

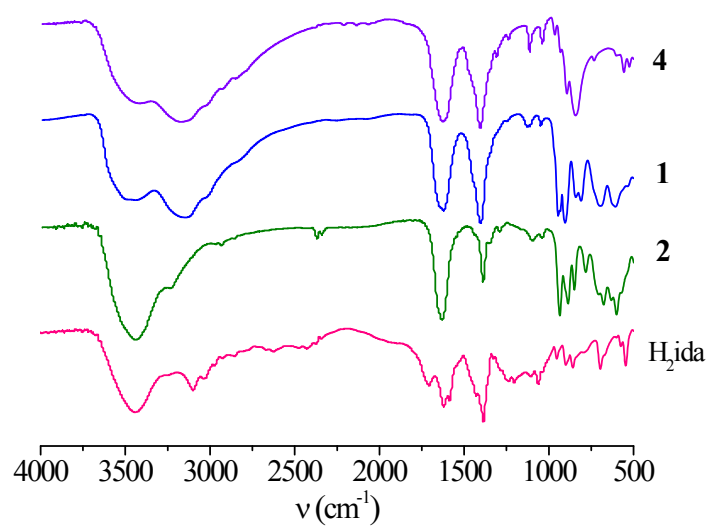
## Supporting Information

### Transformations and reductions of $\gamma$ -octamolybdates with their monomeric and dimeric aminopolycarboxylates

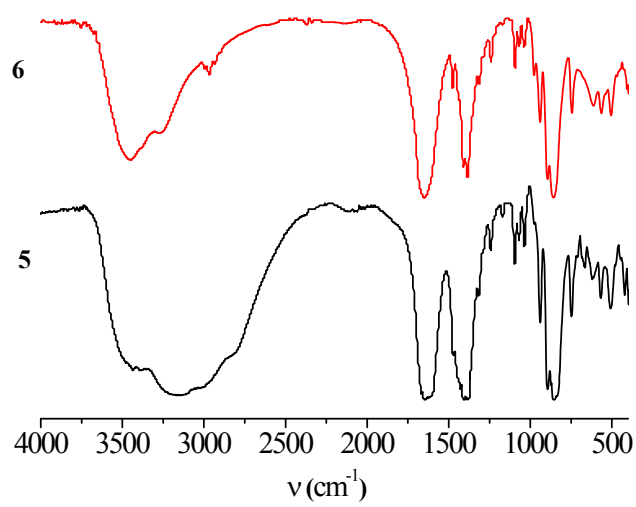
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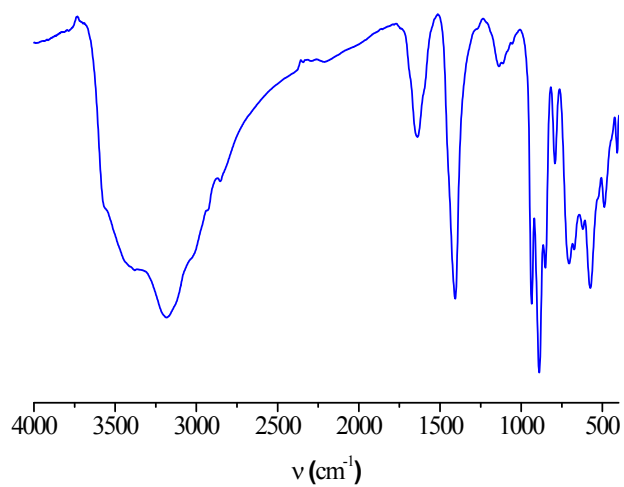
- Figure S1. (a) IR spectra of iminodiacetato molybdates:  $(\text{NH}_4)_4[\text{MoO}_3(\text{ida})]_2 \cdot 3\text{H}_2\text{O}$  (**4**) top;  $(\text{NH}_4)_8\{\gamma\text{-Mo}_8\text{O}_{26}[\text{MoO}_3(\text{ida})]_2\} \cdot 6\text{H}_2\text{O}$  (**1**) middle, and  $\text{K}_8\{\gamma\text{-Mo}_8\text{O}_{24}[\text{MoO}_3(\text{ida})]_2\} \cdot 12\text{H}_2\text{O}$  (**2**) bottom; (b) IR spectra of dimeric propanediaminetetraacetato molybdates:  $(\text{NH}_4)_5[\text{Mo}_2\text{O}_6(1,3\text{-pdta})]\text{Cl} \cdot 2\text{H}_2\text{O}$  (**5**),  $\text{K}_4[\text{Mo}_2\text{O}_6(1,3\text{-pdta})] \cdot 3\text{H}_2\text{O}$  (**6**); (c) IR spectrum of the complex  $(\text{NH}_4)_{6n}(\text{Mo}_8\text{O}_{27})_n \cdot 4n\text{H}_2\text{O}$  (**7**). (d) IR spectrum of the complex  $(\text{NH}_4)_{4n}(\text{Mo}_8\text{O}_{26})_n$  (**8**).
- Figure S2. UV spectra of  $(\text{NH}_4)_8\{\gamma\text{-Mo}_8\text{O}_{26}[\text{MoO}_3(\text{ida})]_2\} \cdot 6\text{H}_2\text{O}$  (**1**) and  $\text{K}_8\{\gamma\text{-Mo}_8\text{O}_{24}[\text{MoO}_3(\text{ida})]_2\} \cdot 12\text{H}_2\text{O}$  (**2**).
- Figure S3. Perspective view of the 3D structure of  $(\text{NH}_4)_4[\text{MoO}_3(\text{ida})]_2 \cdot 3\text{H}_2\text{O}$  (**4**).
- Figure S4. Perspective view of the 2D structure of  $(\text{NH}_4)_8\{\gamma\text{-Mo}_8\text{O}_{26}[\text{MoO}_3(\text{ida})]_2\} \cdot 6\text{H}_2\text{O}$  (**1**).
- Figure S5.  $^1\text{H}$  NMR spectra of the complexes  $(\text{NH}_4)_5[\text{Mo}_2\text{O}_6(1,3\text{-pdta})]\text{Cl} \cdot 2\text{H}_2\text{O}$  (**5**) and  $\text{K}_4[\text{Mo}_2\text{O}_6(1,3\text{-pdta})] \cdot 3\text{H}_2\text{O}$  (**6**); x indicates resonance signals of the coordinated ligand, o for the free ligand.
- Table S1. Bond valence calculations of molybdenum complexes.
- Table S2. Comparisons of the Mo–O and Mo–N distances (Å) in the iminodiacetato complexes.
- Table S3. Comparisons of the Mo–O and Mo–N distances (Å) in the propanediaminetetraacetato molybdenum complexes.



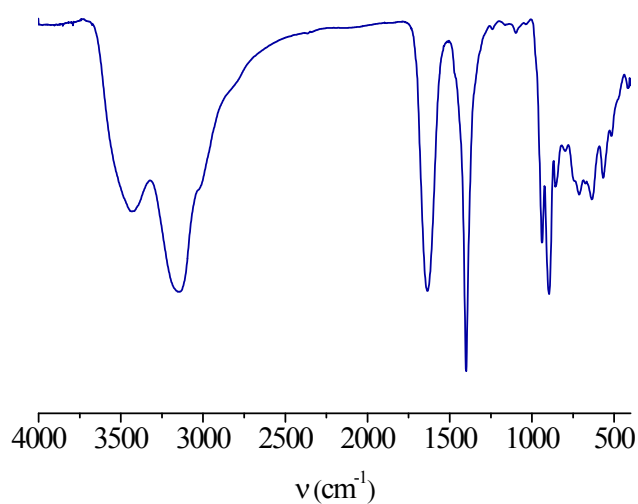
(a)



(b)



(c)



(d)

Figure S1. (a) IR spectra of iminodiacetato molybdates:  $(\text{NH}_4)_4[\text{MoO}_3(\text{ida})]_2 \cdot 3\text{H}_2\text{O}$  (**4**) top;  $(\text{NH}_4)_8\{\gamma\text{-Mo}_8\text{O}_{26}[\text{MoO}_3(\text{ida})]_2\} \cdot 6\text{H}_2\text{O}$  (**1**) middle, and  $\text{K}_8\{\gamma\text{-Mo}_8\text{O}_{24}[\text{MoO}_3(\text{ida})]_2\} \cdot 12\text{H}_2\text{O}$  (**2**) bottom; (b) IR spectra of dimeric propanediaminetetraacetato molybdates:  $(\text{NH}_4)_5[\text{Mo}_2\text{O}_6(1,3\text{-pdta})]\text{Cl} \cdot 2\text{H}_2\text{O}$  (**5**),  $\text{K}_4[\text{Mo}_2\text{O}_6(1,3\text{-pdta})] \cdot 3\text{H}_2\text{O}$  (**6**); (c) IR spectrum of the complex  $(\text{NH}_4)_{6n}(\text{Mo}_8\text{O}_{27})_n \cdot 4n\text{H}_2\text{O}$  (**7**). (d) IR spectrum of the complex  $(\text{NH}_4)_{4n}(\text{Mo}_8\text{O}_{26})_n$  (**8**).

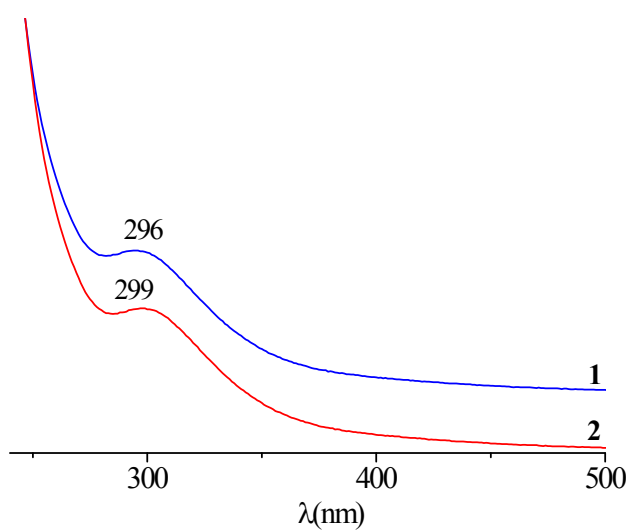


Figure S2. UV spectra of  $(\text{NH}_4)_8\{\gamma\text{-Mo}_8\text{O}_{26}[\text{MoO}_3(\text{ida})]_2\} \cdot 6\text{H}_2\text{O}$  (**1**) and  $\text{K}_8\{\gamma\text{-Mo}_8\text{O}_{26}[\text{MoO}_2(\text{ida})]_2\} \cdot 12\text{H}_2\text{O}$  (**2**).

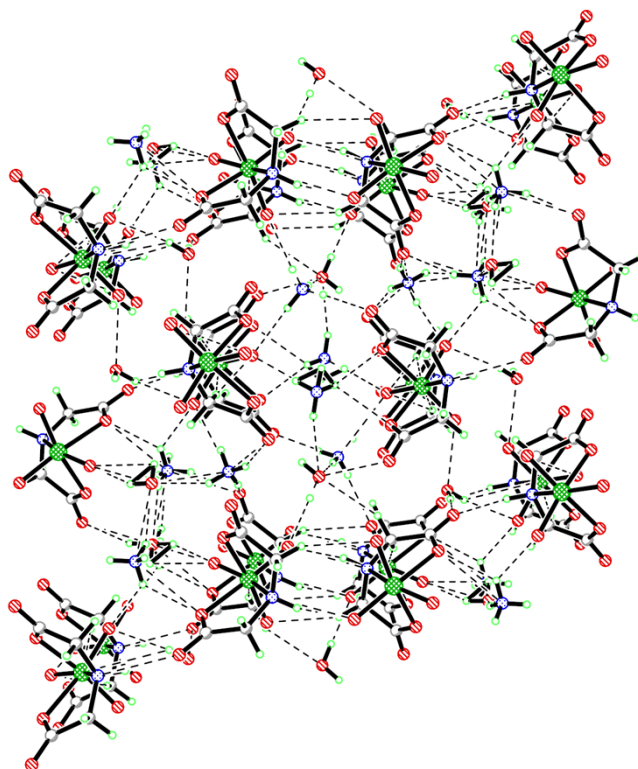


Figure S3. Perspective view of the 3D structure of  $(\text{NH}_4)_4[\text{MoO}_3(\text{ida})]_2 \cdot 3\text{H}_2\text{O}$  (4).

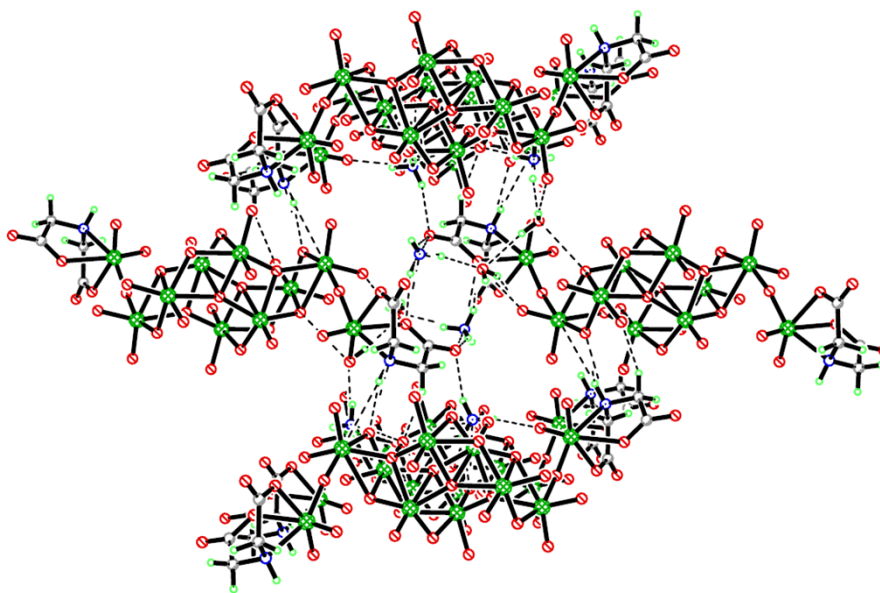


Figure S4. Perspective view of the 2D structure of  $(\text{NH}_4)_8\{\gamma\text{-Mo}_8\text{O}_{26}[\text{MoO}_3(\text{ida})]_2\} \cdot 6\text{H}_2\text{O}$  (1).

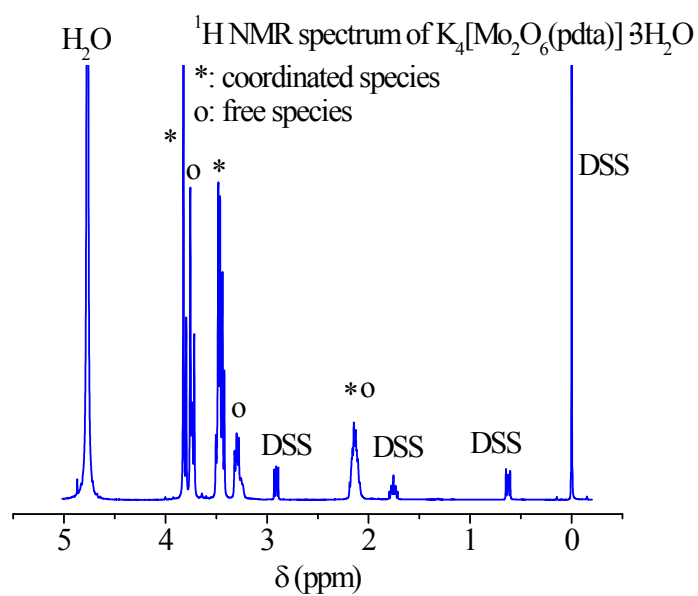
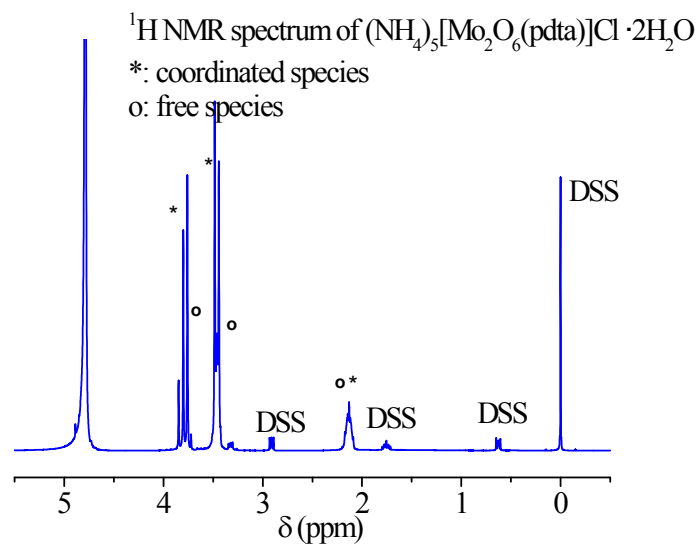


Figure S5.  $^1\text{H}$  NMR spectra of the complexes  $(\text{NH}_4)_5[\text{Mo}_2\text{O}_6(1,3\text{-pdta})]\text{Cl} \cdot 2\text{H}_2\text{O}$  (**5**) and  $\text{K}_4[\text{Mo}_2\text{O}_6(1,3\text{-pdta})] \cdot 3\text{H}_2\text{O}$  (**6**); x indicates resonance signals of the coordinated ligand, o for the free ligand.

Table S1. Bond valence calculations of molybdenum complexes.<sup>18</sup>

| Compounds                                                                                                                                                                                              | Mo1/Mo6 | Mo2/Mo7 | Mo3/Mo8 | Mo4/Mo9 | Mo5/Mo10 |
|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|---------|---------|---------|---------|----------|
| (NH <sub>4</sub> ) <sub>8</sub> { $\gamma$ -Mo <sub>8</sub> O <sub>26</sub> [MoO <sub>3</sub> (ida)] <sub>2</sub> }·6H <sub>2</sub> O ( <b>1</b> )                                                     | 6.142   | 6.018   | 5.924   | 5.865   | 5.933    |
| K <sub>8</sub> { $\gamma$ -Mo <sub>8</sub> O <sub>26</sub> [MoO <sub>2</sub> (ida)] <sub>2</sub> }·12H <sub>2</sub> O ( <b>2</b> )                                                                     | 5.270   | 5.205   | 5.996   | 6.024   | 6.052    |
| (NH <sub>4</sub> ) <sub>8n</sub> { $\gamma$ -Mo <sub>8</sub> O <sub>26</sub> [Mo <sub>2</sub> O <sub>6</sub> (1,3-pdta)] <sub>j</sub> } <sub>n</sub> ·30nH <sub>2</sub> O ( <b>3</b> ) <sup>[13]</sup> | 6.015   | 6.112   | 6.061   | 6.091   | 6.117    |
|                                                                                                                                                                                                        | 6.119   | 6.080   | 6.034   | 6.083   | 6.070    |
| (NH <sub>4</sub> ) <sub>4</sub> [MoO <sub>3</sub> (ida)] <sub>2</sub> ·3H <sub>2</sub> O( <b>4</b> )                                                                                                   | 6.117   | 6.183   |         |         |          |
| (NH <sub>4</sub> ) <sub>5</sub> [Mo <sub>2</sub> O <sub>6</sub> (1,3-pdta)]Cl·2H <sub>2</sub> O( <b>5</b> )                                                                                            | 6.072   | 6.062   |         |         |          |
| K <sub>4</sub> [Mo <sub>2</sub> O <sub>6</sub> (1,3-pdta)]·3H <sub>2</sub> O( <b>6</b> )                                                                                                               | 6.087   | 6.102   |         |         |          |
| (NH <sub>4</sub> ) <sub>6n</sub> (Mo <sub>8</sub> O <sub>27</sub> ) <sub>n</sub> ·4nH <sub>2</sub> O( <b>7</b> )                                                                                       | 5.955   | 6.084   | 6.022   | 6.095   |          |
| (NH <sub>4</sub> ) <sub>4n</sub> (Mo <sub>8</sub> O <sub>26</sub> ) <sub>n</sub> ( <b>8</b> )                                                                                                          | 6.033   | 5.982   | 6.004   | 5.922   |          |

Table S2. Comparisons of the Mo–O and Mo–N distances (Å) in iminodiacetato molybdates.

| Compounds                                                                                                                                          | Mo–O <sub>t</sub> | Mo–O <sub>b</sub> | Mo–O <sub>c1</sub> | Mo–O <sub>c2</sub> | Mo–N      | Ref       |
|----------------------------------------------------------------------------------------------------------------------------------------------------|-------------------|-------------------|--------------------|--------------------|-----------|-----------|
| (NH <sub>4</sub> ) <sub>4</sub> [MoO <sub>3</sub> (ida)] <sub>2</sub> ·3H <sub>2</sub> O( <b>4</b> )                                               | 1.736(2)          |                   | 2.202(2)           | 2.226(2)           | 2.306(2)  |           |
| (NH <sub>4</sub> ) <sub>8</sub> { $\gamma$ -Mo <sub>8</sub> O <sub>26</sub> [MoO <sub>3</sub> (ida)] <sub>2</sub> }·6H <sub>2</sub> O ( <b>1</b> ) | 1.705(5)          | 2.089(2)          | 2.211(5)           | 2.212(4)           | 2.285(5)  | This work |
| K <sub>8</sub> { $\gamma$ -Mo <sub>8</sub> O <sub>24</sub> [MoO <sub>3</sub> (ida)] <sub>2</sub> }·12H <sub>2</sub> O ( <b>2</b> )                 | 1.713(2)          | 2.067(2)          | 2.123(2)           | 2.105(2)           | 2.342(2)  |           |
| Na <sub>2</sub> [MoO <sub>3</sub> (ida)]·2H <sub>2</sub> O                                                                                         | 1.746(2)          |                   | 2.228(2)           | 2.228(2)           | 2.306(4)  | [12]      |
| C <sub>4</sub> H <sub>12</sub> N <sub>2</sub> [UO <sub>2</sub> (ida) <sub>2</sub> (OH) <sub>2</sub> ]·8H <sub>2</sub> O                            | 1.781(8)          |                   | 2.348(8)           | 2.341(9)           | 2.585(10) | [a]       |
| Pd(Hida) <sub>2</sub>                                                                                                                              |                   |                   | 1.988(4)           |                    | 2.061(4)  | [b]       |
| Pt(Hida) <sub>2</sub>                                                                                                                              |                   |                   | 2.014(9)           |                    | 2.066(10) | [b]       |
| (pyH)[Fe(ida) <sub>2</sub> ]                                                                                                                       |                   |                   | 1.980(2)           | 1.984(2)           | 2.161(3)  | [c]       |

- a) M. A. Walters, V. Vapnyar, A. Bolour, C. Incarvito, A. L. Rheingold, *Polyhedron* 2003, **22**, 941–946.  
b) A. J. Chakraborty, H. Mayer-Figge, W. S. Sheldrick, P. Banerjee, *Polyhedron*, 2005, **24**, 771–776.  
c) J. Jiang, M. J. Sarsfield, J. C. Renshaw, F. R Livens, D. Collison, J. M. Charnock, M. Helliwell, H. Eccles, *Inorg. Chem.* 2002, **41**, 2799–2806.

Table S3. Comparisons of the Mo–O and Mo–N distances (Å) for propanediaminetetraacetato molybdates.

| Compounds                                                                                                                                    | Mo–O <sub>t</sub> | Mo–O <sub>b</sub> | Mo–O <sub>c1</sub> | Mo–O <sub>c2</sub> | Mo–N     | Ref       |
|----------------------------------------------------------------------------------------------------------------------------------------------|-------------------|-------------------|--------------------|--------------------|----------|-----------|
| {Na <sub>8</sub> [Mo <sub>10</sub> O <sub>32</sub> (edta)](H <sub>2</sub> O) <sub>35</sub> } <sub>n</sub> <sup>[12]</sup>                    | 1.700(5)          | 1.812(4)          | 2.122(4)           | 2.173(5)           | 2.373(4) | [43]      |
| (NH <sub>4</sub> ) <sub>8n</sub> [Mo <sub>10</sub> O <sub>32</sub> (1,3-pdta)] <sub>n</sub> ·30nH <sub>2</sub> O( <b>3</b> ) <sup>[13]</sup> | 1.711(7)          | 1.825(7)          | 2.118(8)           | 2.231(7)           | 2.342(8) | [43]      |
| (NH <sub>4</sub> ) <sub>5</sub> [Mo <sub>2</sub> O <sub>6</sub> (1,3-pdta)]Cl·2H <sub>2</sub> O( <b>5</b> )                                  | 1.739(4)          |                   | 2.198(4)           | 2.207(3)           | 2.347(3) |           |
| K <sub>4</sub> [Mo <sub>2</sub> O <sub>6</sub> (1,3-pdta)]·3H <sub>2</sub> O( <b>6</b> )                                                     | 1.735(4)          |                   | 2.194(4)           | 2.221(4)           | 2.347(5) | this work |
| Average (Å)                                                                                                                                  | 1.713             | 1.864             | 2.143              | 2.187              | 2.368    |           |