

Supplemental Information for:

Mechanism Studies of Terpolymerization of Phthalic Anhydride, Propylene Epoxide, and Carbon Dioxide Catalyzed by ZnGA

Yulei Liu, Min Xiao*, Shuanjin Wang, Liang Xia, Dongmei Hang, Guofeng Cui,

Yuezhong Meng*

The Key Laboratory of Low-carbon Chemistry & Energy Conservation of Guangdong

Province/State Key Laboratory of Optoelectronic Materials and Technologies, Sun

Yat-Sen University, Guangzhou 510275, PR China

Supplementary figures:

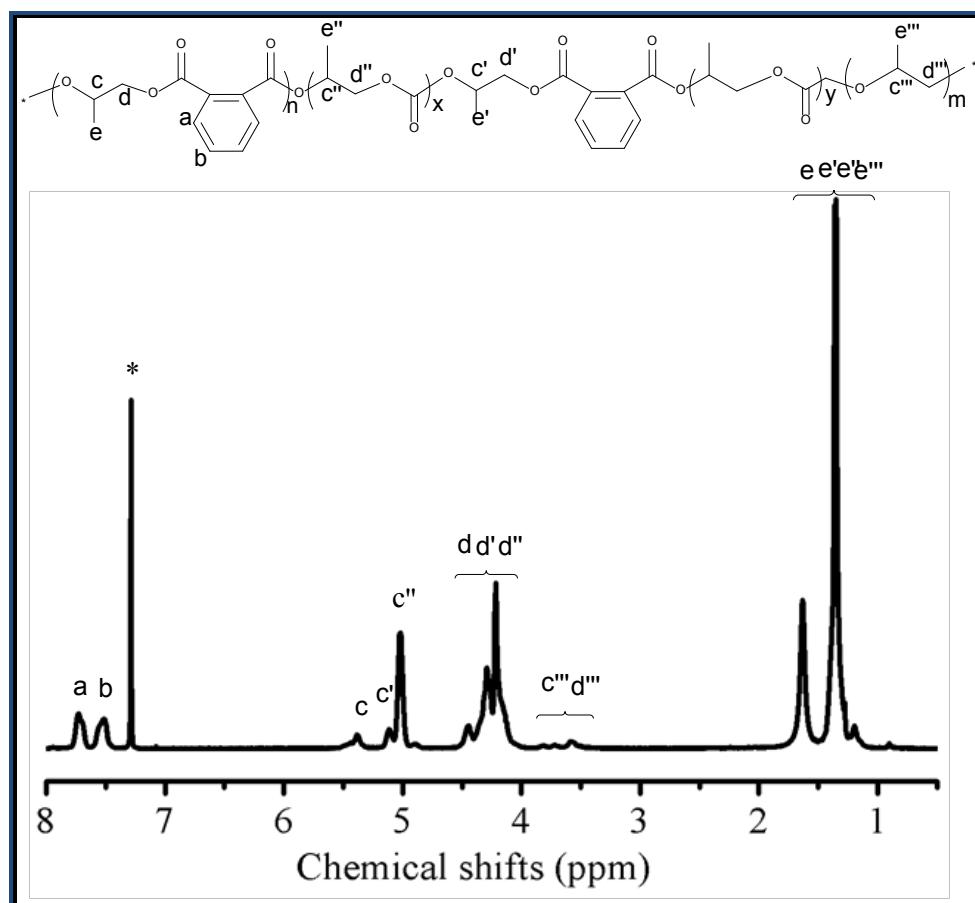


Figure S1. ¹H NMR spectra of PA/PO/CO₂ terpolymer

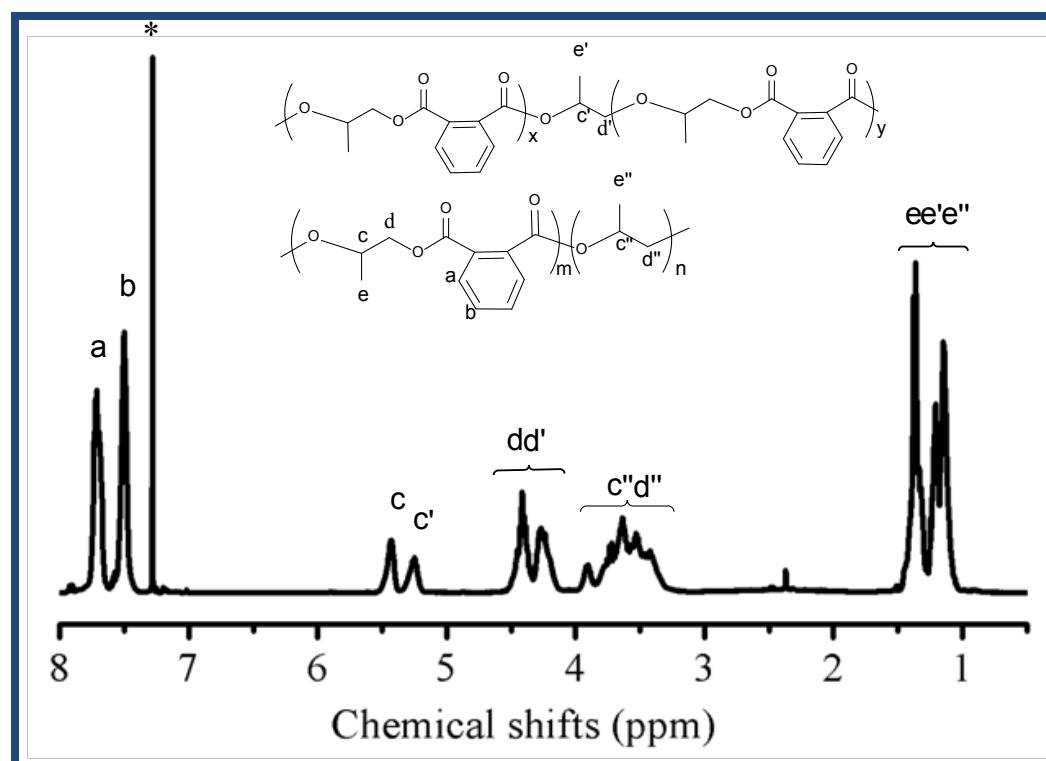


Figure S2. ¹H NMR spectra of PO/PA copolymer

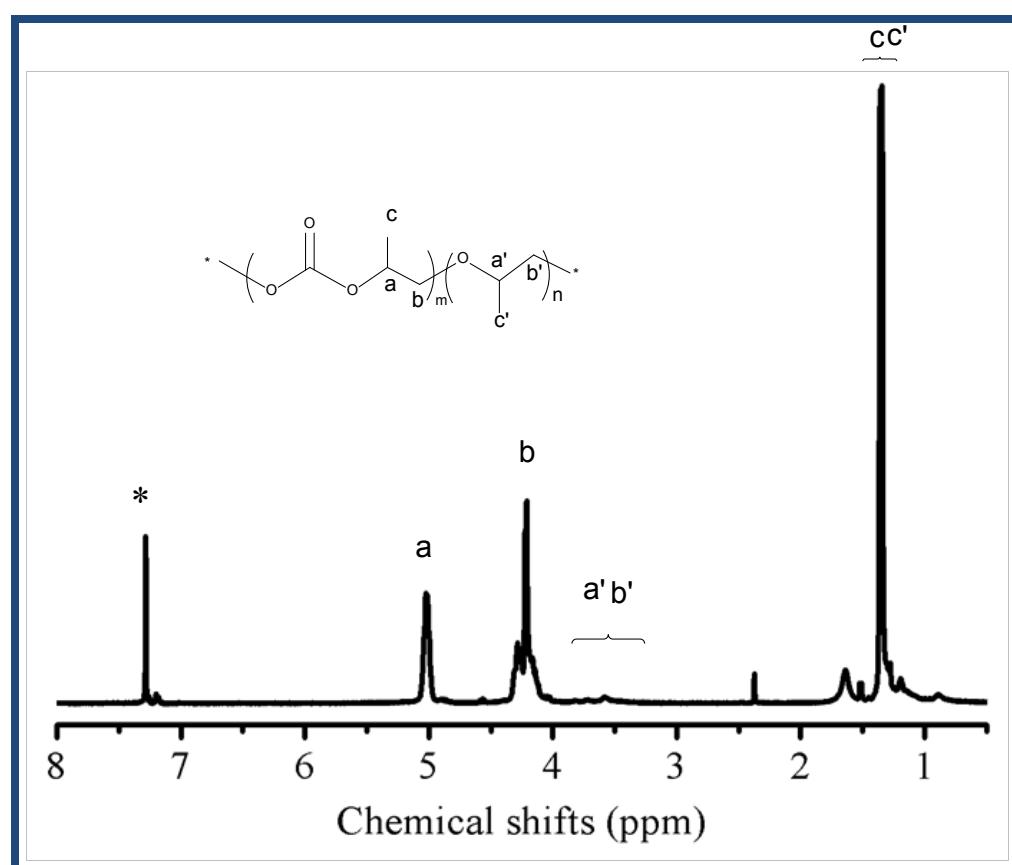


Figure S3. ¹H NMR spectra of PO/CO₂ copolymer

Supplementary equations:

The compositions of PA/PO/CO₂ terpolymers were estimated according to the ¹H NMR spectra by the following equations:

$$\text{PE (polyester) \%} = (A_{\delta(7.4-7.8)}/4)/(A_{\delta(5.2-5.5)}+A_{\delta(4.8-5.2)}+A_{\delta(3.2-4.0)}/3) \quad (\text{Eq.S1})$$

$$\text{PPO(polyether)\%} = (A_{\delta(3.2-3.6)}/3)/(A_{\delta(5.2-5.5)}+A_{\delta(4.8-5.2)}+A_{\delta(3.2-4.0)}/3) \quad (\text{Eq.S2})$$

$$\text{PC (polycarbonate)\%} = 100\% - \text{PE\%} - \text{PPO\%} \quad (\text{Eq.S3})$$

Where $A_{\delta(7.4-7.8)}$ is the total peak area of the two peaks in the range of $\delta=7.4-7.8$ ppm which is the peaks of the four protons at benzene ring; $A_{\delta(5.2-5.5)}$ is the peak area of the peak in the range of $\delta=5.2-5.5$ ppm which is the peak of CH in ester unit; $A_{\delta(4.8-5.2)}$ is the total peak area of the peaks in the range of $\delta=5.2-5.5$ ppm which is the peak of CH in carbonate unit ; $A_{\delta(3.2-4.0)}$ is the total peak area of the peaks in the range of $\delta=3.2-4.0$ ppm which is the peak of -CH-CH₂- in ether unit.