Sample	O (atm %)	Mn (atm %)	K (atm %)	O/Mn (%)
S1	69.48	0.52		2.29
S2	72.66	26.84	0.50	2.71
S3	63.29	34.21	2.50	1.85
S4	75.85	24.15		3.14

Table S1. The atomic ratios of the elements in the samples obtained at different



concentration of precursors.

Fig. S1 Nitrogen adsorption-desorption isotherms of the sample S1 (a), S2 (b), and

S3 (c) obtained at different concentration of precursors.

Sample	BET	Pore volume	Average pore size	
	$(m^2 g^{-1})$	$(cm^3 g^{-1})$	(nm)	
S1	14.21	0.1779	61.74	
S2	20.09	0.1728	34.39	
<b>S</b> 3	31.76	0.2184	27.5	

Table S2 The pore characters of the samples



Fig. S2 The XRD patterns of samples obtained at different concentration of

precursors.



Fig. S3 The XRD patterns of samples obtained at different time: (a) 15 min, (b) 2h,

(c) 6h, (d) 10h.



Fig. S4 O 1s XPS spectra of the sample S1 (a), S2 (b), S3 (c), and S4 (d).

Samples	Binding Energy (eV)				Oxygen distribution (%)		
	Mn 2P 3/2	$O_{\alpha}$	$O_{\beta}$	$O_{\gamma}$	$O_{\alpha}$	$O_{\beta}$	$O_{\gamma}$
S1	642.3	529.5	531.2	532.8	61.34	32.67	5.99
S2	642.2	529.7	531.3	532.8	70.50	25.19	4.31
<b>S</b> 3	642.1	529.6	531.4	532.8	67.02	27.37	5.61
S4	642.3	529.6	531.2	532.8	60.28	33.60	6.12

**Table S3** XPS results of the samples.



Fig. S5 SEM images of commercial MnO<sub>2</sub>



Fig. S6 XRD pattern of commercial MnO<sub>2</sub>



Fig. S7 XRD pattern of the catalyst after the test of the catalytic ability cycle



Fig. S8 SEM images the catalyst after the test of the catalytic ability cycle



*Fig. S9* Nitrogen adsorption-desorption isotherms and pore size distribution (inset) of the catalyst after the test of the catalytic ability cycle



Fig. S10 The catalytic ability cycle of S3 in the catalytic oxidation of benzene