

Supporting Information

Synthesis and adsorption properties of highly monodisperse hollow microporous polystyrene nanospheres

Xinren Kang,^{a,b} Yeru Liang,^{*b} Luyi Chen,^{a,b} Weicong Mai,^b Zhiyong Lin,^{*a} Ruowen Fu,^b and Dingcai Wu^{*b}

^a College of Material Science and Engineering, Huaqiao University, Xiamen, 361021, P. R. China

^b Materials Science Institute, PCFM Laboratory, School of Chemistry and Chemical Engineering, Sun Yat-sen University, Guangzhou 510275, P. R. China

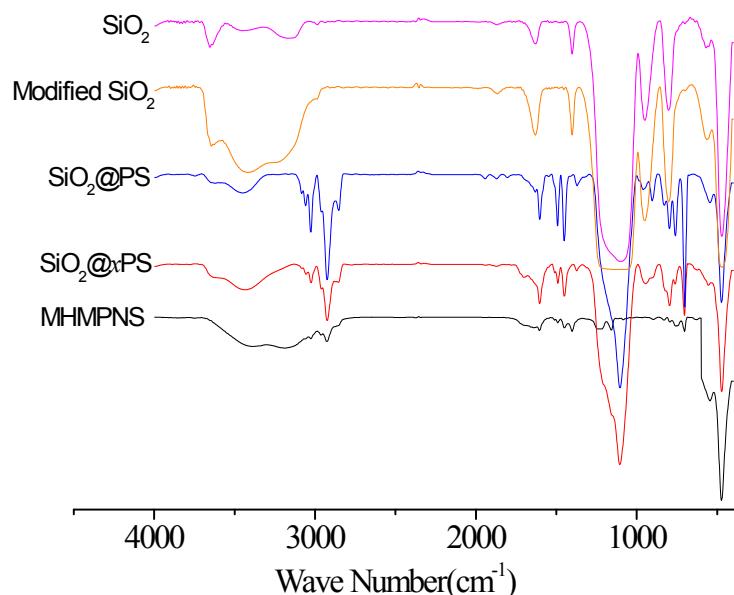


Fig. S1 FTIR spectra of SiO₂, modified SiO₂, SiO₂@PS, SiO₂@xPS and MHMPNS.

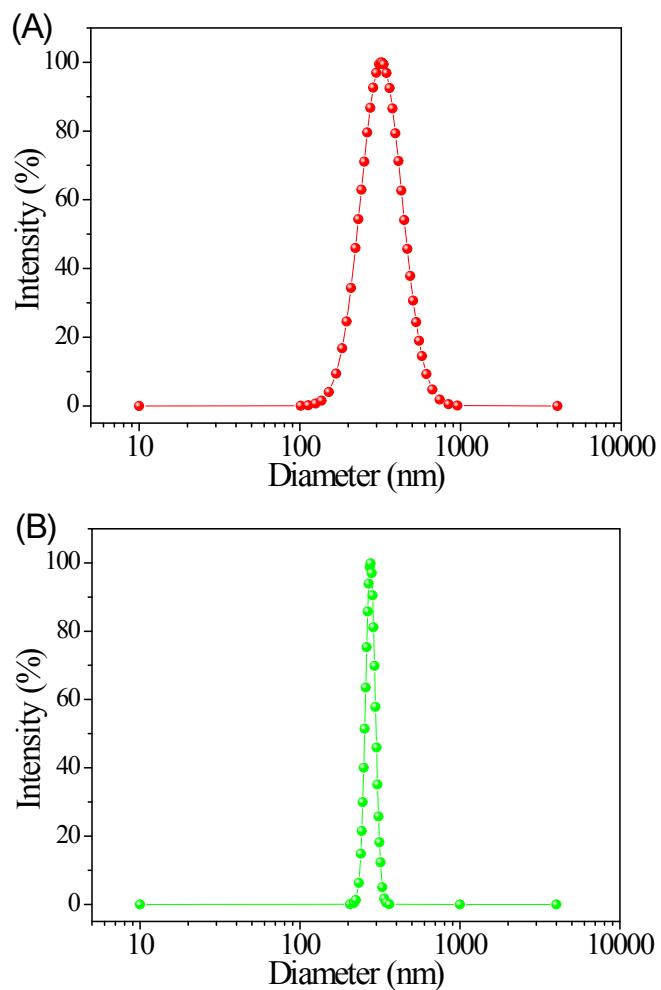
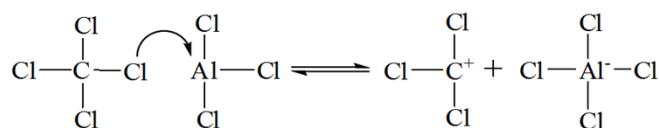
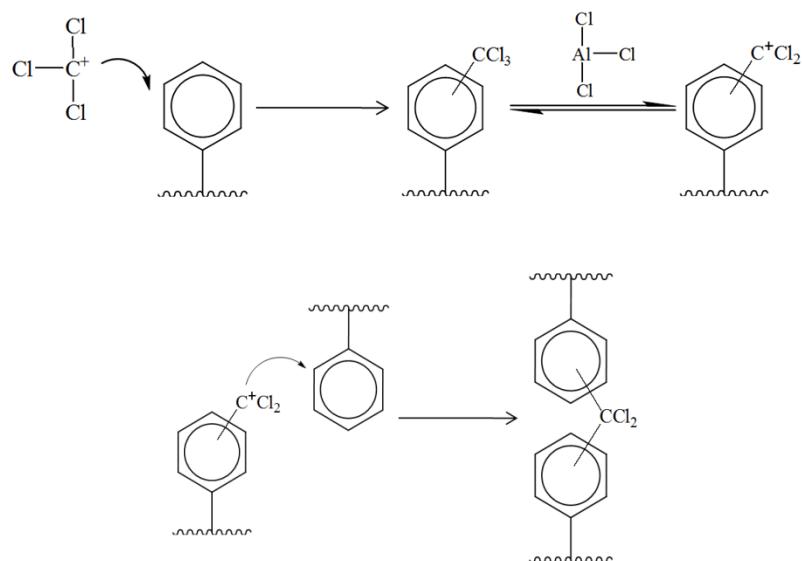


Fig. S2 DLS curves of (A) SiO₂@PS and (B) SiO₂@xPS.

(1) Formation of carbocation $^{+}\text{CCl}_3$



(2) Formation of - CCl_2 - crosslinking bridges



(3) Formation of -CO- crosslinking bridges

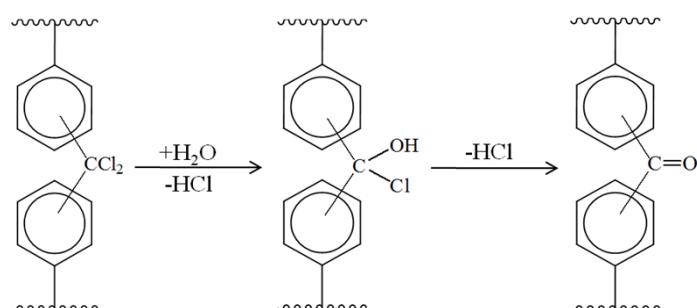


Fig. S3 Formation mechanism of -CO- crosslinking bridges based on the Friedel-Crafts reaction of polystyrene chain and carbon tetrachloride and subsequent hydrolysis.¹

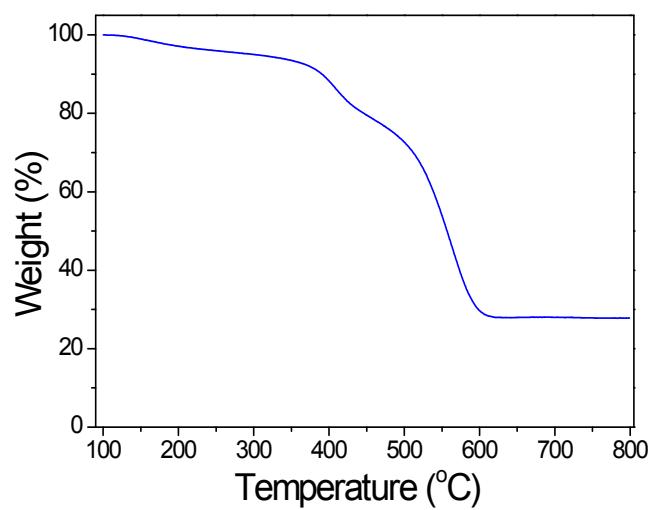


Fig. S4 TGA curve of $\text{SiO}_2@\text{xPS}$ conducted in air from 100 to 800 °C.

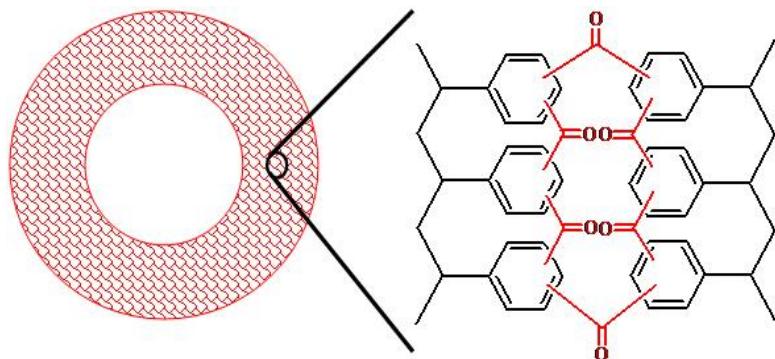


Fig. S5 Scheme of porous structure of MHMPNS.

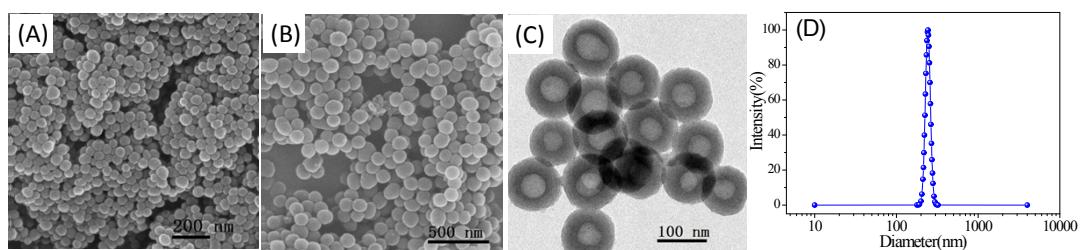


Fig. S6 (A) SEM image of modified SiO_2 nanosphere with particle size of 50 nm. (B) SEM image, (C) TEM image and (D) DLS curve of MHMPNS-2.

Table S1. Adsorption capacities towards organic vapors for SiO₂@xPS and MHMPNS obtained from static adsorption measurement.

Sample	Adsorption capacity (mg/g)			
	Tetrahydrofuran		Toluene	
SiO ₂ @xPS	84 ^a	117 ^b	121 ^a	168 ^b
MHMPNS	235	—	285	—

^a Calculated based on the total weight of SiO₂@xPS composite.

^b Calculated based on the xPS weight of SiO₂@xPS.

References

1. (a) Z. Li, D. Wu, Y. Liang, R. Fu and K. Matyjaszewski, *Journal of the American Chemical Society*, 2014, **136**, 4805; (b) D. Wu, A. Nese, J. Pietrasik, Y. Liang, H. He, M. Kruk, L. Huang, T. Kowalewski and K. Matyjaszewski, *ACS Nano*, 2012, **6**, 6208.