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Figure a) Kinetics of change of difference spectra at the interaction of (I) (1.5·10⁻⁴ M) with Hb (2·10⁻⁵ M). Dotted line (1) is spectrum of Hb. Spectra 2-5 were registered at 0.08 (2), 0.33 (3), 0.6 (4) h after start of reaction; spectra 5-13 were registered with intervals 1 h , spectrum 15 at h after start of reaction. Conditions of reaction: 25 °C, solvent is 0.05 M Tris-HCl buffer, pH 7.



Figure b) Kinetics of change of difference spectra at the interaction of *(II)* $(1.5 \cdot 10^{-4} \text{ M})$ with Hb $(2 \cdot 10^{-5} \text{ M})$. Dotted line (1) is spectrum of Hb. Spectra 2-8 were registered at 0.5 (2), 1 (3), 3 (4), 5 (5), 7 (6), 9 (7), 11(8) h after start of reaction. Conditions of reaction: 25 °C, solvent is 0.05 M Tris-HCl buffer, pH 7.0



Figure c) Kinetics of change of difference spectra at the interaction of (*I*) $(1.5 \cdot 10^{-4} \text{ M})$ with Hb $(2 \cdot 10^{-5} \text{ M})$. Dotted line (1) is spectrum of Hb. Spectra 2-5 were registered at 0.08 (2), 0.33 (3), 0.6 (4) h after start of reaction; spectra 5-13 were registered with intervals 1 h , spectrum 15 at 11 h after start of reaction. Conditions of reaction: 25 °C, solvent is 0.05 M Tris-HCl buffer, pH 7.0.