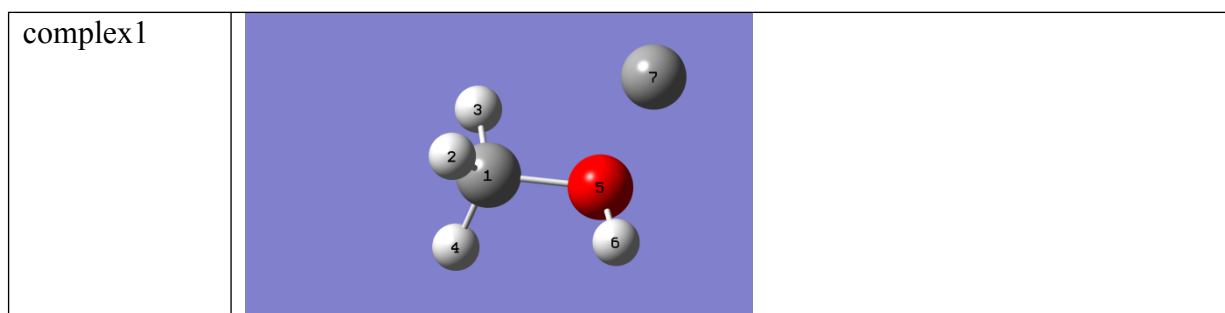


Supporting information file for “The fast C(³P) + CH₃OH reaction as an efficient loss process for gas-phase interstellar methanol.”

Quantum Chemical Calculations

C + CH₃OH theoretical calculations, geometries and frequencies (distances in Angstrom, angles in degrees, Frequencies in cm⁻¹ without corrections factors)



	M06-2X/cc-pVQZ	CCSD(T)-F12/aug-cc-pVTZ
H ₂ -C ₁	1.0872	1.0885
H ₃ -C ₁	1.0838	1.0849
H ₄ -C ₁	1.0870	1.0887
O ₅ -C ₁	1.4362	1.4440
H ₆ -O ₅	0.9636	0.9645
C ₇ -O ₅	1.6895	1.7444
H ₃ -C ₁ -H ₂	106.70	109.91
H ₄ -C ₁ -H ₃	110.70	110.62
O ₅ -C ₁ -H ₄	109.18	108.89
H ₆ -O ₅ -C ₁	110.81	109.61
C ₇ -O ₅ -H ₆	104.29	101.92
Diedre(H ₄ -C ₁ -H ₃ -H ₂)	123.62	123.59
Diedre(O ₅ -C ₁ -H ₄ -H ₃)	-115.85	-115.68
Diedre(H ₆ -O ₅ -C ₁ -H ₄)	-63.05	-63.84
Diedre(C ₇ -O ₅ -H ₆ -C ₁)	119.22	116.71

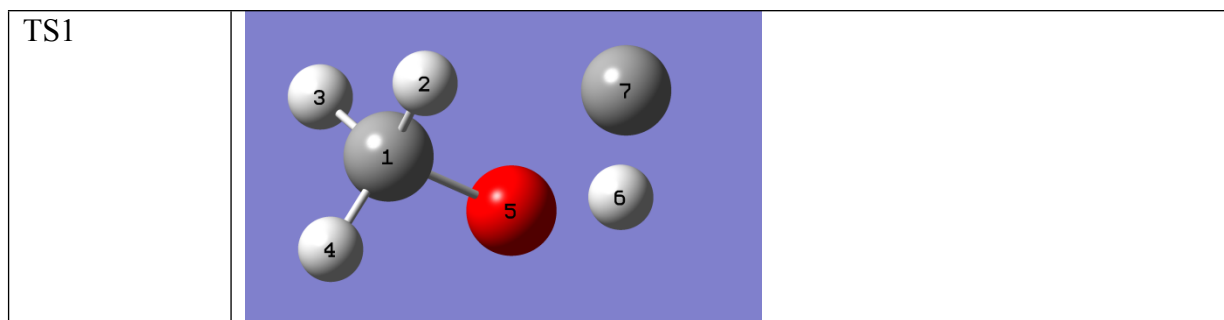
Frequencies (cm⁻¹):

M06-2X/cc-pVQZ:

151, 301, 441, 652, 1041, 1073, 1182, 1342, 1468, 1490, 1509, 3072, 3158, 3185, 3836

CCSD(T)/aug-cc-pVTZ:

158, 259, 368, 633, 1004, 1067, 1176, 1350, 1469, 1500, 1512, 3054, 3142, 3177, 3782



	M06-2X/cc-pVQZ	CCSD(T)-F12/aug-cc-pVTZ
H ₂ -C ₁	1.0884	1.0894
H ₃ -C ₁	1.0864	1.0873
H ₄ -C ₁	1.0857	1.0873
O ₅ -C ₁	1.4353	1.4457
H ₆ -O ₅	1.1673	1.1727
C ₇ -O ₅	1.5588	1.6000
H ₃ -C ₁ -H ₂	109.62	108.89
H ₄ -C ₁ -H ₃	110.81	110.92
O ₅ -C ₁ -H ₄	107.91	107.58
H ₆ -O ₅ -C ₁	112.85	111.22
C ₇ -O ₅ -C ₁	113.00	111.55
Diedre(H ₄ -C ₁ -H ₃ -H ₂)	122.93	123.31
Diedre(O ₅ -C ₁ -H ₄ -H ₃)	-117.42	-117.11
Diedre(H ₆ -O ₅ -C ₁ -H ₄)	-102.96	-101.44
Diedre(C ₇ -O ₅ -C ₁ -H ₆)	-61.50	-58.75

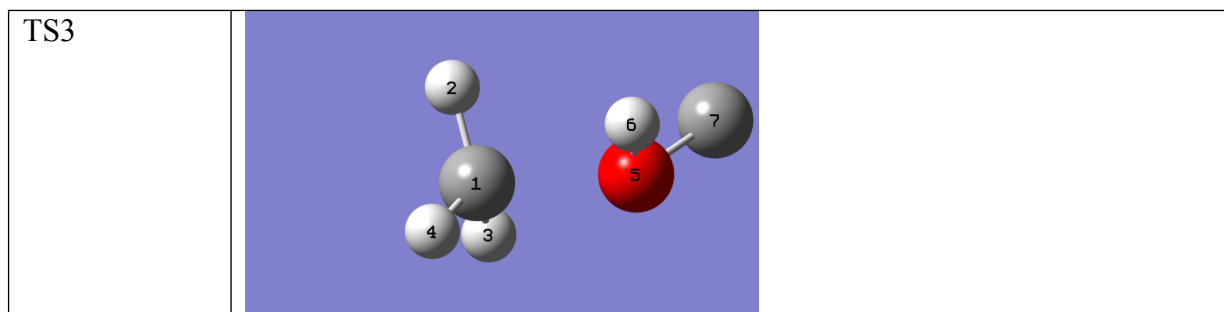
Frequencies (cm⁻¹):

M06-2X/cc-pVQZ:

1495i, 191, 358, 700, 849, 1027, 1188, 1210, 1469, 1494, 1509, 2244, 3072, 3159, 3172

CCSD(T)-F12/aug-cc-pVTZ:

1544i, 115, 333, 625, 841, 982, 1169, 1196, 1456, 1494, 1503, 2213, 3047, 3141, 3155



	M06-2X/cc-pVQZ	CCSD(T)-F12/aug-cc-pVTZ
H ₂ -C ₁	1.0802	1.0827
H ₃ -C ₁	1.0787	1.0806

H ₄ -C ₁	1.0811	1.0833
O ₅ -C ₁	1.7940	1.8078
H ₆ -O ₅	0.9723	0.9735
C ₇ -O ₅	1.3878	1.4035
H ₃ -C ₁ -H ₂	115.30	115.46
H ₄ -C ₁ -H ₃	115.77	115.95
O ₅ -C ₁ -H ₄	105.31	104.88
H ₆ -O ₅ -C ₁	101.65	100.07
C ₇ -O ₅ -C ₁	116.70	117.35
Diedre(H ₄ -C ₁ -H ₃ -H ₂)	140.95	141.66
Diedre(O ₅ -C ₁ -H ₄ -H ₃)	-106.45	-106.34
Diedre(H ₆ -O ₅ -C ₁ -H ₄)	-63.47	-61.95
Diedre(C ₇ -O ₅ -C ₁ -H ₆)	-119.18	-116.49

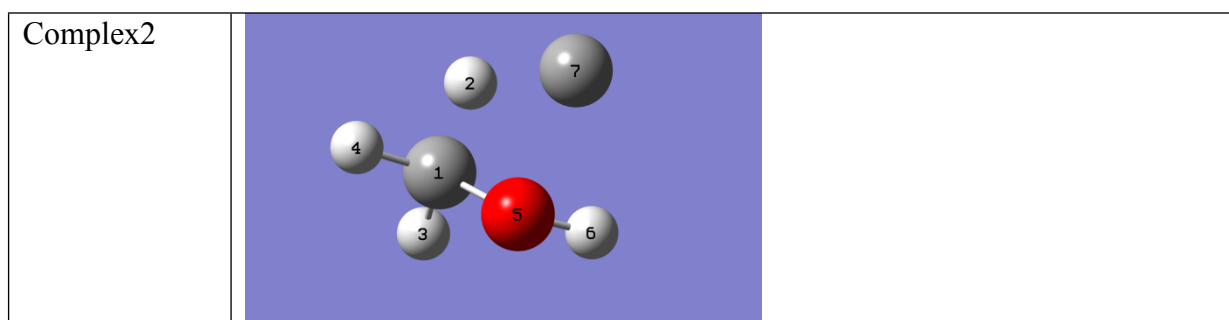
Frequencies (cm⁻¹):

M06-2X/cc-pVQZ:

1050i, 162, 296, 705, 771, 875, 928, 1112, 1174, 1437, 1457, 3101, 3259, 3272, 3626

CCSDT/aug-cc-pVTZ:

1199i, 131, 287, 694, 725, 838, 903, 1097, 1156, 1443, 1454, 3081, 3236, 3253, 3604



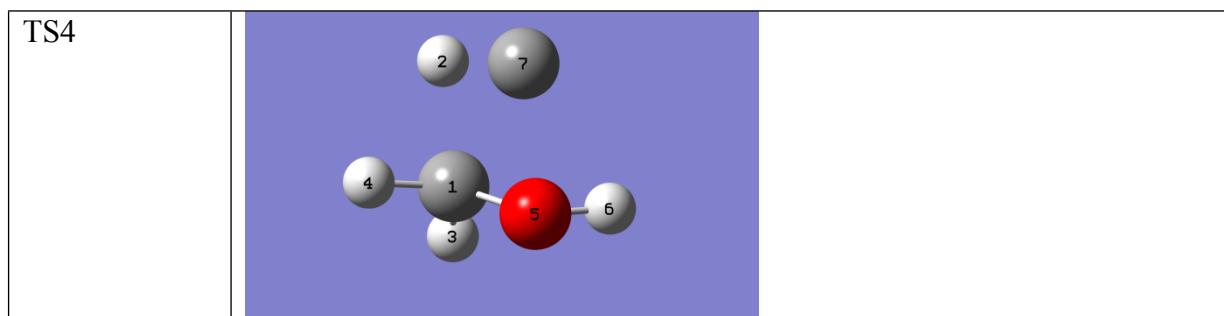
	M06-2X/cc-pVQZ	CCSD(T)-F12/aug-cc-pVTZ
H ₂ -C ₁	1.1661	1.1545
H ₃ -C ₁	1.0876	1.0894
H ₄ -C ₁	1.0892	1.0896
O ₅ -C ₁	1.3761	1.3888
H ₆ -O ₅	0.9626	0.9622
C ₇ -O ₅	2.8177	2.9223
H ₃ -C ₁ -H ₂	108.01	108.06
H ₄ -C ₁ -H ₃	111.16	111.14
O ₅ -C ₁ -H ₄	109.07	108.64
H ₆ -O ₅ -C ₁	108.60	108.00
C ₇ -O ₅ -C ₁	53.21	53.05
Diedre(H ₄ -C ₁ -H ₃ -H ₂)	109.03	111.17
Diedre(O ₅ -C ₁ -H ₄ -H ₃)	-128.13	-126.90
Diedre(H ₆ -O ₅ -C ₁ -H ₄)	-175.85	-177.48
Diedre(C ₇ -O ₅ -C ₁ -H ₆)	33.64	34.80

M06-2X/cc-pVQZ:

183, 241, 365, 577, 1028, 1112, 1162, 1346, 1386, 1484, 1584, 2231, 3089, 3157, 3847

CCSD(T)-F12/aug-cc-pVDZ:

126, 181, 310, 479, 1016, 1105, 1130, 1346, 1290, 1471, 1519, 2287, 3063, 3136, 3803



	M06-2X/cc-pVQZ	CCSD(T)-F12/aug-cc-pVTZ
H ₂ -C ₁	1.4395	1.4682
H ₃ -C ₁	1.0887	1.0903
H ₄ -C ₁	1.0890	1.0899
O ₅ -C ₁	1.3561	1.3623
H ₆ -O ₅	0.9650	0.9661
C ₇ -O ₅	2.5613	2.5699
H ₃ -C ₁ -H ₂	113.25	112.63
H ₄ -C ₁ -H ₃	111.92	112.70
O ₅ -C ₁ -H ₄	109.46	109.48
H ₆ -O ₅ -C ₁	108.42	107.55
C ₇ -O ₅ -C ₁	45.81	45.93
Diedre(H ₄ -C ₁ -H ₃ -H ₂)	97.19	97.18
Diedre(O ₅ -C ₁ -H ₄ -H ₃)	-131.91	-132.48
Diedre(H ₆ -O ₅ -C ₁ -H ₄)	-173.38	-174.20
Diedre(C ₇ -O ₅ -C ₁ -H ₆)	37.40	38.92

Frequencies (cm⁻¹):

M06-2X/cc-pVQZ:

610i, 254, 432, 566, 619, 855, 1136, 1207, 1237, 1387, 1491, 2387, 3091, 3165, 3803

CCSDT/aug-cc-VTZ:

613i, 226, 393, 515, 569, 835, 1117, 1166, 1205, 1390, 1487, 2468, 3055, 3140, 3745

Table S1 Reaction enthalpies in kJ mol^{-1} (at 0 K including ZPE) of species involved in the C + methanol reaction by different methods.

	Complex1	TS1	TS3	Complex2	TS4
MRCI+Q/avtz//					
CASSCF/avtz	-27	+23	+42	+21	+43
CCSDT-F12/avtz	-51	+3	+18	-9	-3
M06/vqz	-69	-17	-1	-26	-20
CASPT2/avqz//					
CASSCF/avqz	-50	-28	-15	-29	-38

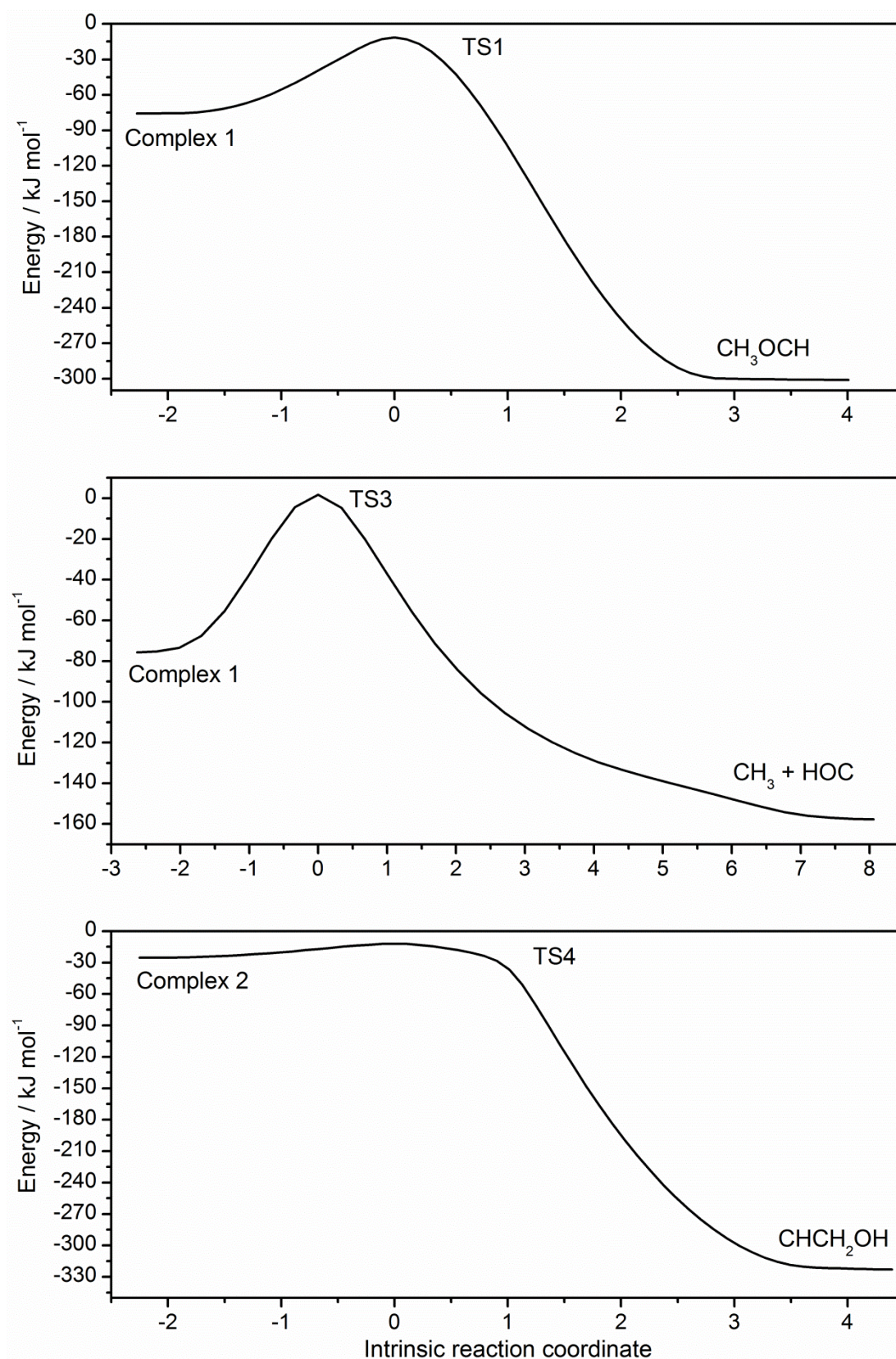


Figure S1 Potentials along the minimum energy path leading from complex1 and complex2 to adducts/products.