Electronic Supplementary Information (ESI)

Synthesis of nano- $Li_4Ti_5O_{12}$ decorated on non-oxidized carbon nanotubes with enhanced rate capability for lithium-ion batteries

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Figure S1. Schematic illustration of the protocol adopted for the synthesis of nano- $Li_4Ti_5O_{12}$ decorated on non-oxidized CNTs.



Figure S2. Schematic diagram of the cell used for measuring the electrical conductivity of the samples by using the two-point probe method. The cell was assembled with a stainless-steel top electrode and a fixed bottom electrode, poly(methyl methacrylate) (PMMA) mould, and $Li_4Ti_5O_{12}/CNTs$ nanocomposite electrode in the form of a pelletized disc of area 1.33 cm².



Figure S3. Raman spectrum of $\text{Li}_4\text{Ti}_5\text{O}_{12}$ in $\text{Li}_4\text{Ti}_5\text{O}_{12}/\text{CNTs}$ nanocomposite. Seven vibrational peaks were observed at 233, 268, 356, 433, 628, 676, and 740 cm⁻¹. These Raman-allowed phonon peaks are the fingerprints of the spinel structure (A1g+Eg+3F2g+F1u+F2u), which confirm the formation of phase-pure spinel $\text{Li}_4\text{Ti}_5\text{O}_{12}$ in the nanocomposite.



Figure S4. (a) SEM and (b) TEM images of the $Li_4Ti_5O_{12}$ nanoparticles prepared by the microwave-solvothermal method in the absence of CNTs. The resulting $Li_4Ti_5O_{12}$ nanoparticles (size < 50 nm) were agglomerated because of the absence of the CNT matrix. (c) The second charge-discharge curves of $Li_4Ti_5O_{12}$ nanoparticles prepared in the absence of CNTs. Rate capability of $Li_4Ti_5O_{12}$ nanoparticles (mixed with 15 wt% CNT) under different current rates (from right to left): 1, 2, 5, 10, 20, and 30 C-rate.



Figure S5. Thermogravimetric analysis (TGA) of $Li_4Ti_5O_{12}/CNT$ nanocomposite. From the thermogravimetric data, the loading amount of $Li_4Ti_5O_{12}$ in the $Li_4Ti_5O_{12}/CNTs$ nanocomposite was evaluated to be 80 wt%.



Figure S6. Comparison of rate capabilities (discharge capacity versus discharge rate) of the $Li_4Ti_5O_{12}/CNT$ nanocomposite prepared in this study with that of the reported $Li_4Ti_5O_{12}/CNT$ composite (refs. 1, 2, 3, 4) and $Li_4Ti_5O_{12}/graphene$ composite (refs. 5, 6, 7, 8, 9, 10, 11).

References

- (1) X. Li, M. Qu, Y. Huai and Z. Yu, *Electrochim. Acta*, 2010, 55, 2978.
- (2) L. Shen, C. Yuan, H. Luo, X. Zhang, K. Xu and F. Zhang, J. Mater. Chem., 2011, 21, 761.
- (3) J. Shu, L. Hou, R. Ma, M. Shui, L. Shao, D. Wang, Y. Ren and W. Zheng, RSC Adv., 2012, 2, 10306.
- (4) B. Zhang, Y. Liu, Z. Huang, S. Oh, Y. Yu, Y. Mai and J. Kim, J. Mater. Chem., 2012, 22, 12133.
- (5) L. Shen, C. Yuan, H. Luo, X. Zhang, S. Yang and X. Lu, *Nanoscale*, 2011, **3**, 572.
- (6) Y. Shi, L. Wen, F. Li and H. M. Cheng, J. Power Sources, 2011, 196, 8610.
- (7) H. Xiang, B. Tian, P. Lian, Z. Li and H. Wang, J. Alloys Comp., 2011, 509, 7205.
- (8) Q. Zhang, W. Peng, Z. G. Wang, X. Li, X. Xiong, H. Guo, Z. Wang and F. Wu, *Ionics*, DOI 10.1007/s11581-012-0813-x.
- (9) Y. Tang, F. G. Huang, W. Zhao, Z. Liua and D. Wana, J. Mater. Chem., 2012, 22, 11257.
- (10) S. Yi Han, I. Y. Kim, K. Y. Jo and S. Hwang, J. Phys. Chem. C, 2012, 116, 7269.
- (11) B. Zhang, Y. Yu, Y. Liu, Z. Huang, Y. He and J. Kim, *Nanoscale*, 2013, DOI: 10.1039/C2NR33099G.