃: C, 63.17; H, 4.80. Found: C, 63.13; H, 4.69.

(2,5-Dichlorophenyl)bis(3-methoxyphenyl)methanol (1d)



Colorless solid; m.p. 42-44 °C; $R_f = 0.53$ (petroleum ether/ethyl acetate = 4/1); product 6.59 g (yield 42.3%).

¹H NMR (300M Hz, CDCl₃, TMS): δ = 7.32 (d, 1H, *J* = 8.4 Hz; H-3), 7.24 (t, 2H, *J* = 8.4 Hz; H-5'), 7.23 (dd, 1H, *J* = 8.4, 2.1 Hz; H-4), 6.89 (t, 2H, *J* = 2.1 Hz; H-2'), 6.85 (dd, 2H, *J* = 8.4, 2.1 Hz; H-4'), 6.76 (d, 2H, *J* = 8.4 Hz; H-6'), 6.74 (d, 2H, *J* = 2.1 Hz, H-6), 3.78 (s, 6H; H-3').

¹³C NMR (75M Hz): δ = 159.4, 146.2, 145.3, 132.5, 132.2, 131.4, 131.3, 129.0, 120.1, 113.4, 112.8, 82.1, 55.1. FT-IR (cm⁻¹): _{max}(KBr/cm⁻¹), 3445, 2956, 1599, 1559, 1487, 1457, 1289, 1250, 1105, 1047, 776, 758, 699, 553. HRMS (EI): m/z calcd for C₂₁H₁₈Cl₂O₃ [M⁺]: 388.0633; found: 388.0659.

Bis(3-methoxyphenyl)(naphthalen-1-yl)methanol (1e)



Colorless solid; m.p. 71-73 °C; $R_f = 0.42$ (petroleum ether/ethyl acetate = 4/1); product 6.04 g (yield 62.7%).

¹H NMR (300M Hz, CDCl₃, TMS): $\delta = 8.11$ (d, 1H, J = 8.7 Hz; H-8), 7.85-7.78 (m, 2H; H-4, 5), 7.39 (td, 1H, J = 8.8, 0.9 Hz; H-6), 7.31-7.27 (m, 1H; naphthalen H), 7.22 (t, 2H, J = 8.1 Hz; H-5') 7.17-7.12 (m, 1H; naphthalen H), 6.95 (t, 2H, J = 2.1 Hz; H-2'), 6.88 (dd, 2H, J = 7.5, 0.9 Hz; naphthalen H), 6.84-6.81 (m, 3H; H-6', naphthalen 2-H), 3.74 (s, 6H; H-3').

¹³C NMR (75M Hz): δ = 159.3, 148.4, 141.7, 134.8, 131.1, 129.3, 128.9, 128.1, 127.8, 125.6, 125.2, 124.2, 120.3, 113.6, 113.5, 112.3, 83.1, 55.1.

FT-IR (cm⁻¹): v_{max}(KBr/cm⁻¹), 3454, 2935, 1597, 1486, 1463, 1432, 1316, 1289, 1250, 1143, 1040, 865, 781, 760, 700.

HRMS (EI): m/z calcd for C₂₅H₂₂O₃ [M]⁺: 370.1569; found: 370.1575.

(3-Methoxyphenyl)dinaphthalen-1-ylmethanol (1g)



Colorless solid; m.p. = 206-207 °C; $R_f = 0.63$ (petroleum ether/ethyl acetate = 4/1); product 4.06 g (yield 52.1%).

¹H NMR (300M Hz, CDCl₃, TMS): $\delta = 8.26$ (brs), 2H; H-8), 7.87 (d, 2H, J = 8.1 Hz; H-5), 7.80 (d, 2H, J = 8.1 Hz; H-4), 7.41 (td, 2H, J = 7.5, 0.9 Hz; H-6), 7.28 (t, 2H, J = 7.5 Hz; H-7), 7.23-7.18 (m, 3H, H-3, 5'), 7.05 (s, 1H; H-2'), 6.88-6.82 (m, 4H, H-2, 4' 6'), 3.75 (s, 6H; H-3').

¹³C NMR (75M Hz): 159.4, 148.6, 141.9, 135.0, 131.2, 129.4, 129.2, 128.9, 128.8, 128.4, 128.2, 125.5, 124.3, 120.5, 113.9, 112.1, 85.1, 55.1.

FT-IR (cm⁻¹): v_{max}(KBr/cm⁻¹), 3412, 3043, 1582, 1504, 1482, 1434, 1393, 1284, 1237, 1160, 1019, 875, 833, 806, 793, 776, 698, 560.

Anal. Calc. for C₂₈H₂₂O₂: C, 86.13; H, 5.68. Found: C, 86.21; H, 5.47.

phenylbis(3-(trifluoromethyl)phenyl)methanol (1j)



Colorless oil; $R_f = 0.30$ (petroleum ether/ethyl acetate = 10/1); product 1.68 g (yield 42.4%).

¹H NMR (300M Hz, CDCl₃, TMS): δ = 7.68 (s, 2H, H-2'), 7.57 (d, 2H; *J* = 6.3 Hz; H-4'), 7.48-7.43 (m, 4H; H-5', 6'), 7.36 (m, 1H; H-4), 7.34 (d, 2H, *J* = 2.4 Hz; H-2, 6) 7.22-7.18 (m, 2H; H-3, 5), 2.91 (s, 1H; -OH).

¹³C NMR (75M Hz): δ = 146.1, 144.5, 130.3, 129.4 (t, *J* = 31.9), 127.5, 127.5, 127.1, 126.7, 124.9, 123.4, 123.3, 121.3, 80.6.

FT-IR (cm⁻¹): v_{max}(KBr/cm⁻¹), 3454, 3076, 2926, 1492, 1446, 1330, 1168, 1120, 1076, 806, 703, 675.

Anal. Calc. for C₂₁H₁₄F₆O: C, 63.64; H, 3.56. Found: C, 63.69; H, 3.47.

(3-methoxyphenyl)bis(3-(trifluoromethyl)phenyl)methanol (1k)



Colorless solid; m.p. = 73-74 °C; R_f = 0.43 (petroleum ether/ethyl acetate = 10/1); product 1.59 g (yield 37.3%). ¹H NMR (300M Hz, CDCl₃, TMS): δ = 7.68 (s, 2H, H-2'), 7.57 (d, 2H; J = 6.3 Hz; H-4'), 7.47-7.41 (m, 4H; H-5', 6'), 7.27 (t, 1H; J = 8.1 Hz; H-5), 6.87 (dd, 1H, J = 8.1, 2.1 Hz; H-4), 6.78 (t, 1H; J = 2.1 Hz; H-2), 6.75 (d, 1H; J = 7.5 Hz; H-6), 3.75 (s, 3H; -OMe), 2.92 (s, 1H; -OH). ¹³C NMR (75M Hz): δ = 159.6, 147.2, 147.0, 131.3, 130.4 (t, J = 31.7), 129.6, 128.8, 128.6, 124.5, 124.3 120.1, 114.0, 113.1, 81.4, 55.2.

FT-IR (cm⁻¹): v_{max}(KBr/cm⁻¹), 3507, 2955, 1597, 1489, 1470, 1437, 1330, 1255, 1164, 1123, 1076, 1034, 809, 771, 705.

Anal. Calc. for C₂₂H₁₆F₆O₂: C, 61.98; H, 3.78. Found: C, 61.93; H, 3.87.

II. Procedure for Preparation of 1i



A mixture of 1-bromo-3-(trifluoromethyl)benzene (4.95, 22.0 mmol) and 40 mL of anhydrous THF is added one third to magnesium (0.53g, 22.0 mmol), and the flask warmed gently until the reaction start. The residual mixture of benzophenone and anhydrous THF is run in at such a rate as to cause vigorous refluxing, the mixture is warmed to refluxing for two hours, in order to dissolved all the magnesium. The reaction was cooled to room temperature, 20.0 mmol of benzophenone is added into the Grinard solution, the mixture is warmed to reflux over night. The reaction mixture is cooled in an ice-salt bath and then added slowly, with constant stirring, into 16 mL saturated NH_4Cl aqueous solution. The mixture is distilled to remove THF and stream-distilled to remove biphenyl and unchanged bromobenzene. The crude product was purified by column chromatography with silica gel using CH_2Cl_2 as eluant to afford pure product.

diphenyl(3-(trifluoromethyl)phenyl)methanol (1i)



Colorless oil; $R_f = 0.41$ (petroleum ether/ethyl acetate = 10/1); product 2.71g (yield 41.3 %).

¹H NMR (300M Hz, CDCl₃, TMS): δ = 7.75 (s, 1H, H-2'), 7.58 (d, 1H; *J* = 7.8 Hz; H-4'), 7.51-7.44 (m, 2H; H-5', 6'), 7.39-7.27 (m, 10H; H-2, 3, 4, 5, 6), 2.98 (s, 1H; -OH).

¹³C NMR (75M Hz): δ = 147.9, 146.3, 131.5, 128.9, 128.8, 128.3, 127.9, 127.7, 124.5, 124.5, 124.2, 124.1, 81.9.

FT-IR (cm⁻¹): v_{max}(KBr/cm⁻¹), 3449, 2925, 2855, 1448, 1167, 1128, 1076, 1018, 804, 755, 702, 661.

Anal. Calc. for C₂₀H₁₅ F₃O: C, 73.16; H, 4.60. Found: C, 73.07; H, 4.71.

References

[S1] Bachmann, W. E., Hetzner, H. P. Org. Synth. 1955, Coll. Vol. 3, 839.

III. NMR Spectral Characterization (¹H and ¹³C NMR for 1c, 1d, 1e, 1g, 1i, 1j, 1k, 2a, 2b, 2c, 2d, 2e, 2f, 2g, and 3a) 1c in CDCl₃



1d in CDCl₃



1e in CDCl₃











• 2a in CDCl₃







2c in CDCl₃



2d in CDCl₃



2e in CDCl₃









2g in benzene-d6





IV. Variable-temperature ¹H NMR and Eyring Plot Analysis of 2g' and 2g" (300 MHz, DMSO)

T (K)	rate constant k/S^{-1}	1/T (K ⁻¹)	$1/T \times 1000$	ln(k)	ln(k/T)	ΔG^{\ddagger} (J/mol)
3.630E+02	4.000E+00	2.7548E-03	2.7548209	1.3863	-4.50811	8.5313E+04
3.730E+02	8.000E+00	2.6810E-03	2.6809651	2.0794	-3.84214	8.5598E+04
3.780E+02	1.200E+01	2.6455E-03	2.6455026	2.4849	-3.44999	8.5513E+04
3.830E+02	2.000E+01	2.6110E-03	2.6109661	2.9957	-2.9523	8.5059E+04
3.880E+02	3.500E+01	2.5773E-03	2.5773196	3.5553	-2.40566	8.4406E+04
3.930E+02	5.000E+01	2.5445E-03	2.5445293	3.9120	-2.06179	8.4370E+04



ln(k)=lnA-Ea/RT	R Ea/R -∆H ^{‡/} R	8.314E+0 12364 -1.199E+04	Ea (J/mol)	102.794E+3
$ln(k/T) = 23.76 + \Delta S^{*}_{*}/8.31 - \Delta H^{*}/(8.31T)$	ΔH^{\ddagger} (J/mol)	99651.60		
$\Delta G^{\ddagger} = (23.76 - \ln(k/T)) * R * T$ $\Delta G^{\ddagger} = aT(10.319 - \log(k/T))$ $a = 1.914 * 0.01 (\Delta G^{\ddagger} \text{ in } kJ/\text{mol})$ $a = 1.914 * 10 (\Delta G^{\ddagger} \text{ in } J/\text{mol})$ $\Delta G^{\ddagger}_{\ddagger} = \Delta H^{\ddagger} - T\Delta S^{\ddagger}$	ΔS^{\ddagger} (J/mol*K)	37.6783112	(T = 373K)	

V. MS Spectra

1d



1e







2g



VI. HPLC of 2g.

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VII. H-HCOSY/HMBC/HMQC 2D NMR Spectroscopy of 2g (500M Hz, in CDCl₃).





S27



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Reference

[S1]. Bachmann, W. E.; Hetzner H. P. Org. Synth. 1955, Coll. Vol. 3, 839-840.