

## **Electronic Supplementary Information**

### **Cyclic Arginyl-Glycyl-Aspartic Acid (RGD) Peptide-Induced Synthesis of Uniform and Stable One-Dimensional CdTe Nanostructures in Aqueous Solution**

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## 1. Experimental Section

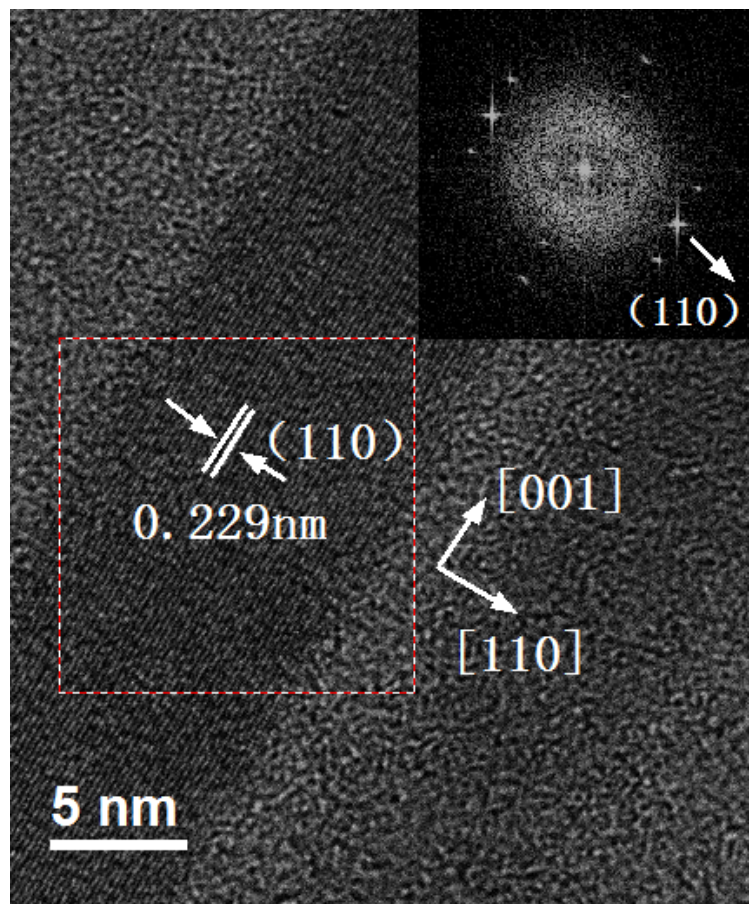
### 1.1. Materials

Tellurium powder (99.997%),  $\text{CdCl}_2$  (99.99+%), and  $\text{NaBH}_4$  ( $\geq 96\%$ ) were purchased from Aldrich. Cyclo(RGDfC), cylco(GGDfC), cylco(KGDfC), cylco(EGDfC) and acetyl-RGDfC-amide were purchased from GL Biochem Ltd. (Shanghai, China) with purities of  $> 95\%$ . All chemicals were used without additional purification. All solutions were prepared with ultra-pure water ( $18.2 \text{ M}\Omega\cdot\text{cm}$ ) purified on Millipore System (Millipore, USA).

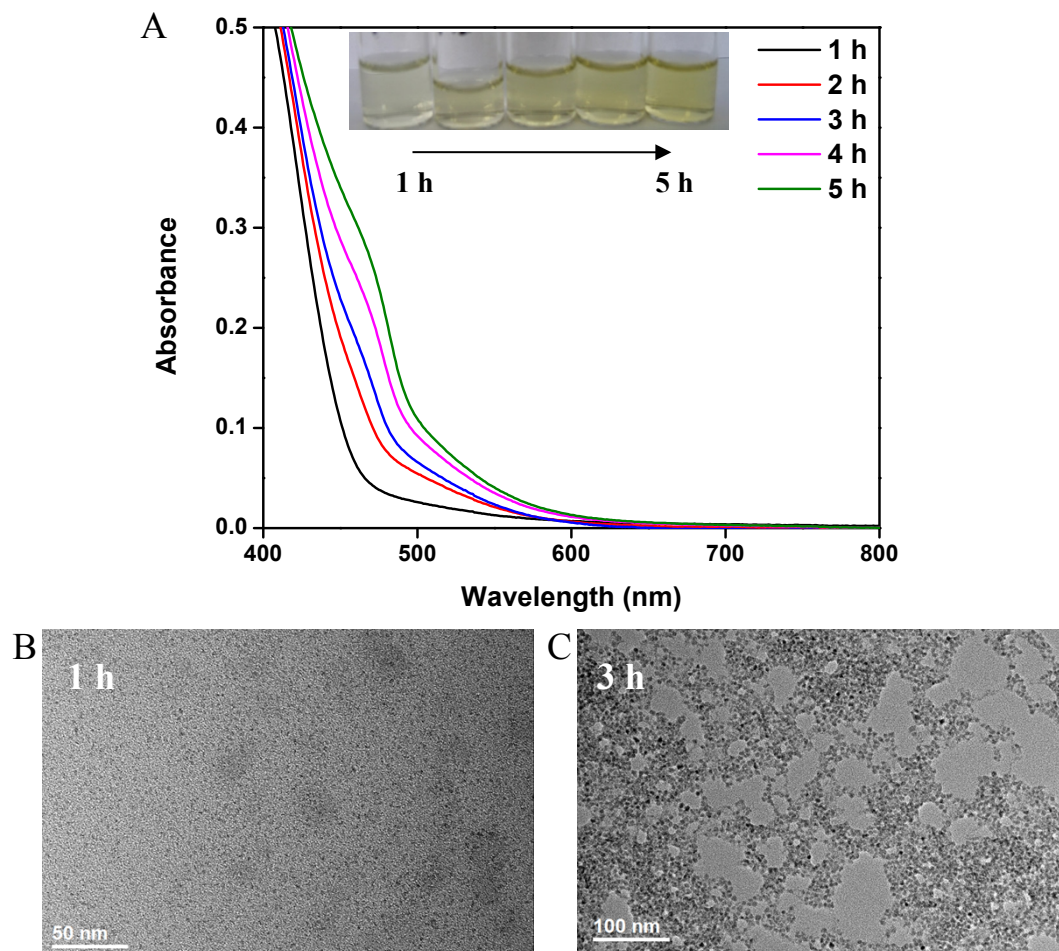
### 1.2. Synthesis and characterization of 1D CdTe nanocrystals

$\text{NaHTe}$  solution was freshly prepared by dissolving 0.05 g  $\text{NaBH}_4$  in 3 ml water and then 0.04 g Te powder was added into the  $\text{NaBH}_4$  solution. This reaction was conducted at room temperature overnight in a syringe with a needle to help release the gas generated during the reaction, and the resulting  $\text{NaHTe}$  solution was then diluted by injecting into 122 ml ultrapure water prior to use.  $\text{Cd}^{2+}$ -CRGDS precursor solution was prepared by dissolving  $\text{CdCl}_2$  and a given RGD peptide in ultrapure water, and then adjusted to pH 8.5 with 1 M  $\text{NaOH}$ . The  $\text{NaHTe}$  solution was then injected into a  $\text{N}_2$ -saturated precursor solution under vigorous stirring. The typical molar ratio of  $\text{Cd}^{2+}$ ,  $\text{HTe}^-$  and peptide introduced was 2 : 1 : 5 in a total volume of 12 ml with 1 mM peptide. The resulting mixture was heated to  $98^\circ\text{C}$  and refluxed at different times to control the growth of the CdTe nanocrystals. Ultraviolet-visible (UV-Vis) absorption spectra were recorded on a Shimadzu UV-2450 spectrophotometer performed at room temperature under ambient conditions. Fourier transform infrared (FTIR) spectra were obtained using Nicolet 6700 FTIR (Nicolet Instrument Company, USA) between 500 and  $4000 \text{ cm}^{-1}$ . TEM samples were prepared by dropping the aqueous nanocrystals onto carbon-coated copper grids with excess solvent evaporated. TEM and HRTEM images and SAED were recorded on a JEM-2100 electron microscope operating at 200 kV.

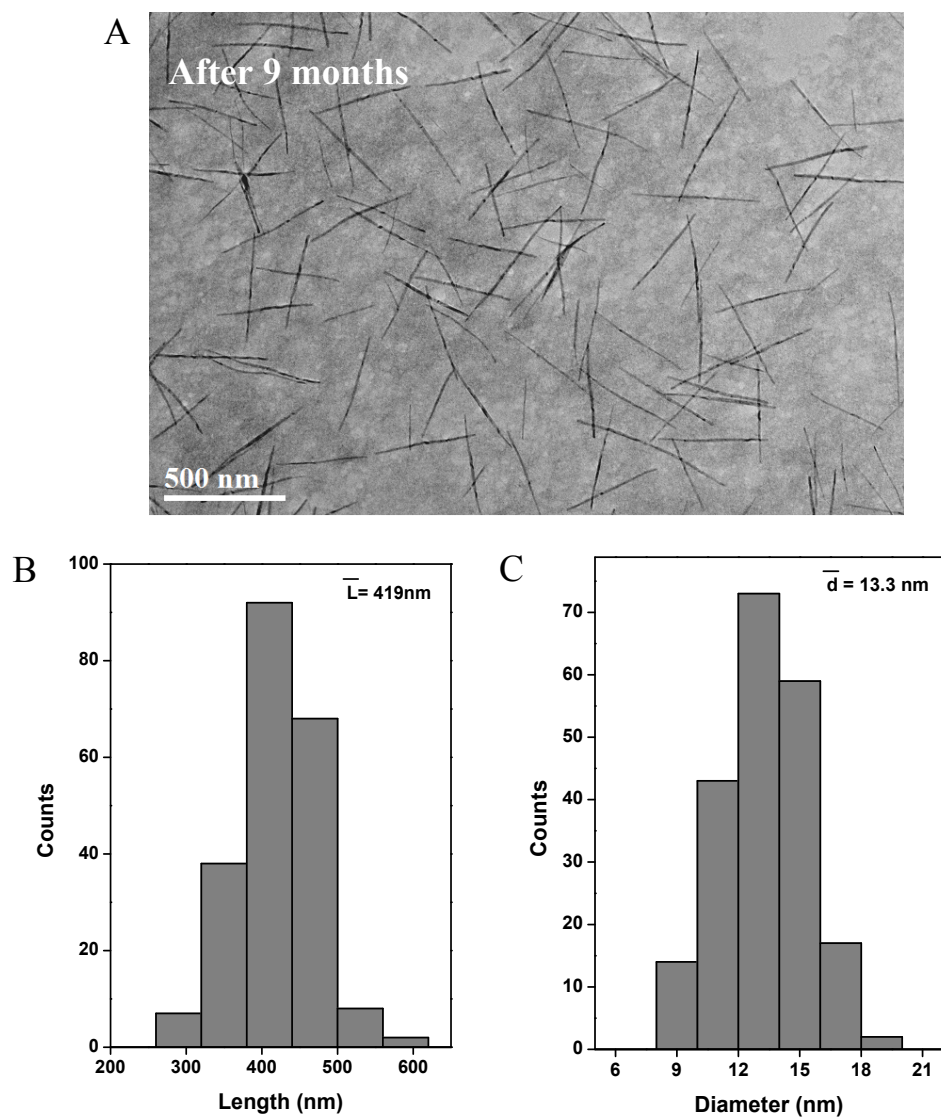
## 2. Figures S1-S6



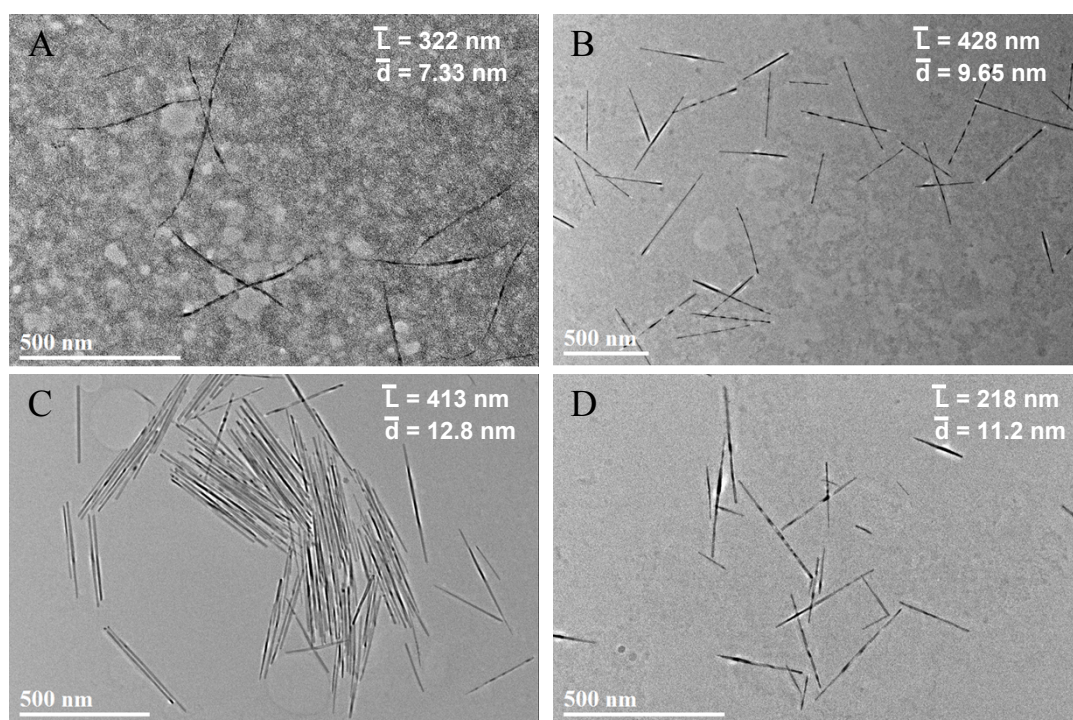
**Fig. S1** HRTEM image taken from part of an individual 1D CdTe formed from a precursor solution with  $0.4\text{ mM Cd}^{2+}$  and  $\text{Cd:Te:cylco(RGDfC)} = 2:1:5$  after 1 h of heating. The inset shows the FFT pattern of the indicated square area. Note that the  $[110]$  orientation of the  $(110)$  plane is perpendicular with respect to the  $[001]$  direction.



**Fig. S2** (A) Absorption spectra and optical photographs (the inset) of the crude CdTe solutions prepared with the precursor Cd/Te ratio of 4/0.5 ( $\text{Cd}^{2+} = 0.4 \text{ mM}$ ,  $\text{cyclo(RGDfC)}/\text{Cd}^{2+} = 2.5/1$ ) after different periods of heating. (B and C) TEM images of spherical CdTe nanostructures formed after 1 and 3 h of heating, respectively.



**Fig. S3** (A) Representative TEM image of 1D CdTe nanocrystals after 9 months of storage at 4 °C. The nanocrystal solution was prepared with the precursor Cd/Te ratio of 2/1 ( $\text{Cd}^{2+} = 0.4\text{ mM}$ ,  $\text{cyclo(RGDfC)}/\text{Cd}^{2+} = 2.5/1$ ) and 1 h of heating. (B) Length and (C) diameter distribution histograms of 1D CdTe nanocrystals, which were derived from TEM imaging.

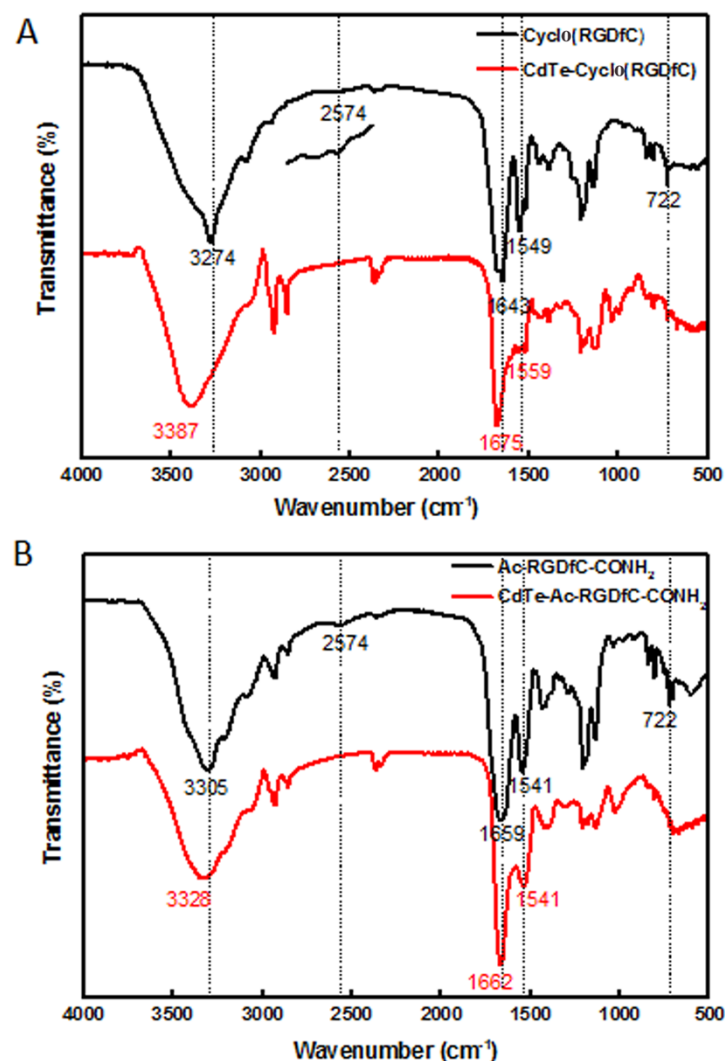


**Fig. S4** TEM images of CdTe nanostructures prepared with the precursor Cd/Te ratios of (A) 4/0.75, (B) 4/1, (C) 4/2, and (D) 4/4 ( $\text{Cd}^{2+} = 0.4 \text{ mM}$ ,  $\text{cyclo(RGDfC)}/\text{Cd}^{2+} = 2.5/1$ ) after 1 h of heating.





**Fig. S5** Optical photographs of Cd-acetyl-RGDfC-amide, Cd-cyclo(KGDfC), Cd-cyclo(GGDfC) Cd-cyclo(EGDfC) and Cd-cyclo(RGDfC) precursor solutions at pH 8.5.



**Fig. S6** FTIR spectra of (A) cyclo(RGDfC) and cyclo(RGDfC)-derived 1D CdTe nanostructures; (B) acetyl-RGDfC-amide and acetyl-RGDfC-amide-derived CdTe nanostructures. The weak bands at 2574  $\text{cm}^{-1}$  and at 722  $\text{cm}^{-1}$  corresponded to the thiol (-SH) in the cyclo(RGDfC) and acetyl-RGDfC-amide spectrum disappeared in peptide-derived CdTe nanostructures, suggesting the formation of thiolate (-S-CdTe).<sup>[1,2]</sup> However, , -N-H stretching ( $\sim 3274 \text{ cm}^{-1}$ ), amide I ( $\sim 1643 \text{ cm}^{-1}$ , C=O stretching) and amide II ( $\sim 1549 \text{ cm}^{-1}$ , N-H bending) regions in CdTe-cyclo(RGDfC) spectrum were shifted to higher energy with respect to the free cyclo(RGDfC) peptide, while these characteristic bands in CdTe- acetyl-RGDfC-amide spectrum exhibit a little or no shifts with respect to the free acetyl-RGDfC-amide. These results showed that the guanidine and amide besides the thiol in cyclo(RGDfC) were involved in the formation and stabilization of 1D CdTe nanostructures.<sup>[2-4]</sup>



## Reference:

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