

Supporting Information

Investigation of the Reactions of U, U⁺ and U²⁺ with Ammonia: Mechanisms and Topological Analysis

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1. The ELF localization domains corresponding to the lowest-energy minima and transition states of U⁺⁽²⁺⁾ + NH₃ reaction are shown in Fig. S1 and S2.

Fig. S1. ELF localization domains ($\eta=0.70$) of the lowest energy minima and transition states corresponding to the U⁺ + NH₃ reaction pathway.

Fig. S2. ELF localization domains ($\eta=0.70$) of the lowest energy minima and transition states corresponding to the U²⁺ + NH₃ reaction pathway.

2. IRC energy and Wiberg bond order along the reaction coordinate s calculated at the PW91/SDD levels corresponding to the U⁺⁽²⁺⁾ + NH₃ are shown in Fig. S3 and S4.

Fig. S3. IRC energy and Wiberg bond order along the reaction coordinate s calculated at the PW91/SDD levels corresponding to the U⁺ + NH₃. Black solid curves are the energy and coloured curves are Wiberg bond order.

Fig. S4. IRC energy and Wiberg bond order along the reaction coordinate s calculated at the PW91/SDD levels corresponding to the U²⁺ + NH₃. Black solid curves are the energy and coloured curves are Wiberg bond order.

3. *Ab Initio* molecular dynamics simulation of the product H₂UNH.

Fig. S5. The RMSD at different temperatures for the product H₂UNH.

1. Fig. S1 and Fig. S2

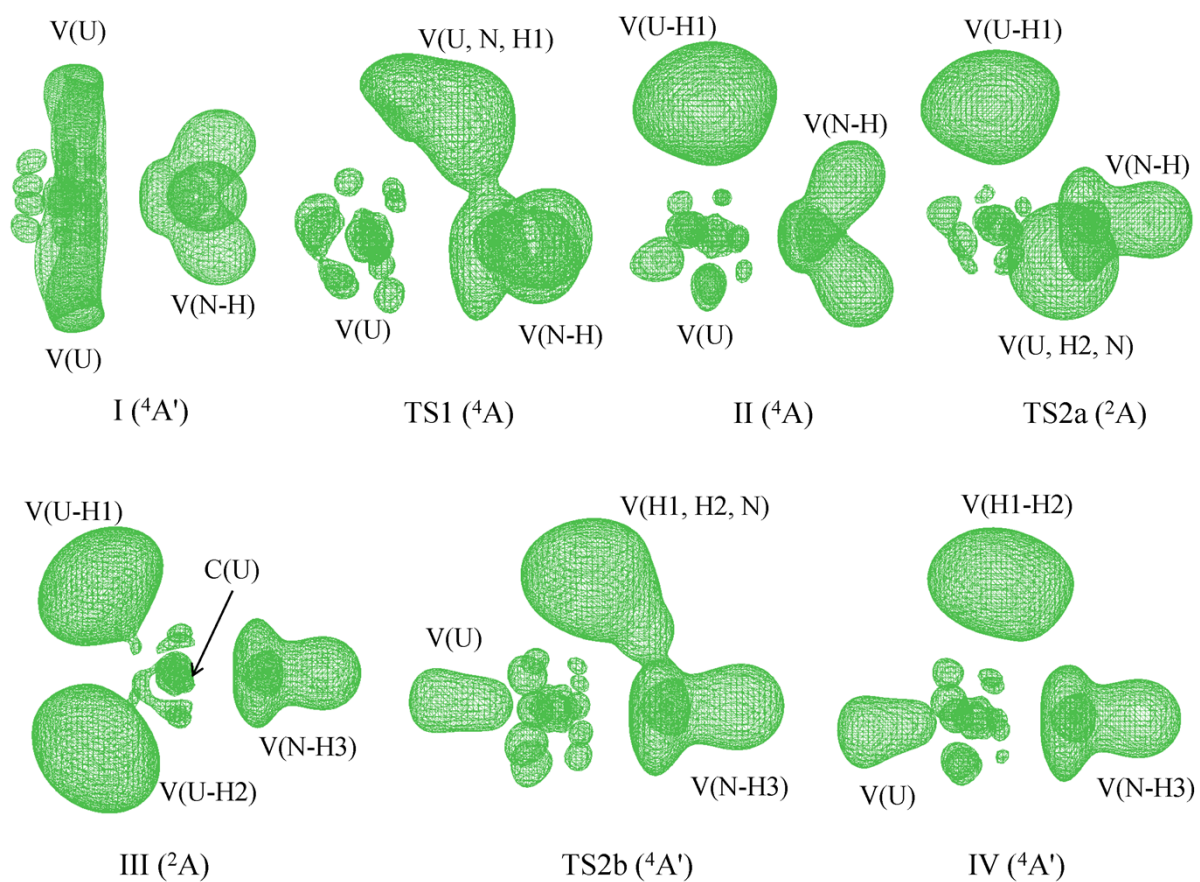


Fig. S1 ELF localization domains ($\eta=0.70$) of the lowest energy minima and transition states corresponding to the $U^+ + NH_3$ reaction pathway.

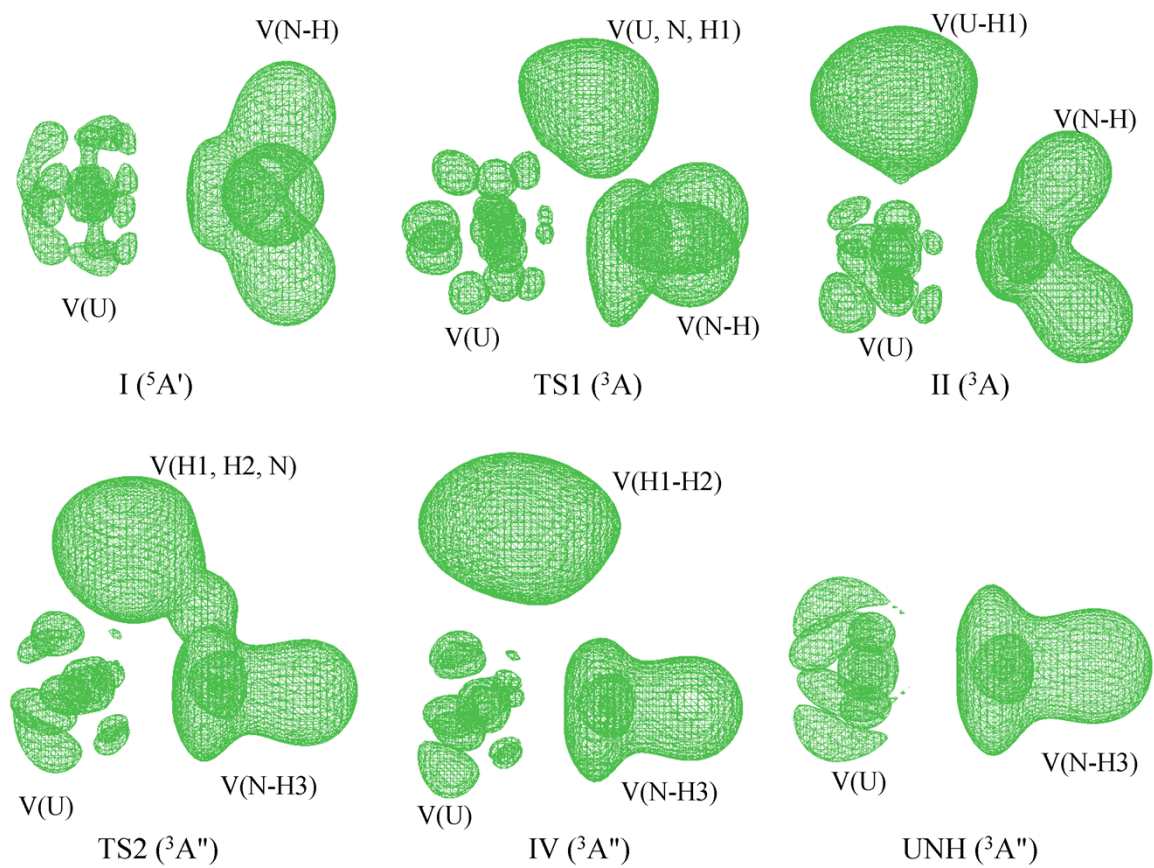


Fig. S2 ELF localization domains ($\eta=0.70$) of the lowest energy minima and transition states corresponding to the $U^{2+} + NH_3$ reaction pathway.

2. Fig. S3 and Fig. S4

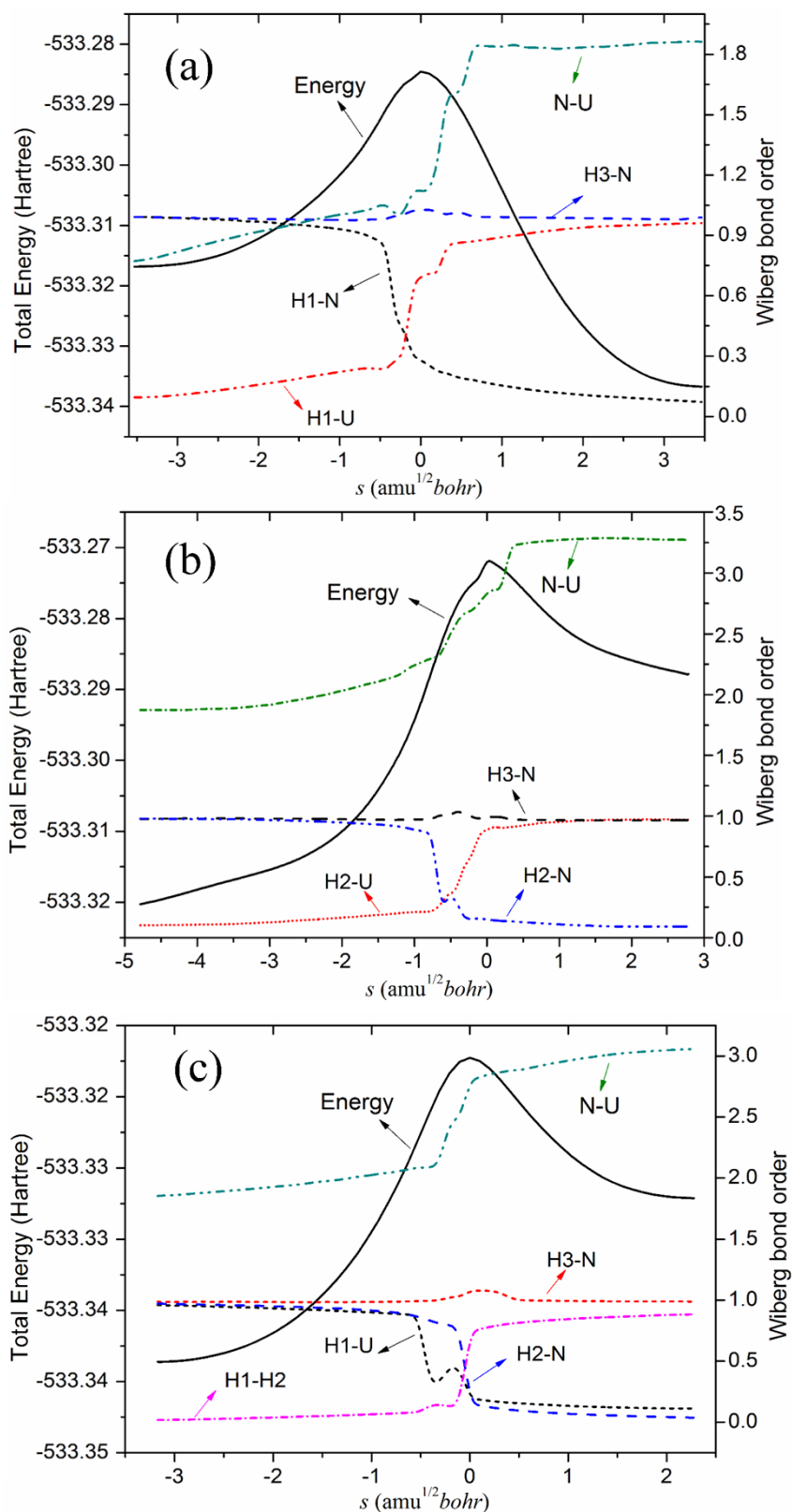


Fig. S3 IRC energy and Wiberg bond order along the reaction coordinate s calculated at the PW91/SDD levels corresponding to the $\text{U}^+ + \text{NH}_3$. Black solid curves are the energy and coloured curves are wiberg bond order. (a) $\text{U}^+ - \text{NH}_3 \rightarrow \text{TS1} \rightarrow \text{HUNH}_2^+$; (b) $\text{HUNH}_2^+ \rightarrow \text{TS2a} \rightarrow \text{HNUH}_2^+$; (c) $\text{HUNH}_2^+ \rightarrow \text{TS2b} \rightarrow \text{H}_2 - \text{UNH}^+$.

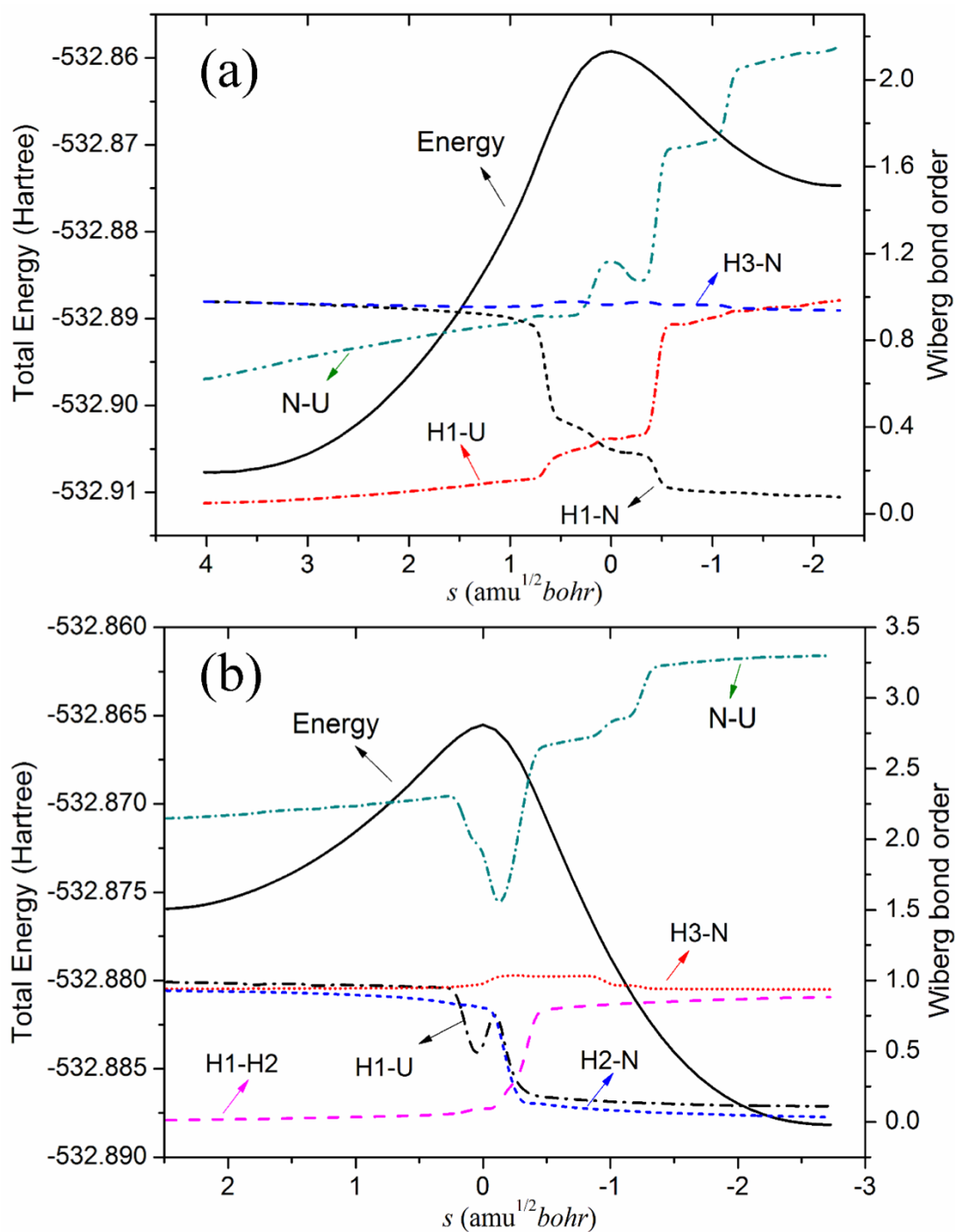


Fig. S4 IRC energy and Wiberg bond order along the reaction coordinate s calculated at the PW91/SDD levels corresponding to the $U^{2+} + NH_3$. Black solid curves are the energy and coloured curves are wiberg bond order. (a) $U^{2+}-NH_3 \rightarrow TS1 \rightarrow HUNH_2^{2+}$; (b) $HUNH_2^{2+} \rightarrow TS2 \rightarrow H_2-UNH_2^+$.

3. *Ab Initio* molecular dynamics simulation of the product H₂UNH.

We performed *ab Initio* molecular dynamics simulation to confirm the thermal stability of the product H₂UNH. Simulations are carried out for almost 5 ps using the BOMD method in Gaussian03. Our result indicates that the H₂UNH is stabilized. Using the dynamics simulations results, we plotted the root mean square displacement (RMSD) through time at different temperature (300 and 500K). The results are shown in the following figures. Our MD runs indicate that the H₂UNH structure keeps its identity at both room temperature (300 K) and high temperature (500 K).

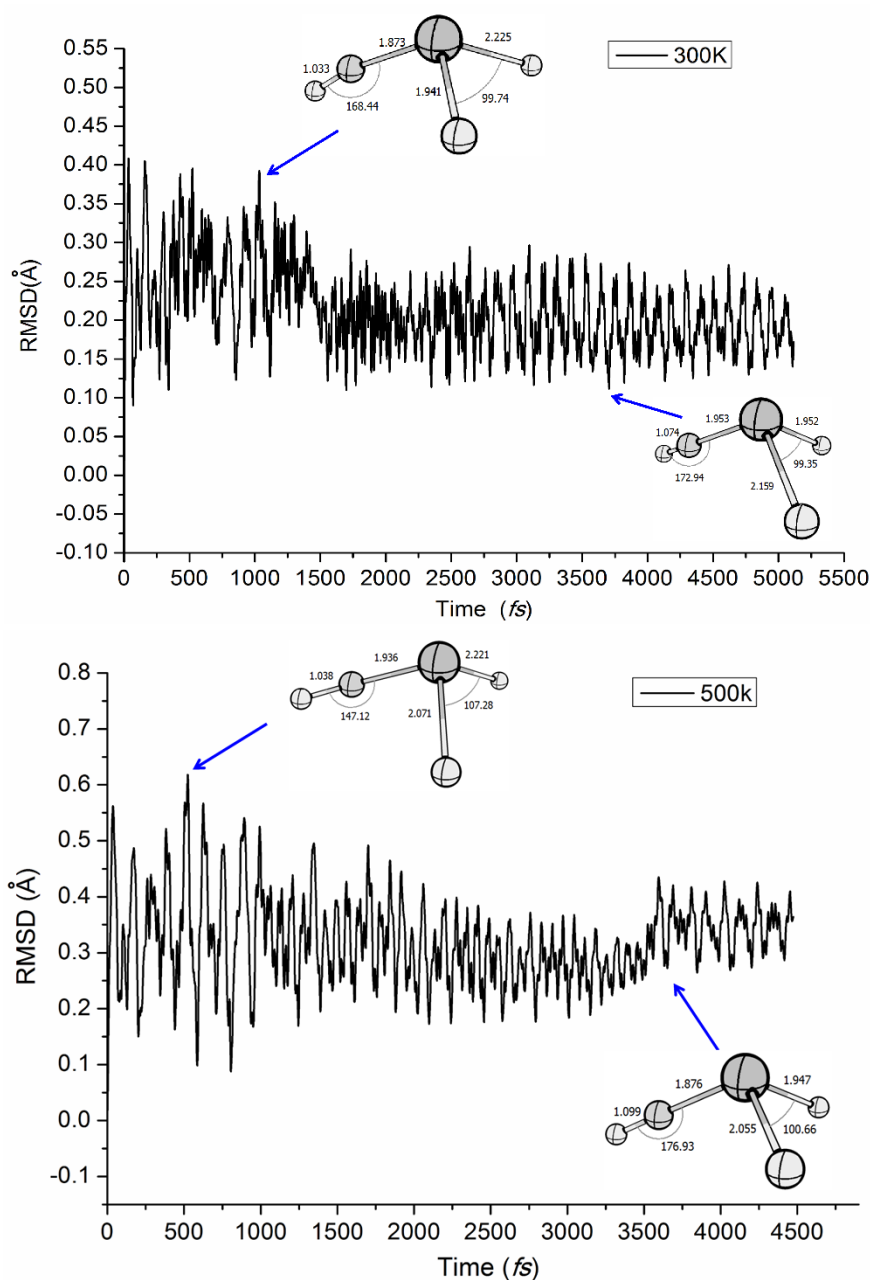


Fig. S5. The RMSD at different temperatures for the product H₂UNH.