

Copper(I)-Catalyzed Homo-Coupling of Terminal Alkynes at Room Temperature under Solvent and Base Free Conditions Using O₂ as an Oxidant

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Page S2-S8: Materials and Methods.

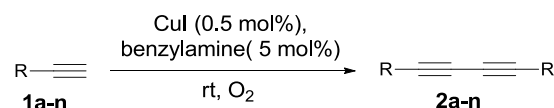
Page S9-S24: the ¹H NMR and ¹³C NMR spectra for compounds **2a-p**:

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Page S11	¹ H NMR, ¹³ C NMR of 2c
Page S12	¹ H NMR, ¹³ C NMR of 2d
Page S13	¹ H NMR, ¹³ C NMR of 2e
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Experimental Section

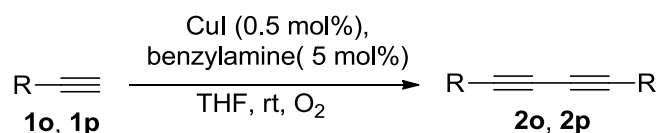
All reagents were used directly without further purification. Silica gel was purchased from Qing Dao Hai Yang Chemical Industry Co. All melting points were determined on a Beijing Science Instrument Dianguang Instrument Factory XT4B melting point apparatus and are uncorrected. ^1H and ^{13}C NMR spectra were recorded at 400 and 100 MHz for CDCl_3 solutions. HRMS-ESI spectra were obtained on Agilent 6450 spectrometer. IR data were recorded on a Nicolet iS10 spectrometer. The products listed below were determined by ^1H , ^{13}C NMR. PE is petroleum ether (60–90 °C).

General Procedure for the Preparation of symmetric 1,3-diynes **2a–n**:



A mixture of CuI (2.9 mg, 0.015 mmol), benzylamine (16 mg, 0.15 mmol) and alkyne **1** (3 mmol) was stirred at room temperature under 1 atm O_2 for 12–48 h. After **1** was exhausted completely (monitored by TLC or by GC for **2l**, **2m**), the crude products was purified by chromatography (silica gel, 1% EtOAc in PE or 20% EtOAc in PE for **2j**) to give **2**.

General Procedure for the Preparation of symmetric 1,3-diynes **2o**, **2p**:



A mixture of CuI (2.9 mg, 0.015 mmol), benzylamine (16 mg, 0.15 mmol) and alkyne **1** (3 mmol) in THF (3 mL) was stirred at room temperature under 1 atm O_2 for 16–20 h. After **1** was exhausted completely (monitored by TLC), the crude products was purified by chromatography (silica gel, 30% EtOAc in PE) to give **2**.

In 2008, Mizuno et al. reported that monomeric dicopper-substituted silicotungstate was an effective oxidative alkyne homo-coupling catalyst with a TON of 468, which is the highest one among copper-catalyzed oxidative homocoupling catalysts (see the reference 1 of Table S1). Other results reported after 2008 are listed in Table S1.

Table S1. Results of the Homocoupling Reaction of Phenylacetylene

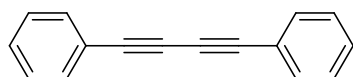
Catalyst (mol%)	Additive (mol%)	Oxidant	Temp. (°C)	Time (h)	Yield (%)	TON ^a	Ref.
CuI (0.05)	Benzylamine (5)	O ₂	r.t.	24	88	1760	This work
		air	r.t.	48	67	1340	
Cu-Silicotungstate (0.2)	–	O ₂	100	18	93	468	1
CuI (5)	Na ₂ CO ₃ (200)	I ₂	80	24	99	20	2
CuCl (2)	DBU ^b (100)	O ₂	r.t.	18	99	50	3
	TMEDA ^c (1.5)						
Pd(NH ₃) ₂ Cl ₂ (1)	Et ₃ N (300)	I ₂	r.t.	3	99	99	4
CuI (1)	TBAB ^d (50)						
CuI (50)	DIPEA ^e (50)	NBS	r.t.	2	93	2	5
CuNPs (400)	–	–	66	8	90	<1	6
Cu(OH) _x /TiO ₂ (5)	–	O ₂	100	0.5	90	18	7
NiCl ₂ 6H ₂ O (5)	Et ₃ N (300)	air	r.t.	20	99	20	8
CuI (5)	TMEDA (20)						
Cu(OAc) ₂ H ₂ O (0.2)	Piperidine (100)	air	r.t.	3	60	299	9
Cu-Zeolite (30)	–	air	110	15	97	3	10
Fe(acac) ₃ (10)	K ₂ CO ₃ (200)	air	50	8	94	940	11
Cu(acac) ₂ (0.1)							
CuCl ₂ (2)	Et ₃ N (3)	air	60	6	96	48	12
CuCl (2)	Piperidine (10)	air	60	5	96	48	13
CuNPs/TiO ₂ (1)	Piperidine (30)	air	65	6	96	96	14
CuCl ₂ (10)	DBU (120)	air	r.t.	24	92	9	15
Cu(OAc) ₂ H ₂ O (20)	–	air	90	10	90	5	16
CuCl (10)	TMEDA (10)	air	r.t.	1	93	9	17
CuCl ₂ H ₂ O (10)	NaOAc (100)	O ₂	120	1.5	99	10	18
Cu(OH) _x /OMS-2 (2)	–	O ₂	100	0.17	99	50	19
						666 ^f	
Cu(OAc) ₂ (5)	AFIL 1 (10) ^g	O ₂	60	6	99	20	20
	NFIS 1 (10) ^h						
CuI (5)	KF/Al ₂ O ₃ (60 w%)	air	r.t.	0.17	99	19	21
Cu ₂ SO ₄ 5H ₂ O (5)	KOAc (100)	I ₂	120	24	99	20	22
CuCl (5)	–	air	90	7	96	19	23
CuI (5)	NaOAc (100)	air	90	30	97	19	24
CuI (5)	Et ₃ N (300)	air	r.t.	20	91	18	25
	TMEDA (10)						
Cu(OAc) ₂ H ₂ O (10)	–	air	160	7	85	9	26
CuI complex (1)	Piperidine (10)	air	r.t.	2	94	94	27
CuCl (10)	TMEDA (30)	air	20	1.5	97	10	28

^a TON = (phenylacetylene consumed (mol))/(Cu catalyst used (mol)). ^b DBU = 1,8-diazabicyclo[5.4.0]undec-7-ene. ^c TMEDA = *N,N,N',N'*-tetramethylethylenediamine. ^d TBAB = tetrabutyl ammonium bromide. ^e DIPEA = diisopropylethylamine. ^f For 14 repeated runs. ^g Amine-functionalized ionic liquid. ^h 1-Methyl-3-(4-nitrobenzyl)imidazolium hexafluorophosphate.

Reference of Table S1:

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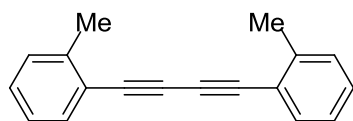
Data of diynes



2a

1,4-Diphenyl-1,3-butadiyne (2a)¹:

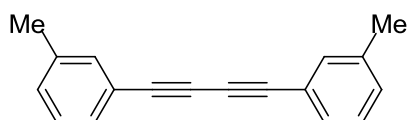
¹H NMR (400 MHz, CDCl₃) δ 7.60-7.46 (m, 4H), 7.45-7.27 (m, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 132.5, 129.2, 128.4, 121.8, 81.6, 73.9.



2b

1,4-Bis(2-methylphenyl)-1,3-butadiyne (2b)¹:

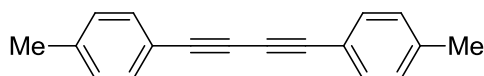
¹H NMR (400 MHz, CDCl₃) δ 7.49 (d, *J* = 7.6 Hz, 2H), 7.30-7.05 (m, 6H), 2.48 (s, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 141.6, 132.9, 129.5, 129.1, 125.6, 121.7, 81.1, 77.5, 20.7.



2c

1,4-Bis(3-methylphenyl)-1,3-butadiyne (2c)¹:

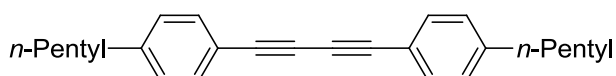
¹H NMR (400 MHz, CDCl₃) δ 7.36-7.28 (m, 4H), 7.25-7.12 (m, 4H), 2.32 (s, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 138.1, 132.9, 130.1, 129.6, 128.3, 121.6, 81.6, 73.6, 21.2.



2d

1,4-Bis(4-methylphenyl)-1,3-butadiyne (2d)¹:

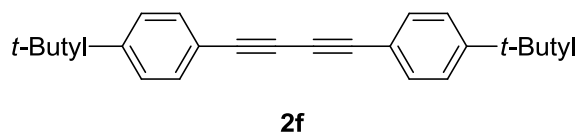
¹H NMR (400 MHz, CDCl₃) δ 7.41 (d, *J* = 8.0 Hz, 4H), 7.13 (d, *J* = 8.1 Hz, 4H), 2.36 (s, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 139.5, 132.4, 129.2, 118.8, 81.5, 73.4, 21.6.



2e

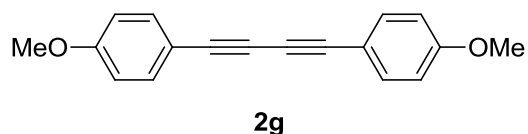
1,4-Bis(4-*n*-pentylphenyl)-1,3-butadiyne (2e)²:

^1H NMR (400 MHz, CDCl_3) δ 7.42 (d, J = 7.7 Hz, 4H), 7.12 (d, J = 7.9 Hz, 4H), 2.66-2.52 (m, 4H), 1.65-1.53 (m, 4H), 1.37-1.25 (m, 8H), 0.88 (m, 6H); ^{13}C NMR (101 MHz, CDCl_3) δ 144.4, 132.4, 128.5, 119.0, 81.6, 73.5, 35.9, 31.4, 30.8, 22.5, 14.0.



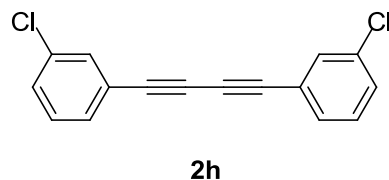
1,4-Bis(4-*t*-butylphenyl)-1,3-butadiyne (2f)³:

^1H NMR (400 MHz, CDCl_3) δ 7.50-7.42 (m, 4H), 7.40-7.30 (m, 4H), 1.31 (s, 18H); ^{13}C NMR (101 MHz, CDCl_3) δ 152.5, 132., 125.5, 118.8, 81.5, 73.5, 34.9, 31.0.



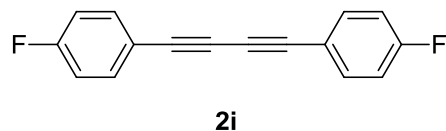
1,4-Bis(4-methoxyphenyl)-1,3-butadiyne (2g)¹:

^1H NMR (400 MHz, CDCl_3) δ 7.45 (d, J = 8.6 Hz, 4H), 6.84 (d, J = 8.6 Hz, 4H), 3.81 (s, 6H); ^{13}C NMR (101 MHz, CDCl_3) δ 160.2, 134.0, 114.1, 113.9, 81.2, 72.9, 55.3.



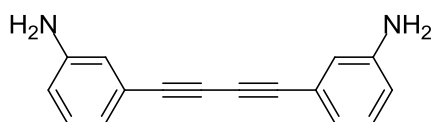
1,4-Bis(3-chlorophenyl)-1,3-butadiyne (2h)¹:

^1H NMR (400 MHz, CDCl_3) δ 7.49 (s, 2H), 7.43-7.36 (m, 2H), 7.36-7.32 (m, 2H), 7.25 (dd, J = 10.0, 5.8 Hz, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 134.3, 132.2, 130.6, 129.7, 123.2, 80.5, 74.7.



1,4-Bis(4-fluorophenyl)-1,3-butadiyne (2i)¹:

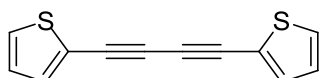
^1H NMR (400 MHz, CDCl_3) δ 7.56-7.47 (m, 4H), 7.08-6.98 (m, 4H); ^{13}C NMR (101 MHz, CDCl_3) δ 163.0 (d, J = 251.6 Hz), 134.5 (d, J = 8.7 Hz), 117.8 (d, J = 3.6 Hz), 115.9 (d, J = 22.4 Hz), 80.4, 73.5.



2j

1,4-Bis(3-aminophenyl)-1,3-butadiyne (2j)⁴:

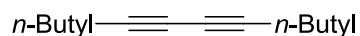
¹H NMR (400 MHz, CDCl₃) δ 7.10 (t, *J* = 7.8 Hz, 2H), 6.97-6.89 (m, 2H), 6.84-6.77 (m, 2H), 6.68 (ddd, *J* = 8.0, 2.4, 0.9 Hz, 2H), 3.67 (s, 4H); ¹³C NMR (101 MHz, CDCl₃) δ 146.3, 129.3, 122.9, 122.4, 118.4, 116.2, 81.6, 73.4.



2k

1,4-Bis(4-thienylphenyl)-1,3-butadiyne (2k)²:

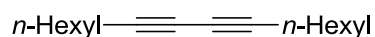
¹H NMR (400 MHz, CDCl₃) δ 7.33 (d, *J* = 3.6 Hz, 2H), 7.30 (d, *J* = 5.1 Hz, 2H), 7.01-6.96 (m, 2H); ¹³C NMR (101 MHz, CDCl₃) δ 134.4, 128.9, 127.2, 121.8, 77.8, 76.6.



2l

5,7-Dodecadiyne (2l)⁵:

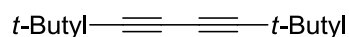
¹H NMR (400 MHz, CDCl₃) δ 2.18 (t, *J* = 6.9 Hz, 4H), 1.49-1.28 (m, 8H), 0.84 (t, *J* = 7.2 Hz, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 77.3, 65.2, 30.4, 21.9, 18.8, 13.5.



2m

7,9-Hexadecadiyne (2m)¹:

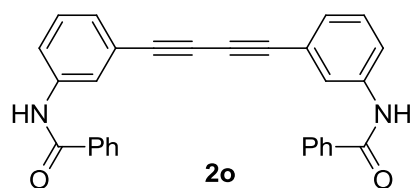
¹H NMR (400 MHz, CDCl₃) δ 2.28-2.06 (m, 4H), 1.60-1.07 (m, 16H), 0.96-0.74 (m, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 77.3, 65.3, 31.3, 28.5, 28.3, 22.5, 19.1, 13.9.



2n

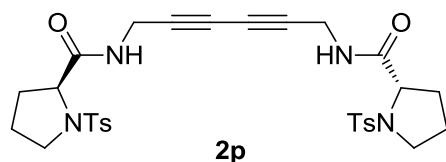
2,2,7,7-tetramethylocta-3,5-diyne (2n)⁵:

¹H NMR (400 MHz, CDCl₃) δ 7.73 (s, 18H); ¹³C NMR (101 MHz, CDCl₃) δ 86.2, 63.7, 30.6, 27.9.



***N,N'*-(buta-1,3-diyne-1,4-diylbis(3,1-phenylene))dibenzamide (2o):**

gray solid; m.p. 265-266 °C; IR (KBr): ν 3287, 1645, 1580, 1424, 1307, 1534, 693 cm^{-1} ; ^1H NMR (400 MHz, DMSO) δ 10.38 (s, 2H), 8.14-7.76 (m, 8H), 7.71-7.19 (m, 10H); ^{13}C NMR (101 MHz, DMSO) δ 166.3, 140.1, 136.0, 132.3, 129.9, 128.9, 128.2, 124.0, 122.5, 121.0, 82.3, 73.7; HRMS m/z (ESI) calcd for $\text{C}_{30}\text{H}_{21}\text{N}_2\text{O}_2(\text{M} + \text{H})^+$ 441.1603, found 441.1601.



(2*S*,2'*S*)-*N,N'*-(hexa-2,4-diyne-1,6-diyl)bis(1-tosylpyrrolidine-2-carboxamide) (2p):

white solid; m.p. 161-162 °C; $[\alpha]_{\text{D}}^{20} = -100^\circ \text{cm}^3 \text{g}^{-1} \text{dm}^{-1}$ ($c = 0.2 \text{ g cm}^{-3}$ in CHCl_3); IR (KBr): ν 3340, 1661, 1519, 1347, 666 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.73 (d, $J = 8.2 \text{ Hz}$, 4H), 7.36 (d, $J = 8.0 \text{ Hz}$, 4H), 7.20 (t, $J = 5.0 \text{ Hz}$, 2H), 4.27 (dd, $J = 17.9, 6.2 \text{ Hz}$, 2H), 4.12-4.02 (m, 4H), 3.58 (m, 2H), 3.20 (m, 2H), 2.45 (s, 6H), 2.24-2.11 (m, 2H), 1.77 (m, 2H), 1.66-1.51 (m, 4H); ^{13}C NMR (101 MHz, CDCl_3) δ 170.9, 144.5, 132.7, 130.0, 127.8, 73.9, 67.5, 62.4, 49.8, 29.9, 29.8, 24.3, 21.5; HRMS m/z (ESI) calcd for $\text{C}_{30}\text{H}_{34}\text{N}_4\text{O}_6\text{S}_2\text{Na}(\text{M} + \text{Na})^+$ 633.1817, found 633.1816.

References

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