

From iridoids to dyes: a theoretical study on genipin reactivity - Supporting information

Stefania Di Tommaso¹, Hervé David², Jérôme Gomar², Frédéric Leroy², Carlo Adamo^{1, 3}

¹*Laboratoire d'Electrochimie, Chimie des Interfaces et Modélisation pour l'Energie, CNRS UMR 7575, Chimie ParisTech, 11, rue Pierre et Marie Curie, F-75231 Paris Cedex 05*

²*L'Oréal Research and Innovation, 1, Avenue Eugène Schueller, 93601 Aulnay-sous-Bois*

³*Institut Universitaire de France, 103 Boulevard Saint Michel, F-75005 Paris, France*

Table S1. Activation enthalpies ($\Delta H^\ddagger$, kcal/mol) and products stabilization (ΔH , kcal/mol) calculated for the water assisted open-ring reaction and tautomerization in gas phase and in solvent (PCM, solvent=ethanol).

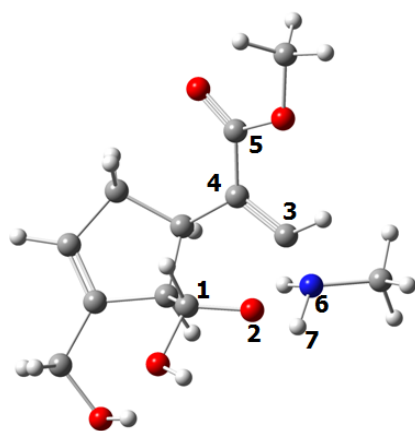
REACTION	GAS PHASE		SOLVENT	
	$\Delta H^\ddagger$	ΔH	$\Delta H^\ddagger$	ΔH
1→2' (H ₂ O assisted)	16.8	5.6	16.6	5.6
2'→2 (H ₂ O assisted)	19.2	4.1	16.5	2.3

Table S2 and S3. Activation barriers ($\Delta G^\ddagger$, kcal/mol) and products stabilization (ΔG , kcal/mol) calculated in the gas phase for each step of the reaction pathway producing **7** and starting from nucleophilic attack of methylamine on genipin in the open-ring form **2** dubbed **C** and **D** in Figure 2. Activation barriers and products stabilizations are also reported for the same reactions catalyzed by one or two water molecules.

REACTIONS	$\Delta G^\ddagger$	ΔG
C	32.1	0.5
2A $\rightarrow$ 3 + H ₂ O	58.7	2.9
+ H ₂ O	29.4	1.6
+ 2 H ₂ O	24.9	3.4
3 $\rightarrow$ 4	76.5	6.7
+ H ₂ O	30.5	-1.2
+ 2 H ₂ O	27.2	1.4
4 $\rightarrow$ 5cis	33.6	-22.6
+ H ₂ O	3.7	-17.2
+ 2 H ₂ O	2.6	-14.8
5cis $\rightarrow$ 7 + H ₂ O + H ₂ O	44.5	3.9

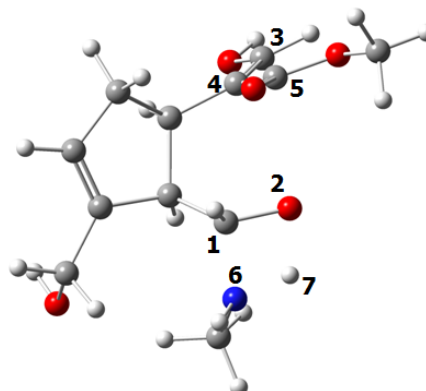
REACTIONS	$\Delta G^\ddagger$	ΔG
D	33.7	4.3
2A' $\rightarrow$ 3' + H ₂ O	49.6	2.0
+ 2 H ₂ O	22.7	0.5
3' $\rightarrow$ 4'	57.3	3.8
+ 2 H ₂ O	31.2	4.8
4' $\rightarrow$ 5'cis	23.2	-10.8
+ H ₂ O	5.2	-14.6
+ 2 H ₂ O	3.0	-13.9
5'cis $\rightarrow$ 7 + H ₂ O + H ₂ O	36.5	5.1

Figure S1. Optimized structure of the transition states involved in the reactions described in the paper.



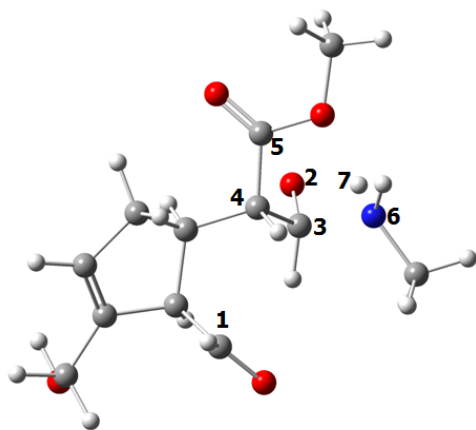
reaction A

C3-N6= 1.50 Å
O2-H7= 1.69 Å
C5-C4-C3-N6= -147.7°
C1-O2-C3-N6= -110.7 °



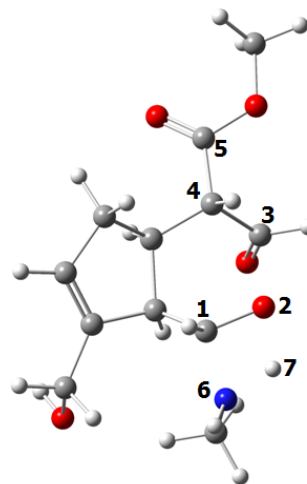
reaction B

C1-N6= 1.56 Å N6-H7= 1.21 Å
O2-H7= 1.38 Å O2-C3= 3.56 Å
N6-C1-O2= 95.4°
C1-O2-H7= 75.2 °
C3-C4-C5-O2= -106.8 °



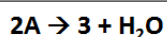
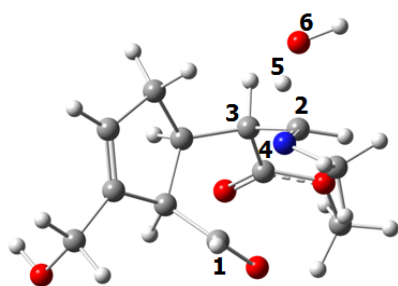
reaction C

C3-N6= 1.53 Å N6-H7= 1.22 Å
O2-H7= 1.37 Å C1-O2= 4.51 Å
N6-C3-O2= 97.2°
C3-O2-H7= 75.4 °
C5-C4-C3-N6= 67.5 °

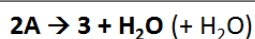
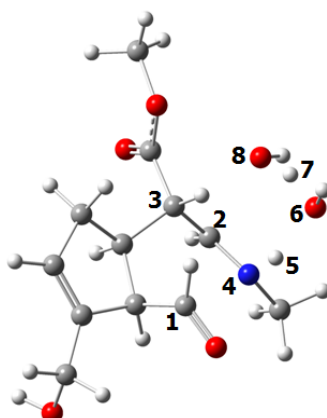


reaction D

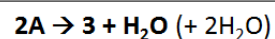
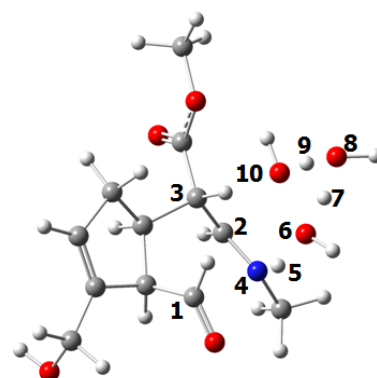
C1-N6= 1.54 Å N6-H7= 1.22 Å
O2-H7= 1.35 Å O2-C3= 2.45 Å
N6-C1-O2= 95.4°
C1-O2-H7= 76.0 °
C5-C4-C3-O2= 162.4 °



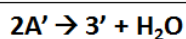
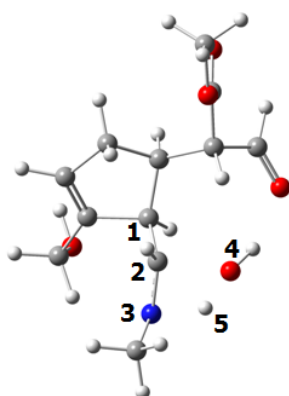
C1-N4= 2.64 Å N4-H5= 1.28 Å
 H5-O6= 1.27 Å C2-O6= 1.93 Å
 C2-N4-H5= 81.1°
 N4-H5-O6= 129.9°
 C3-C2-N4-H5= -96.0°



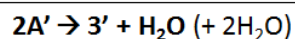
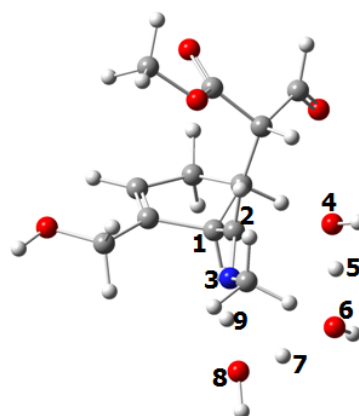
C1-N4= 3.16 Å N4-H5= 1.10 Å
 H5-O6= 1.51 Å O6-H7= 1.29 Å
 H7-O8= 1.15 Å C2-O8= 2.19 Å
 C2-N4-H5-O6= -59.1°
 H5-O6-H7-O8= -2.0°



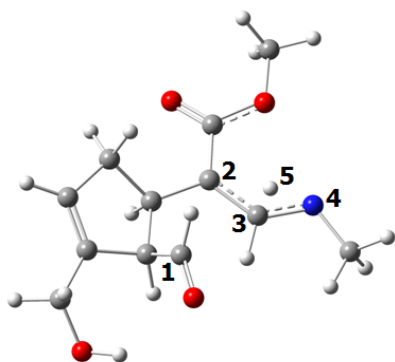
C1-N4= 3.16 Å N4-H5= 1.07 Å
 H5-O6= 1.60 Å O6-H7= 1.13 Å
 H7-O8= 1.32 Å O8-H9= 1.38 Å
 H9-O10= 1.09 Å C2-O10= 2.18 Å
 C2-N4-H5-O6= -31.4°
 O6-H7-O8-H9= -20.5°
 H5-N4-C2-O10= 83.1°



C2-N3= 1.35 Å C2-O4= 1.82 Å
 N3-H5= 1.34 Å O4-H5= 1.22 Å
 C1-C2-N3-H5= 100.4°
 C2-N3-H5-O4= 16.0°

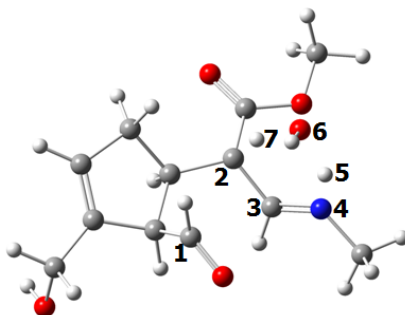


C2-N3= 1.29 Å C2-O4= 2.29 Å
 N3-H9= 1.07 Å O4-H5= 1.09 Å
 H5-O6= 1.38 Å O6-H7= 1.35 Å
 H7-O8= 1.10 Å O8-H9= 1.59 Å
 C1-C2-N3-H9= 21.3°
 C2-N3-H9-O8= 47.4°
 N3-H9-O8-H7= 18.4°
 H9-O8-H7-O6= -5.1°
 O8-H7-O6-H5= -16.1°
 H7-O6-H5-O4= 22.7°



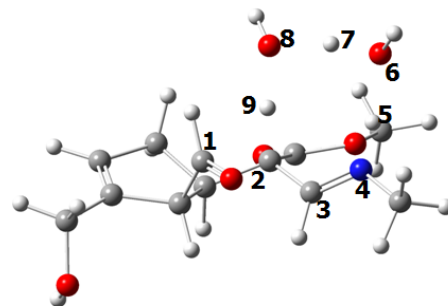
$3 \rightarrow 4$

C1-C3= 3.07 Å N4-H5= 1.64 Å
 C2-H5= 1.84 Å C3-H5= 1.17 Å
 C2-C3-N4-H5= 81.1°



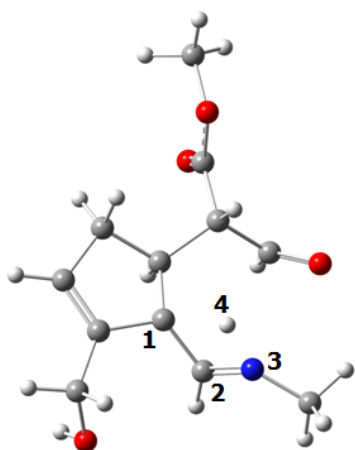
$3 \rightarrow 4 (+ H_2O)$

C1-C3= 3.05 Å N4-H5= 1.12 Å
 H5-O6= 1.46 Å O6-H7= 1.31 Å
 C2-H7= 1.34 Å
 C2-C3-N4-H5= 9.2°
 N4-H5-O6-H7= -10.3°



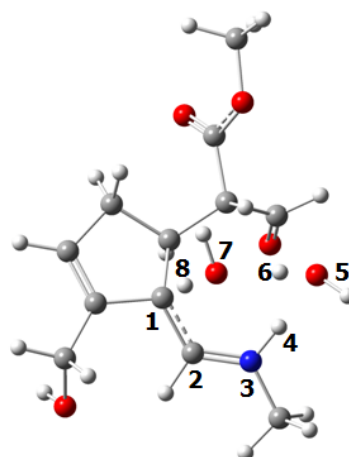
$3 \rightarrow 4 (+ 2H_2O)$

C1-C3= 2.95 Å N4-H5= 1.09 Å
 H5-O6= 1.53 Å O6-H7= 1.10 Å
 H7-O8= 1.36 Å O8-H9= 1.48 Å
 C2-H9= 1.22 Å
 C2-C3-N4-H5= 6.4°
 N4-H5-O6-H7= -44.1°
 O6-H7-O8-H9= -22.1°



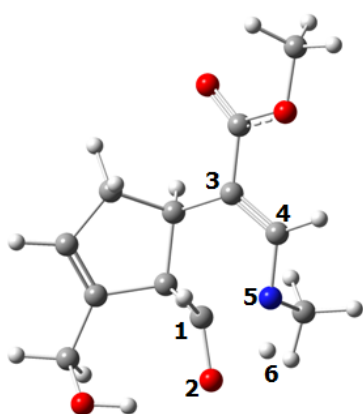
$3' \rightarrow 4'$

C1-C2= 1.47 Å C2-N3= 1.29 Å
 N3-H4= 1.25 Å C1-H4= 1.63 Å
 C1-C2-N3-H4= 2.3°



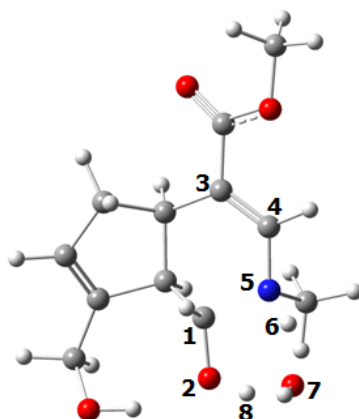
$3' \rightarrow 4' (+ 2H_2O)$

C1-C2= 1.44 Å C2-N3= 1.30 Å
 N3-H4= 1.04 Å H4-O5= 1.81 Å
 O5-H6= 1.03 Å H6-O7= 1.55 Å
 C1-H8= 1.29 Å
 C1-C2-N3-H4= 4.6°
 C2-N3-H4-O5= 71.4°
 N3-H4-O5-H6= -33.8°
 H4-O5-H6-O7= -21.4°
 O5-H6-O7-H8= -7.3°



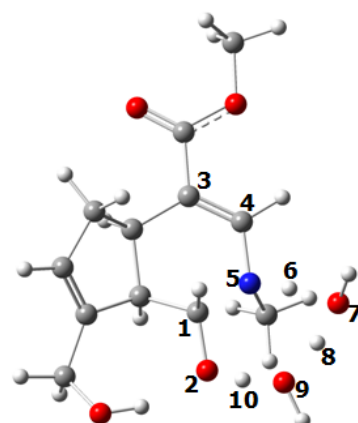
4 → 5*cis*

C1-N5= 1.52 Å	N5-H6= 1.25 Å
O2-H6= 1.42 Å	
C1-O2-H6= 72.9°	
C4-N5-H6= 139.8°	
C3-C4-N5-H6= -122.7°	



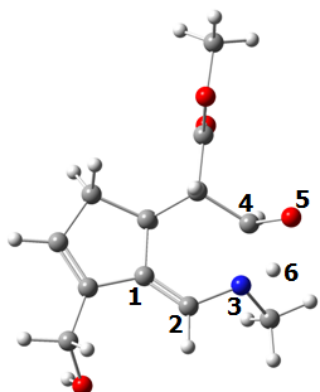
4 → 5*cis* (+ H₂O)

C1-N5= 1.55 Å	N5-H6= 1.15 Å
H6-O7= 1.42 Å	O7-H8= 1.24 Å
H8-O2= 1.20 Å	
C1-O2-H8= 101.4°	
C4-N5-H6= 116.7°	
C3-C4-N5-H6= -139.1°	
C4-N5-H6-O7= 162.6°	
N5-H6-O7-H8= -18.1°	



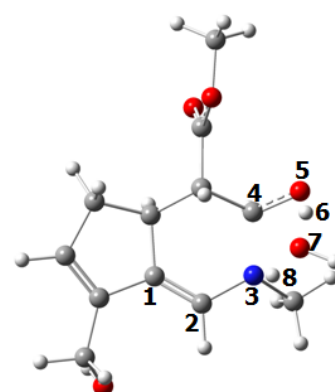
4 → 5*cis* (+ 2H₂O)

C1-N5= 1.55 Å	N5-H6= 1.14 Å
H6-O7= 1.41 Å	O7-H8= 1.16 Å
H8-O9= 1.26 Å	O9-H10= 1.26 Å
H10-O2= 1.17 Å	
C1-O2-H10= 109.7°	
C4-N5-H6= 107.3°	
C3-C4-N5-H6= -138.3°	
C4-N5-H6-O7= -167.0°	
N5-H6-O7-H8= -49.1°	
H6-O7-H8-O9= -4.5°	
O7-H8-O9-H10= 16.1°	



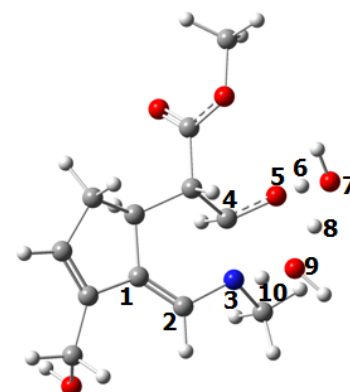
4' → 5'*cis*

C2-N3= 1.43 Å	N3-C4= 1.56 Å
C4-O5= 1.35 Å	N3-H6= 1.21 Å
O5-H6= 1.39 Å	
C4-O5-H6= 75.8°	
C1-C2-N3-H6= 105.4°	



4' → 5'*cis* (+ H₂O)

C2-N3= 1.44 Å	N3-C4= 1.57 Å
C4-O5= 1.34 Å	O5-H6= 1.21 Å
H6-O7= 1.23 Å	O7-H8= 1.46 Å
N3-H8= 1.12 Å	
C4-O5-H6= 103.9°	
C1-C2-N3-H8= 104.2°	
C2-N3-H8-O7= -133.3°	
N3-H8-O7-H6= -13.8°	



4' → 5'*cis* (+ 2H₂O)

C2-N3= 1.45 Å	N3-C4= 1.56 Å
C4-O5= 1.34 Å	O5-H6= 1.15 Å
H6-O7= 1.28 Å	O7-H8= 1.27 Å
H8-O9= 1.15 Å	O9-H10= 1.47 Å
N3-H10= 1.12 Å	
C4-O5-H6= 113.4°	
C1-C2-N3-H10= 98.5°	
C2-N3-H10-O9= -85.4°	
N3-H10-O9-H8= -60.9°	
H10-O9-H8-O7= 0.1°	
O9-H8-O7-H6= 1.3°	

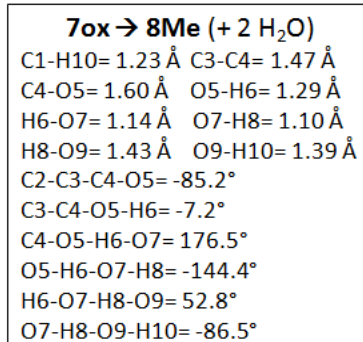
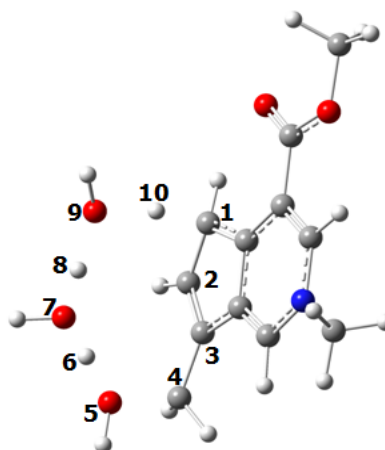
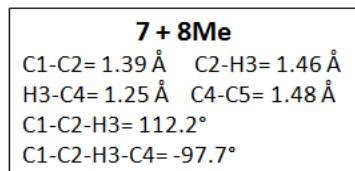
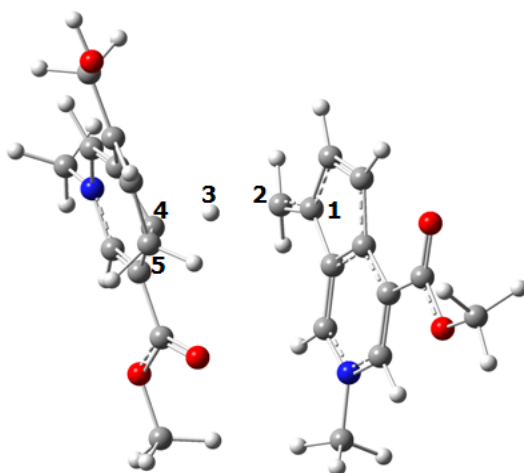
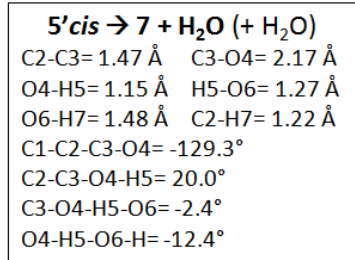
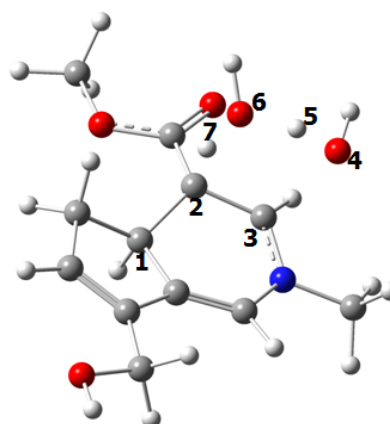
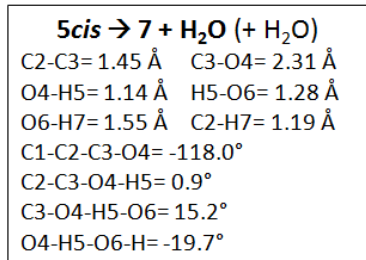
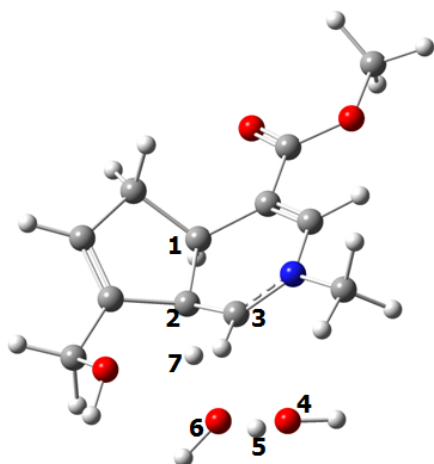


Figure S2. UV-visible spectrum simulated (PBE0/6-31+G(d)) for the intermediate 7. Optimized structure of the intermediate and his calculated emission color are also reported.

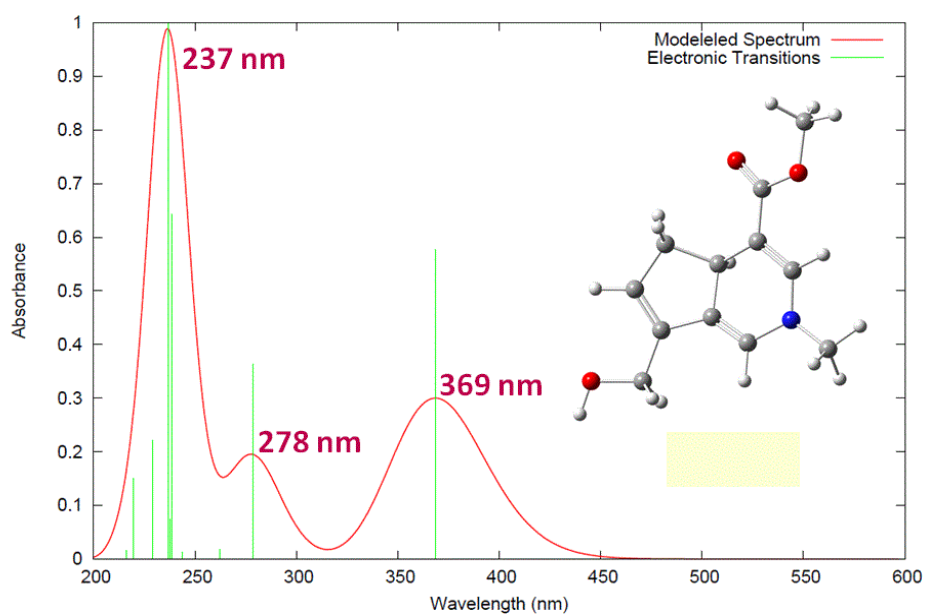


Figure S3. Orbitals involved in the HOMO-LUMO electronic transition of **8** intermediate.

