

Electronic Supplementary Information

A facile hard-templating synthesis of mesoporous spinel CoFe₂O₄ nanostructures as promising electrocatalysts for H₂O₂ reduction reaction

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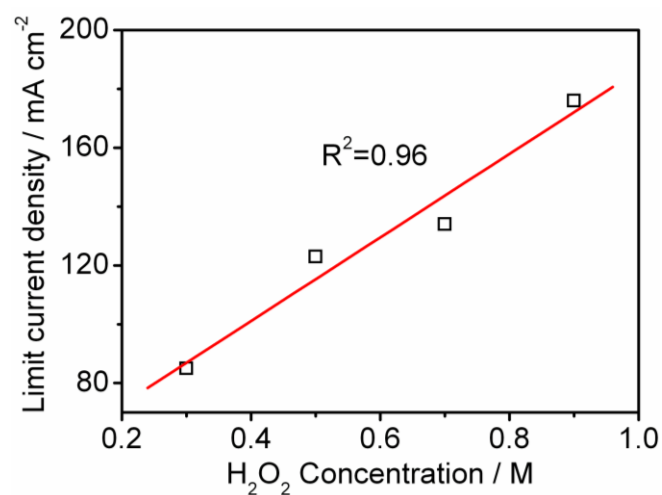


Fig. S1 Limit (peak) current density as a function of H₂O₂ concentration.

The limit (peak) current density and H₂O₂ concentration comply with the following relationship:

$$j_l = nFm_O C_O \quad (1)$$

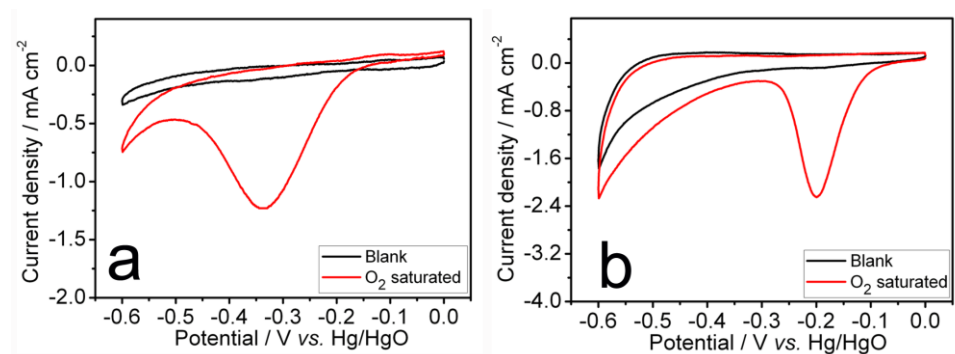


Fig. S2 CV plots of TAB (a) and CoFe₂O₄ (b) electrodes in blank and O₂ saturated 3 M NaOH solutions at a scan rate of 10 mV s⁻¹.

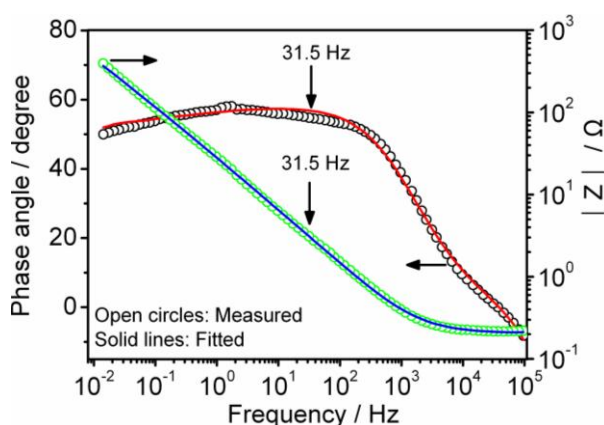


Fig. S3 Bode plots of CoFe_2O_4 electrode in 3 M NaOH solutions.

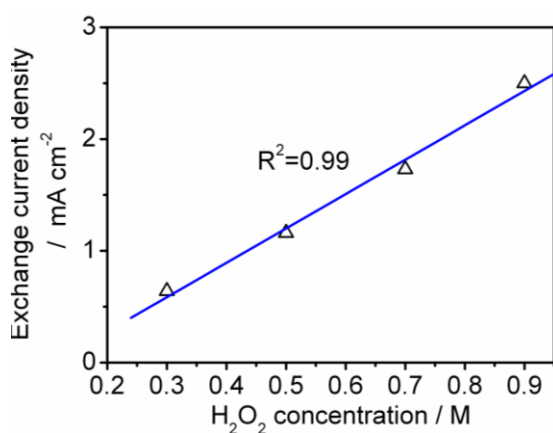


Fig. S4 Exchange current density as a function of H_2O_2 concentration.

Considering the EIS spectra were tested under open circuit potentials (quasi-equilibrium state), the exchange current density (j^0) can be calculated with the fitted values of charge transfer resistance (R_{ct}) according to following equation:

$$j^0 = \frac{RT}{nFR_{ct}} \quad (2)$$

Table S1 The fitted values of impedimetric parameters for the CoFe₂O₄ electrode in 3 M NaOH with different H₂O₂ concentrations.

H ₂ O ₂ Conc.	Impedimetric parameters										χ^2 (E-3)
	<i>L</i> (H cm ²)	<i>R_e</i> (Ω cm ²)	<i>Q1, Y₀</i> (Ω ⁻¹ s ⁿ cm ⁻²)	<i>Rct1</i> (Ω cm ²)	<i>T, Y₀; B</i> (Ω ⁻¹ s ^{0.5} cm ⁻² ; s ^{0.5})	<i>C</i> (F cm ⁻²)	<i>Q2, Y₀</i> (Ω ⁻¹ s ⁿ cm ⁻²)	<i>Rct2</i> (Ω cm ²)	<i>n1</i>	<i>n2</i>	
Blank	7.1E-8	0.21	5.7E-3	5.64	6.3E-3; 10.65	1.42	–	–	0.74	–	1.2
0.3 M	3.5E-8	0.31	7.3E-3	0.80	–	–	7.6E-3	19.9	0.84	0.74	1.9
0.5 M	3.5E-8	0.38	7.5E-3	0.71	–	–	6.9E-3	11.1	0.83	0.75	4.0
0.7 M	8.6E-8	0.17	6.3E-3	0.74	–	–	8.1E-3	7.41	0.85	0.74	5.3
0.9 M	8.3E-8	0.16	5.1E-3	0.70	–	–	8.1E-3	5.13	0.88	0.72	4.0