

## Supporting Information for

# Adsorption of Hydrogen Sulfide on Metal-Organics Frameworks

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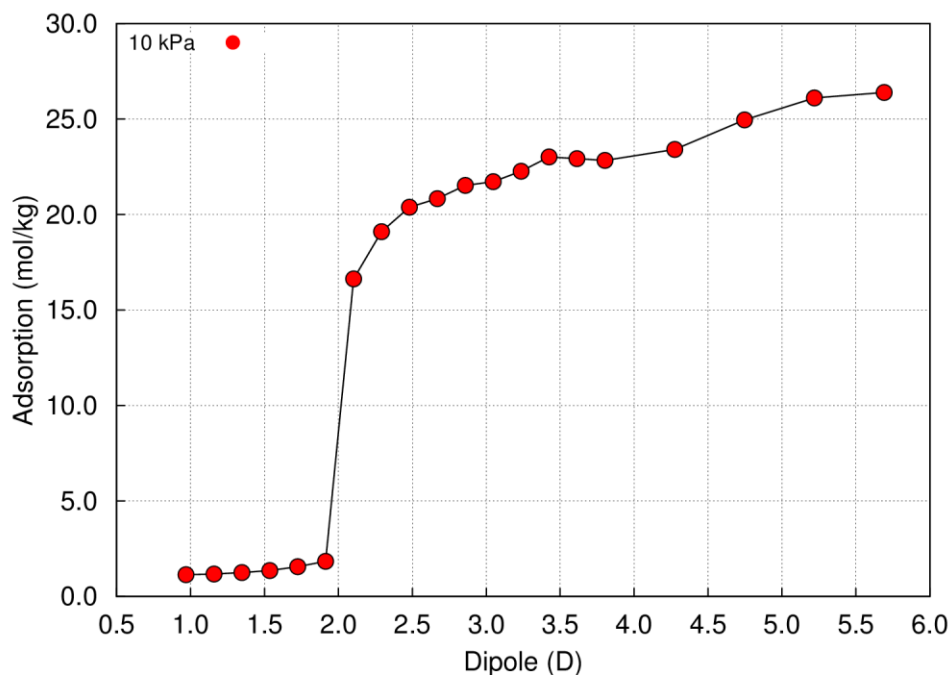
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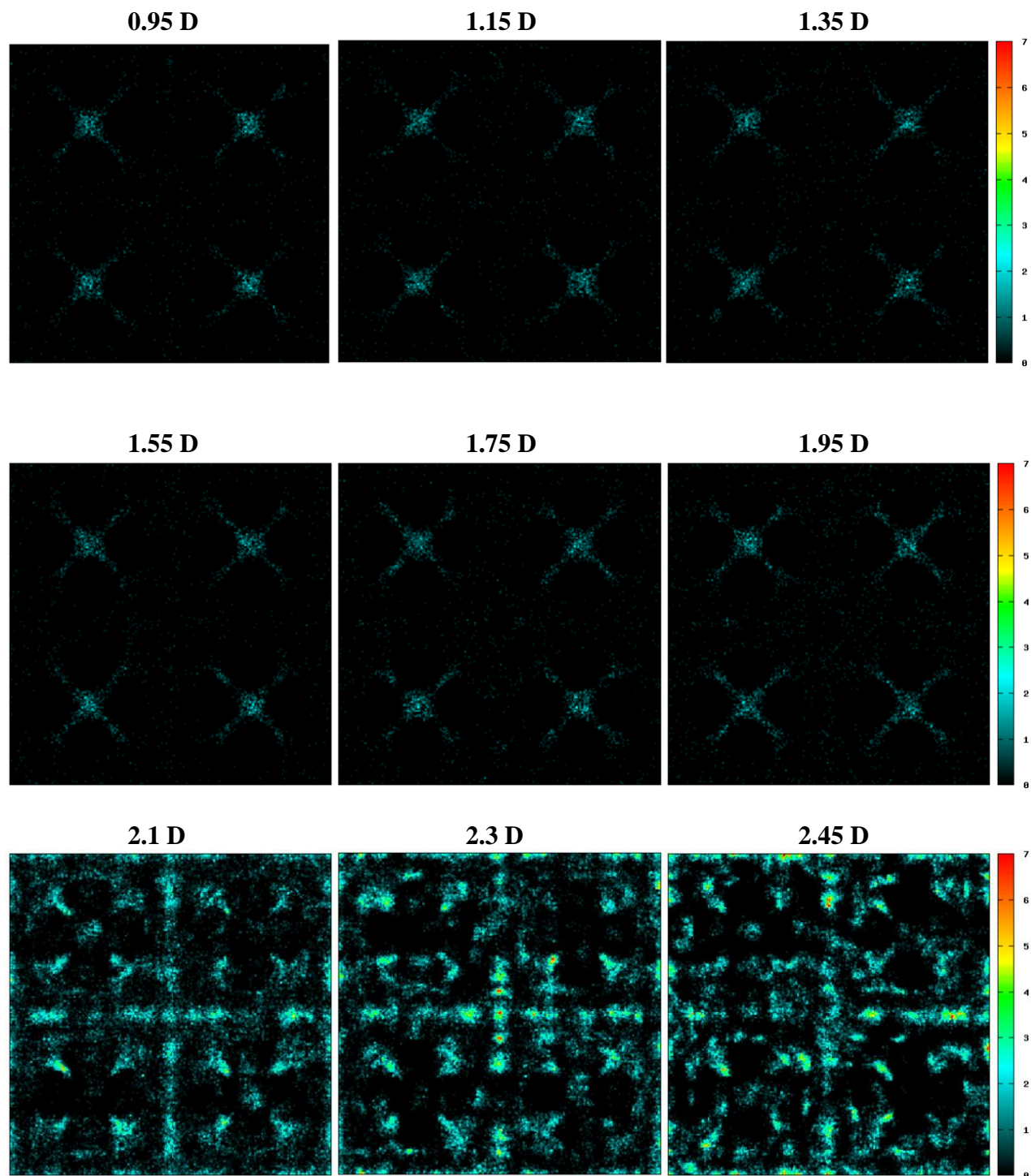
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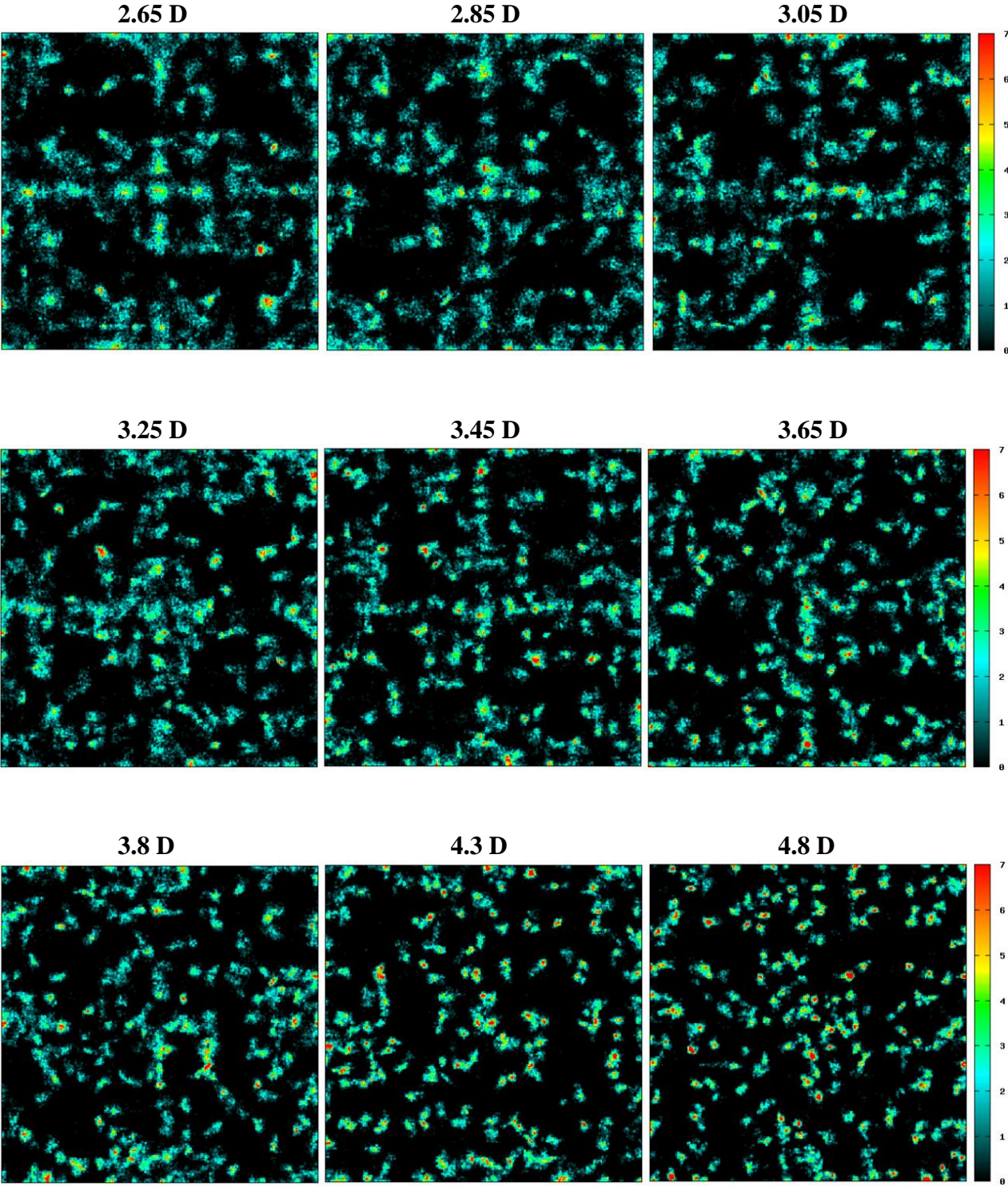
## S1. Influence of the molecular dipole on the adsorption



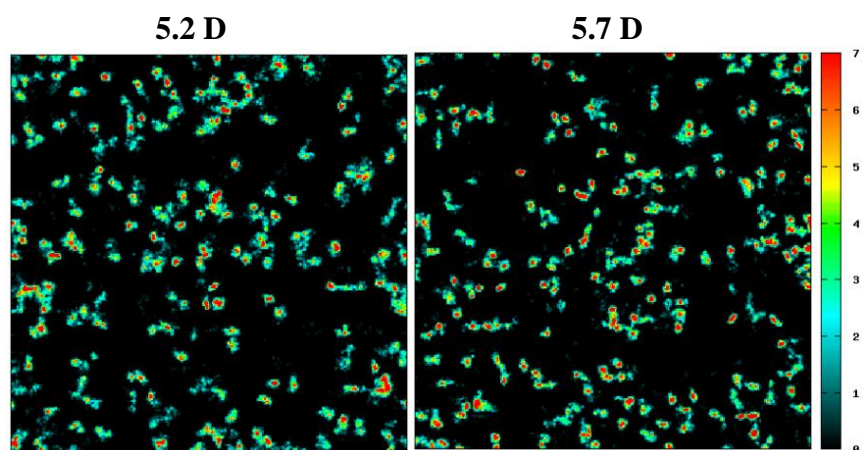
Adsorption of H<sub>2</sub>S in Cu-BTC as a function of the dipole moment of the molecule, employing the 5S force field. The temperature is 300 K and the pressure is 10 kPa.

## S2. Variation of the average occupation profile with the molecular dipole



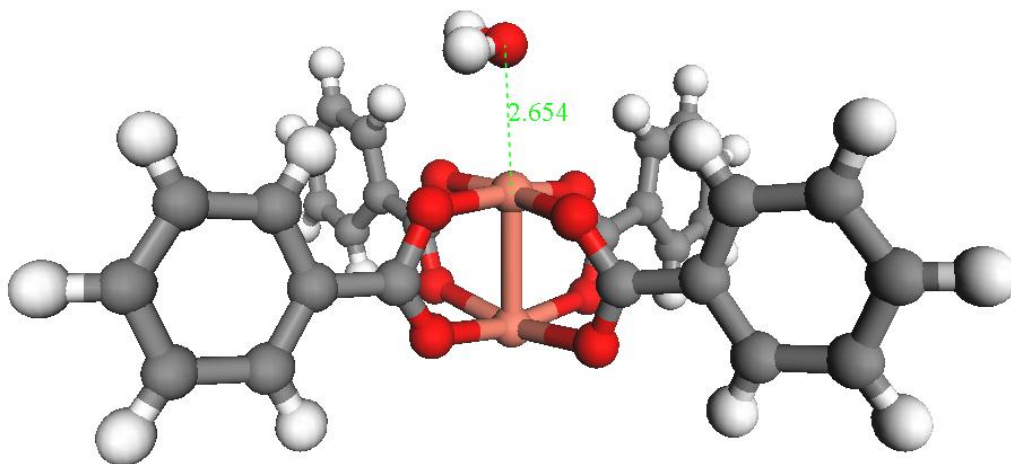




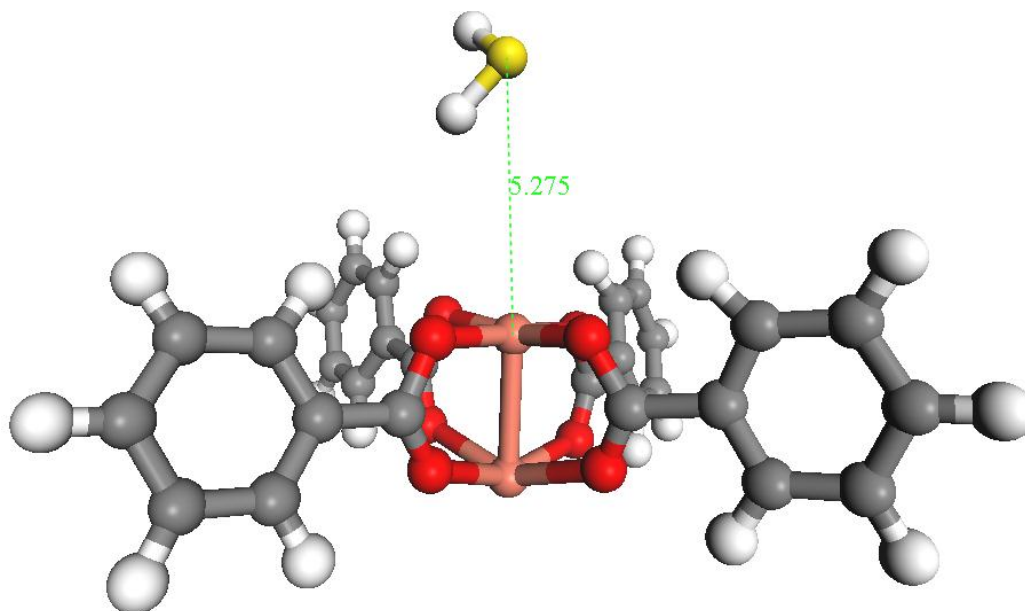


**Variation of the average occupation profiles of the H<sub>2</sub>S centers of mass (at 10 kPa and 300 K) in Cu-BTC, as the molecular dipole changes from 0.95 D to 5.7 D. The force field employed is the five-sites 5S.**

### S3. Snapshots of *ab initio* Molecular Dynamics simulations of H<sub>2</sub>S and H<sub>2</sub>O adsorbed on a cluster model of Cu-BTC



a)



b)

Snapshots of *ab initio* Molecular Dynamics simulations of H<sub>2</sub>O (a) and H<sub>2</sub>S (b) adsorbed on a cluster model of Cu-BTC. The simulations are carried out employing the Gaussian 09 code, with the Atom Centered Density Matrix Propagation molecular dynamics method. The level of theory used is HF/LanL2MB. The calculations were 3 ps long, with a time step was 0.1 fs, and the temperature was 300 K.