Supporting Information

Development of a cell permeable ratiometric chemosensor and biomarker for hydrogen sulphate ions in aqueous solution

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Fig. S1a FTIR spectrum of L.



Fig. S1b 1 H NMR spectrum of the probe (L) in DMSO-d₆.



Fig. S1c ESI-MS spectrum of probe (L).



Fig. S1d 13 C NMR spectrum of the probe (L) in DMSO-d₆.



Fig. S2a FTIR spectrum of L-HSO₄⁻ ensembled species.



Fig. S2b ¹H NMR spectrum of the species L-HSO₄⁻ in DMSO-d₆, where TBA= \underline{T} etra \underline{B} utyl <u>A</u>mmonium group.



Fig. S2c ESI-MS spectrum of L-HSO₄⁻ species.



Fig. S2d 13 C NMR spectrum of the species L-HSO₄⁻ in DMSO-d₆.



Fig. S3 Visual colour change of L, A) in absence and B) in presence of HSO_4^- ions.



Fig. S4 Absorbance - emission overlay spectra of L (10 μ M) in HEPES buffer (1 mM, pH 7.4; 2% EtOH) at 25 °C.



Fig. S5 Fluorescence response of the L in absence / presence of HSO_4^- in HEPES buffer (1 mM, 2% EtOH) at 25 °C in different pH at λ_{em} = 483nm.



Fig. S6 Visual fluorescence colour change of L, A) in absence and B) in presence of HSO_4^- ions in UV light.



Fig. S7 Ratiometric signal output of L.



Fig. S8 Detection limit of HSO₄⁻ ions (5.5×10⁻⁷ M) in HEPES buffer (1 mM, pH 7.4; 2% EtOH) at 25 °C at λ_{em} = 483nm.



Fig.S9 Interference of different anions (30 μM) in presence of L (10.0 μM) and HSO₄⁻ (10.0 μM) in HEPES buffer (1 mM, pH 7.4; 2% EtOH) at 25 °C where a) F^- , b) Cl⁻, c) Br⁻, d) Γ , e) CN⁻, f) N₃⁻, g) NO₃⁻, h) ClO₄⁻, i) H₂PO₄⁻, j) HPO₄²⁻, k) H₂AsO₄⁻, l) HAsO₄²⁻, m) AsO₃³⁻, n) OAc⁻, o) SO₄²⁻, p) S²⁻ at λ_{em} = 483nm.



Fig. S10 Job's plot for stoichiometry determination between L and HSO_4^- ions from emission intensity showing maximum emission at 1:1 ratio in HEPES buffer (1 mM, pH 7.4; 2% EtOH) at 25 °C at λ_{em} = 483nm.



Fig. S11 Partial ¹H NMR spectra for L (10 mM) in presence of varying $[HSO_4^-]$ [A) 0 mM, B) 3.33 mM, C) 6.67 mM, and D) 10 mM] in DMSO-d₆.



Fig. S12 Binding constant (*K*) value $4.13 \times 10^6 \text{ M}^{-1}$ determined from the interaction of **L** with HSO₄⁻ ions in HEPES buffer (1 mM, pH 7.4; 2% EtOH) at 25 °C.



Fig. S13 Optimized structure of L and L-HSO₄ species.



Fig. S14 Cytotoxic effect of L (5, 10, 20, and 50 μ M) in HeLa cells incubated for 12 h by MTT assay. Results are expressed as mean of three independent experiments.

Empirical Formula	$C_{15}H_{12}N_4O_4$
Formula Weight	312.09
Crystal System	triclinic
Space group	P-1
a (Å)	7.187(5)
b (Å)	8.453(5)
c (Å)	12.382(5)
α (°)	93.382(5)
β (°)	95.364(5)
γ (°)	107.885(5)
Volume (Å ³)	709.7(7)
Temperature, K	293(2)
Z	2
ρ_{calc} (g/cm ³)	1.283
F (000)	284
θ range(deg)	1.66 to 28.19°
Collected reflns	12449
Independent reflns	3426
R flns with $I > 2\sigma(I)$	2801
R1 [I > 2.0 $\sigma(I)$]	0.0421
wR1 [I > 2.0 σ (I)]	0.1262
Goodness-of-fit	0.92

Table S1 Crystal data and details of refinements for $C_{15}H_{12}N_3O.NO_3$

Bond distances (Å)					
C8-N3	1.3887(18)	C1-N1	1.3557(18)		
C9-N3	1.3351(18)	C11-N1	1.4114(17)		
C9-N2	1.3455(17)	C1-C2	1.5134(19)		
C3-N2	1.3947(17)	C3-C8	1.3920(20)		
C2-N2	1.4602(18)	C10-C11	1.4080(20)		
Bond angles	(°)				
N1-C1-C2	115.95(12)	N3 C9 C10	126.70(12)		
N2-C2-C1	109.73(12)	N2 C9 C10	124.04(12)		
C8-C3-N2	106.01(11)	C1 N1 C11	128.78(12)		
N3-C8-C3	107.12(11)	C9 N2 C3	108.84(11)		
N3-C9-N2	109.26(12)	C3 N2 C2	127.96(11)		
01-C1-N1	122.09(13)	C9 N3 C8	108.74(11)		

Table S2 Selected bond distances (Å) and bond angles (°) for $C_{15}H_{12}N_3O.NO_3$

Table S3 Fluorescence quantum yield (Φ_{f}) and life time (τ_f in ns) of the corresponding singlet excited states

	B ₁	B ₂	$ au_{ m av}(m ns)$	χ^2	φ	$k_r(10^8 s^{-1})$	$k_{nr} (10^9 s^{-1})$
L	39.69	60.31	5.77	1.07	0.12	0.2079	0.1525
L+HSO ₄ ⁻ (1:0.5)	10	00	11.79	1.01	0.26	0.2205	0.0628
L+HSO ₄ ⁻ (1:1)	10	00	12.15	1.02	0.45	0.3704	0.0453

Table S4 HOMO-LUMO energy of L and L-HSO ₄ species

Component	номо	LUMO	Difference(a.u.)	Difference(eV)
L	-0.21452	-0.04806	0.16646	4.53
L + HSO ₄ (Chain)	-0.04630	+0.04798	0.09428	2.57
L + HSO ₄ (Ring)	-0.09893	+0.03975	0.13868	3.77