Nanoaggregates of Benzothiazole-Based Amidine-Coupled Chemosensors: A Chemosensor for Ag^+ and the Resultant Complex as a Secondary Sensor for CI^-

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Figure S1. The absorption spectra of sensors (A) 1 and N1, (B) 2 and N2, and (C) 3 and N3. The absorption spectra of compounds 1-3 were recorded in pure methanol while the absorption spectra of sensors N1-N3 were taken in aqueous medium with a concentration of 20 μ M.



Figure S2. Distribution of particle size of (A) N1, (B) N2 and (C) N3 measured with a DLS-based particle size analyzer.

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Figure S3. Changes in UV-vis absorption spectra of sensors (A) N2 and (B) N3 (20 μ M) varying the pH in aqueous medium.



Figure S4. Distribution of particle size of sensor N2 upon addition of Ag^+ . The measurements were done with a DLS-based particle size analyzer.



Figure S5. Fluorescence responses of sensors (A) N2 and (B) N3 in water with a concentration of 20 μ M.



Figure S6. Calculation of the binding constants of sensors (A) N2 and (B) N3 with Ag⁺.



Figure S7. Distribution of particle size of sensor $N2.Ag^+$ upon addition of Cl⁻. The measurements were done with a DLS-based particle size analyzer.

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Figure S8. Calculation of the binding constant of complex N2.Ag⁺ with Cl⁻.



Figure S9. Bar diagram showing the absorbance of complex N2.Ag⁺ (20 μ M) binding to Cl⁻ (200 μ M) in the presence of other halides (200 μ M) in HEPES aqueous solution (20 mM, pH = 7.0) at 355 nm.



Figure S10. ¹H NMR of compound **2**.



Figure S11. ¹³C NMR of compound 2.



Figure S12. Mass spectrum of complex N2.Ag⁺.



Figure S13. ¹H NMR of compound **3**.



Figure S14. ¹³C NMR of compound **3**.