

Biocatalytic Promiscuity of HRP in Resveratrol Oxidation: A Function Guided Design with Metal Binding

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I. HPLC yields of the products in HRP catalyzed resveratrol oxidation.

Enzyme used	conversion of 1 (%)	Yield of 2 (%)	Yield of 3&4 (%)	Yield of 5 (%)	Yield of 6 (%)	Yield of 7 (%)
native HRP	95.9	52.5	35.7	7.7	-	-
HRP-Zn ²⁺	96.7	67.6	23.1	6.1	-	-
HRP-Ni ²⁺	96.3	61.7	27.4	7.2	-	-
HRP-Co ²⁺	94.1	51.2	34.2	8.8	-	-
HRP-Ca ²⁺	57.1	34.0	19.8	3.3	-	-
HRP-Cu ²⁺	>99	-	-	-	70.5 ^a	48.2 ^a
HRP-Fe ²⁺	>99	-	-	-	74.2 ^a	51.6 ^a
HRP-Mn ²⁺	>99	94.0 (91.2 ^a)	-	-	-	-

Table S1. HPLC yields of the products in HRP catalyzed resveratrol oxidation

^a isolated yields.

II. UV spectra of synthesized resveratrol dimers **2**, **3&4** and **5**.

1, **2**, **3&4**, and **5** were prepared as 0.001 M acetone solution, and their spectra were recorded using a Shimadzu. Spectra were recorded in the range of 200 - 400 nm. The spectra were shown in Figure S1.

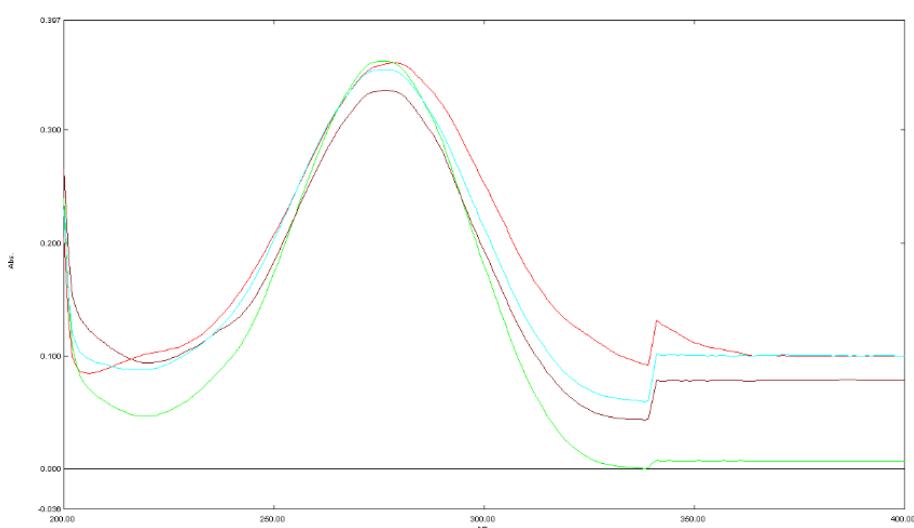


Figure S1. UV spectra of **1**, **2**, **3&4** and **5** (0.001 M acetone solution)

III. Characterization of HRP-metal complex's conformation with circular dichroism spectrometer.

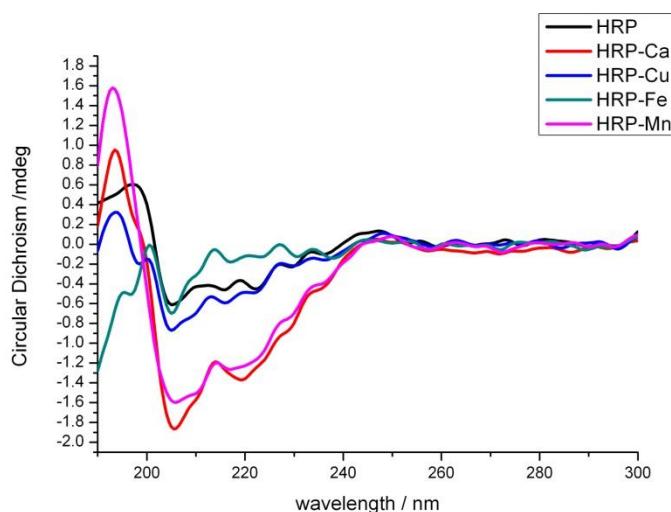


Figure S2. CD spectra of native HRP, [HRP-Ca], [HRP-Mn], [HRP-Fe] and [HRP-Cu]

enzyme/complex	α -helix	β -sheet	β -turn	random
native HRP	0.0	82.8	0.0	17.2
[HRP-Ca]	0.0	82.0	0.0	18.0
[HRP-Cu]	0.0	82.5	0.0	17.5
[HRP-Fe]	0.0	82.7	0.0	17.3
[HRP-Mn]	0.0	82.1	0.0	17.9

Table S2. Prediction of secondary structure constitution by Yang's method

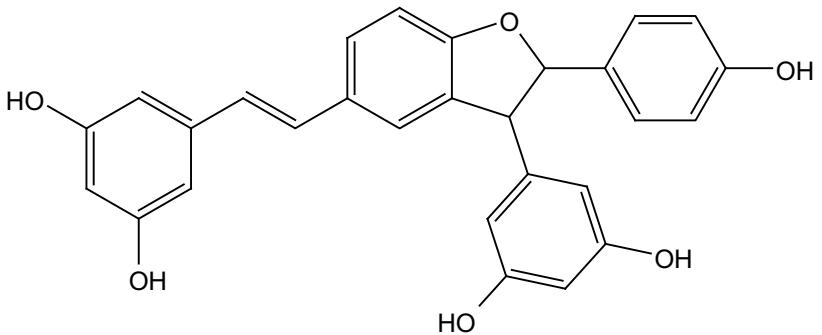
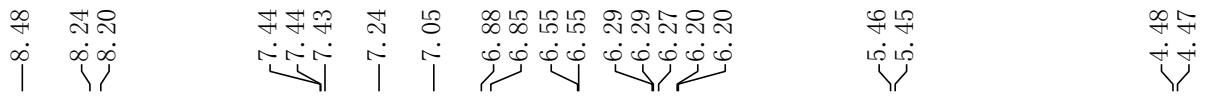
III. Determination of the HRP-metal complex constitution with Atomic Absorption Spectrometer (AAS)

Sample	Concentration of Fe (ppm)	Concentration of binded metal (ppm)	Number of binded metal
Native HRP	13.5564	-	-
HRP-Cu	13.5782	83.1007	6.12
HRP-Fe	52.4633	52.4633	2.87
HRP-Mn	13.5773	150.5723	11.09

Table S3. AAS suggested concentration of Fe and binded metals in 2 mg/mL native HRP, [HRP-Cu], [HRP-Fe] and [HRP-Mn]

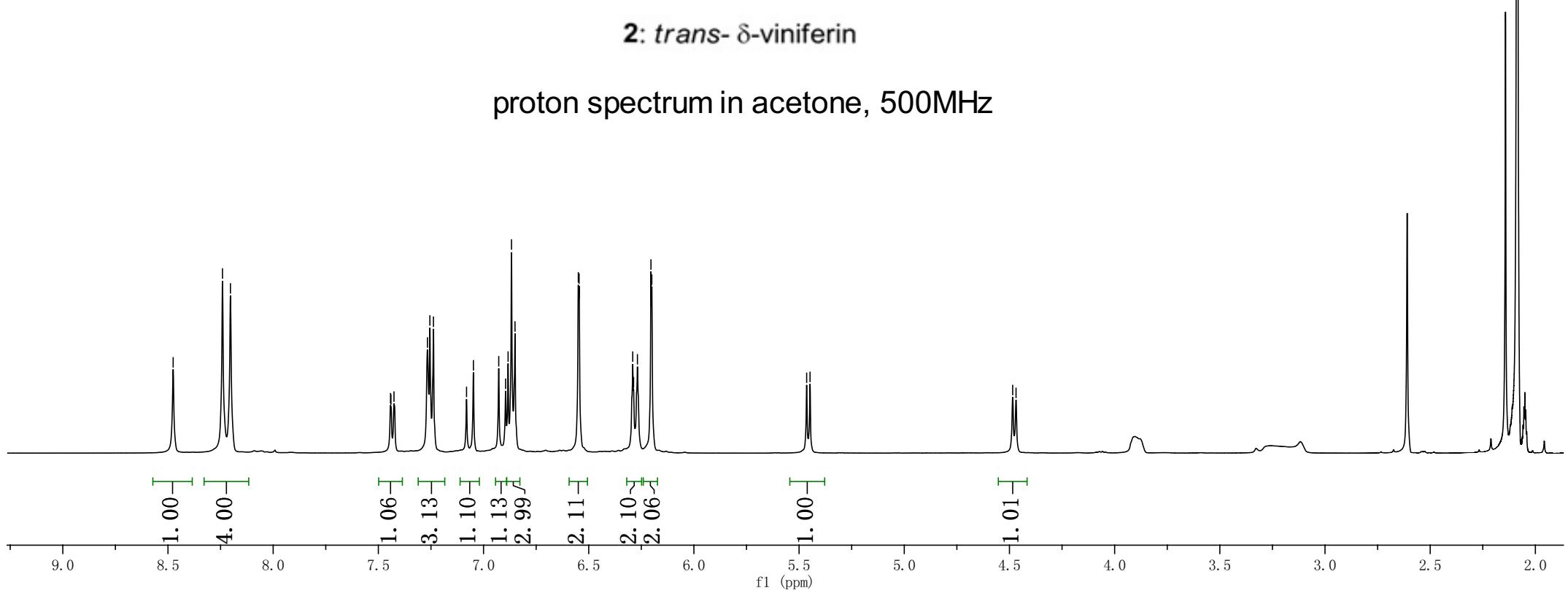
III. NMR Spectra of Synthesized Compounds

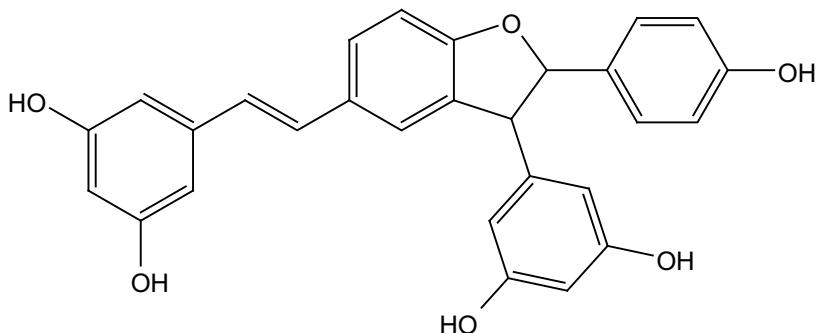
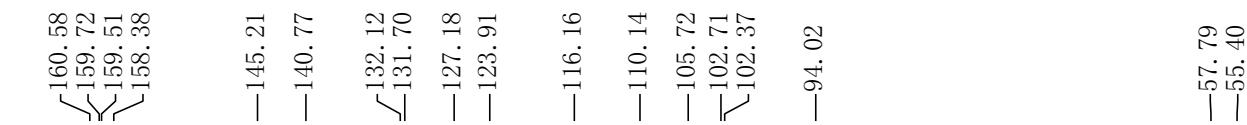
NMR Spectra of 2 (1H, 13C and DEPT-135)	7-9
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NMR Spectra of 6 (1H and 13C)	16-17
NMR Spectra of 7 (1H and 13C)	18-19



2: trans- δ -viniferin

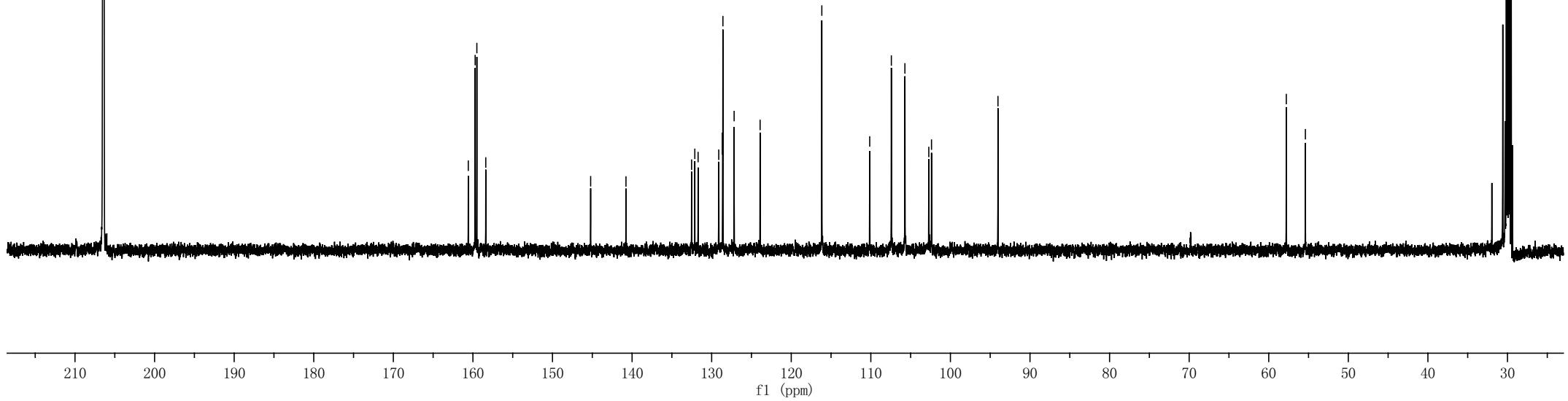
proton spectrum in acetone, 500MHz





2: *trans*- δ -viniferin

carbon spectrum in acetone, 125MHz



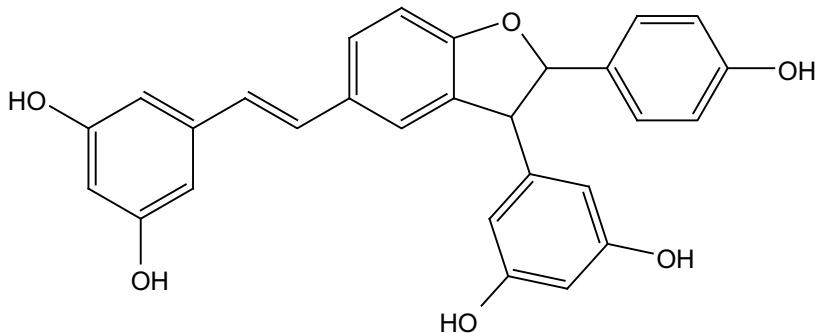
—129.11
—128.58
—127.17
—123.91

—116.15

—110.14
—107.42
—105.71
—102.71
—102.37

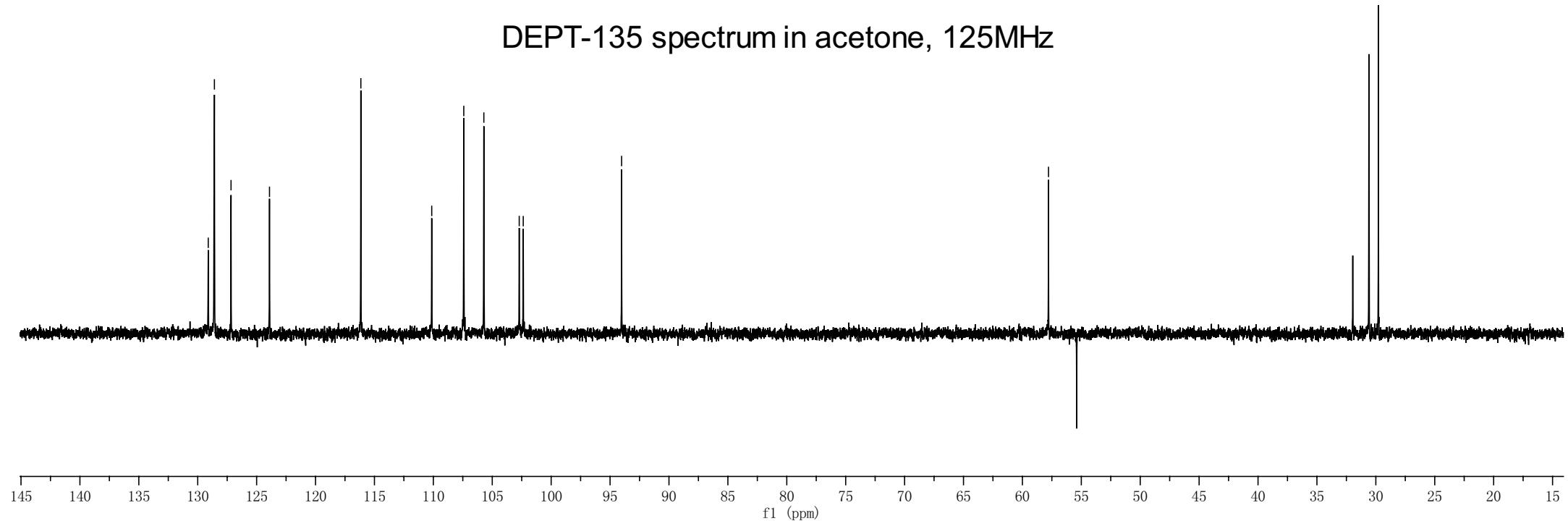
—94.02

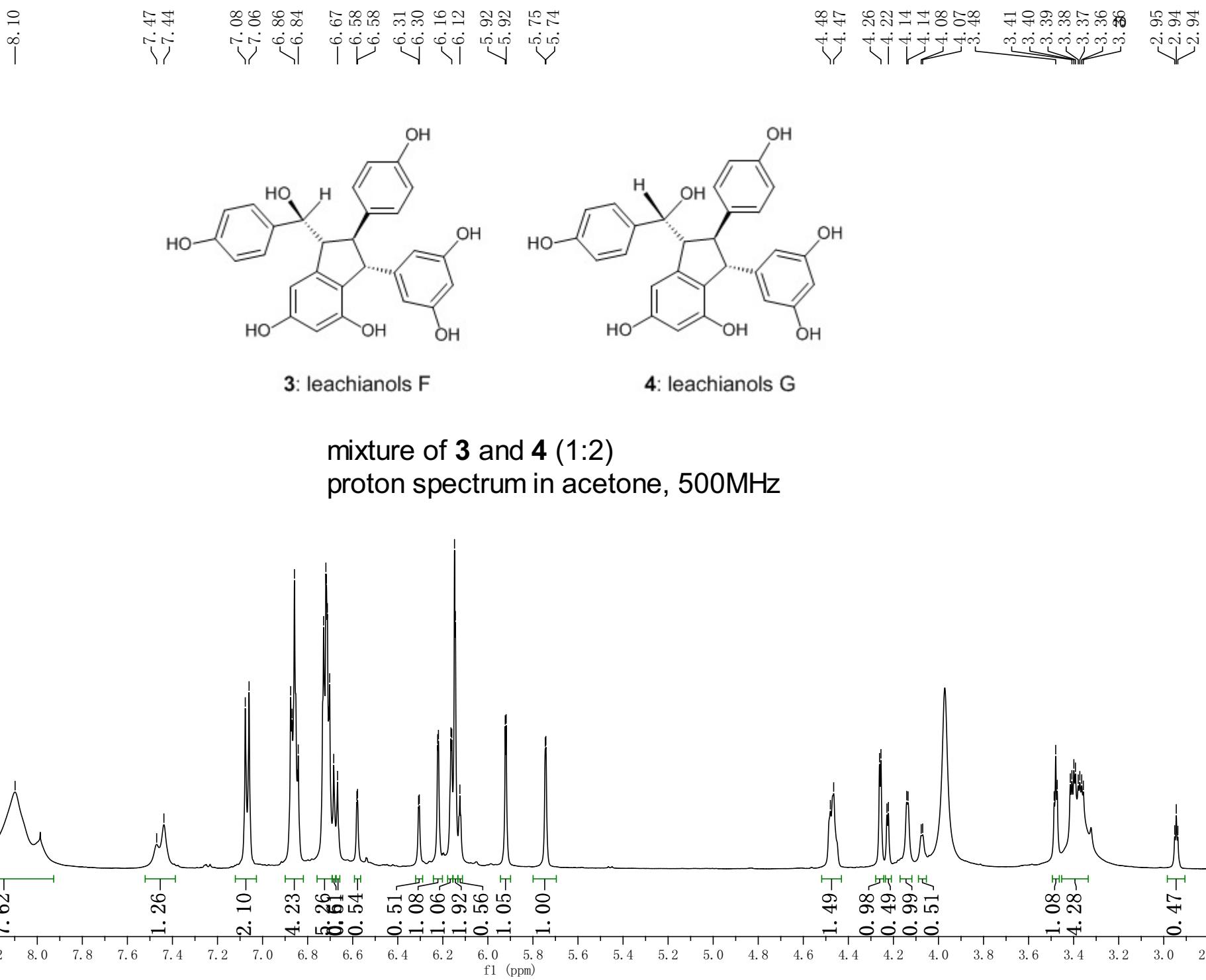
—57.78



2: *trans*- δ -viniferin

DEPT-135 spectrum in acetone, 125MHz

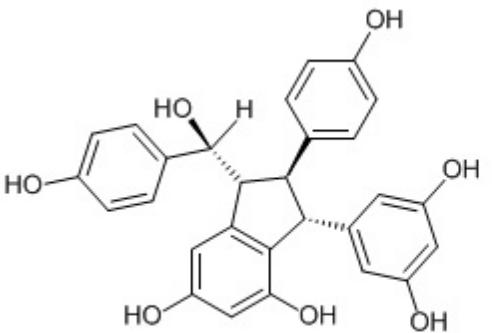




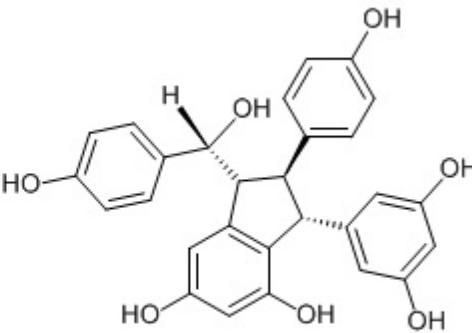
—159.20
—159.12
—158.49
—157.24
—156.18
—154.85
—151.43

—147.27

—137.97
—135.67
—129.43
—129.29
—129.26
—128.81
—123.09
—115.63
—115.59
—115.40
—115.25

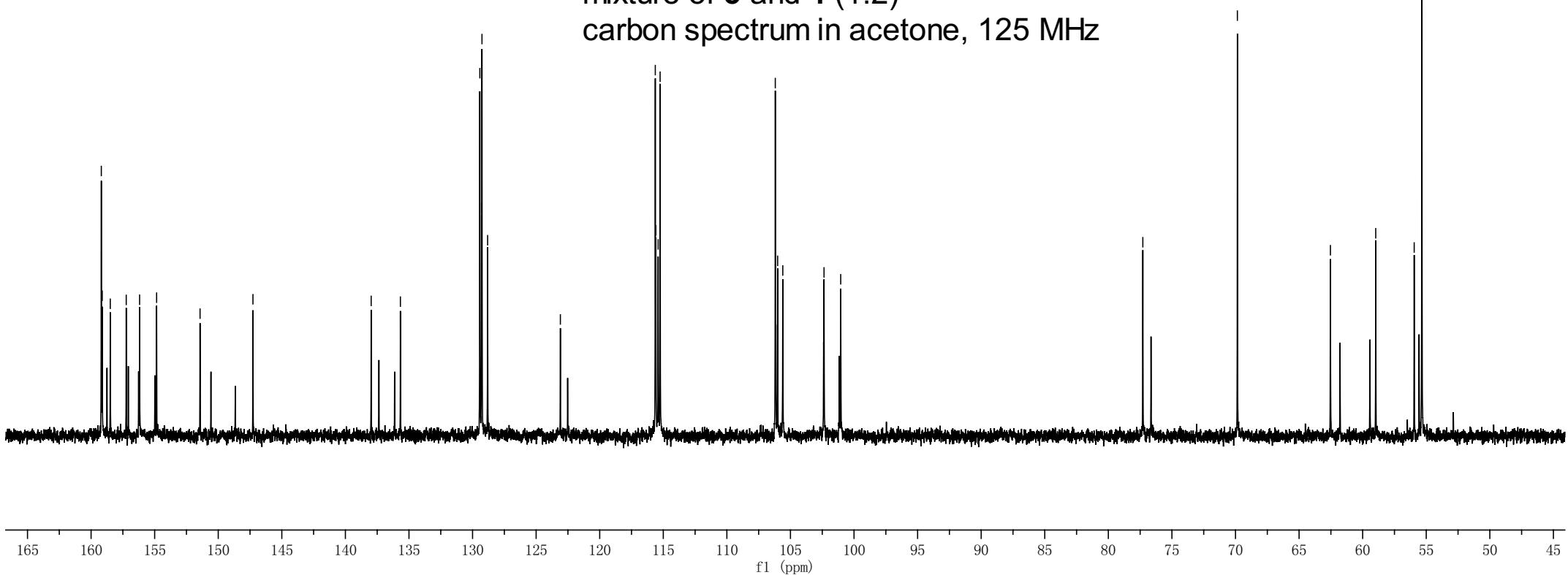


3: leachianols F



4: leachianols G

mixture of 3 and 4 (1:2)
carbon spectrum in acetone, 125 MHz



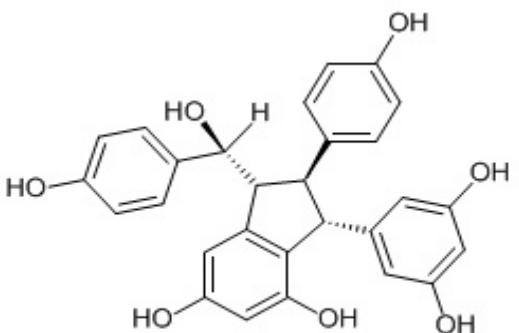
128.66
128.53
128.49
128.05

114.86
114.82
114.64
114.49

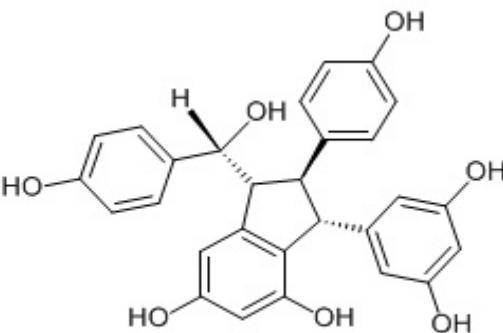
105.42
105.24
104.84
101.59
100.38
100.28

76.53
75.88

61.78
61.02
58.21
55.18
54.78

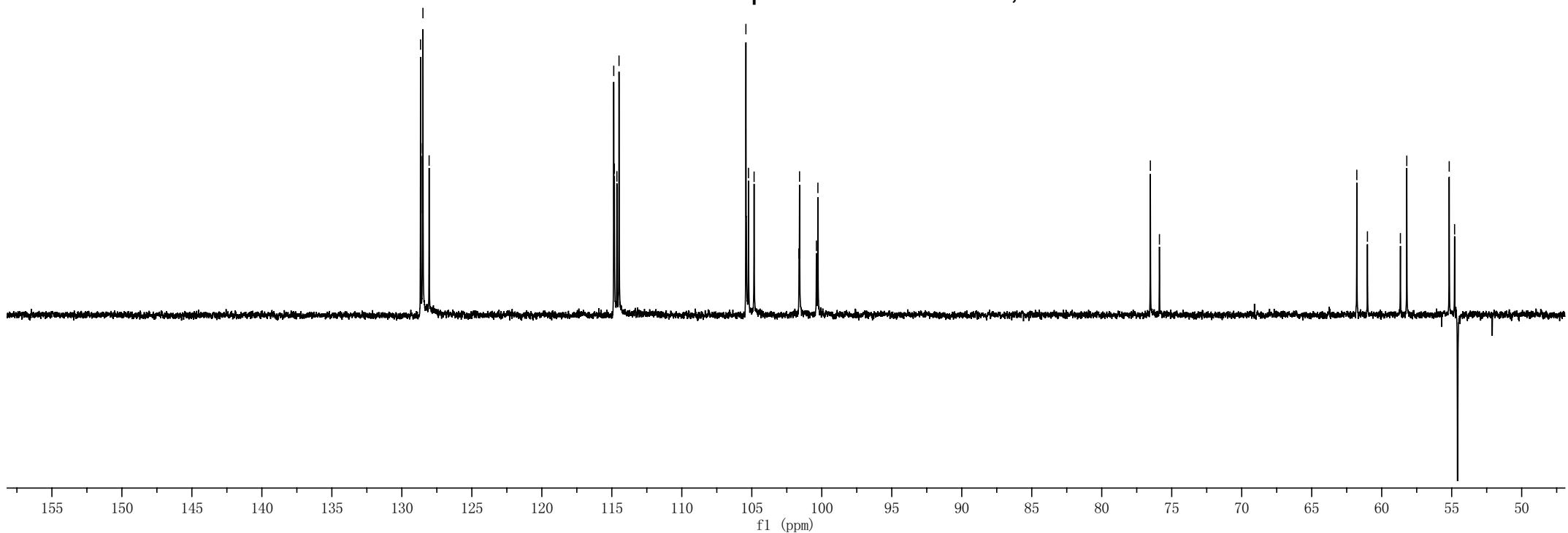


3: leachianols F



4: leachianols G

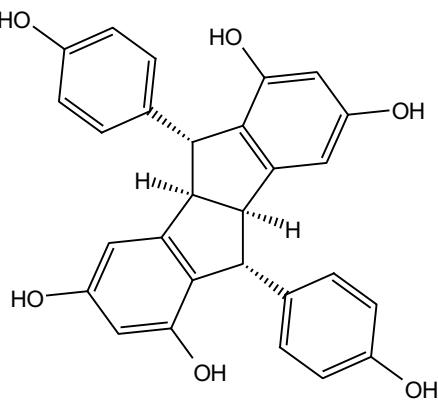
mixture of **3** and **4** (1:2)
DEPT-135 spectrum in acetone, 125 MHz



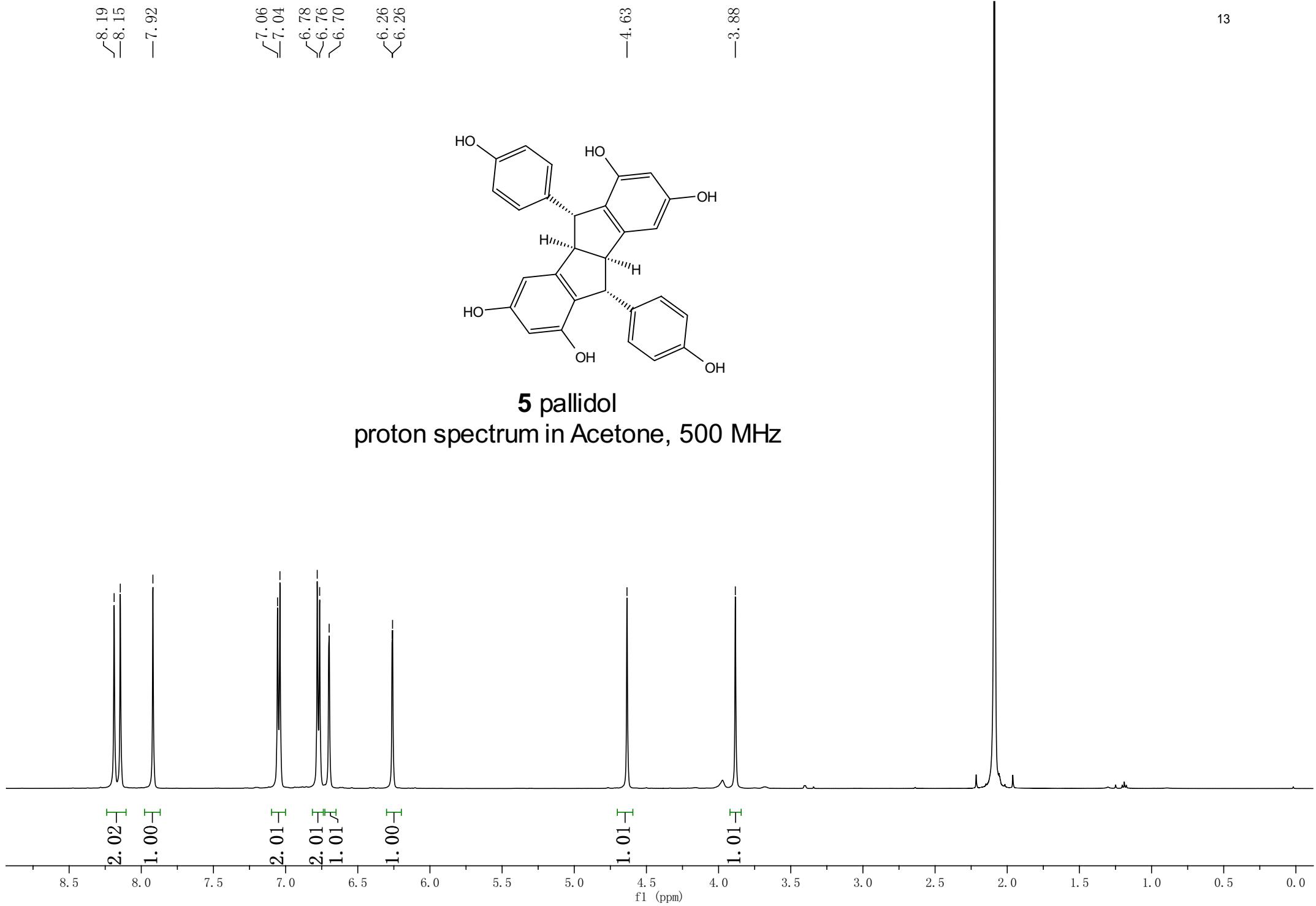
8.19
8.15
7.92

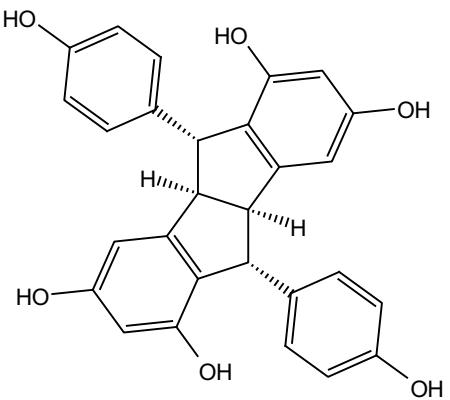
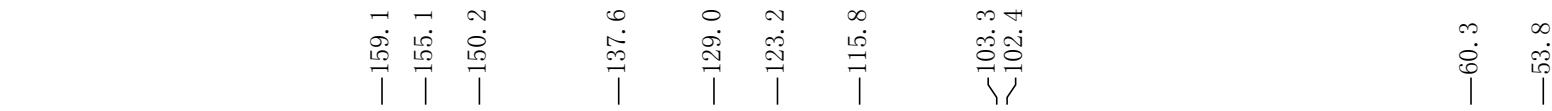
7.06
7.04
6.78
6.76
6.70
6.26
6.26

4.63
3.88

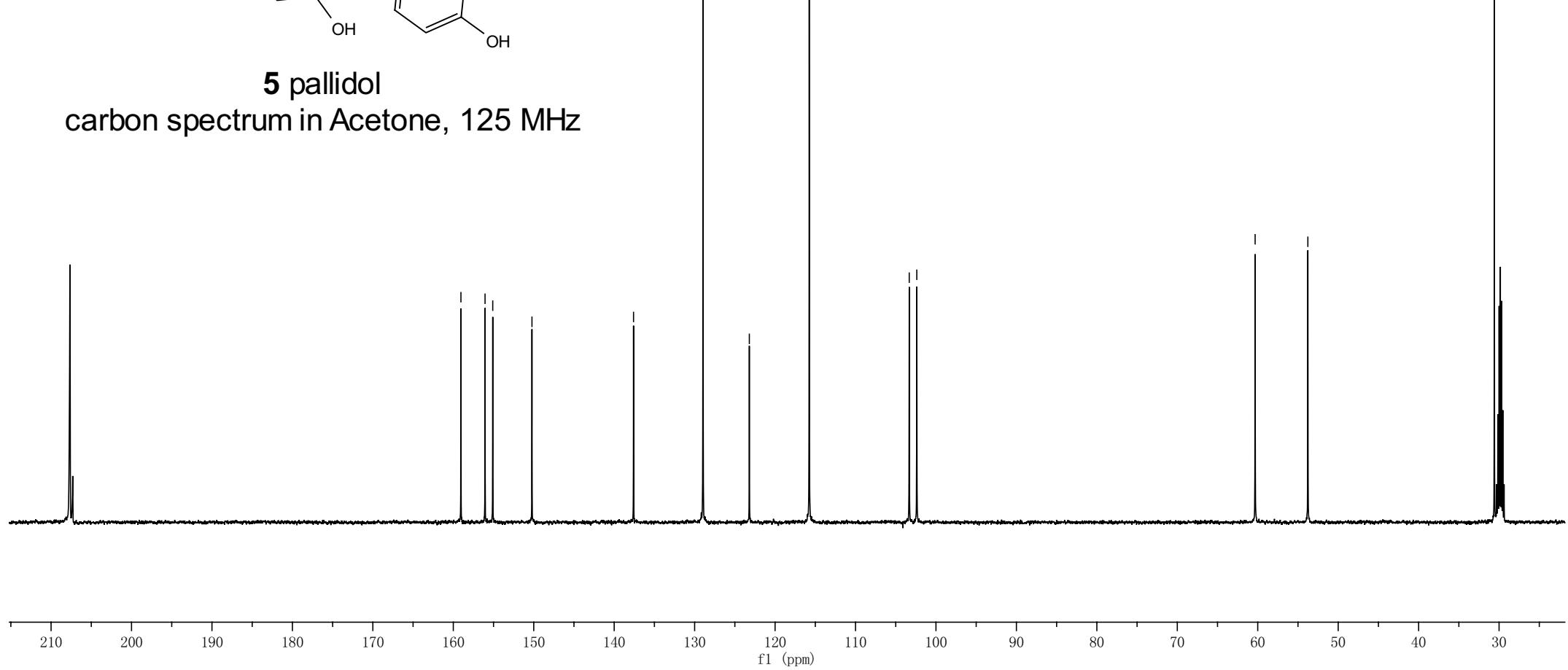


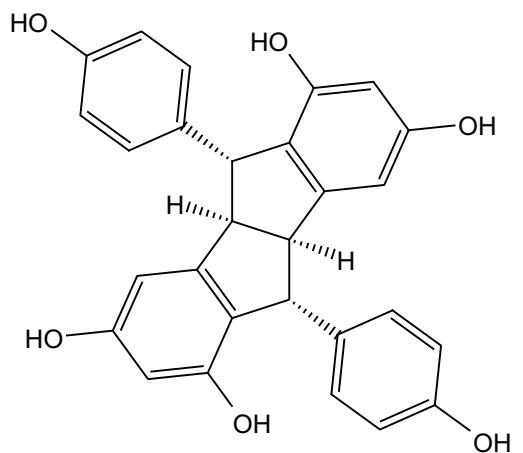
5 pallidol
proton spectrum in Acetone, 500 MHz



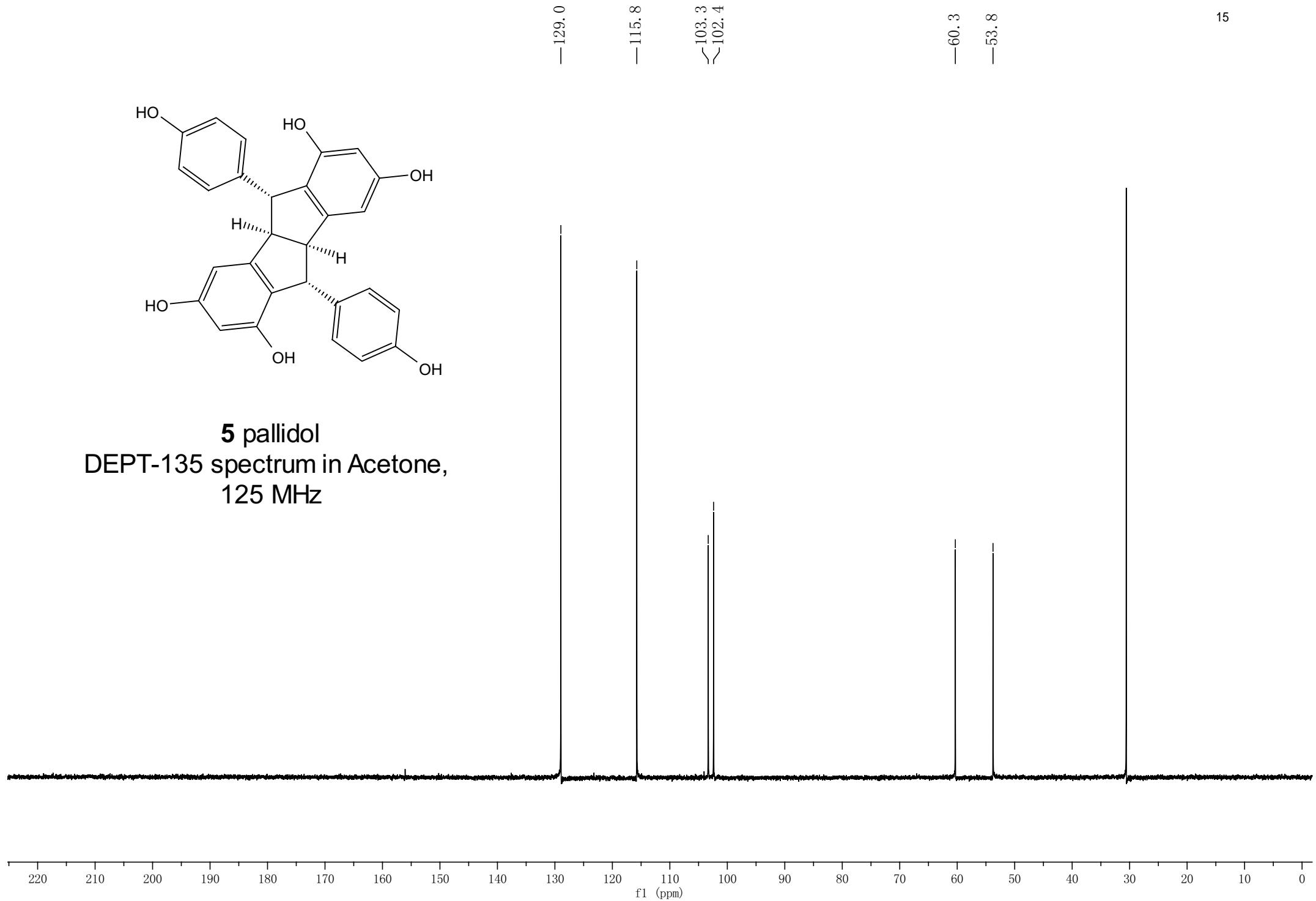


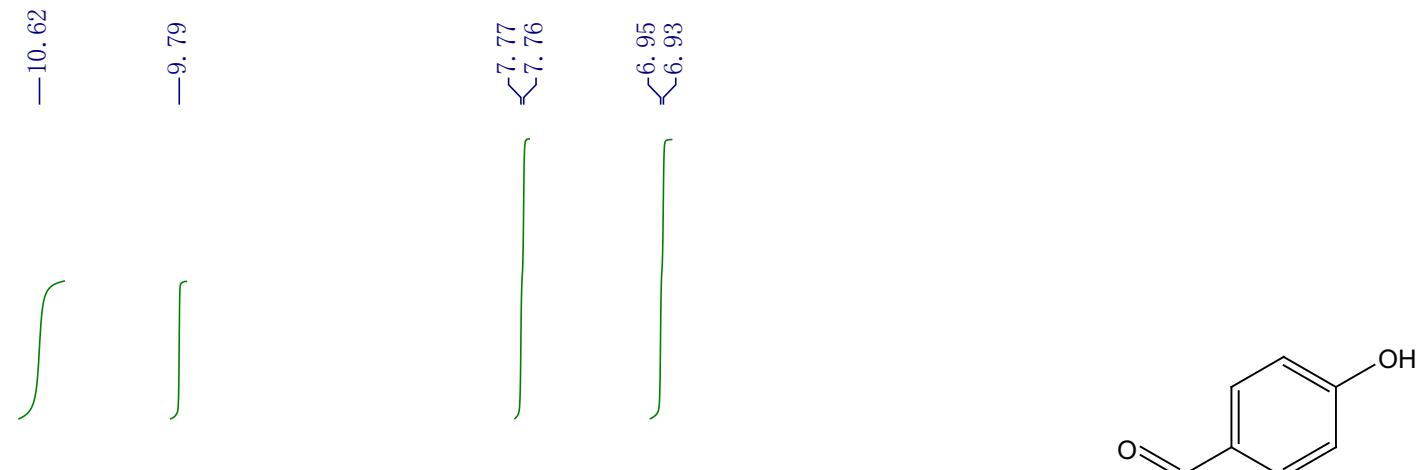
5 pallidol
carbon spectrum in Acetone, 125 MHz



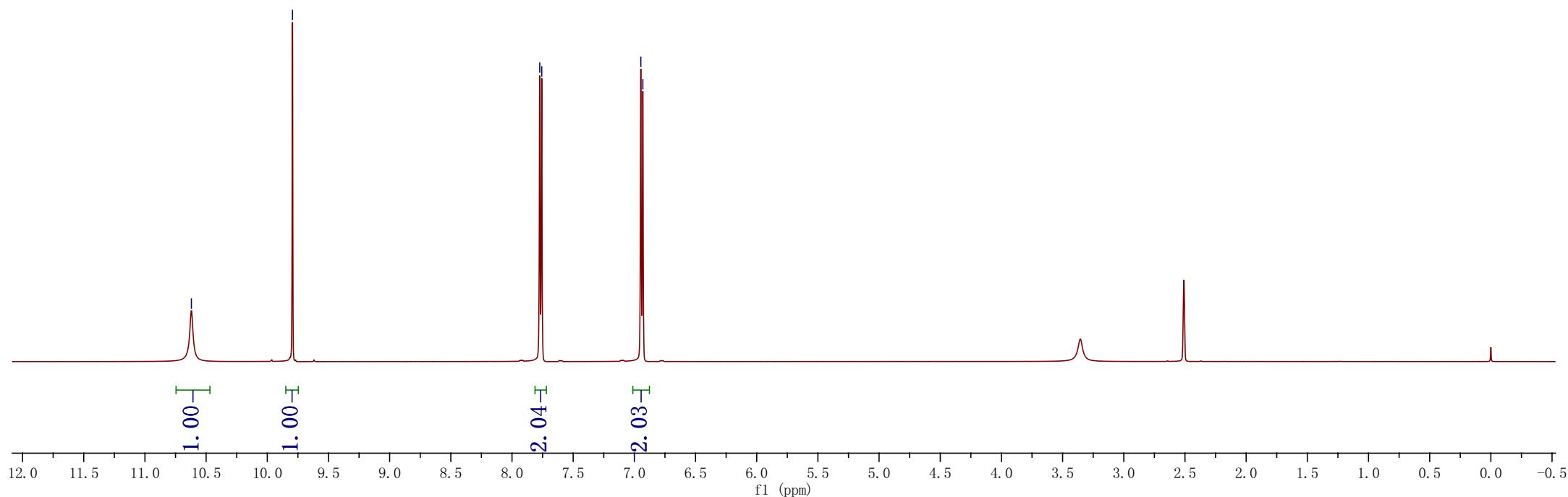


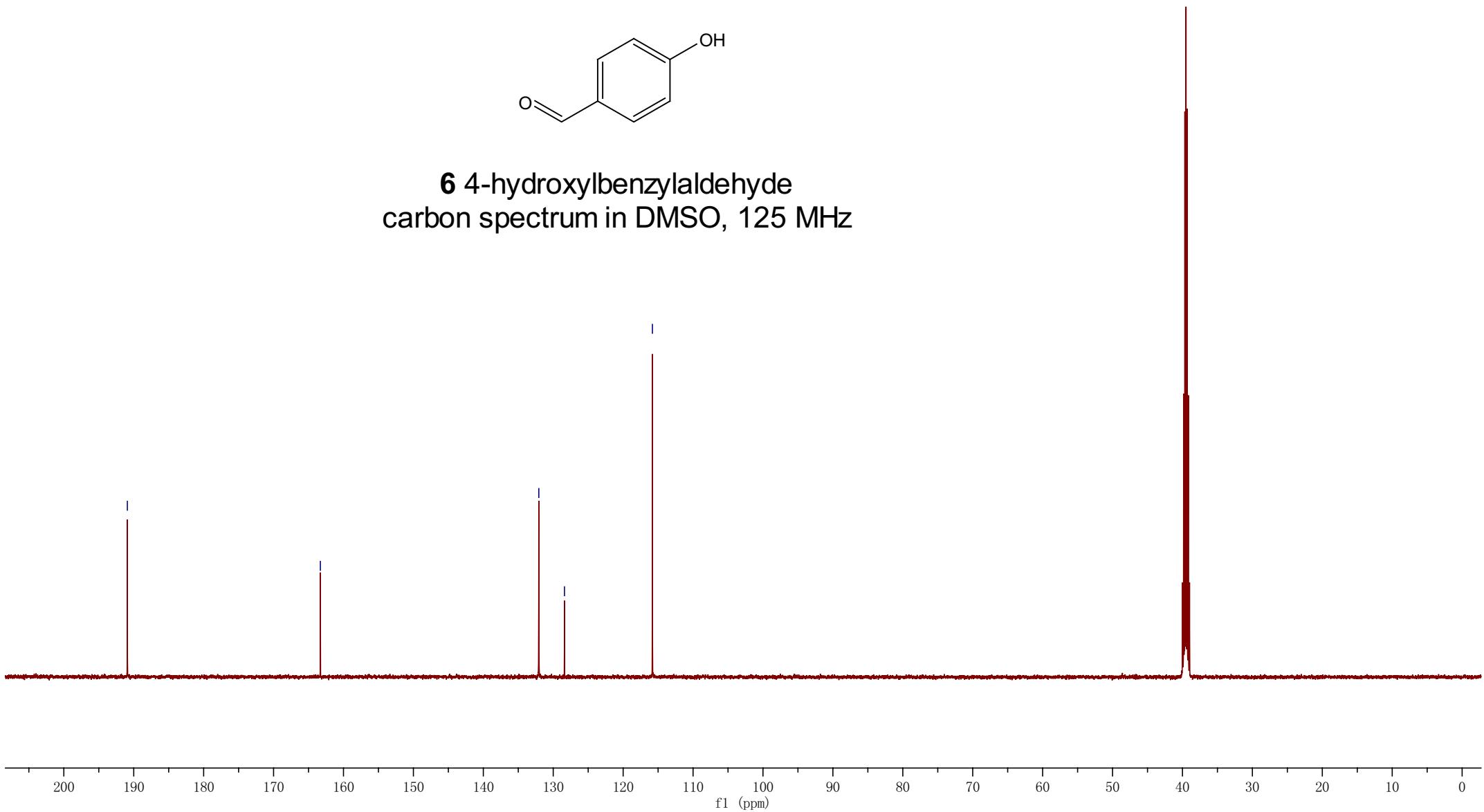
5 pallidol
DEPT-135 spectrum in Acetone,
125 MHz

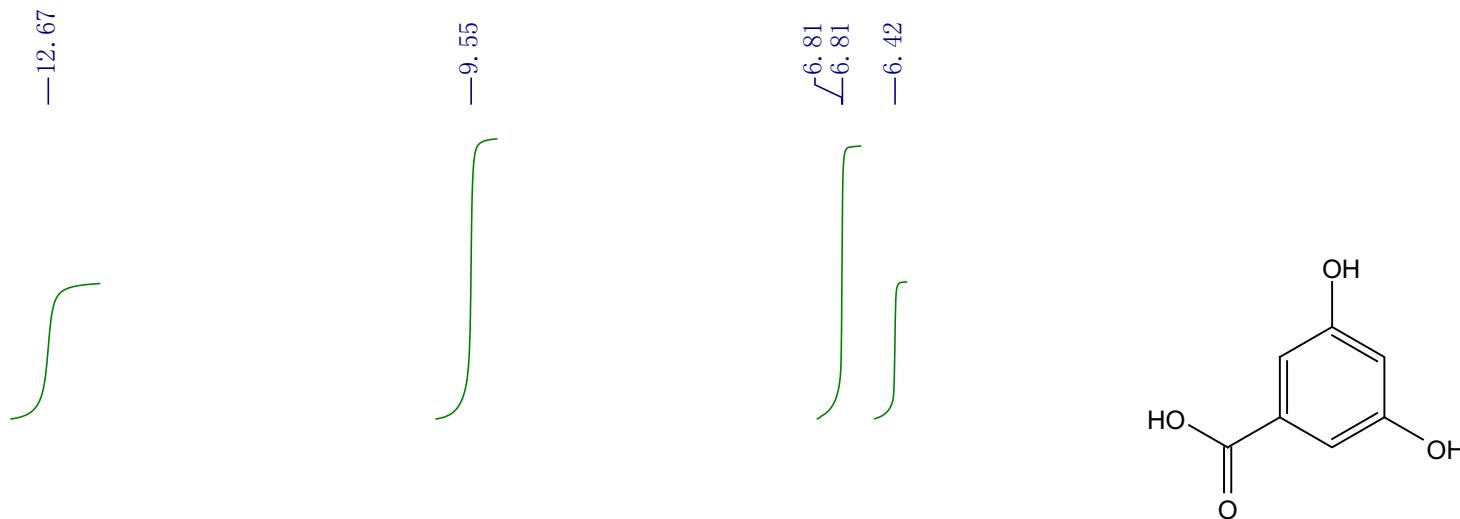




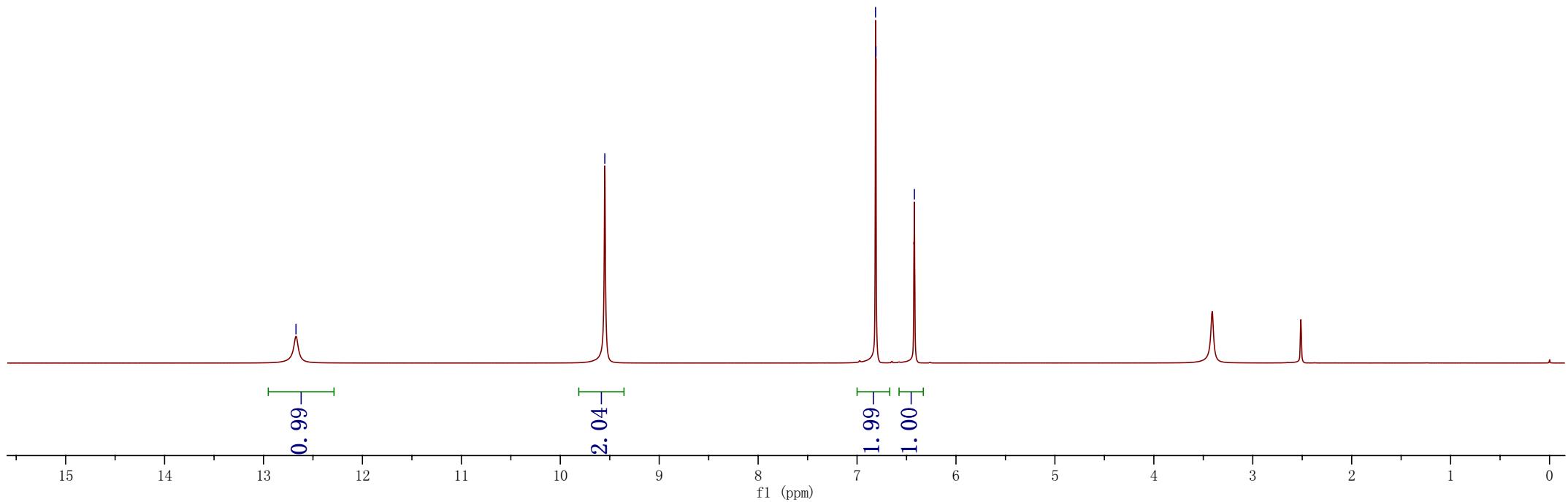
6 4-hydroxylbenzylaldehyde
proton spectrum in DMSO, 500 MHz





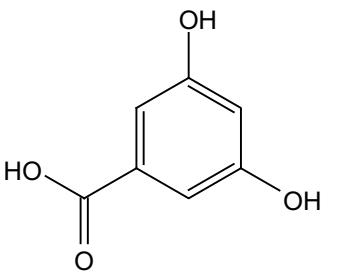


7 3,5-dihydroxylbenzylacid
proton spectrum in DMSO, 500 MHz



—167.32
—158.35

—132.46
 $\begin{array}{c} ^{107.26} \\ \diagdown \\ ^{106.75} \end{array}$



7 3,5-dihydroxybenzylacid
carbon spectrum in DMSO, 125 MHz

