

Supporting information

Synthesis, Characterization and Gas Permeation Properties of Anthracene Maleimide-Based Polymers of Intrinsic Microporosity

Muntazim Munir Khan ^a, Gisela Bengtson ^a, Silvio Neumann ^a, Md. Mushfequr Rahman ^a,
Volker Abetz ^{a,b}, Volkan Filiz ^{a*}

^a - Helmholtz-Zentrum Geesthacht, Institute of Polymer Research, Max-Planck-Str. 1, 21502
Geesthacht, Germany.

E-mail : volkan.filiz@hzg.de

^b - University of Hamburg, Institute of Physical Chemistry, Grindelallee 117, 20146
Hamburg, Germany.

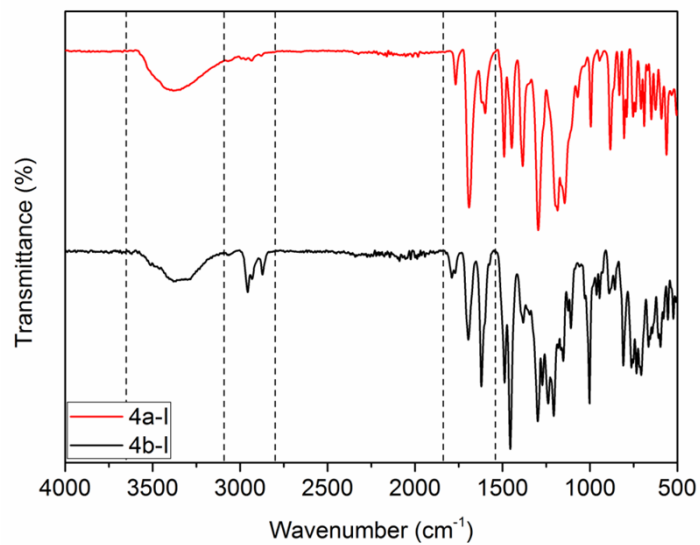
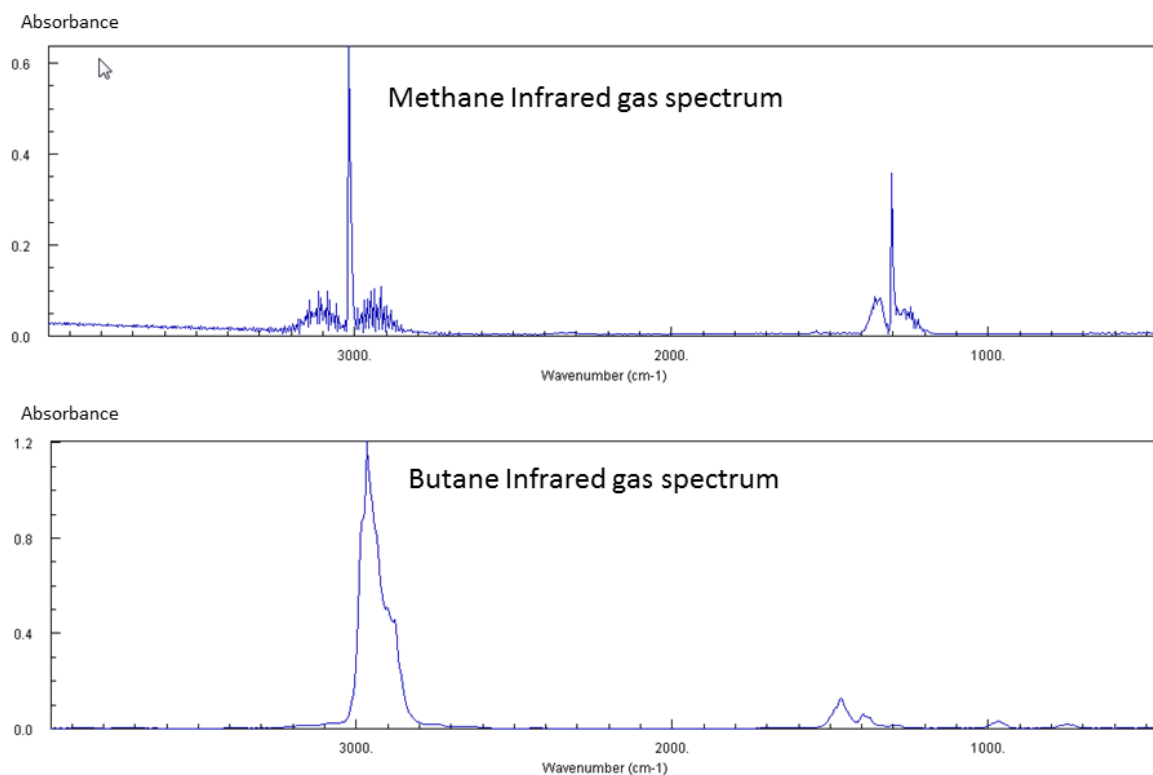


Figure S-1: Comparative FTIR spectra of 4a-I and 4b-I.



NIST Chemistry WebBook (<http://webbook.nist.gov/chemistry>)

Figure S-2: Reference gas spectra of ~~Methane~~-methane and ~~Butane~~-butane obtained from NIST chemistry WebBook.

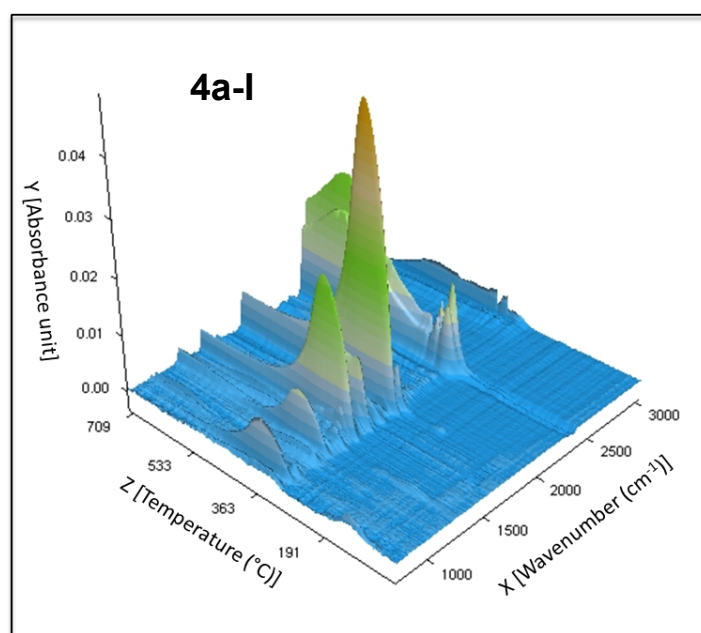


Figure S-3: 3D TGA-FTIR of 4a-I.

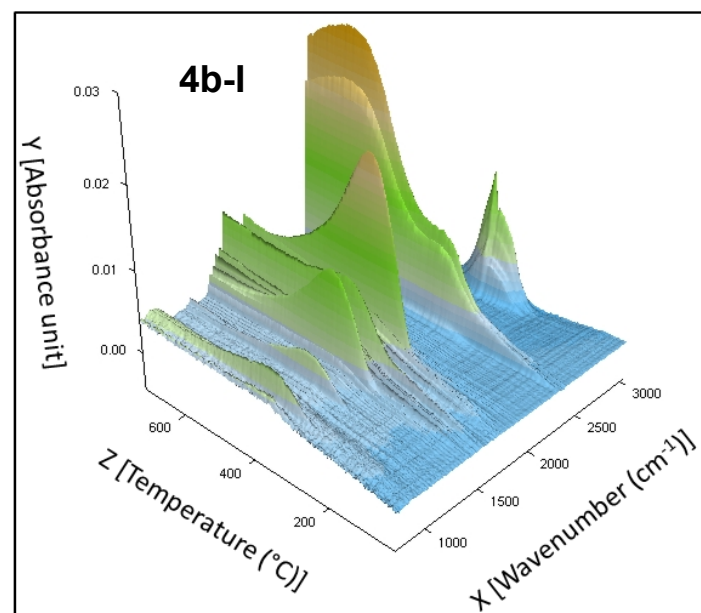


Figure S-4: 3D TGA-FTIR of 4b-I

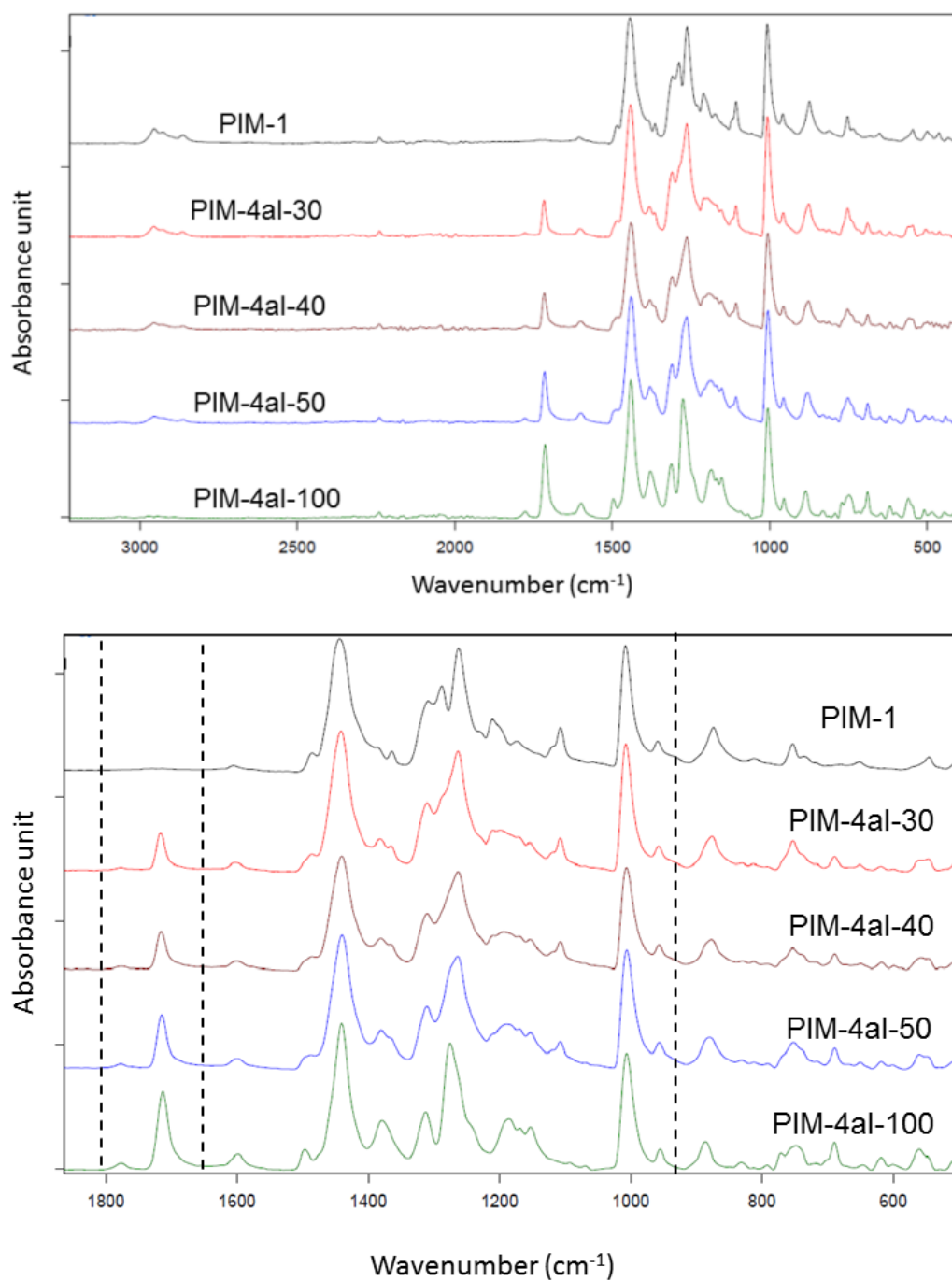


Figure S-5: Comparative FT-IR of copolymer derived from comonomer 4al.

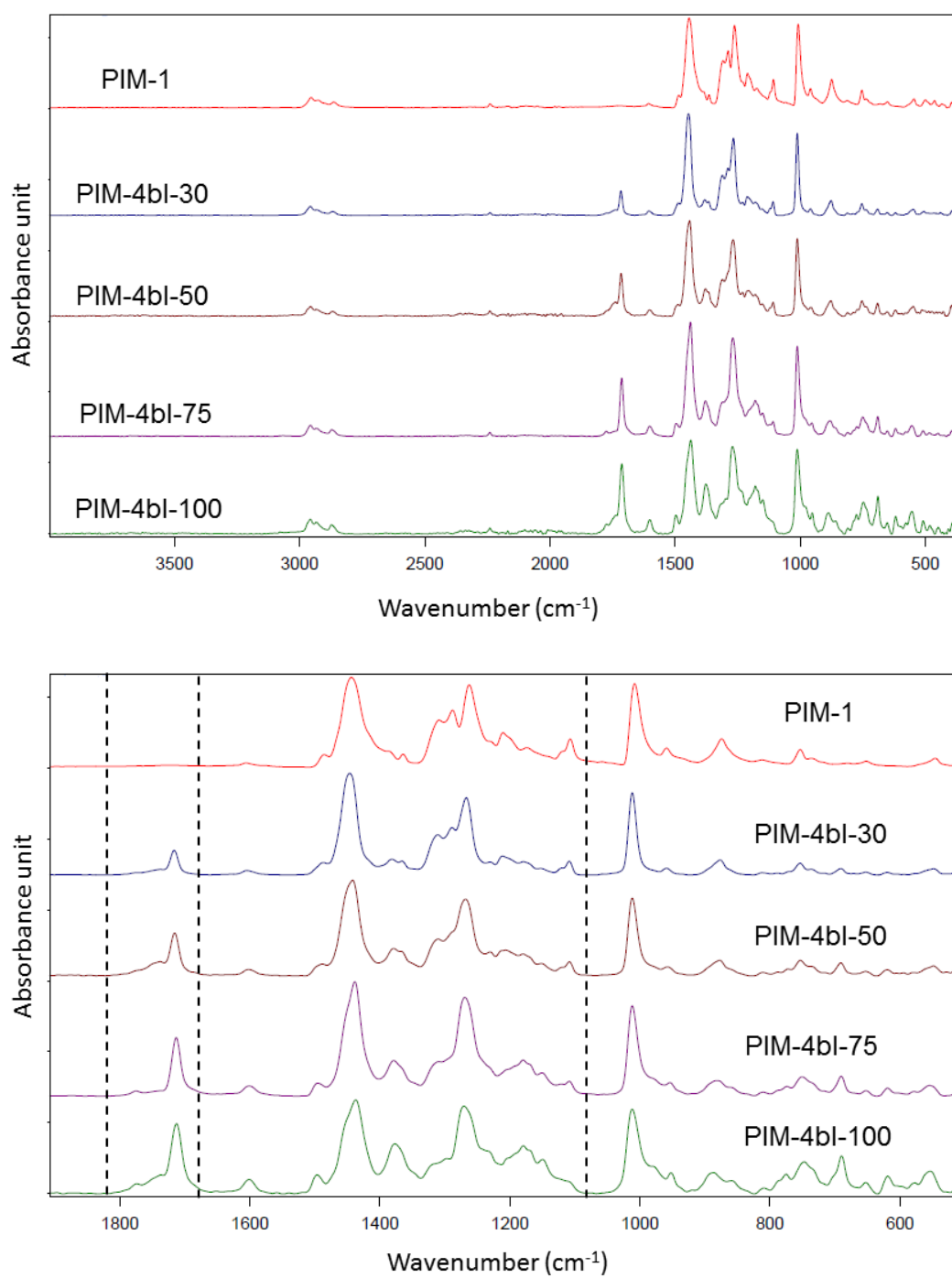


Figure S-6: Comparative FT-IR of copolymer derived from comonomer 4bI.

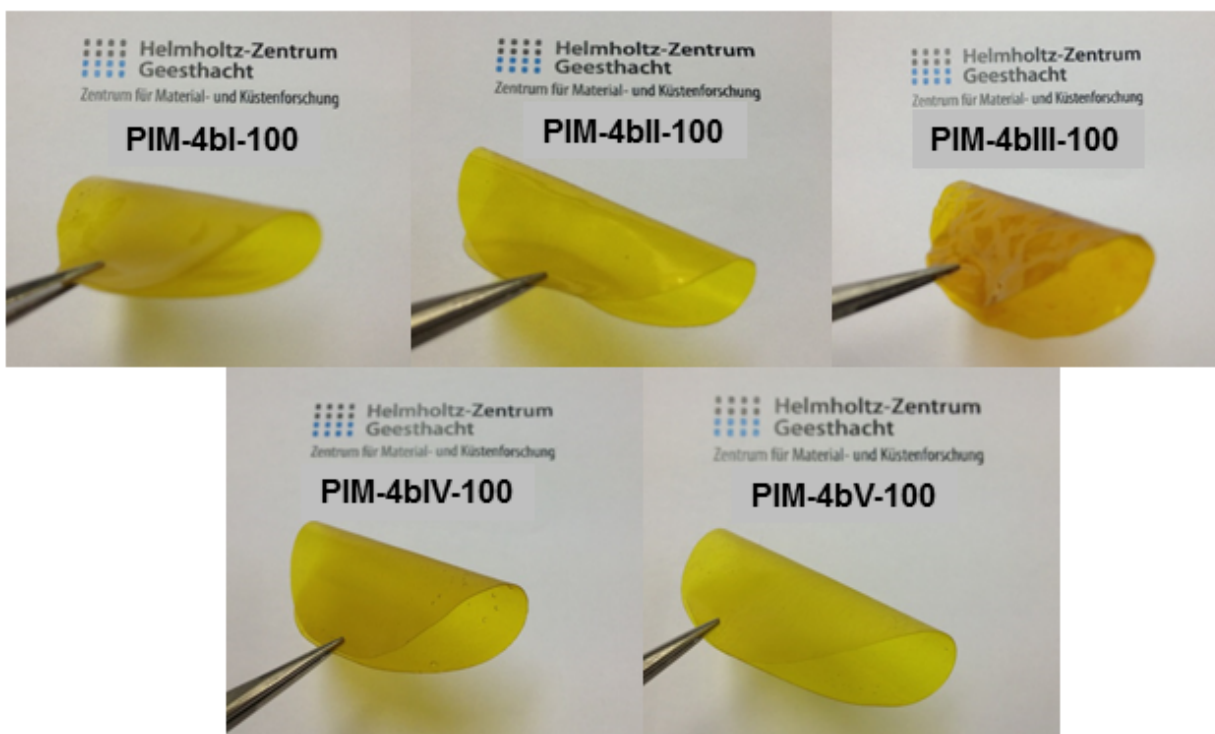


Figure S-7: Time lag films of homopolymers derived from comonomer 4b(I-V).

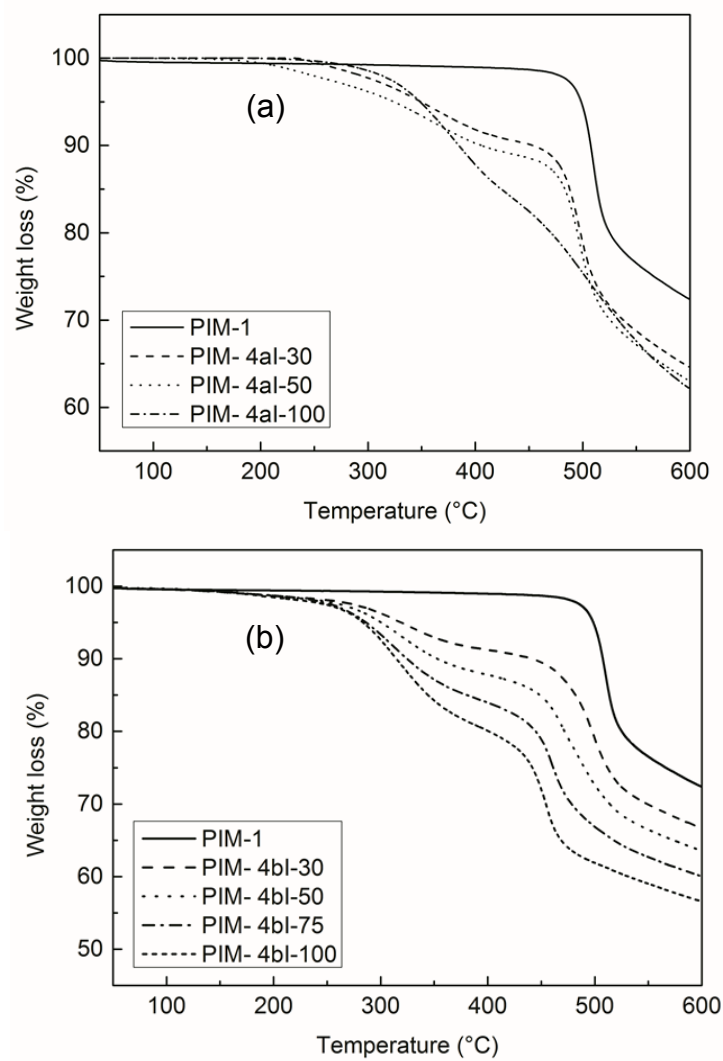
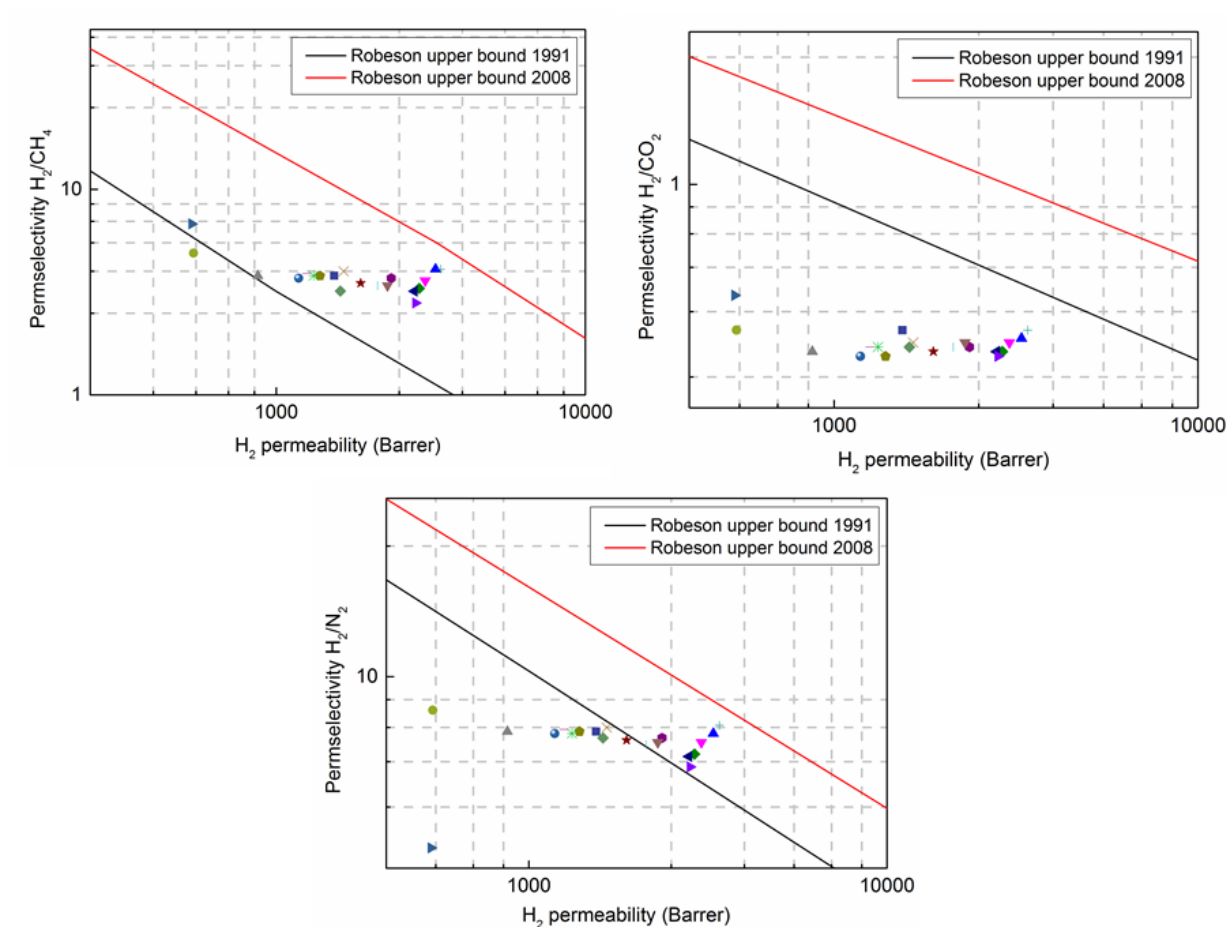


Figure S-8: Comparative TGA spectra of different copolymer composition derived from comonomer (a) 4a-I and (b) 4b-I.



▲ PIM-1^a, ● PIM-4bI-30, ★ PIM-4bI-50, ◆ PIM-4bI-75, ● PIM-4bI-100, + PIM-4bII-30, × PIM-4bII-50, ✱ PIM-4bII-75,
 — PIM-4bII-100, ▼ PIM-4bIII-30, ◆ PIM-4bIII-50, ◀ PIM-4bIII-75, ▶ PIM-4bIII-100, ▽ PIM-4bIV-30, ◆ PIM-4bIV-50, ◀ PIM-4bIV-75,
 ▶ PIM-4bIV-100, | PIM-4bV-30, ■ PIM-4bV-50, ● PIM-4bV-75 ▲ PIM-4bV-100, ● PIM-1^b ^a - this work ; ^b - Ref.

Figure S-9: Robeson plot for (a) H₂/CH₄ (b) H₂/CO₂ and (c) H₂/N₂ gas pairs showing data for methanol treated PIM-1 and CO-PIMs. The black and red lines represent the 1991 and 2008 upper bound, respectively.

Polymer code	Permeability (Barrer)				
	H ₂	N ₂	O ₂	CO ₂	CH ₄
PIM-4bI-100	1182	176	465	4010	316
PIM-4bI-75	1385	204	551	4707	365
PIM-4bI-50	1875	291	773	6308	530
PIM-4bI-30	2355	361	981	7608	641
PIM-4bII-100	1255	180	491	4093	322
PIM-4bII-75	1321	196	521	4233	346
PIM-4bII-50	1651	238	642	5089	401
PIM-4bII-30	3404	482	1328	9803	828
PIM-4bIII-100	2822	532	1272	9772	998
PIM-4bIII-75	2797	490	1233	9376	886
PIM-4bIII-50	2902	499	1279	9538	883
PIM-4bIII-30	3032	485	1313	9519	834
PIM-4bIV-100	533	51	158	1172	78
PIM-4bIV-75	1827	329	769	5882	622
PIM-4bIV-50	1612	245	758	5271	510
PIM-4bIV-30	2290	363	956	7155	669
PIM-4bV-100	872	127	337	2900	226
PIM-4bV-75	540	68	185	1524	110
PIM-4bV-50	1541	225	598	4689	406
PIM-4bV-30	2126	342	881	6806	634
PIM-1 ^a	3274	483	1396	9896	789
PIM-1 ^b	3300	500	1610	12600	740

^a This work. ^b Ref - [2]

Table S-1: Single gas permeability of PIM-1, homopolymer and copolymer derived from the monomer 4b(I-V).

Polymer code	Permselectivity					
	O ₂ /N ₂	H ₂ /N ₂	H ₂ /CH ₄	H ₂ /CO ₂	CO ₂ /N ₂	CO ₂ /CH ₄
PIM-4bI-100	2.6	6.7	3.7	0.29	22.8	12.7
PIM-4bI-75	3.1	6.8	3.8	0.29	22.2	12.6
PIM-4bI-50	2.6	6.4	3.5	0.30	21.7	11.9
PIM-4bI-30	2.7	6.5	3.7	0.31	21.1	11.9
PIM-4bII-100	2.7	6.9	3.9	0.31	22.7	12.7
PIM-4bII-75	2.6	6.7	3.8	0.31	21.6	12.2
PIM-4bII-50	2.7	7.0	4.0	0.32	21.4	12.7
PIM-4bII-30	2.7	7.1	4.1	0.35	20.3	11.8
PIM-4bIII-100	2.4	5.3	2.8	0.29	18.4	9.8
PIM-4bIII-75	2.5	5.7	3.2	0.3	19.1	10.6
PIM-4bIII-50	2.6	5.8	3.3	0.3	19.1	10.8
PIM-4bIII-30	2.7	6.3	3.6	0.32	19.6	11.4
PIM-4bIV-100	3.1	10.4	6.8	0.45	22.9	15.0
PIM-4bIV-75	2.3	5.6	2.9	0.31	17.9	9.4
PIM-4bIV-50	3.1	6.5	3.2	0.31	21.5	10.3
PIM-4bIV-30	2.6	6.3	3.4	0.32	19.7	10.7
PIM-4bV-100	2.6	6.8	3.8	0.3	22.8	12.8
PIM-4bV-75	2.7	7.9	4.9	0.35	22.3	13.9
PIM-4bV-50	2.6	6.8	3.8	0.35	20.8	11.5
PIM-4bV-30	2.6	6.2	3.4	0.31	19.9	10.7
PIM-1 ^a	2.9	6.7	4.1	0.33	20.5	12.5
PIM-1 ^b	2.5	6.6	4.4	0.26	18.4	9.6

^a This work. ^b Ref – [2]

Table S-2: Various gas pairs permselectivity of PIM-1, homopolymer and copolymer derived from the monomer 4b(I-V).