Hierarchical porous NiCo₂O₄ nanograss arrays grown on Ni foam as

electrode material for high-performance supercapacitors

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The calculation of mass loading

The ICP analysis was used to calculate the mass loading on the Ni foam. In a typical

procedure, the NiCo₂O₄ combined with Ni foam obtained after the synthesis reaction

was dissolved in 25 ml of concentrated nitric acid, and diluted with deionized

water to 250 ml. Then the samples were analyzed by ICP-AES. In the ICP-AES

analysis, the content of cobalt was tested. As the content of Cobalt in the NiCo₂O₄ is

constant, the mass loading of NiCo₂O₄ on the Ni foam could be calculated through the

Equation (1). The results were shown in the table S1.

 $M=a\times250 \text{ (ml)}\times M_{\text{NiCo2O4}}/1000 \text{ (ml)}\times 2M_{\text{Co}}$ Equation (1)

Where M (mg) represents the mass loading of NiCo₂O₄, a (ppm) represents the

content of cobalt, M_{NiCo2O4} represents the molar mass of NiCo₂O₄, M_{Co} represents the

molar mass of cobalt.

Table S1. The Cobalt content of the samples.

No	Content of cobalt (ppm)	Mass loading (mg)
1	0	0
2	4.3	2.20
3	4.2	2.14
4	4.3	2.20
5	4.3	2.20
6	4.4	2.25
Average	4.3	2.20

To eliminate the influence of Nickel foam that may import the Cobalt into the solution, we have tested the pure Nickel foam (No 1) to detect the content of Cobalt. Fortunately, the content of Cobalt in Nickel foam is 0, showing that the Nickel foam did not import the impurity. Through the calculation, we got that the mass loading of $NiCo_2O_4$ was 2.2 mg.

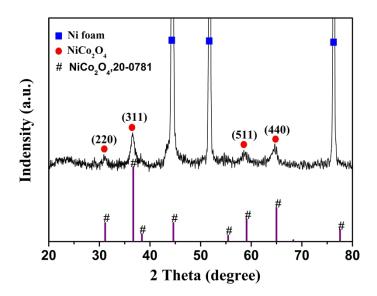


Fig.S1. XRD pattern of hierarchical porous NiCo₂O₄ nanograss arrays grown on Ni foam.

Fig.S1 shows the typical XRD spectrum of the product. The diffraction peaks except for the three typical peaks originating from the Ni foam can be indexed to spinel NiCo₂O₄ phase (JCPDS no 20-0781), which demonstrate the material obtained after the thermal treatment was NiCo₂O₄.

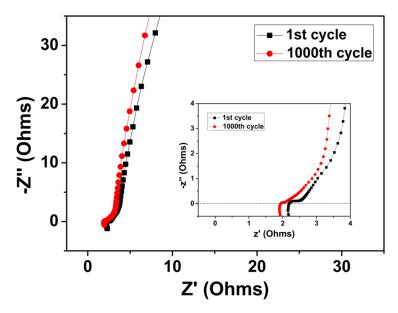


Fig.S2 EIS spectrum 1st and 1000th cycle of the NiCo₂O₄ electrode. The inset is an enlarged curve of the high frequency region.

To explore the evidence for the capacitance increase in the first 1000 cycles, the EIS analysis was adopted. Fig. S2 shows the EIS spectrum of hierarchical porous $NiCo_2O_4$ nanograss arrays on Ni foam electrode obtained after 1^{st} and 1000^{th} cycle. Both the impedance spectra are almost similar in form with a quasi-semicircle at high-frequency and a linear component at the low-frequency. The internal resistance (R_b) , which includes the total resistances of the ionic resistance of electrolyte, intrinsic resistance of active materials and contact resistance at the active material/current collector interface, can be obtained from the intercept of the plots on real axis. As can be seen from the inset, the R_b of 1000^{th} cycle is 1.8, smaller than 2.2 of the 1^{st} cycle, which agrees well with the result shown in the long-term stability test, indicating the enhancement of the ion diffusion and effective electron transfer in the electrode reaction. In addition, in the low frequency region, the slope of the impedance plot of the 1000^{th} cycle shows more vertical than the 1^{st} cycle, indicating

the enhancement of electrochemical capacitance the $NiCo_2O_4$ nanograss/Ni foam electrode in the KOH aqueous solution through the activation after 1000 cycles.²

References

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- 2. C. Yuan, J. Li, L. Hou, X. Zhang, L. Shen and X. W. D. Lou, *Adv. Funct. Mater.*, 2012, 22, 4592.