Supporting Information

for

Noncovalent Functionalization of Electrospun Fibrous Network with





Figure S1. a. Chemical structure of PMP b. Mass spectrum of PMP after substracting mass spectra of water sample at that time interval. Mass data $[M-H]^-$ (calculated) =1116.43; $[M-H]^-$ (observed) = 1115.37 c. Liquid chromatogram of PMP after dialysis.



Figure S2. SEM images of PMP-CDNF a.before and after incubation in b. Cd, c. Ni and d. Cr solutions (scale bars: $10 \ \mu m$).

Table S1. Atom percentages of PMP-CDNF in terms of XPS spectrum of PMP-CDNF.

O1s	C1s	N1s	P2p	S2p
26.97	61.72	5.9	3.38	2.04

Table S2. Atom weight percentages of PMP, CDNF, and CDNF treated with TCEP and TRISand PMP-CDNF in terms of CHNS-O analyzer.

Compound	wt % C	wt % H	wt % N	wt % S
PMP	46.119± 0.5075	5.8676± 0.2769	11.156± 0.1063	7.9138± 0.0842
CDNF	46.888±0.648 3	6.7914± 0.1449	0	0
PMP-CDNF	43.572±0.283 9	6.4800± 0.0123	1.2238 ± 0.0546	0.4751 ± 0.0349
CDNF treated with TCEP&TRIS	44.670±1.089 4	6.8153±0.0326	0.8670 ± 0.2824	0

Table S3. Comparison of experimental and theoretical [C]/[S], [N]/[S], [H]/[S], and [C]/[N] of PMP.

Compound	[C]/[S]	[N]/[S]	[H]/[S]	[C]/[N]
Experimental- PMP	15.540	3.2222	23.726	4.823
Theoretical - PMP	15.333	3.3333	23.000	4.6000
% Difference	1.0135	0.9667	1.0315	1.0484

$\mathbf{PMP}^{[a,b,c]}(\mathbf{g})$	CDNF ^[c] (g)	$\mathbf{PMP}^{[a,c]}$ (mol)	CDNF ^[c] (mol)
5.5080	94.492	0.0049	0.0647

Table S4. PMP and HPβCD amount in 100 g PMP-CDNF.

^aCalculated in terms of S amount in PMP immobilized fiber and experimental S content in PMP elemental analysis. ^bCalculated based on deprotonated PMP molecular weight. ^cAmount of molecules in 100 g PMP-CDNF.



Figure S3. Absorbance of PMP in 50 mM TRIS buffer at pH 8.0.



Figure S4. a. Isothermal titration curve of HP β CD molecule with CdCl₂ solution. XPS spectra of b. PMP-CDNF and c. CDNF after incubation in Cd^{II} solution.



Figure S5. Isothermal titration curve of a. PMP and b. HP β CD molecule with Ni(NO₃)₂ solution. c. Absorption change in 20 μ M Ni(NO₃)₂ titration with PMP solution in 50 mM TRIS buffer at pH 8.0. d. XPS spectrum of PMP-CDNF after incubation in Ni^{II} solution.



Figure S6. Isothermal titration curve of a. PMP and b. HP β CD molecule with K₂Cr₂O₇ solution. c. Absorption change in 20 μ M K₂Cr₂O₇ titration with PMP solution in 50 mM TRIS buffer at pH 8.0. XPS spectra of d. PMP-CDNF after incubation in Cr^{VI} solution.



Figure S7. The amount of metal ions a. Cd, b. Ni and c. Cr in μ mol concentrations bound to per mg of the PMP-CDNF system and bare CDNF from different metal solutions.