

Supplementary Information

High Rate Capacity Retention of Binder-free, Tin Oxide Nanowire Arrays Using Thin Titania and Alumina Coatings

Tu Quang Nguyen^{a,b}, Arjun Kumar Thapa^a, Venkat Kalyan Vendra^{a,b}, Jacek B. Jasinski^a, Gamini U. Sumanasekera^{a,c} and Mahendra K. Sunkara^{a,b,*}

^a*Conn Center for Renewable Energy Research*

^b*Department of Chemical Engineering*

^c*Department of Physics*

University of Louisville

Louisville, KY 40292

**Email: mahendra@louisville.edu*

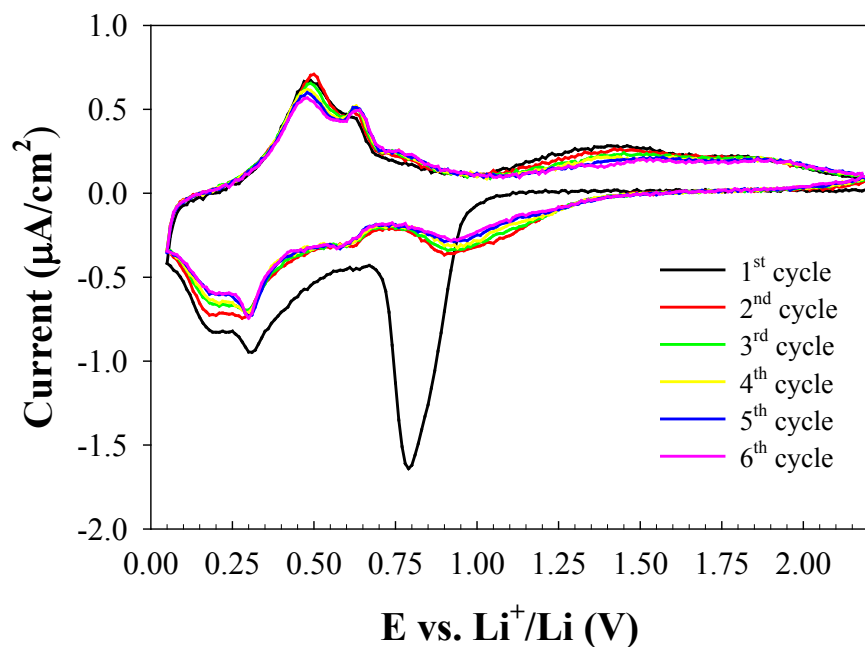


Fig. S1: Cyclic voltammetry of SnO₂ NWs at the voltage range of 2.2 ÷ 0.005 V using scan speed of 5 mV/min

Table S1: 1st cycle electrochemical performance comparison of pure SnO₂ NWs and titania-coated SnO₂ NWs

Sample	Discharge capacity, mAhg ⁻¹	Charge capacity, mAhg ⁻¹	Columbic efficiency, %	ICL, mAhg ⁻¹	ICL due to reduction of tin oxide to tin, mAhg ⁻¹	ICL due to SEI formation, mAhg ⁻¹
Pure SnO ₂ NWs	1680	832	49.5	848	400	448
titania-coated SnO ₂	1705	1238	72.6	467	520	-

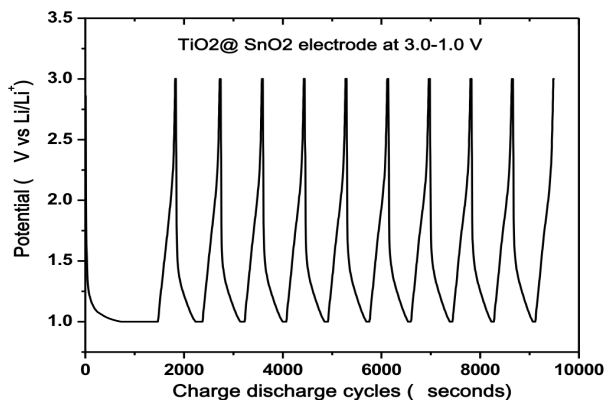


Fig. S2: The charge-discharge capacities with times vs. voltage profiles of titania-coated SnO₂ NWs electrode at 3.0-1.0 V

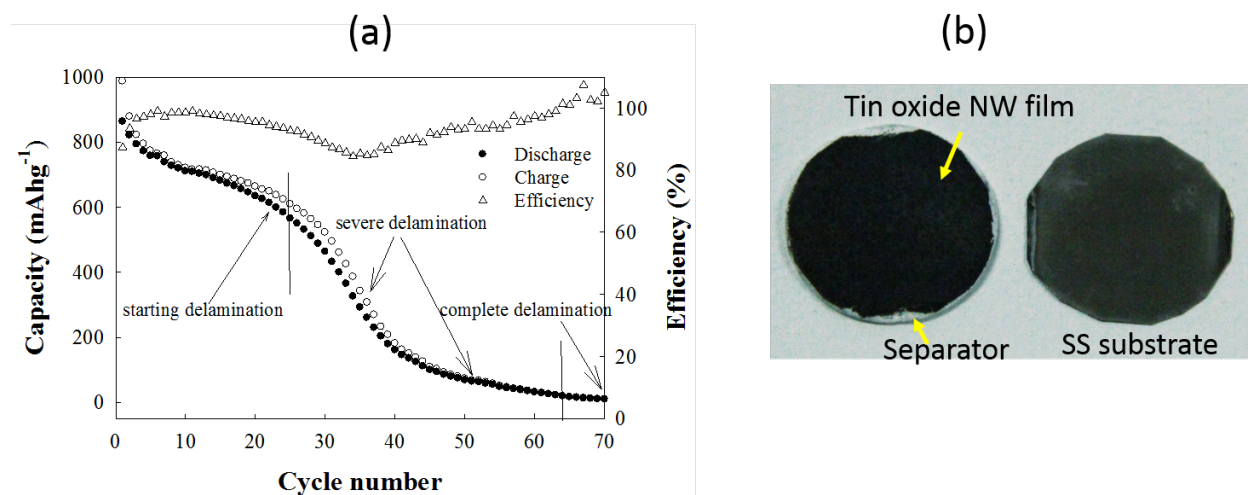


Fig. S3: (a) Charge and discharge capacities vs. cycle number of titania-coated SnO₂ NWs at current density of 1500 mA/g; (b) the photograph of delaminated electrode: the tin oxide film (black in color) was completely delaminated from stainless substrate and adhered onto the separator (white in color).

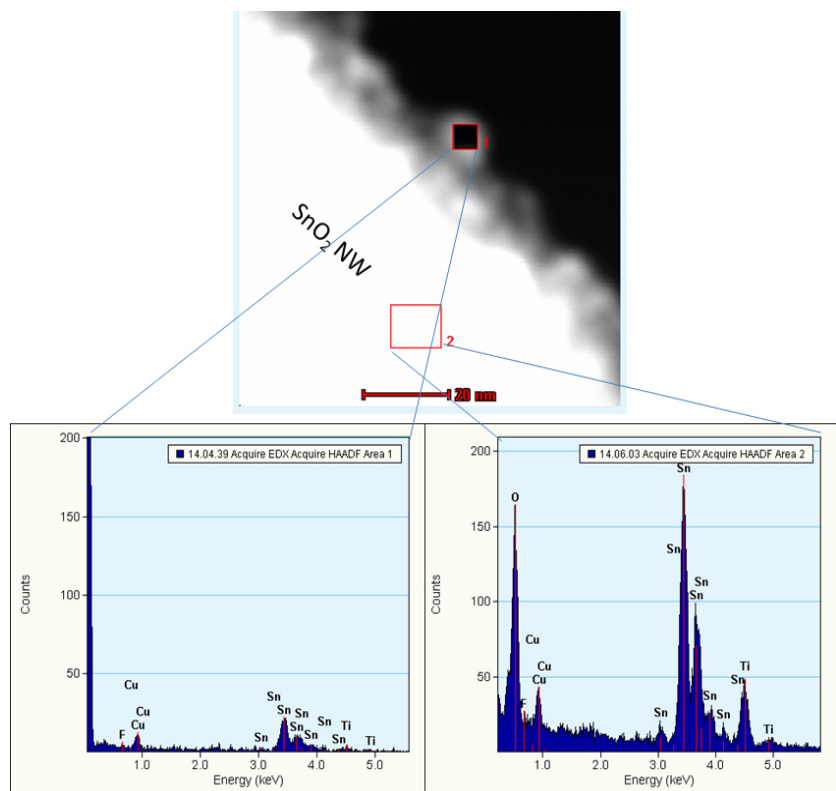


Fig. S4: EDS data of 15 nm titania-coated SnO₂ NWs after cycling

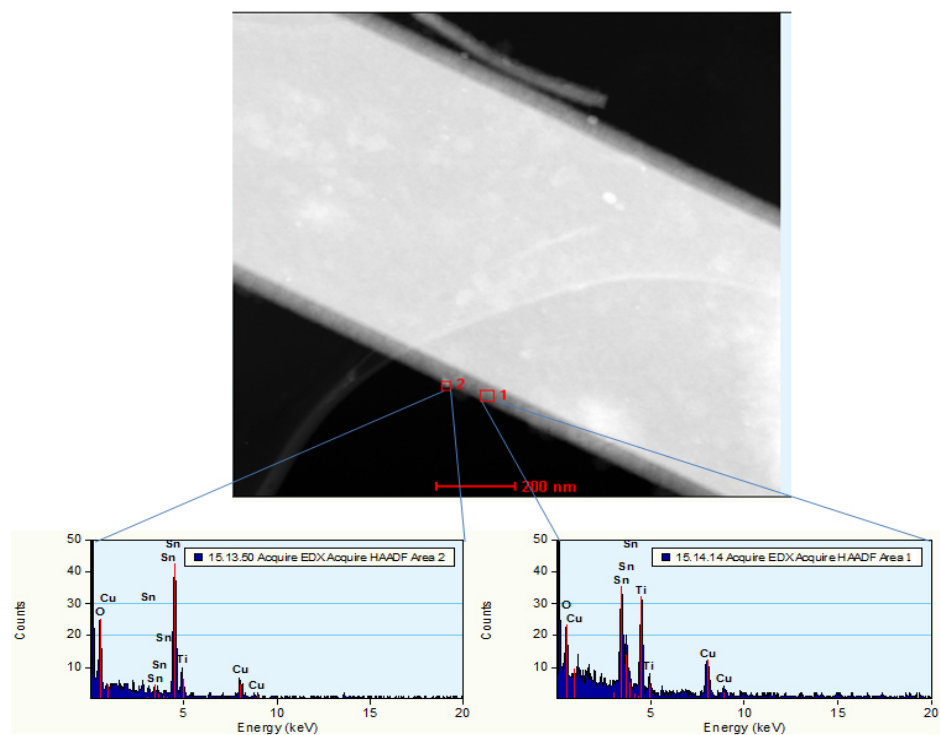


Fig. S5: EDS data of titania-coated SnO_2 NWs after cycling at voltage of 1.0 – 3.0V

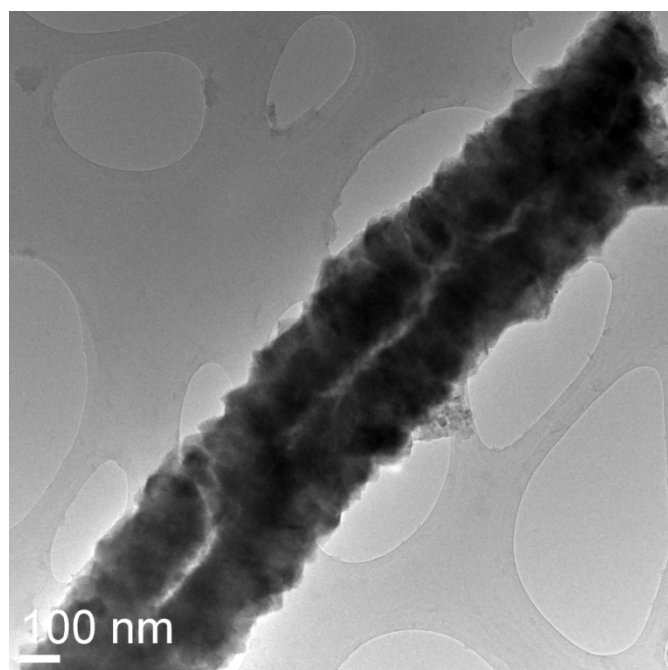


Fig. S6: TEM image of titania-coated SnO_2 NWs after cycling at showing nanowire morphology with hollow structure

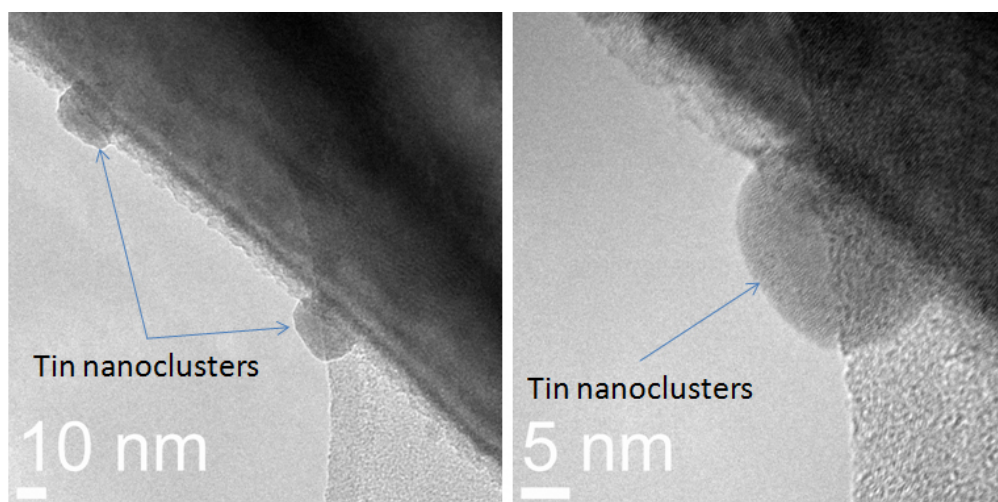


Fig. S7: HR-TEM of titania-coated SnO_2 NWs after 1st cycle showing presence of tin nanoclusters on nanowire surface

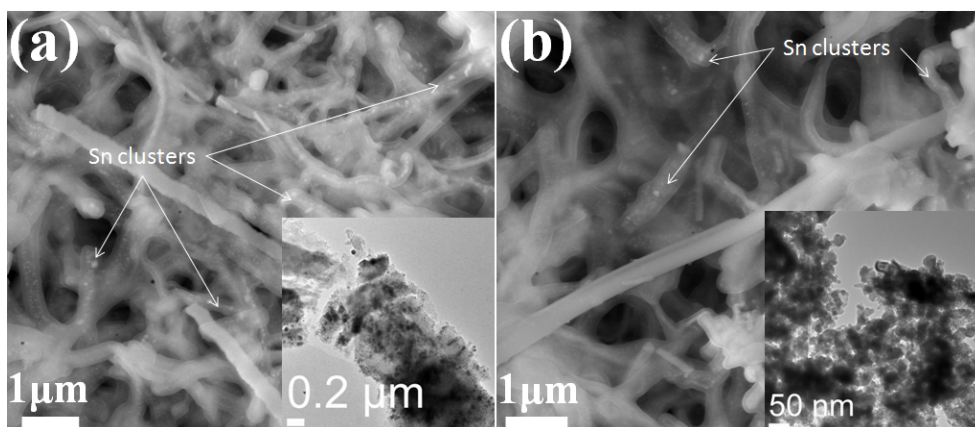


Fig. S8: SEM, TEM (inset) images of thin layer of titania-(a), alumina-(b) coated SnO_2 NWs after cycling. The white spot as Sn clusters are presence in either titania- or alumina-coated SnO_2 NWs

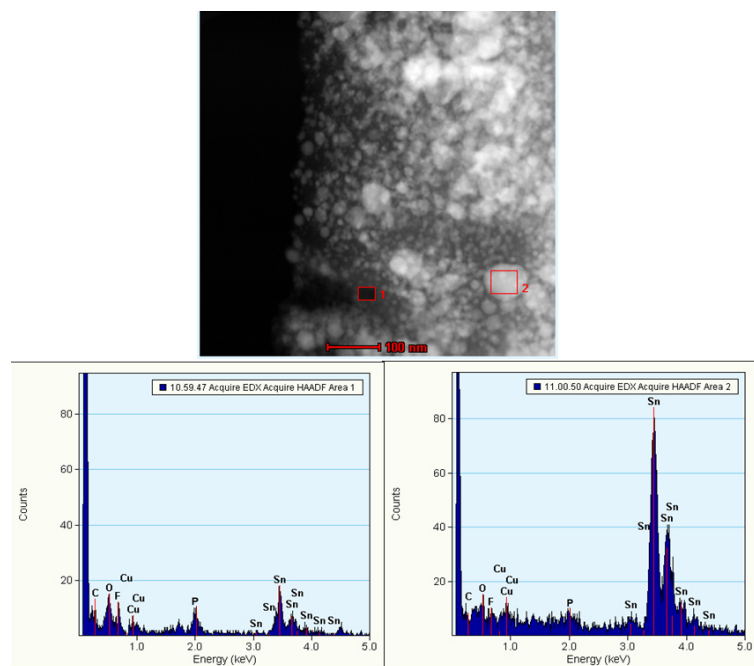


Fig. S9: EDS data of 5 nm titania-coated SnO₂ NWs after cycling

Table S2: Young's and bulk modulus of alumina and titania

	E, GPa	G, GPa	Structure	Ref.
	36-43		10-18 nm wall thickness, 35-70 nm diameter, nanotube	[1]
Titania	23		10 nm wall thickness, 65 nm diameter, nanotube	[2]
	44		30 nm wall thickness, 80 nm diameter, nanotube	[2]
	151		200 nm thickness, anatase film	[3]
	146		280 nm thickness film	[4]
		140		[5]
Alumina	168-		50-300 nm thickness film	[6]
	182			
		235		[7]

Titania nanotube deformation calculations

For a thin wall titania tube with external and internal nominal diameters of ~75 nm and ~65 nm, **Shokuhfar** et al. has reported that the maximum axial strain of 5%. Assuming that the radial strain is the same as axial strain (it should be actually smaller), we can calculate the volume expansion of a titania shell of 10 nm wall thickness, 300 nm inner diameter as following:

$$\begin{aligned} \%Volume &= \frac{V_2}{V_1} = \frac{\pi R_2^2 L_2}{\pi R_1^2 L_1} = \frac{\pi (R_1 + 0.05R_1)^2 (L_1 + 0.05L_1)}{\pi R_1^2 L_1} \\ &= 115.7\% \end{aligned}$$

where V_1, V_2 are volume of titania nanotube before and after deformation, respectively

R_1, R_2 are inner radius of titania nanotube before and after deformation, respectively

L_1, L_2 are length of titania nanotube before and after deformation, respectively.

References

1. Crawford, G.A., N. Chawla, and J.E. Houston, Nanomechanics of biocompatible TiO₂ nanotubes by Interfacial Force Microscopy (IFM). *Journal of the Mechanical Behavior of Biomedical Materials*, 2009. 2(6): p. 580-587.
2. Shokuhfar, T., et al., Direct Compressive Measurements of Individual Titanium Dioxide Nanotubes. *ACS Nano*, 2009. 3(10): p. 3098-3102.
3. Borgese, L., et al., Young modulus and Poisson ratio measurements of TiO₂ thin films deposited with Atomic Layer Deposition. *Surface and Coatings Technology*, 2012. 206(8–9): p. 2459-2463.
4. Anderson, O., et al., Density and Young's modulus of thin TiO₂ films. *Fresenius' Journal of Analytical Chemistry*, 1997. 358(1-2): p. 315-318.
5. Fischer-Cripps, A.C., 2004. *Nanoindentation*, second ed. Springer, New York.
6. Tripp, M.K., et al., The mechanical properties of atomic layer deposited alumina for use in micro- and nano-electromechanical systems. *Sensors and Actuators A: Physical*, 2006. 130–131(0): p. 419-429.
7. Lang et al., *Properties of High-Temperature Ceramics and Cermets, Elasticity and Density at Room Temperature*, Monograph 6, National Bureau of Standards, Washington, D.C. (1960).