

Electronic Supplementary Information

The impact of surface structure and band gap on the optoelectronic properties of Cu₂O nanoclusters of varying size and symmetry

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Table S1: Optimised structural parameters of Cu₂O clusters

Clusters	Symmetry	Bond distances (d)/Å	Reported (d)/Å	Angles	Reported	
(Cu ₂ O) ₁	C _{2v}	Cu ₂ CUS—O ₁ CUS ² Cu ₃ CUS—O ₁ CUS ²	1.765 _L 1.788 _G	^(a) 1.75	∠ Cu ₂ CUS—O ₁ CUS ² —Cu ₃ CUS	90.9° ^L 93.1° ^G
		Cu ₂ CUS—Cu ₃ CUS	2.516 _L 2.61 _G	^(a) 2.41	∠ O ₁ CUS ² —Cu ₂ CUS—Cu ₃ CUS	44.5° ^L 43° ^G
(Cu ₂ O) ₂	C _{2v}	Cu ₂ CSA—O ₁ CUS ³ Cu ₃ CSA—O ₁ CUS ³ Cu ₂ CSA—O ₄ CUS ³ Cu ₃ CSA—O ₄ CUS ³	1.999 _L 2.026 _G	^(b) 1.91	∠ Cu ₂ CSA—O ₁ CUS ³ —Cu ₃ CSA	72.4° ^L 74.2° ^G
		Cu ₆ CUS—O ₁ CUS ³ Cu ₅ CUS—O ₄ CUS ³	1.830 _L 1.830 _G	^(b) 1.76	∠ Cu ₂ CSA—O ₄ CUS ³ —Cu ₃ CUS ∠ O ₁ CUS ³ —Cu ₂ CSA—O ₄ CUS ∠ O ₁ CUS ³ —Cu ₃ CSA—O ₄ CUS ∠ O ₁ CUS ³ —Cu ₂ CSA—Cu 3CSA ∠ O ₁ CUS ³ —Cu ₃ CSA—Cu 2CSA ∠ O ₄ CUS ³ —Cu ₂ CSA—Cu 3CSA ∠ O ₄ CUS ³ —Cu ₃ CSA—Cu 2CSA	107.5° ^L 105.3° ^G
		Cu ₂ CSA—Cu ₃ CSA	2.361 _L 2.44 _G		∠ Cu ₂ CSA—O ₁ CUS ³ —Cu ₆ CUS ∠ Cu ₃ CSA—O ₁ CUS ³ —Cu ₆ CUS ∠ Cu ₂ CSA—O ₄ CUS ³ —Cu ₅ CUS ∠ Cu ₃ CSA—O ₄ CUS ³ —Cu ₅ CUS	82.6° ^L 87.4° ^G
(Cu ₂ O) ₃	C _{2v}	Cu ₂ CSA—O ₁ CSA Cu ₂ CSA—O ₄ CSA Cu ₃ CSA—O ₁ CSA Cu ₃ CSA—O ₄ CSA	2.026 _L 2.057 _G		∠ Cu ₂ CSA—O ₁ CSA—Cu ₃ CSA ∠ Cu ₂ CSA—O ₄ CSA—Cu ₃ CSA ∠ O ₁ CSA—Cu ₂ CSA—O ₄ CSA ∠ O ₁ CSA—Cu ₃ CSA—O ₄ CSA ∠ O ₁ CSA—Cu ₂ CSA—Cu 3CSA ∠ O ₁ CSA—Cu ₃ CSA—Cu 2CSA ∠ O ₄ CSA—Cu ₂ CSA—Cu 3CSA ∠ O ₄ CSA—Cu ₃ CSA—Cu 2CSA	75.1° ^L 74.4° ^G

	$\text{Cu}_8 \text{CUS}^{\text{—}}\text{O}_1 \text{CSA}$	1.822 ^L	^(b) 1.78	$\angle \text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_1 \text{CSA}^{\text{—}}\text{Cu}_8$	141.1 ^L	
	$\text{Cu}_9 \text{CUS}^{\text{—}}\text{O}_4 \text{CSA}$	1.828 ^G		$\angle \text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_1 \text{CSA}^{\text{—}}\text{Cu}_8$	138.7 ^G	
				$\angle \text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_4 \text{CSA}^{\text{—}}\text{Cu}_9$		
				$\angle \text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_4 \text{CSA}^{\text{—}}\text{Cu}_9$		
	$\text{Cu}_6 \text{CSA}^{\text{—}}\text{O}_1 \text{CSA}$	1.956		$\angle \text{Cu}_5 \text{CSA}^{\text{—}}\text{O}_4 \text{CSA}^{\text{—}}\text{Cu}_9$	112.8 ^L	
	$\text{Cu}_5 \text{CSA}^{\text{—}}\text{O}_4 \text{CSA}$	^L 1.96 ^G		$\angle \text{Cu}_6 \text{CSA}^{\text{—}}\text{O}_1 \text{CSA}^{\text{—}}\text{Cu}_8$	119.9 ^G	
	$\text{Cu}_6 \text{CSA}^{\text{—}}\text{O}_7 \text{CUS}^{\text{—}}_2$	1.832 ^L		$\angle \text{O}_4 \text{CSA}^{\text{—}}\text{Cu}_5 \text{CSA}^{\text{—}}\text{O}_7$	147.7 ^L	
	$\text{Cu}_5 \text{CSA}^{\text{—}}\text{O}_7 \text{CUS}^{\text{—}}_2$	1.813 ^G		$\angle \text{O}_1 \text{CSA}^{\text{—}}\text{Cu}_6 \text{CSA}^{\text{—}}\text{O}_7$	147.8 ^G	
	$\text{Cu}_2 \text{CSA}^{\text{—}}\text{Cu}_3 \text{CSA}$	2.470 ^L		$\angle \text{Cu}_5 \text{CSA}^{\text{—}}\text{O}_7 \text{CUS}^{\text{—}}\text{Cu}_6$	85.8° ^L	
		2.488 ^G			87.2° ^G	
	$\text{Cu}_5 \text{CSA}^{\text{—}}\text{Cu}_6 \text{CSA}$	2.493 ^L				
		2.5 ^G				
$(\text{Cu}_2\text{O})_3$	C_{3v}	$\text{Cu}_1 \text{CSA}^{\text{—}}\text{O}_4 \text{CUS}^{\text{—}}_3$	1.832 ^L	^(b) 1.91	$\angle \text{Cu}_1 \text{CSA}^{\text{—}}\text{O}_4 \text{CSA}^{\text{—}}\text{Cu}_8$	136.1 ^L
		$\text{Cu}_1 \text{CSA}^{\text{—}}\text{O}_5 \text{CUS}^{\text{—}}_3$			$\angle \text{Cu}_1 \text{CSA}^{\text{—}}\text{O}_5 \text{CSA}^{\text{—}}\text{Cu}_7$	101.2 ^G
		$\text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_4 \text{CUS}^{\text{—}}_3$	1.912 ^G		$\angle \text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_4 \text{CSA}^{\text{—}}\text{Cu}_8$	
		$\text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_6 \text{CUS}^{\text{—}}_3$			$\angle \text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_5 \text{CSA}^{\text{—}}\text{Cu}_7$	
		$\text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_5 \text{CUS}^{\text{—}}_3$			$\angle \text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_6 \text{CSA}^{\text{—}}\text{Cu}_9$	
		$\text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_6 \text{CUS}^{\text{—}}_3$			$\angle \text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_6 \text{CSA}^{\text{—}}\text{Cu}_9$	
		$\text{Cu}_7 \text{CUS}^{\text{—}}\text{O}_5 \text{CUS}^{\text{—}}_3$	1.762 ^L	^(b) 1.77	$\angle \text{Cu}_1 \text{CSA}^{\text{—}}\text{O}_4 \text{CSA}^{\text{—}}\text{Cu}_2$	87.7° ^L
		$\text{Cu}_8 \text{CUS}^{\text{—}}\text{O}_4 \text{CUS}^{\text{—}}_3$			$\angle \text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_6 \text{CSA}^{\text{—}}\text{Cu}_3$	83.2° ^G
		$\text{Cu}_9 \text{CUS}^{\text{—}}\text{O}_6 \text{CUS}^{\text{—}}_3$	1.811 ^G		$\angle \text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_5 \text{CSA}^{\text{—}}\text{Cu}_1$	
		$\text{Cu}_1 \text{CSA}^{\text{—}}\text{Cu}_2 \text{CSA}$	2.539 ^L		$\angle \text{O}_4 \text{CSA}^{\text{—}}\text{Cu}_1 \text{CSA}^{\text{—}}$	46° ^L
		$\text{Cu}_2 \text{CSA}^{\text{—}}\text{Cu}_3 \text{CSA}$			$\angle \text{O}_4 \text{CSA}^{\text{—}}\text{Cu}_2 \text{CSA}^{\text{—}}$	48.4° ^G
		$\text{Cu}_1 \text{CSA}^{\text{—}}\text{Cu}_3 \text{CSA}$	2.539 ^G		$\angle \text{O}_5 \text{CSA}^{\text{—}}\text{Cu}_1 \text{CSA}^{\text{—}}$	
					$\angle \text{O}_5 \text{CSA}^{\text{—}}\text{Cu}_3 \text{CSA}^{\text{—}}$	
					$\angle \text{O}_6 \text{CSA}^{\text{—}}\text{Cu}_2 \text{CSA}^{\text{—}}$	
					$\angle \text{O}_6 \text{CSA}^{\text{—}}\text{Cu}_3 \text{CSA}^{\text{—}}$	
$(\text{Cu}_2\text{O})_3$	C_s	$\text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_1 \text{CUS}^{\text{—}}_2$	1.776		$\angle \text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_1 \text{CUS}^{\text{—}}_2\text{—}\text{Cu}_3$	114.2°
		$\text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_1 \text{CUS}^{\text{—}}_2$			$\angle \text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_4$	
		$\text{Cu}_2 \text{CSA}^{\text{—}}\text{O}_4 \text{CUS}^{\text{—}}_3$	1.839		$\angle \text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_7$	108.3°
		$\text{Cu}_3 \text{CSA}^{\text{—}}\text{O}_7 \text{CUS}^{\text{—}}_3$			$\angle \text{Cu}_8 \text{CUS}^{\text{—}}\text{O}_7$	

$\text{Cu}_6 \text{CUS}^{\text{3}}-\text{O}_4 \text{CUS}^{\text{3}}$	1.857	$\angle \text{Cu}_2 \text{CSA}-\text{O}_4 \text{CUS}^{\text{3}}-\text{Cu}_5$	96.6°
$\text{Cu}_8 \text{CUS}^{\text{3}}-\text{O}_7 \text{CUS}^{\text{3}}$		cus	
$\text{Cu}_5 \text{CUS}^{\text{3}}-\text{O}_4 \text{CUS}^{\text{3}}$	1.872	$\angle \text{Cu}_3 \text{CSA}-\text{O}_7 \text{CUS}^{\text{3}}-\text{Cu}_9 \text{CUS}$	86.2°
$\text{Cu}_9 \text{CUS}^{\text{3}}-\text{O}_7 \text{CUS}^{\text{3}}$		cus	

(a) is from reference 100.

(b) is from reference 101.

Table S2: Partial DOS (PDOS) data of Cu₂O clusters with increasing size at C_{2v} symmetry

Clusters	Sy mm	P e	MOs. involved in Transition	MOS.	Contribution of AOs of Cu atom in forming MOs. (%)	Contribution of AOs of O atom in forming MOs. (%)
					s p d s p	
			Occupied Virtual			
(Cu ₂ O) ₁	C _{2v}	1	5A1 5B1	5A1 5B1	30 50 2.1 35.6	30 55.5 1.1 28.2
						34.08 13.04 21.01
(Cu ₂ O) ₂	C _{2v}	1	10B2 17A1	10B2 17A1	8 61.6 25.5	6.4
						25 44.41
(Cu ₂ O) ₃	C _{2v}	1	14A1 13B2	14A1 13B2	26 57.17 20.27	25 4.59
						13.81

Table S3: Partial DOS (PDOS) data of Cu₂O clusters with fixed size at different symmetries

clusters	Sy m	p e	MOS. involved in Transition	MOS.	Contribution of AOs of Cu atom in forming MOs. (%)	Contribution of AOs of O atom in forming MOs. (%)
				.		
					s p d s p	
			Occupied Virtual			
(Cu ₂ O) ₃	C _s	2	20AAA 22AAA	20AAA 22AAA	17.9 59.6 2.21 28	35.9 25 4.59
						9.06
(Cu ₂ O) ₃	C _{2v}	2	14A1 13B2	14A1 13B2	26 57.17 20.27	25 4.59
						44.41 13.81
(Cu ₂ O) ₃	C _{3v}	2	95A 100A	95A 100A	2.58 62.1	40.7 13.6
						44.3 14.7

Table S4. Vibrational frequencies of clusters of different size and symmetry calculated using GGA as exchange and BLYP as correlation functional.

Clusters	Symmetry	Frequencies (cm ⁻¹)
(Cu ₂ O) ₁	<i>C</i> _{2v}	v ₁ = 99.011, v ₂ = 566.589, v ₃ = 594.664
(Cu ₂ O) ₂	<i>C</i> _{2v}	v ₁ = 15.04, v ₂ = 49.45, v ₃ = 59.767, v ₄ = 74.66, v ₅ = 123.785, v ₆ = 150.66, v ₇ = 165.108, v ₈ = 241.094, v ₉ = 317.499, v ₁₀ = 413.617, v ₁₁ = 525.733, v ₁₂ = 547.473
(Cu ₂ O) ₃	<i>C</i> _{2v}	v ₁ = i13.422, v ₂ = 17.442, v ₃ = 25.247, v ₄ = 45.538, v ₅ = 60.975, v ₆ = 97.623, v ₇ = 99.626, v ₈ = 103.112, v ₉ = 116.452, v ₁₀ = 153.383, v ₁₁ = 195.809, v ₁₂ = 197.193, v ₁₃ = 199.533, v ₁₄ = 206.058 v ₁₅ = 221.477, v ₁₆ = 316.122, v ₁₇ = 334.313, v ₁₈ = 506.849, v ₁₉ = 597.349, v ₂₀ = 607.34, v ₂₁ = 612.904.
(Cu ₂ O) ₃	<i>C</i> _{3v}	v ₁ = i17.692, v ₂ = 10.035, v ₃ = 11.067, v ₄ = 21.395, v ₅ = 27.751, v ₆ = 30.145, v ₇ = 103.919, v ₈ = 104.917, v ₉ = 133.134, v ₁₀ = 147.902, v ₁₁ = 148.521, v ₁₂ = 200.889, v ₁₃ = 342.961, v ₁₄ = 344.991, v ₁₅ = 408.415, v ₁₆ = 449.01, v ₁₇ = 484.134, v ₁₈ = 486.374, v ₁₉ = 563.883 v ₂₀ = 566.764, v ₂₁ = 584.341.
(Cu ₂ O) ₃	<i>C</i> _s	v ₁ = i11.756, v ₂ = i1.781, v ₃ = 7.733, v ₄ = 12.833, v ₅ = 26.312, v ₆ = 34.469, v ₇ = 43.984 v ₈ = 86.646, v ₉ = 107.612, v ₁₀ = 119.286, v ₁₁ = 141.953, v ₁₂ = 144.674, v ₁₃ = 193.337, v ₁₄ = 421.759, v ₁₅ = 425.993, v ₁₆ = 442.952, v ₁₇ = 472.448, v ₁₈ = 517.353, v ₁₉ = 529.809, v ₂₀ = 590.120, v ₂₁ = 714.755.

Table S5: Partial DOS (PDOS) data of larger Cu₂O clusters with different size at fixed symmetries

clusters	Symmetry	p e a k s	MOs. involved in Transition		MOs .	Contribution of AOs of Cu atom in forming MOs. (%)			Contribution of AOs of O atom in forming MOs.(%)	
			Occupied	Virtual		s	p	d	s	p
[Cu ₂₈ O ₁₅] ⁶⁺	T_d	1	71T2	38A1	71T2			40		47
					38A1	74.8 6		15.4 5		10.75
[Cu ₄₄ O ₁₅] ⁶⁺	T_d	1	19A1	43T2	19A1	76.0 7		20.9		
					43T2	62.0 8		2.24		15.01

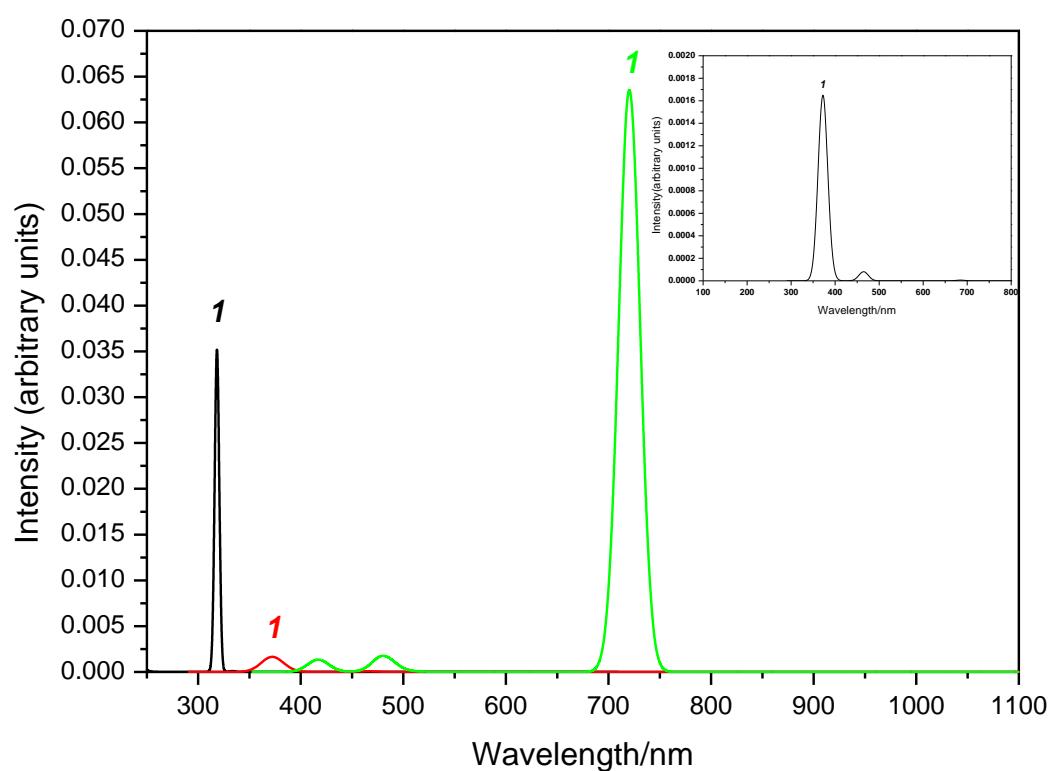


Figure S1: TDDFT valence excitation spectra (intensity vs wavelength) of $(\text{Cu}_2\text{O})_1$ (black), $(\text{Cu}_2\text{O})_2$ (red) and $(\text{Cu}_2\text{O})_3$ (green) clusters having same symmetry (C_{2v}). Only the most intense discrete transitions are reported and labelled as 1. In the inset the intensity of peak **I** of $(\text{Cu}_2\text{O})_2$ is displayed distinctly.

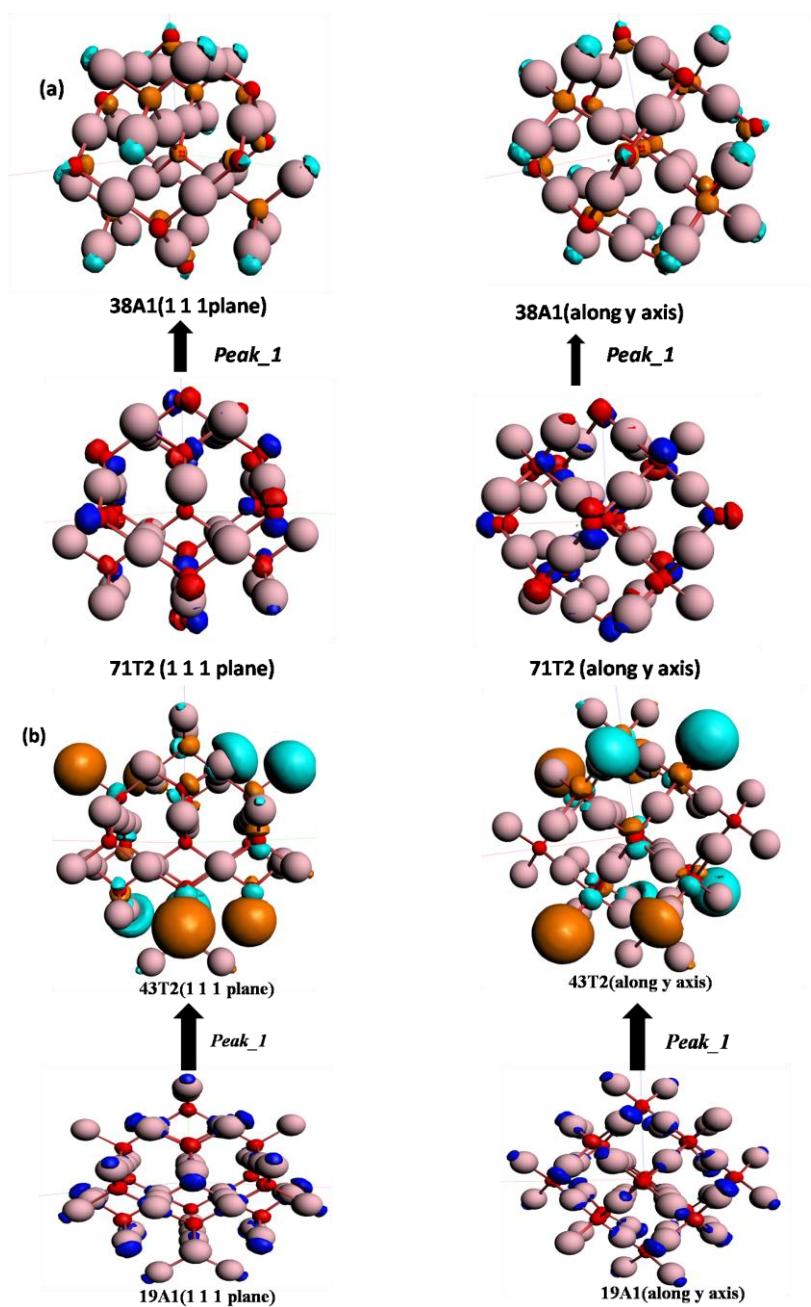


Figure S2: The representation of MOs participating in the electronic transitions to form the most intense peaks labelled as **I** (a) $[Cu_{28}O_{15}]^{6+}$ and (b) $[Cu_{44}O_{30}]^{6+}$ in T_d symmetry.