

## Supporting Information

For

### A Cr<sup>3+</sup> Selective Dual Turn-on Chemosensor And Its Application In Living Cell Imaging

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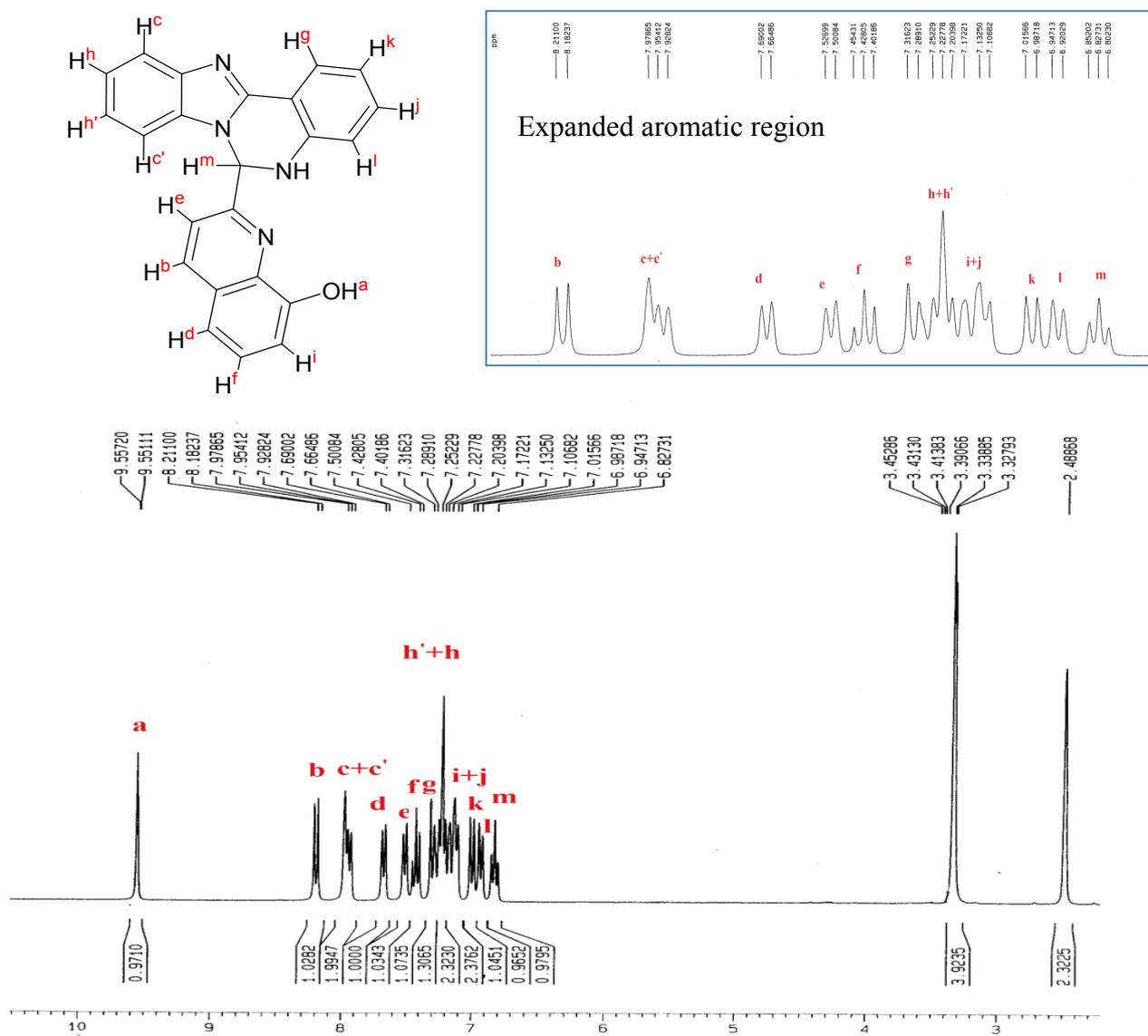
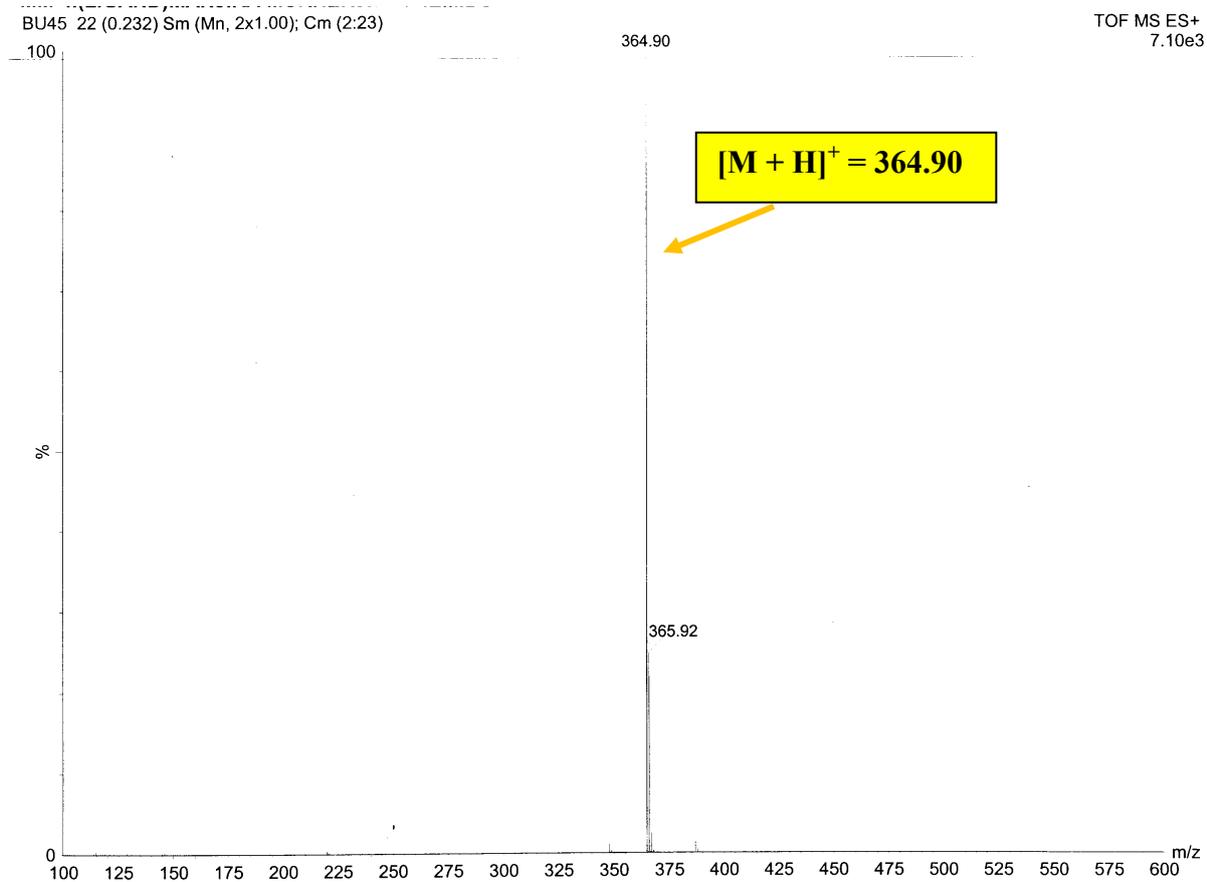


Fig. s1  $^1\text{H}$  NMR spectrum of  $\text{H}_2\text{L}^1$  in  $\text{DMSO-d}_6$



**Fig. s2** Mass spectrum of 2-(5,6-dihydro-benzo[4,5]imidazo[1,2-c]quinazolin-6-yl)-quinolin-8-ol (H<sub>2</sub>L<sup>1</sup>)

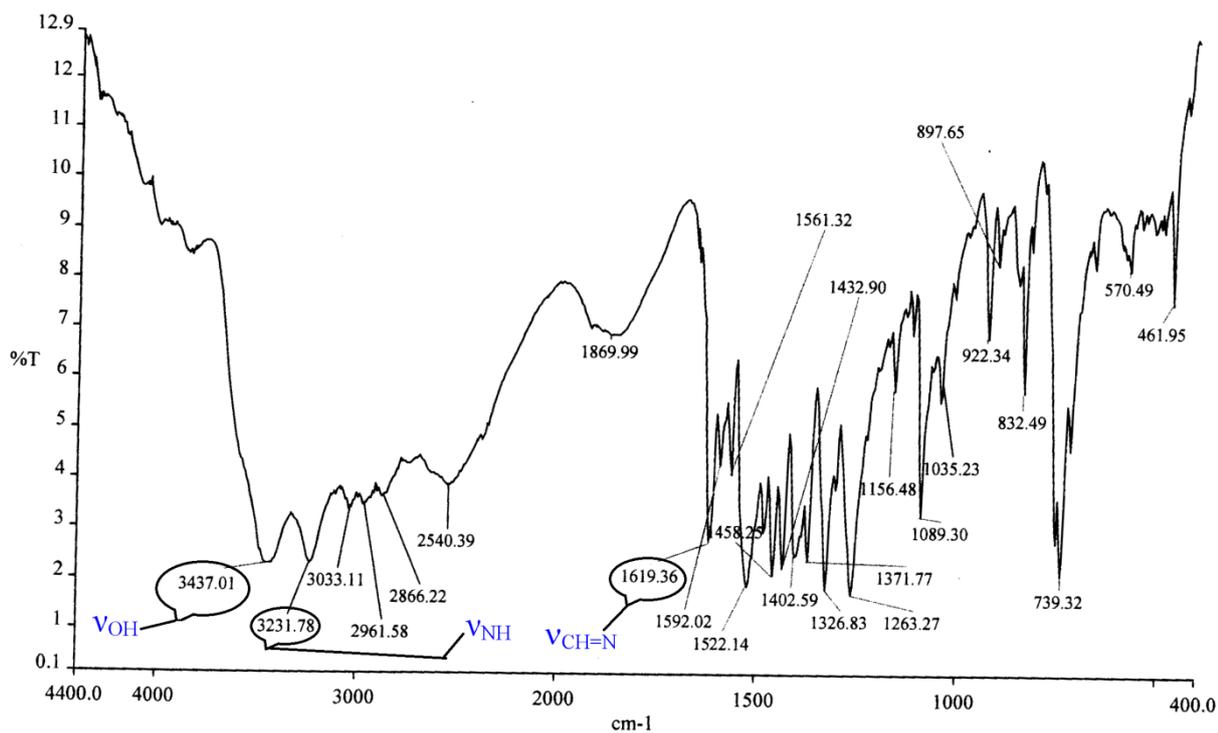


Fig. s3 IR spectrum of H<sub>2</sub>L<sup>1</sup>

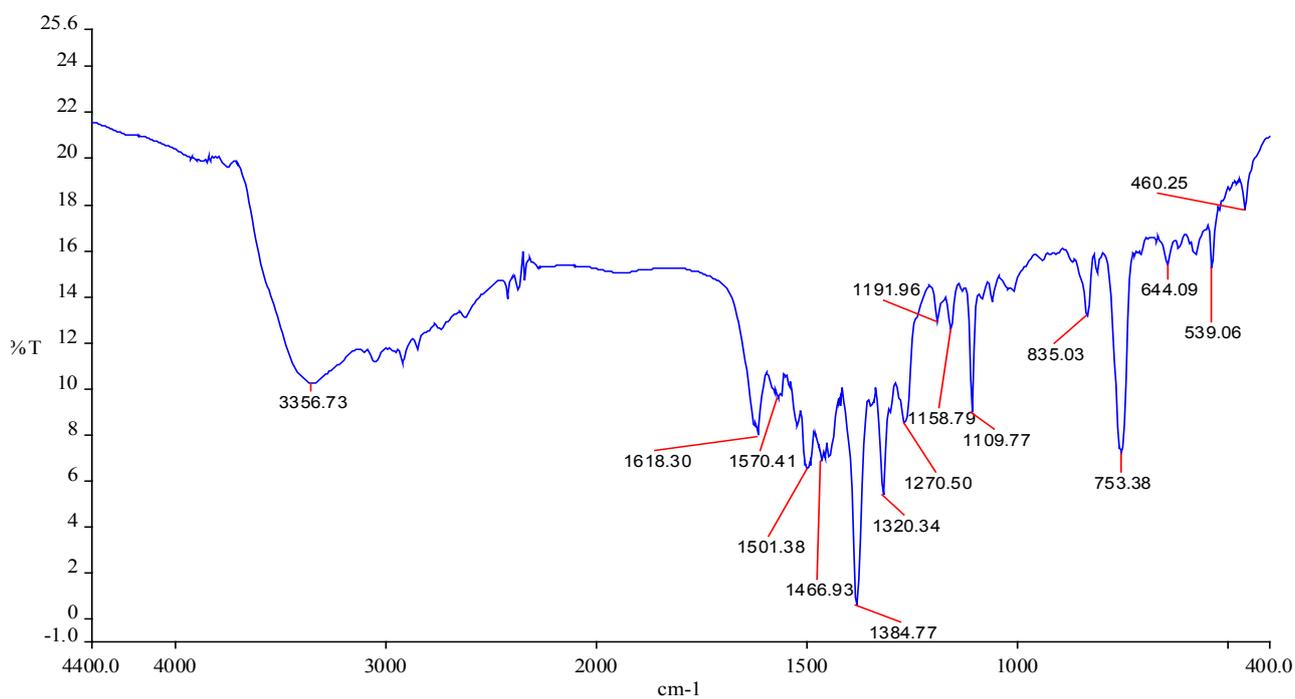


Fig. s4 IR spectrum of the chromium(III) complex, [Cr(L)(NO<sub>3</sub>)(H<sub>2</sub>O)]

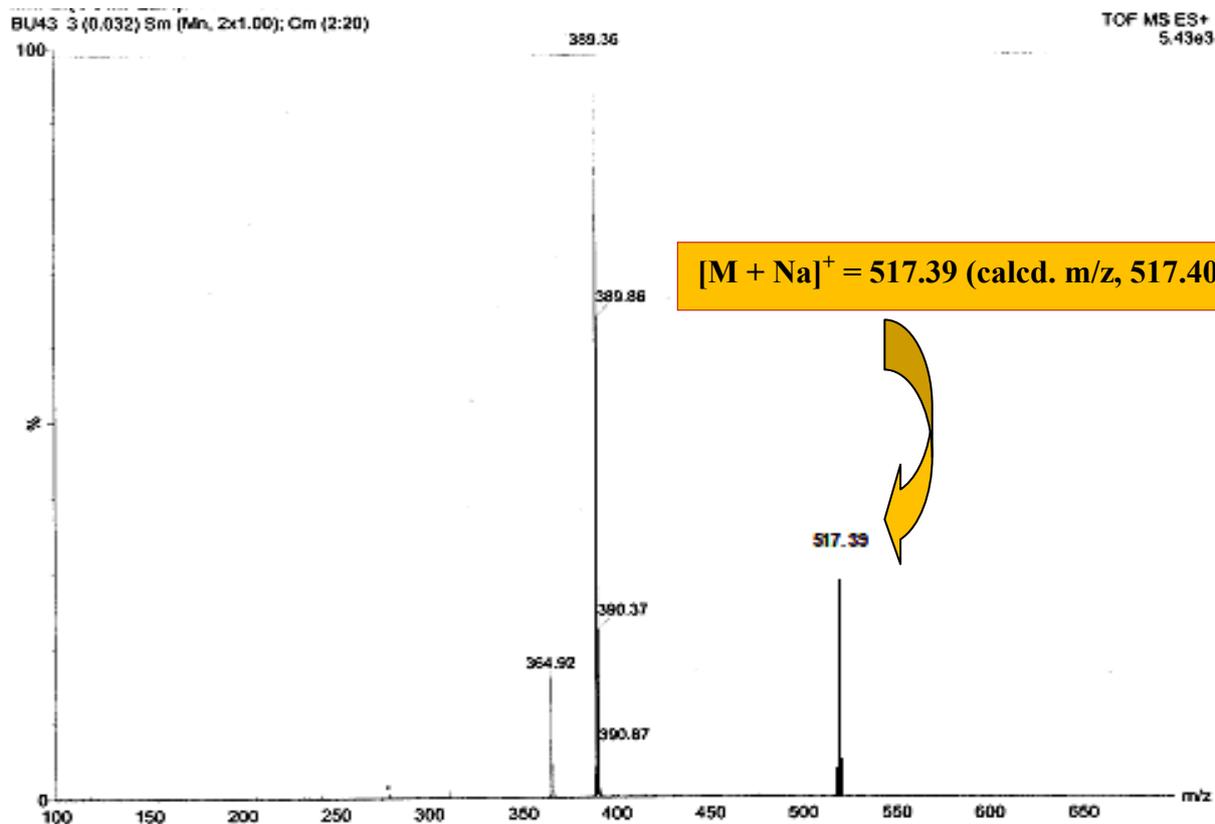


Fig. s5 Mass spectrum of the chromium(III) complex,  $[Cr(L)(NO_3)(H_2O)]$

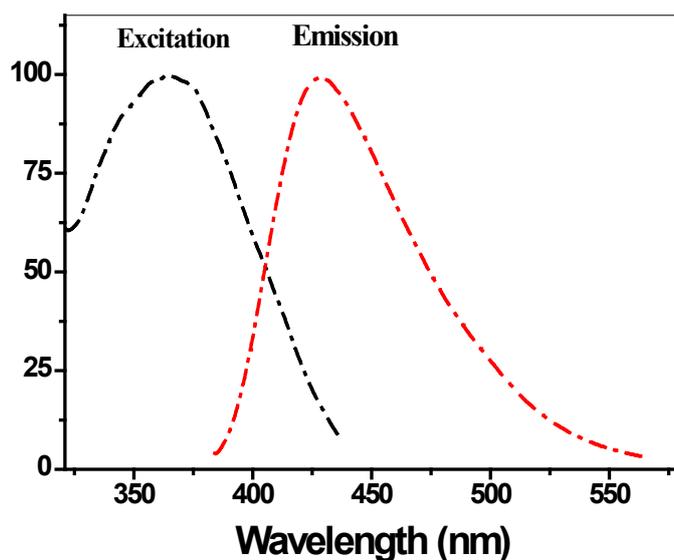
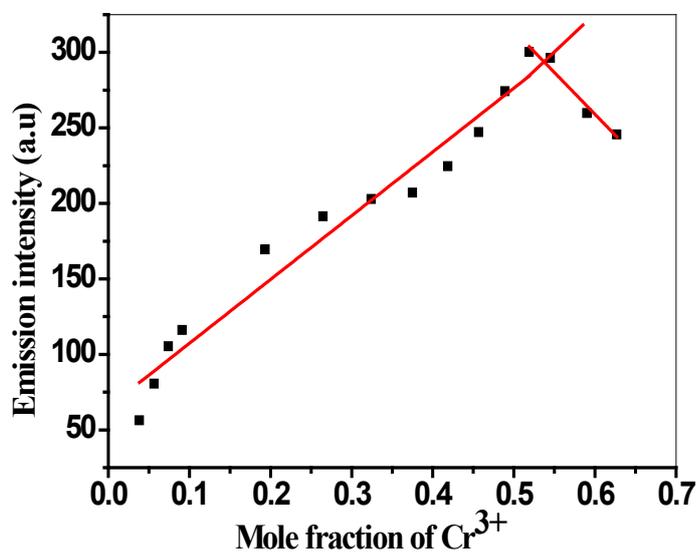


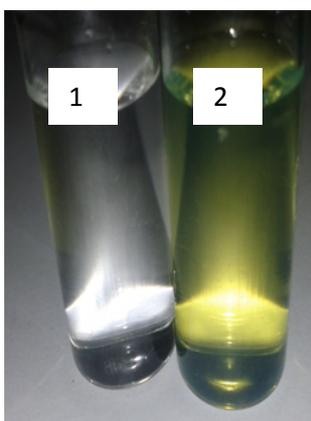
Fig.s6 Absorption and emission spectra of 12.5  $\mu$ M of the probe in 100 mM HEPES buffer (ethanol/water 1:5, v/v) at 25°C



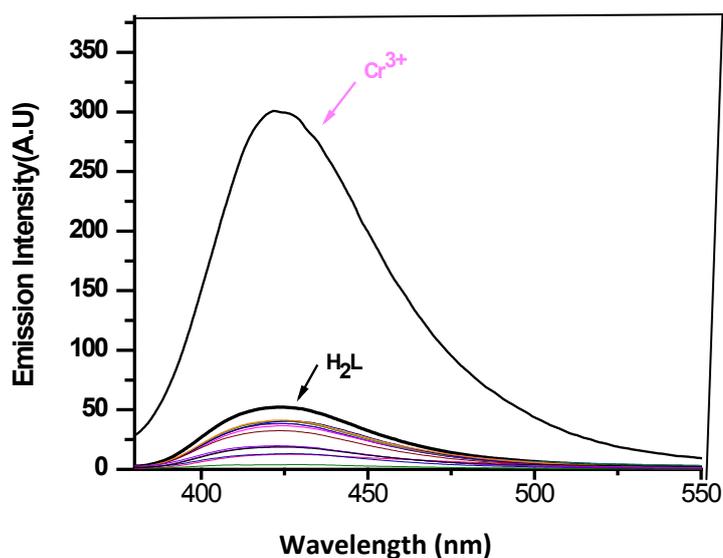
**Fig. s7** Fluorescence colour of the probe in absence (1) and presence (2) of  $\text{Cr}^{3+}$  ion



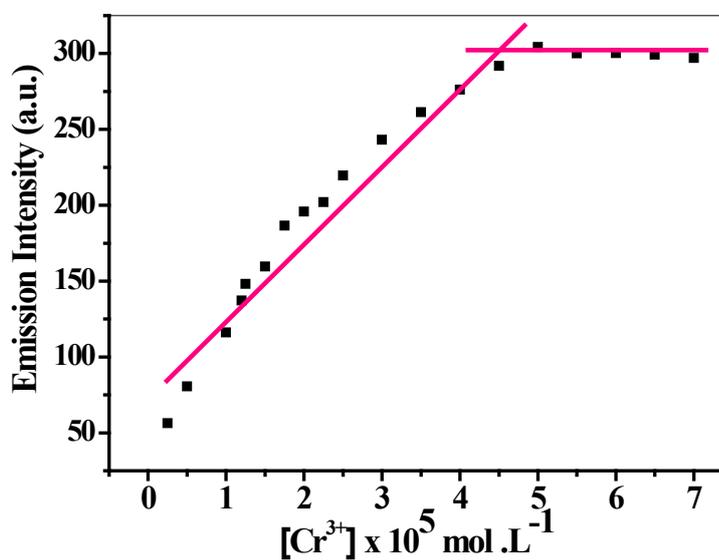
**Fig.s8** Job's plot analysis showing maximum emission at 1:1 ratio [ $\text{H}_2\text{L} : \text{Cr}^{3+}$ ]



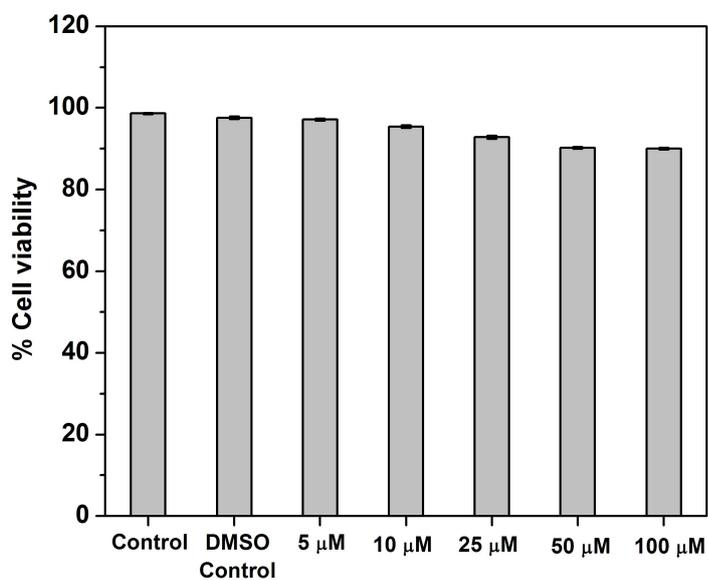
**Fig. s9** Visual colour change of the probe due to the addition of  $\text{Cr}^{3+}$  ion: colour of the probe in absence (1) and presence (2) of  $\text{Cr}^{3+}$  ion



**Fig.s10** Fluorescence spectra of H<sub>2</sub>L<sup>1</sup> (10 μM), H<sub>2</sub>L<sup>1</sup> + Cr<sup>3+</sup> (10 μM) and H<sub>2</sub>L<sup>1</sup> (10 μM) + M (200 μM), where M = Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Ag<sup>+</sup>, Mn<sup>2+</sup>, Hg<sup>2+</sup>, Fe<sup>3+</sup>, Co<sup>2+</sup>, Ni<sup>2+</sup>, Cu<sup>2+</sup>, Zn<sup>2+</sup>, Cd<sup>2+</sup> and Pb<sup>2+</sup> ( $\lambda_{em}$ : 426 nm,  $\lambda_{ex}$ : 370 nm, in ethanol-HEPES buffer (1:5)).



**Fig. s11** Linearity range:  $3.6 \times 10^{-7}$  to  $4.5 \times 10^{-5}$  mol.L<sup>-1</sup>



**Fig. s12** Cytotoxic effect of  $H_2L^1$  (5, 10, 25, 50 and 100  $\mu M$ ) in HeLa cells incubated for 8 h

Table s1 The resulting data of time-resolved fluorescence lifetime measurements

System	$\tau_1$ (ns)	$B_1$	$\tau_2$ (ns)	$B_2$	$\tau_{av}$ (ns)*	$\chi^2$
$H_2L^1$	0.107	78.49	5.231	21.51	1.209	1.001
$H_2L^1 + Cr^{3+}$ (0.5eq)	0.086	28.54	6.554	71.46	4.708	1.006
$H_2L^1 + Cr^{3+}$ (1eq)	0.118	12.43	6.264	87.57	5.500	1.011

\* $\tau_{av} = [(\tau_1 \times B_1) + \tau_2 \times B_2] / 100$