# A highly sensitive and selective resonance Rayleigh scattering method for bisphenol A based on the aptamer-nanogold catalysis of HAuCl<sub>4</sub>-vitamine C particle reaction

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# Absorption spectra

There is only one surface plasmon resonance (SPR) absorption peak at 518 nm for the GNs. Figure 1S indicated that Apt-GN was aggregated nonspecifically in the condition of pH 7.6  $Na_2HPO_4$ - $NaH_2PO_4$  buffer solution, which showed a weak SPR peak at 518 nm. Upon addition of BPA, Apt-GN specifically combined with BPA to form dispersed BPA-Apt-GN conjugates, the SPR absorption peak increased at 518 nm.



Figure 1S Absorption spectra of the BPA-Apt-GN system a: 2.7×10<sup>-5</sup> mol/L Apt-GN-pH7.6 PBS;b:a-66.7 ng/mL BPA;c:a-133.3 ng/mL BPA; d:a-333.3 ng/mL BPA;e:a-666.7 ng/mL BPA

## Scanning electron microscope

The reaction solution of BPA-Apt-GN system was obtained by the procedure, and centrifugation and ultrasonic

dispersion of the reaction solution were made twice to prepare sample solution. Then, a 5.0 µL of the sample solution was added to a clean silicon wafer, natural drying, the sample was placed in a scanning electron microscope, to obtain the scanning electron micrograph (Figure 2S). Figure 2Sa showed that Apt-GN particles dispersed in solution. In the absent of BPA, the Apt-GNs were aggregated and formed big aggregation in the pH 7.6 Na<sub>2</sub>HPO<sub>4</sub>-NaH<sub>2</sub>PO<sub>4</sub> buffer solution (Figure 2Sb). When the concentration of BPA increased, Apt-GN reacted specifically with BPA to form BPA-Apt-GN conjugates that were dispersed in PBS buffer solution stably (Figure 2Sc), which led to the aggregates reduced, that is agreement with the RRS results.



 $\label{eq:Figure 2S} Figure 2S \ Scanning \ electron \ microscope \ images $a: 2.7 \times 10^{-5} \ mol/L \ Apt-GN; \ b: 5.17 \mu g/mL \ Apt-GN-pH7.6 \ PBS; \ c: \ a-1 \mu g/mL \ BPA. $a= 1 \ \mu g/mL \ BPA.$ 

#### Preparation conditions of the Apt-GN probe

# Effect of NaCl concentration

The effect of NaCl concentration on the GN labeled-aptamer was examined. A 200  $\mu$ L of 47.3 $\mu$ g/mL GN was added into each 5mL marked tube, a certain amount of 2.0 mol/L NaCl was added, after 10min, the solution was diluted to 2.0mL. The RRS intensity at 380 nm ( $I_{380nm}$ ) was recorded. Table 1S showed that when the NaCl concentration was more than 0.013 mol/L, the  $I_{380 nm}$  increased slightly. Thus, a 0.013 mol/L NaCl concentration was chosen for use.

Table 1S Effect of NaCl concentration on the  $I_{380nm}$ 

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NaCl(µL)	0	5	10	15	20	30	40
I <sub>380nm</sub>	188	617	929	808	911	834	812

## Selection of labeling pH

The effect of pH on the GN labeled aptamer was examined. A 200  $\mu$ L of 47.3  $\mu$ g/mL GN was added into each 5mL marked tube and the pH adjusted to 3.0-9.0 using 0.1mol/L HCl and 0.1mol/L K<sub>2</sub>CO<sub>3</sub>, and 100  $\mu$ L of 0.5 $\mu$ mol/L aptamer was added, after 10min, 10 $\mu$ L of 2.0 mol/L NaCl was added. After 30min, the solution was diluted to 1.5mL. The RRS intensity at 520 nm ( $I_{380nm}$ ) was recorded. Table 2S showed that when the pH was more than 5.0, the  $I_{380nm}$  decreased slightly, owing to GN being coated by aptamer. Thus the GN could not aggregate in NaCl solution. A pH 6.5 was chosen for use.

Table 2S Effect of pH on the  $I_{380nm}$ 

pН	3.0	3.5	4.0	4.5	5.0	6.0	6.5	7.5	8.0	9.0
I <sub>380nm</sub>	544	344	282	316	246	279	240	244	231	271

#### Selection of aptamer amount

A 200  $\mu$ L of 2.4×10<sup>-4</sup> mol/L GN was added into each 5mL marked tube and the pH adjusted to 6.5. Then, different amounts of 0.5 $\mu$ mol/L aptamer were added, then ultrasound 10min, 10 $\mu$ L of 2.0 mol/L NaCl was added. After 30min, the solution was diluted to 1.5mL. The RRS intensity at 520 nm ( $I_{380nm}$ ) was recorded. Table 3S showed that when aptamer amount was more than 100 $\mu$ L, the  $I_{520nm}$  decreased slightly. Thus, 100 $\mu$ L of aptamer was chosen for use.

Table 3S Effect of aptamer amount on the  $I_{380nm}$ 

$Apt(\mu L)$	0	20	50	80	100	130	160	290	250	300
<i>I</i> <sub>380nm</sub>	910	438	254	370	253	234	215	263	214	279

Table 4S The effect of ultrasonic and standing form on the  $I_{380nm}$ 

Reaction form	I <sub>380 nm</sub>	RSD(%)
Standing	1362, 1240, 1327, 1275, 1119, 1391, 1579, 1677, 1705	14.4
Utrasonic	1238, 1250, 1256, 1237, 1252, 1375, 1368, 1238, 1330	4.5