Supporting Information for:

New organometallic ruthenium(II) complexes containing chelidonic acid (4-oxo-4*H*-pyran-2,6-dicarboxylic acid): Synthesis, structure and in vitro biological activity

Thangavel Sathiya Kamatchi,^a Palaniappan Kalaivani,^a Paramasivan Poornima,^b Viswanadha Vijaya Padma,^b Frank R. Fronczek,^c Karuppannan Natarajan^{*a}

^aDepartment of Chemistry, Bharathiar University, Coimbatore 641046, India. E-mail: k_natraj6@yahoo.com; Tel.: +91 422 2428319; Fax: +91 422 2422387.

^bDepartment of Biotechnology, Bharathiar University, Coimbatore 641 046, India.

^cDepartment of Chemistry, Louisiana State University, Baton Rouge, LA 70803, USA.

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Fig. S1 ORTEP diagram of 3 with hydrogen bonding



Fig. S2 Packing diagram of the unit cell for complex 3

Complex 3				
Interatomic distances (Å)				
Ru(1)–C(2)	2.033(2)			
Ru(1)–O(1)	2.143(2)			
Ru(1)–C(8)	1.845(2)			
Ru(1)-N(1)	2.119(2)			
Ru(1) - P(1)	2.3897(3)			
C(5) - C(4)	1.351(3)			
C(1) - C(2)	1.359(3)			
C(3) –O(6)	1.244(3)			
Bond angles(°)				
C(8)-Ru(1)-N(1)	94.53(9)			
C(8)-Ru(1)-P(1)	91.71(7)			
C(8)-Ru(1)-O(1)	174.35(8)			
C(8)-Ru(1)-C(2)	95.83(10)0			
C(2)-Ru(1)-O(1)	78.52(7)			
C(2)-Ru(1)-P(1)	91.30(6)			
C(2)-Ru(1)-N(1)	169.65(8)			
O(1)-Ru(1)-P(1)	88.45(5)			
O(1)-Ru(1)-N(1)	91.13(7)			
N(1)-Ru(1)-P(1)	88.39(6)			
P(1)-Ru(1)-P'(1)	175.48(2)			

Table S1 Selected bond lengths (Å) and angles (°) for ${\bf 3}$

Hydrogen-bonding distances (Å) and angles (°) for **3**

Complex	D-HA	D-H	НА	DA	D-HA
3	$O5$ — $H5$ ··· $O2^{i}$	0.85	1.79	2.639 (2)	180
	O1 <i>W</i> —H1 <i>W</i> ····O4	0.94 (5)	1.91 (5)	2.832 (3)	170 (4)
	O1 <i>W</i> —H2 <i>W</i> ···O2	1.03 (5)	1.85 (5)	2.867 (3)	167 (4)
a .	1 (1) 1/0				

Symmetry code: (i) -*x*, -*y*, *z*-1/2.



Fig. S3 The emission spectra of the DNA–EB system ($\lambda_{exc} = 515 \text{ nm}$, $\lambda_{em} = 530-750 \text{ nm}$), in the presence of **H**²**L** and complexes **1-4**. [DNA] = 10 μ M, [Complex] = 0–50 μ M, [EB] = 10 μ M. The arrow shows the emission intensity changes upon increasing complex concentration.



Fig. S4 Synchronous spectra of BSA (1 μ M) in the presence of increasing amounts of H^2L and complexes **1-4** (0 – 40 μ M) for a wavelength difference of $\Delta\lambda = 15$ nm. The arrow shows the emission intensity changes upon increasing concentration of compound



Fig. S5 Plausible mechanisms for DPPH radical scavenging activity