

Electronic Supplementary information

Structure and thermal properties of composites with RE-borohydrides (RE = La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Er, Yb or Lu) and LiBH₄

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Table S1 The refined unit cell dimensions, RE-B and RE-Cl distances in the LiRE(BH₄)₃Cl phases, RE = La, Ce, Pr, Nd, at RT. Estimated standard deviations in parenthesis.

Atom in LiRE(BH ₄) ₃ Cl	Unit cell dimension, <i>a</i> (Å)	RE-B distance, (Å)	RE-Cl distance, (Å)
La	11.6853(1)	2.714(7)	2.978(1)
Ce	11.6243(1)	2.720(8)	2.952(2)
Pr	11.5784(1)	2.72(1)	2.932(2)
Nd	11.5480(2)	2.69(3)	2.926(5)

Table S2 Experimentally found IR active modes for the different LiRE(BH₄)₃Cl phases

	RE = La	RE = Ce	RE = Pr	RE = Nd
B-H bending	1176	1175	1176	1179
	1202	1202	1203	1206
B-H stretching	2325	2324	2324	2322
	2422	2420	2416	2427
	2446	2447	2443	2452

Table S3 Crystallographic data for LiLa(BH₄)₃Cl at RT from SR-PXD data

Phase data						
Compound	LiLa(BH ₄) ₃ Cl					
Space group	<i>I</i> -43 <i>m</i> (217)					
Unit cell parameters	<i>a</i> = 11.6853(1) Å					
Unit cell volume	1595.57(3) Å ³					
Z	8					
Calculated density	1.88009 g/cm ³					
Atomic parameters						
Atom	Wyck.	Occ.	x/a	y/b	z/c	<i>U</i> [Å ²]
La	8 <i>c</i>	1	0.85762(3)	0.85762(3)	0.85762(3)	0.0140(1)
B	24 <i>g</i>	1	0.1271(6)	0.6224(5)	0.6224(5)	0.018(1)
H1	24 <i>g</i>	1	0.028(2)	0.619(2)	0.619(2)	0.024(6)
H2	48 <i>h</i>	1	0.164(2)	0.659(3)	0.537(2)	0.024(6)
H3	24 <i>g</i>	1	0.160(3)	0.691(1)	0.691(1)	0.024(6)
Cl	8 <i>c</i>	1	0.8923(2)	0.1077(2)	0.8923(2)	0.0155(7)
Li	12 <i>d</i>	0.6667	1/4	1/2	0	0.036(6)

Table S4 - Crystallographic data for LiCe(BH₄)₃Cl at RT from SR-PXD data

Phase data						
Compound		LiCe(BH ₄) ₃ Cl				
Space group		<i>I</i> -43 <i>m</i> (217)				
Unit cell parameters		<i>a</i> = 11.6243(1) Å				
Unit cell volume		1570.74(4) Å ³				
Z		8				
Calculated density		1.92004 g/cm ³				
Atomic parameters						
Atom	Wyck.	Occ.	<i>x/a</i>	<i>y/b</i>	<i>z/c</i>	<i>U</i> [Å ²]
Ce	8 <i>c</i>	1	0.85825(4)	0.85825(4)	0.85825(4)	0.0133(2)
B	24 <i>g</i>	1	0.1260(7)	0.6218(6)	0.6218(6)	0.009(2)
H1	24 <i>g</i>	1	0.026(3)	0.615(2)	0.615(2)	0.005(7)
H2	48 <i>h</i>	1	0.155(3)	0.667(3)	0.552(3)	0.005(7)
H3	24 <i>g</i>	1	0.161(4)	0.687(2)	0.687(2)	0.005(7)
Cl	8 <i>c</i>	1	0.8925(2)	0.1075(2)	0.8925(2)	0.0153(9)
Li	12 <i>d</i>	0.6667	1/4	1/2	0	0.011(6)

Table S5 - Crystallographic data for LiPr(BH₄)₃Cl at RT from SR-PXD data

Phase data						
Compound		LiPr(BH ₄) ₃ Cl				
Space group		<i>I</i> -43 <i>m</i> (217)				
Unit cell parameters		<i>a</i> = 11.5784(1) Å				
Unit cell volume		1552.17(5) Å ³				
Z		8				
Calculated density		1.94977 g/cm ³				
Atomic parameters						
Atom	Wyck.	Occ.	<i>x/a</i>	<i>y/b</i>	<i>z/c</i>	<i>U</i> [Å ²]
Pr	8 <i>c</i>	1	0.85854(4)	0.85854(4)	0.85854(4)	0.0152(2)
B	24 <i>g</i>	1	0.1258(9)	0.6215(8)	0.6215(8)	0.010(2)
H1	24 <i>g</i>	1	0.024(4)	0.613(3)	0.613(3)	0.012(9)
H2	48 <i>h</i>	1	0.149(4)	0.668(3)	0.562(3)	0.012(9)
H3	24 <i>g</i>	1	0.161(5)	0.686(3)	0.686(3)	0.012(9)
Cl	8 <i>c</i>	1	0.8929(2)	0.1071(2)	0.8929(2)	0.016(1)
Li	12 <i>d</i>	0.6667	1/4	½	0	0.010(7)

Table S6 Crystallographic data for LiNd(BH₄)₃Cl at RT from SR-PXD data

Phase data						
Compound		LiNd(BH ₄) ₃ Cl				
Space group		<i>I</i> -43 <i>m</i> (217)				
Unit cell parameters		<i>a</i> = 11.5480(2) Å				
Unit cell volume		1540.00(7) Å ³				
Z		8				
Calculated density		1.9939 g/cm ³				
Atomic parameters						
Atom	Wyck.	Occ.	<i>x/a</i>	<i>y/b</i>	<i>z/c</i>	<i>U</i> [Å ²]
Nd	8 <i>c</i>	1	0.8585(1)	0.8585(1)	0.8585(1)	0.0141(4)
B	24 <i>g</i>	1	0.127(3)	0.620(2)	0.620(2)	0.0250
H1	24 <i>g</i>	1	0.028(4)	0.627(7)	0.627(7)	0.0250
H2	48 <i>h</i>	1	0.161(7)	0.634(9)	0.533(5)	0.0250
H3	24 <i>g</i>	1	0.180(8)	0.680(5)	0.680(5)	0.0250
Cl	8 <i>c</i>	1	0.8927(6)	0.1073(6)	0.8927(6)	0.020(3)
Li	12 <i>d</i>	0.6667	1/4	1/2	0	0.0250

Table S7 Crystallographic data for Sm(BH₄)₂ at about 200 °C from *in situ* SR-PXD data. Estimated standard deviations in parenthesis

Phase data						
Compound		Sm(BH ₄) ₂				
Space group		<i>Pbcn</i> (60)				
Unit cell parameters		<i>a</i> = 6.9848(1) Å <i>b</i> = 8.4464(1) Å <i>c</i> = 7.5891(1) Å				
Unit cell volume		447.73(1) Å ³				
Z		4				
Calculated density		2.672 g/cm ³				
Atomic parameters						
Atom	Wyck.	Occ.	<i>x/a</i>	<i>y/b</i>	<i>z/c</i>	<i>U</i> [Å ²]
Sm	4 <i>c</i>	1	1/2	0.1522(1)	1/4	0.0352(2)
B	8 <i>d</i>	1	0.7320(11)	0.1144(11)	0.9206(9)	0.003(2)

H1	<i>8d</i>	1	0.6131(11)	0.0423(11)	0.8400(9)	=UB1
H2	<i>8d</i>	1	0.7989(11)	0.0307(11)	1.0320(9)	=UB1
H3	<i>8d</i>	1	0.8560(11)	0.1567(11)	0.8222(9)	=UB1
H4	<i>8d</i>	1	0.6601(11)	0.2277(11)	0.9881(9)	=UB1

Table S8. Some selected bond distances (\AA) and bond angles ($^\circ$) in $\text{Sm}(\text{BH}_4)_2$. Estimated standard deviations in parenthesis.

Sm_B	(2x) 2.996(7)
Sm_B	(2x) 3.011(8)
Sm_B	(2x) 3.061(8)
B_Sm_B	(2x) 81.7(2)
B_Sm_B	85.3(2)
B_Sm_B	(2x) 88.3(2)
B_Sm_B	(2x) 89.3(2)
B_Sm_B	(2x) 92.7(2)
B_Sm_B	(2x) 95.3(2)
B_Sm_B	98.2(2)
B_Sm_B	167.8(2)
B_Sm_B	(2x) 173.0(2)
Sm_B_Sm	98.3(2)
Sm_B_Sm	128.7(2)
Sm_B_Sm	129.0(3)

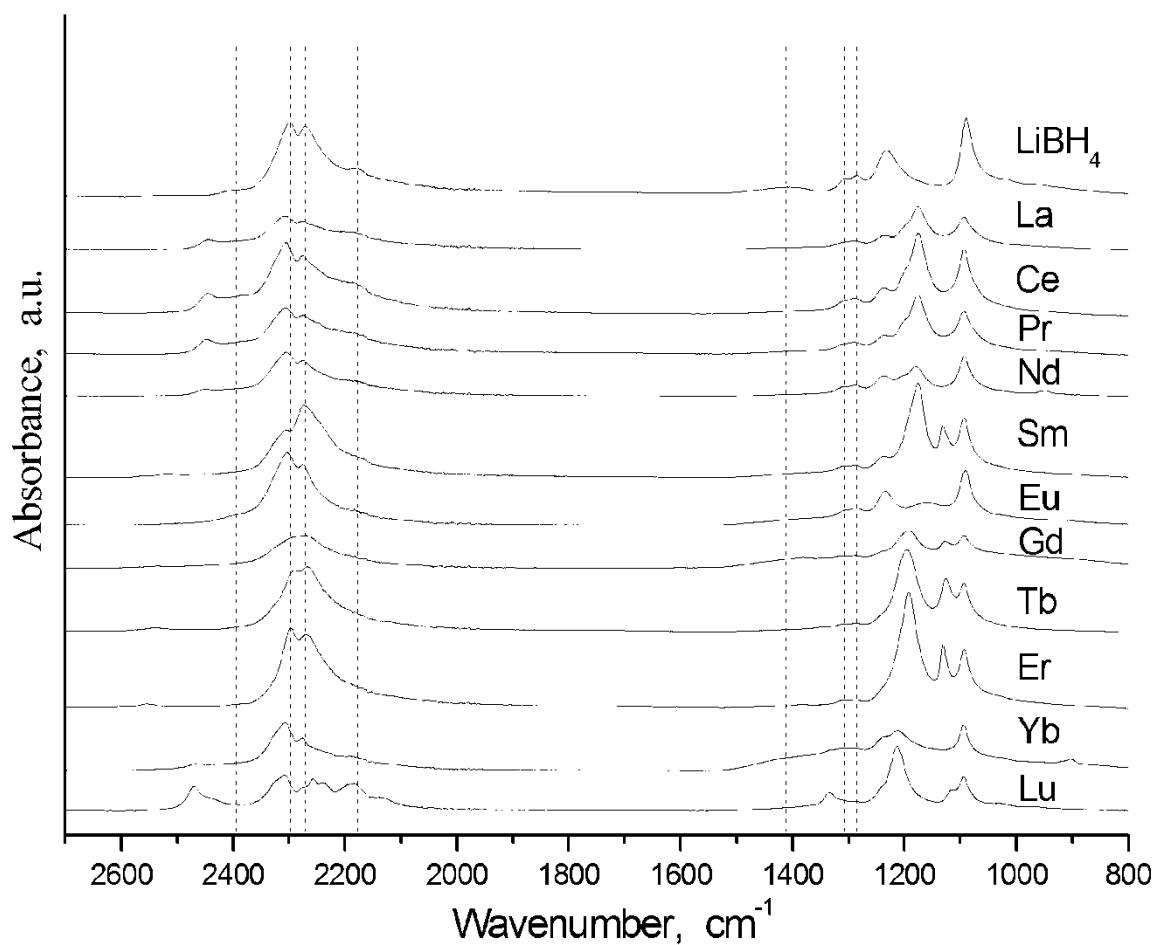


Fig. S1 IR spectra of all samples compared to LiBH_4

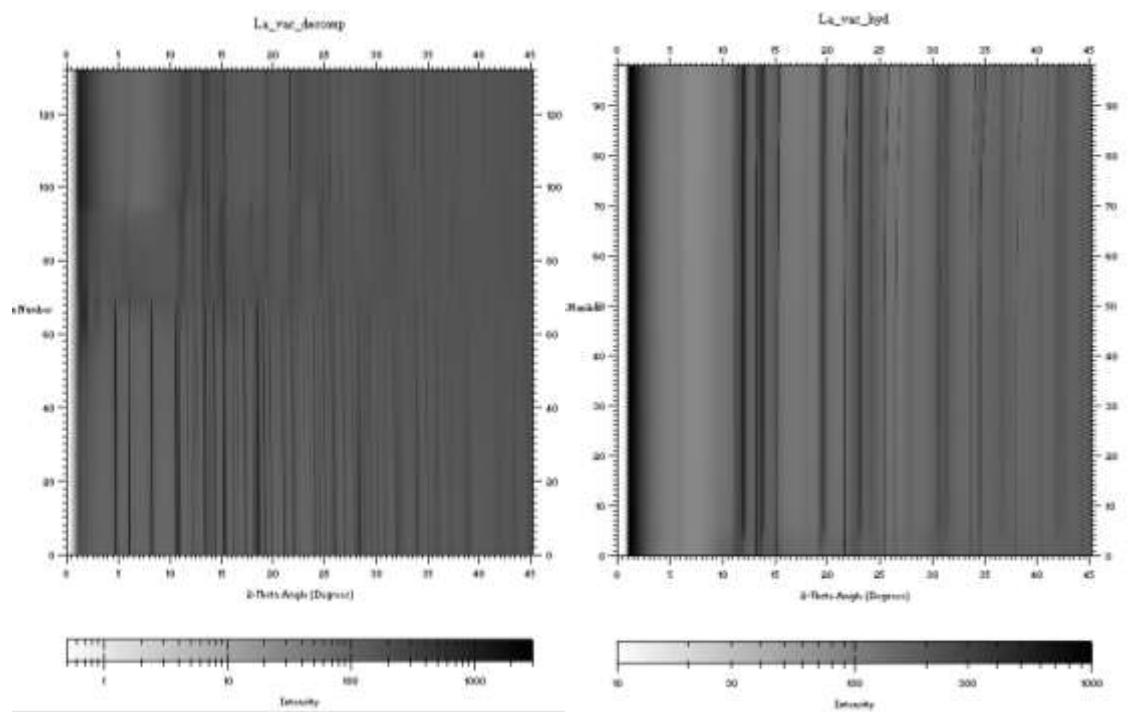


Figure S1 – *In situ* SR-PXD measurement of the ball milled composite of $\text{LaCl}_3 + 6 \text{ LiBH}_4$. Decomposition in vacuum (left) and subsequent hydrogenation in 10 MPa H_2 (right).

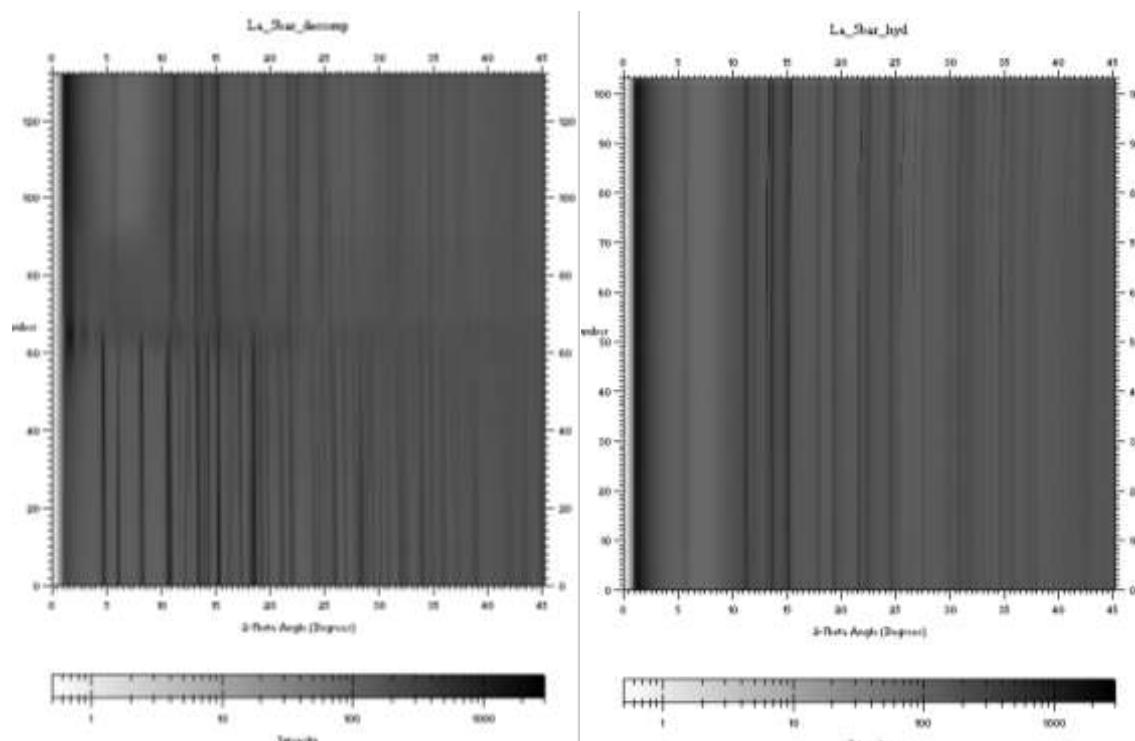


Figure S3 – *In situ* SR-PXD measurement of the ball milled composite of $\text{LaCl}_3 + 6 \text{ LiBH}_4$. Decomposition in 0.5 MPa H_2 backpressure (left) and subsequent hydrogenation in 10 MPa H_2 (right).

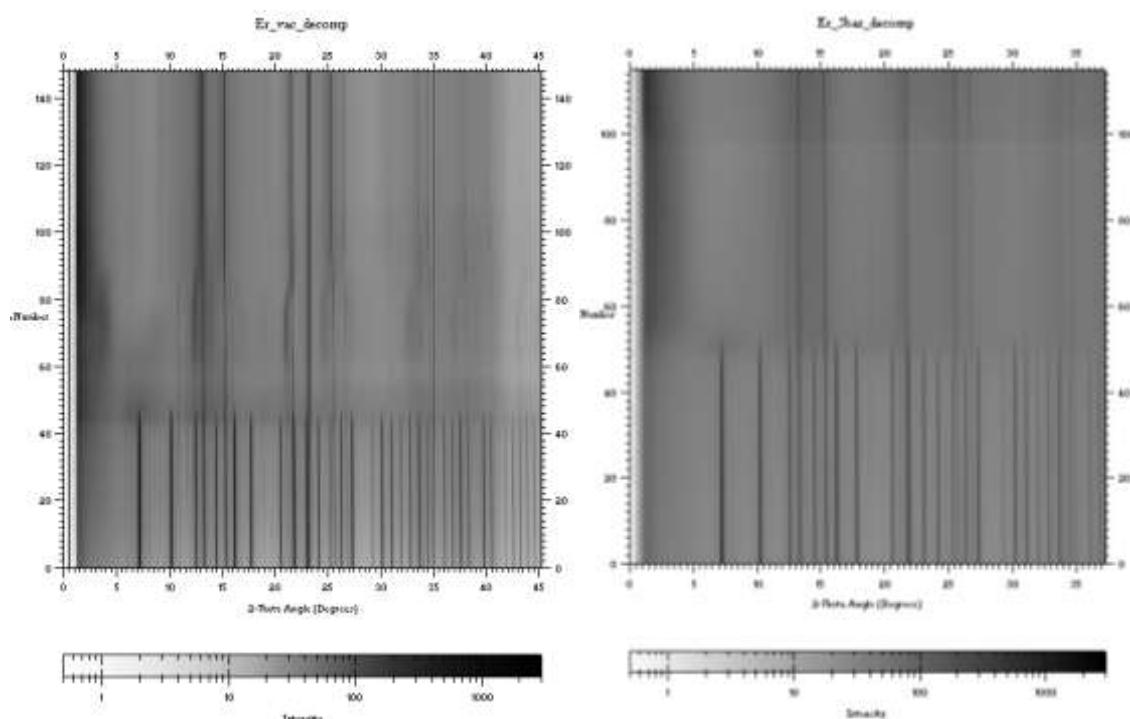


Figure S4. Contour plots from *in situ* SR-PXD during decomposition of the Er-sample, vacuum (left) and 0.5MPa (right). $\lambda = 0.694118 \text{ \AA}$

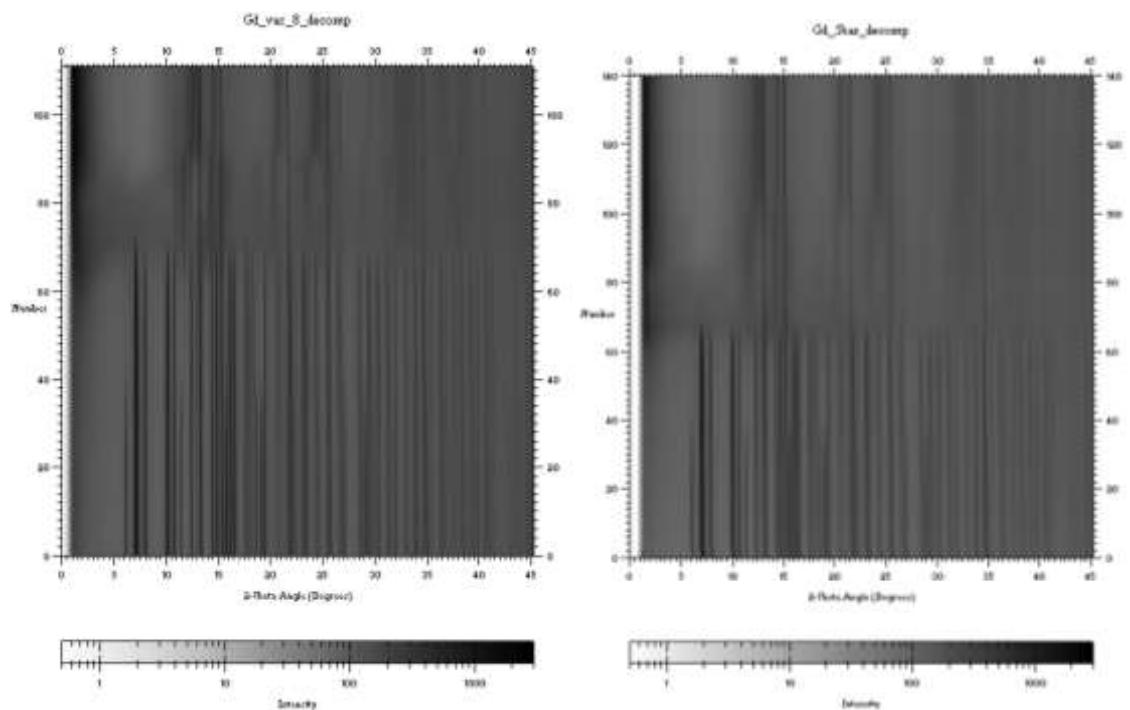


Figure S5. Contour plots from *in situ* SR-PXD during decomposition of the Gd-sample, vacuum (left) and 0.5MPa (right). $\lambda = 0.694118 \text{ \AA}$

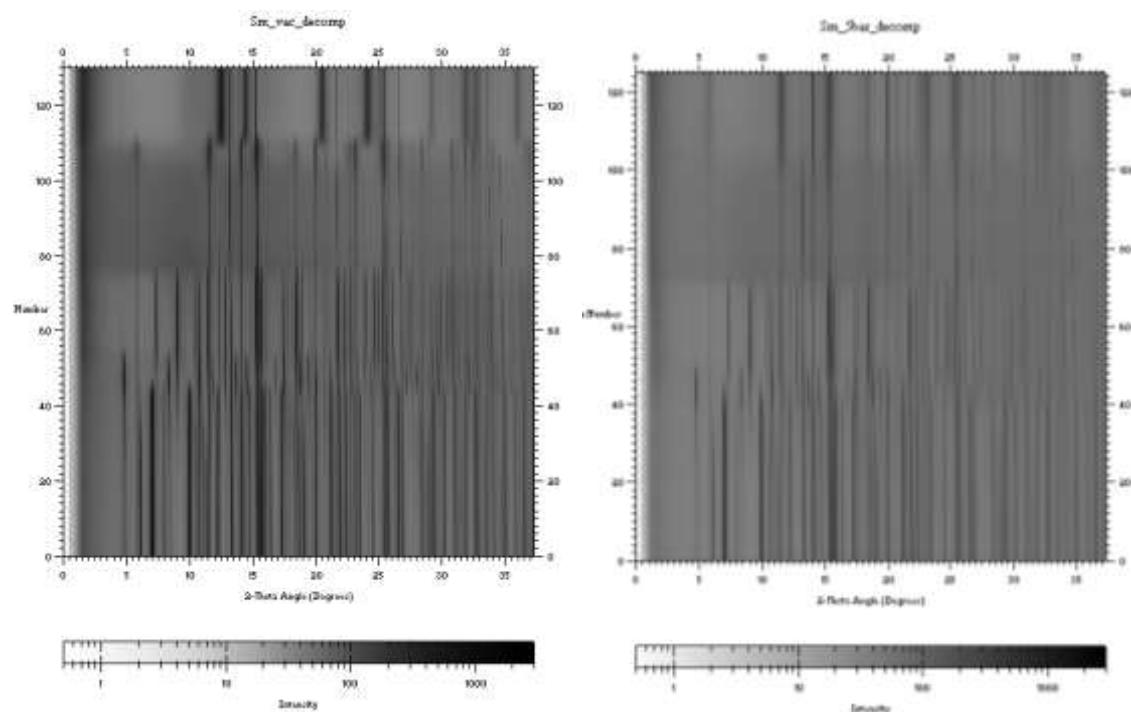


Figure S6. Contour plots from in situ SR-PXD during decomposition of the Sm-sample, vacuum (left) and 0.5MPa (right). $\lambda = 0.694118 \text{ \AA}$

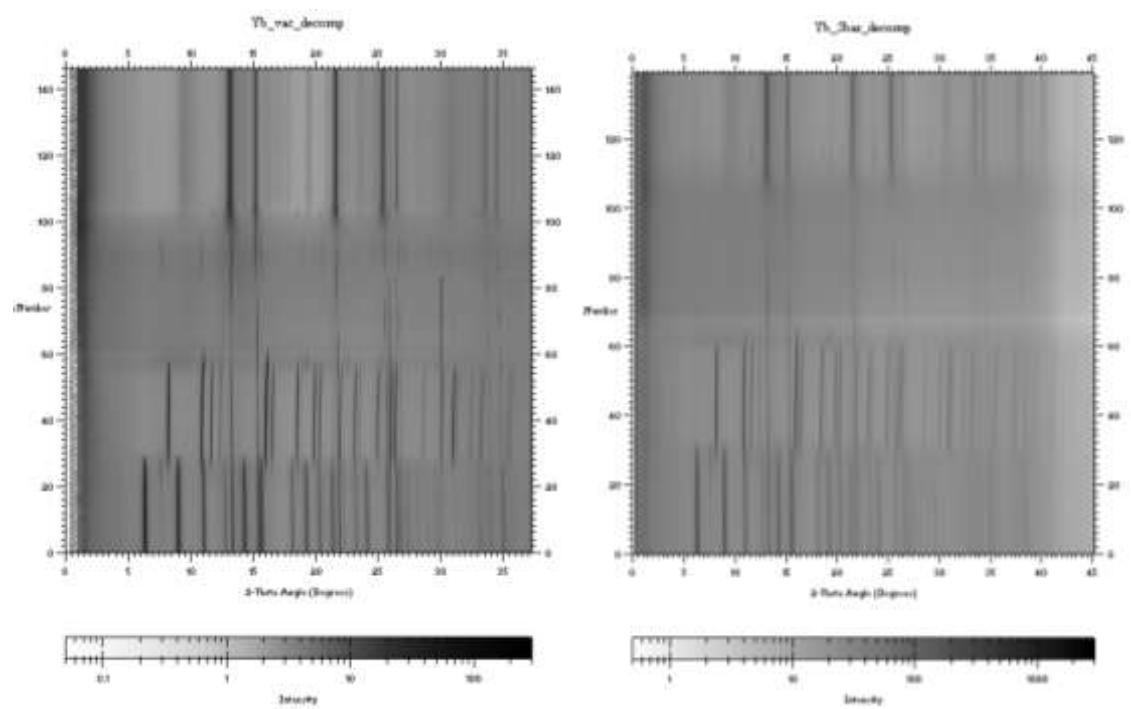


Figure S7. Contour plots from in situ SR-PXD during decomposition of the Yb-sample, vacuum (left) and 0.5MPa (right). $\lambda = 0.694118 \text{ \AA}$

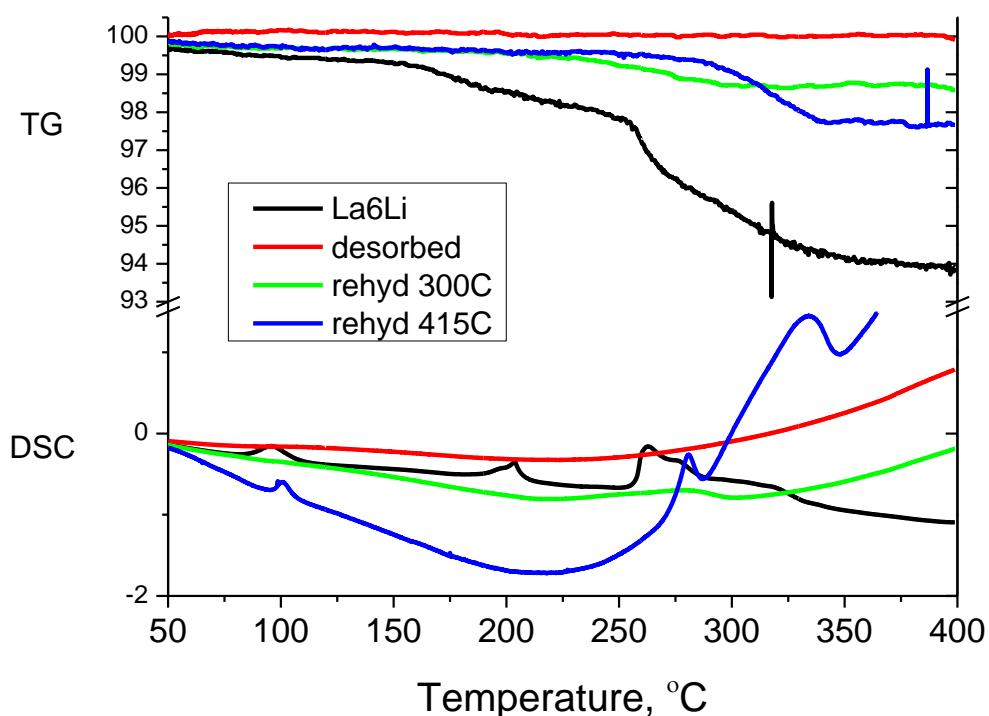


Figure S8 - TG/DSC measurement for the rehydrogenation samples of $\text{LaCl}_3 + 6 \text{ LiBH}_4$. The spectra of the ball milled sample (black), the desorbed sample (red), the sample rehydrogenated at $300\text{ }^\circ\text{C}$ (green) and the sample rehydrogenated at $415\text{ }^\circ\text{C}$ (blue).

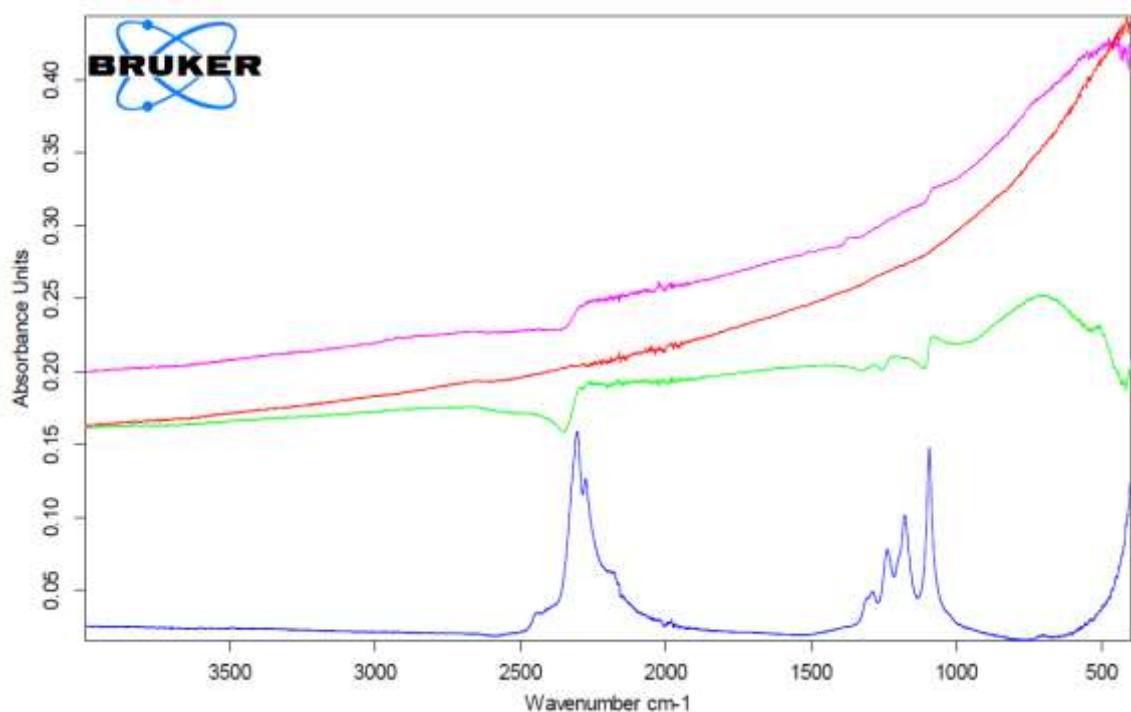


Figure S9 - IR spectra for rehydrogenation samples of $\text{LaCl}_3 + 6 \text{ LiBH}_4$. The spectra of the ball milled sample (blue), the desorbed sample (red), the sample rehydrogenated at $300\text{ }^\circ\text{C}$ (purple) and the sample rehydrogenated at $415\text{ }^\circ\text{C}$ (green).

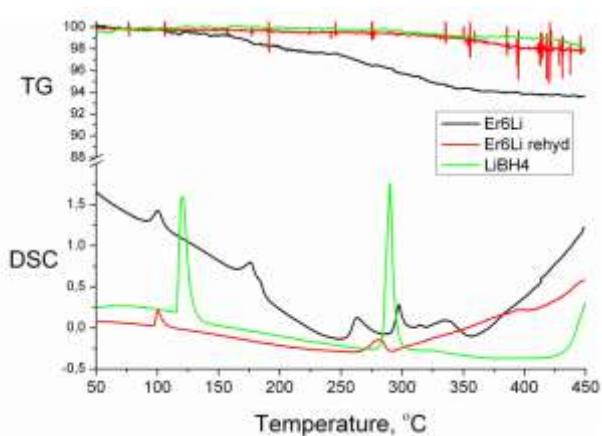


Fig. S10 TG/DSC measurement of $\text{ErCl}_3 + 6 \text{ LiBH}_4$ during decomposition (black line) and after first rehydrogenation (red line), TG/DSC data from LiBH_4 are added as reference (green line).

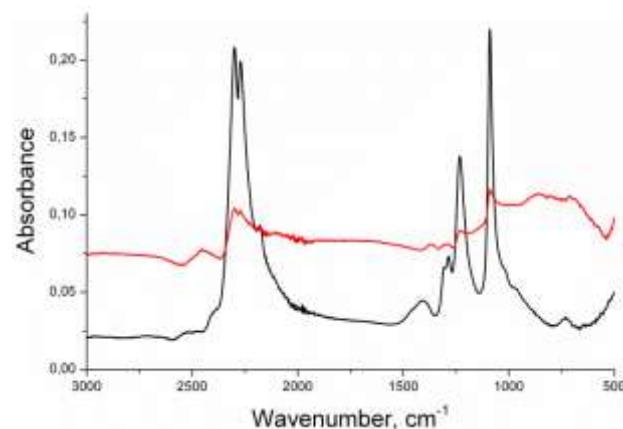


Fig. S11 IR spectra from LiBH₄ (black line) and the rehydrogenated Er-sample (red line)