

## **Degradation of aniline with zero-valent iron as an activator of persulfate in aqueous solution**

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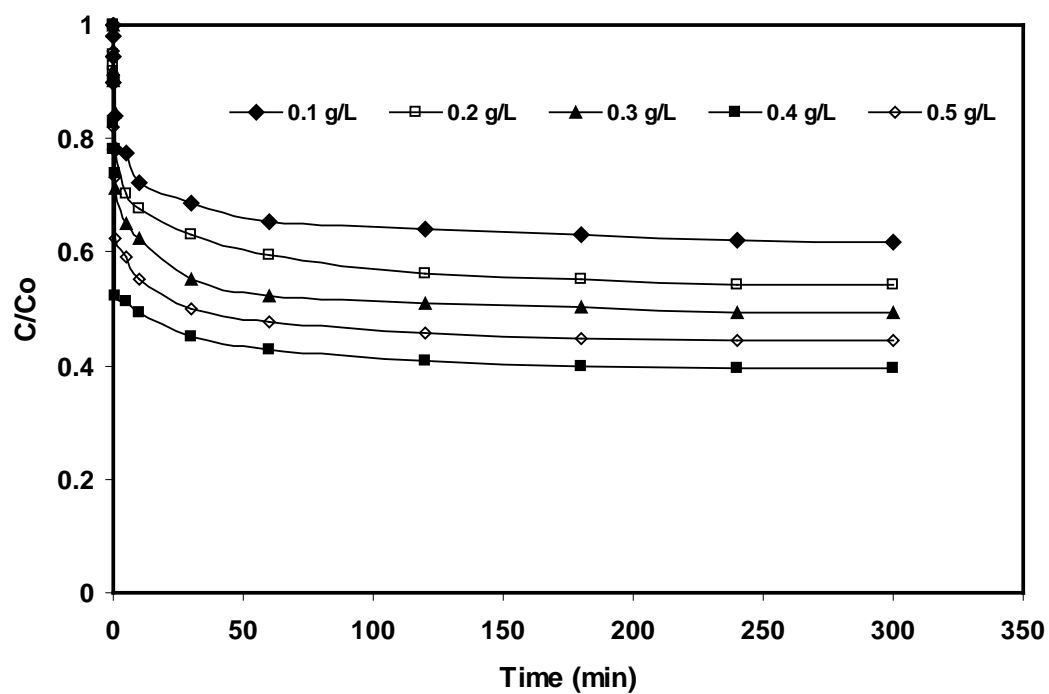
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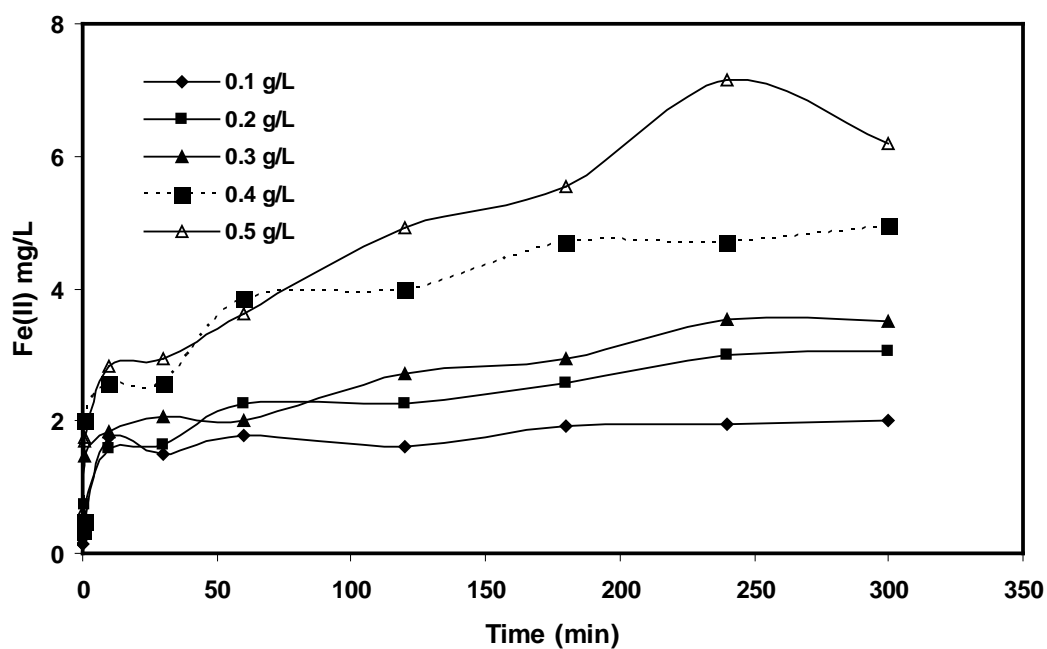
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### Supporting Information



**Fig. 1.** Effect of  $Fe^{2+}$  dosages on aniline degradation in persulfate-  $Fe^{2+}$  system.  $[Aniline]_0 = 0.05$  mM; Temp. =  $25^\circ C$ ;  $[PS]_0 = 2.5$  mM; pH = 7.0.



**Fig. 2.**  $\text{Fe}^{2+}$  concentration in the persulfate-ZVI system.  $[\text{Aniline}]_0 = 0.05 \text{ mM}$ ;  $[\text{PS}]_0 = 2.5 \text{ mM}$ ; Temp. =  $25^\circ\text{C}$ ;  $[\text{ZVI}] = 0.1\text{--}0.5 \text{ g L}^{-1}$ ; pH = 7.0.

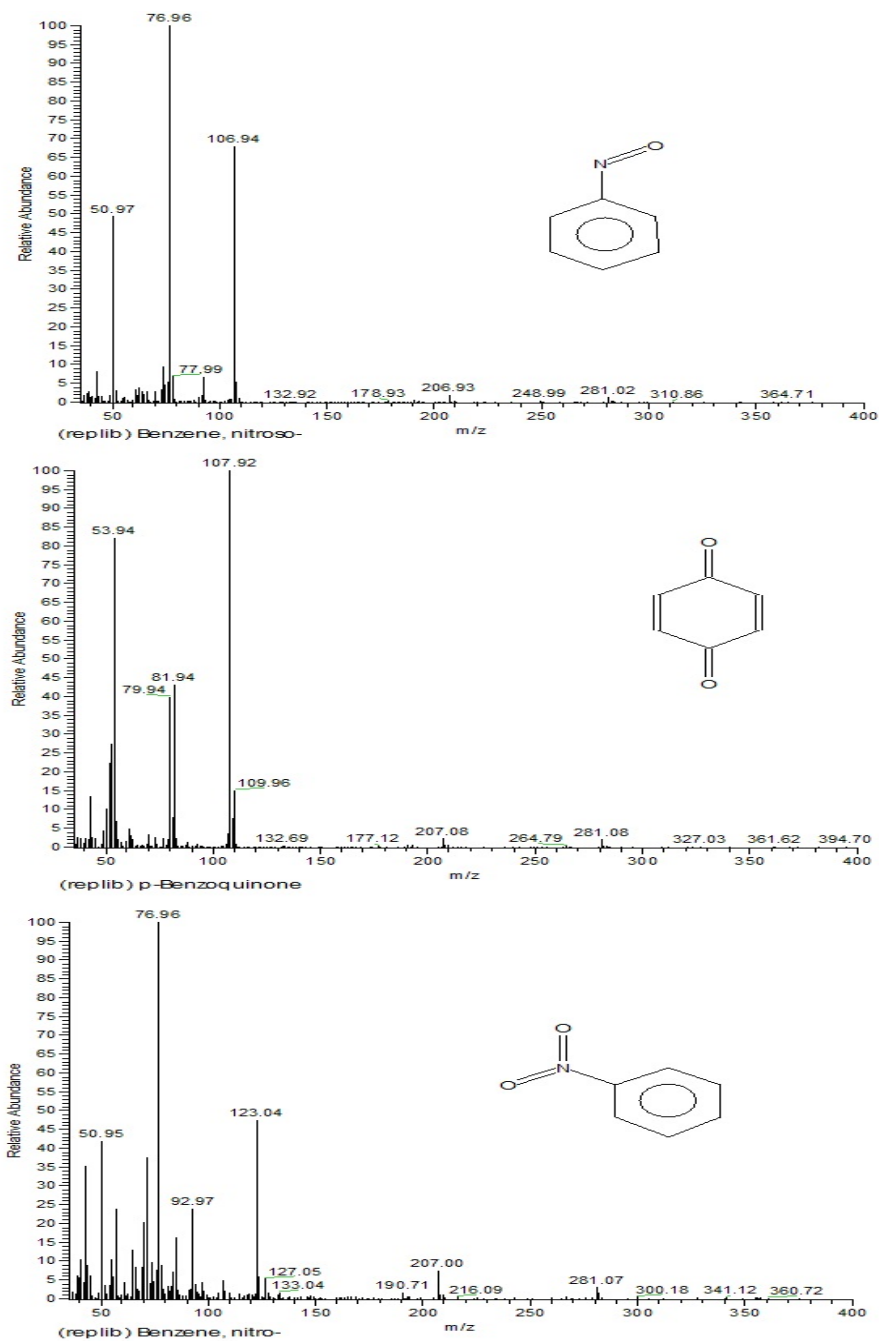


Fig.3 EI mass spectrum of main intermediates of aniline by persulfate-ZVI system with  
GC /MS