

## Supporting Information

### **Structural and electronic properties of Pd-decorated graphene oxides and their effects on the adsorption of nitrogen oxides: insights from density functional calculations**

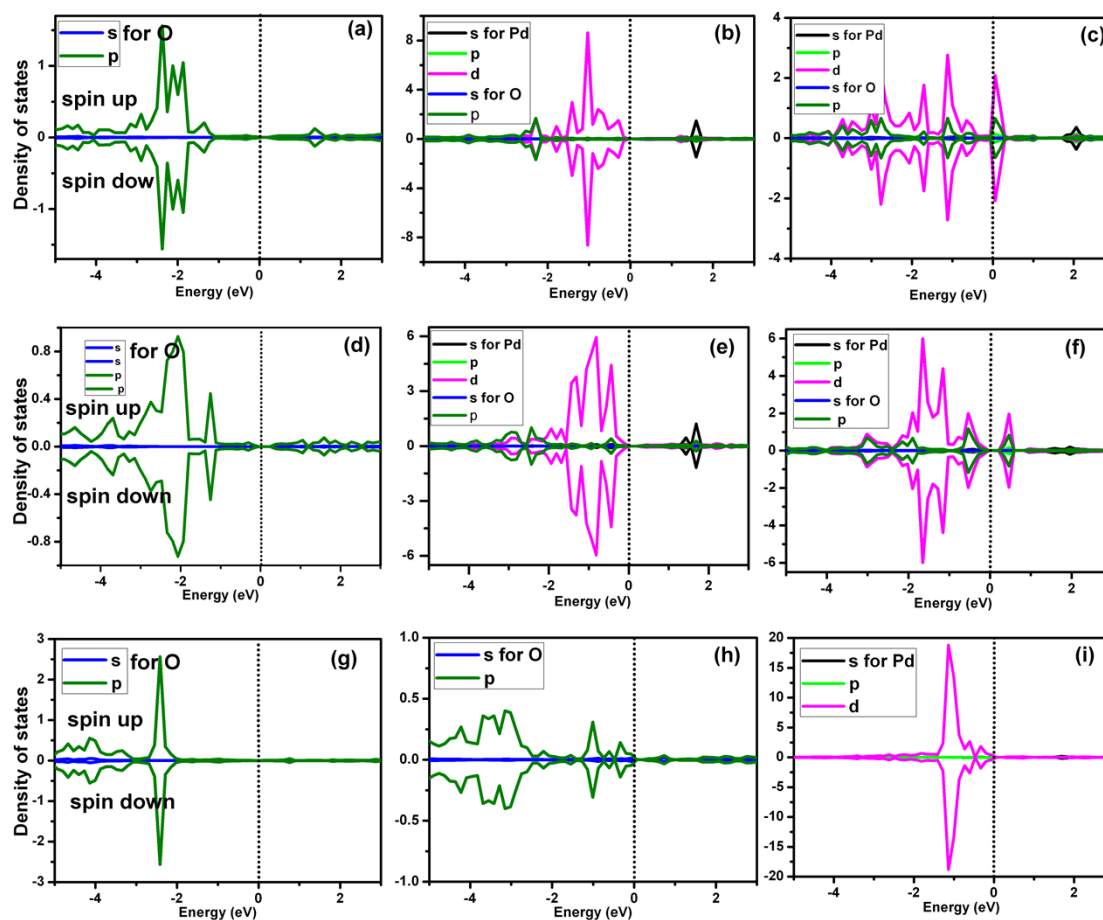
Shaobin Tang\* and Jiayi Zhu

*Key Laboratory of Organo-Pharmaceutical Chemistry of Jiangxi Province, Gannan Normal University, Ganzhou 341000, China*

\* E-mail address: [tsb1980@xmu.edu.cn](mailto:tsb1980@xmu.edu.cn)

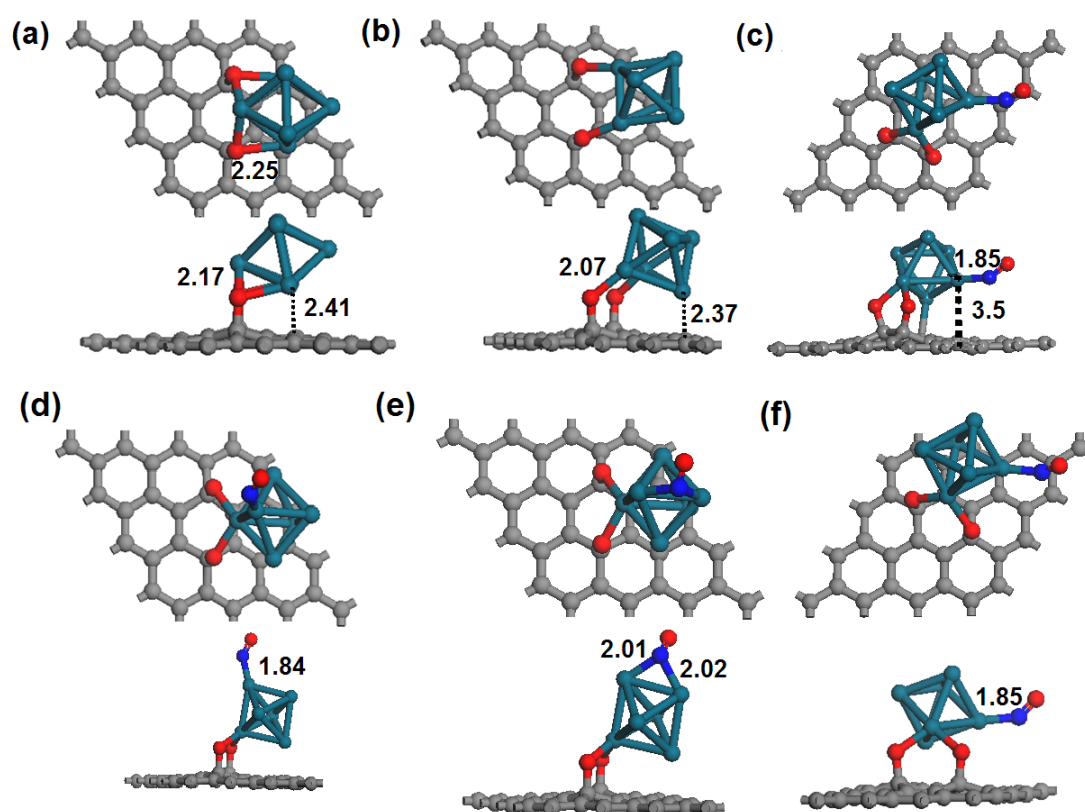
**Table S1: Summary of calculated results for the adsorption of Pd on rGOs with both hydroxyl and epoxy groups: the binding energy ( $E_b$ ), the distance between Pd and its neighboring C ( $d_1$ ), and O ( $d_2$ ) atoms.**

Structure	binding site	$E_b$ (eV)	$d_1$ (Å)	$d_2$ (Å)
Pd@G-O-OH-1a	C	1.45	2.12	2.42
Pd@G-O-OH-1b	C and O	2.25	2.13	2.06
Pd@G-O-OH-2a	C	1.06	2.25	
Pd@G-O-OH-2b	C and O	1.36	2.13	2.07
Pd@G-O-OH-2c	O	0.08	2	
Pd@G-2O-OH-1a	C	1.63	2.1	2.69
Pd@G-2O-OH-1b	O	2.25		2.04 (2.3)
Pd@G-2O-OH-1c	O	2.66		1.97
Pd@G-2O-OH-1d	O	2.58		1.96
Pd@G-2O-OH-2	C and O	1.78	2.08	2.05

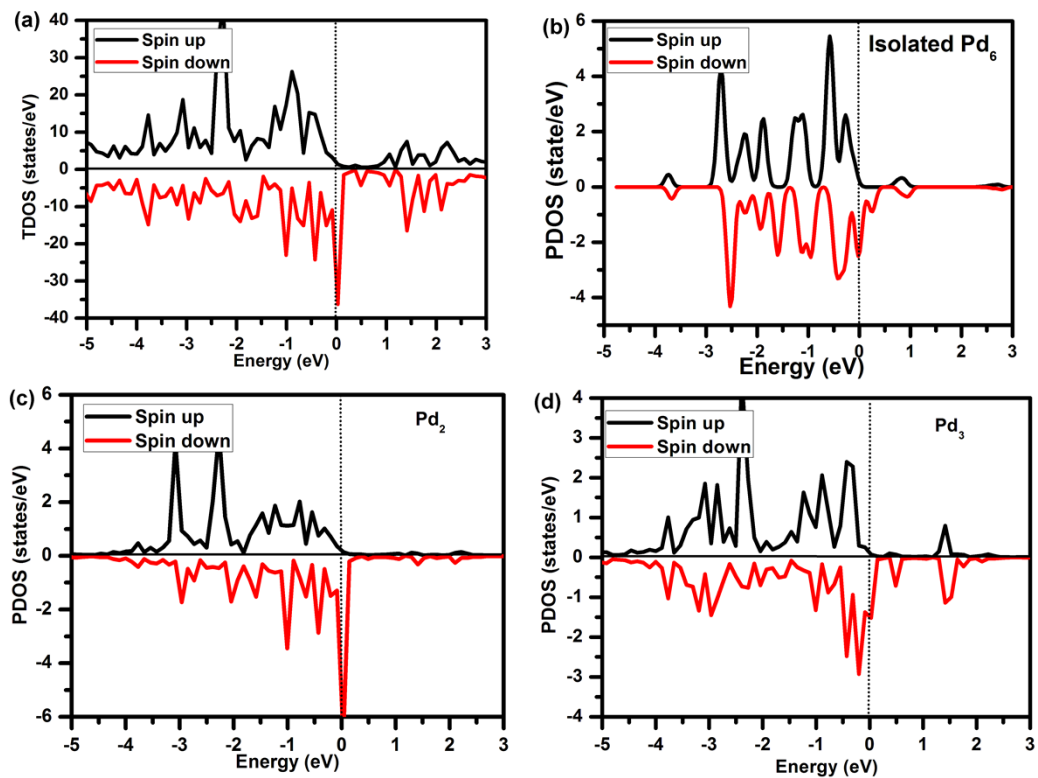


**Figure S1.** The spin-polarized Projected density of states (PDOS) of O atom from GO before and after Pd interaction and the PDOS of adsorbed Pd for Pd-decorated GOs. (a) G-2O-1, (b) Pd@G-2O-1a, (c) Pd@G-2O-1c, (d) G-2O-2, (e) Pd@G-2O-2a, (f)

Pd@G-2O-2b, (g) G-2OH-2, and (h)-(i) Pd@G-2OH-2. The Fermi level is set to 0.



**Figure S2.** Top and side views of optimized structure (distance in Å) for Pd<sub>6</sub> deposited on G-2O-1 (a)-(b) and NO adsorbed on Pd<sub>6</sub>-rGO hybrid (e)-(h). (a) Pd<sub>6</sub>@G-2O-1-c and (b) Pd<sub>6</sub>@G-2O-1-d. Adsorption of NO at (c) Pd<sub>5</sub> site of Pd<sub>6</sub>@G-2O-1-a and (d)-(f) various Pd sites of Pd<sub>6</sub>@G-2O-1-b.



**Figure S3.** The spin-polarized TDOS and PDOS of Pd atom for isolated Pd<sub>6</sub> cluster and Pd cluster adsorbed rGO. (a) Pd<sub>6</sub>@ G-2O-1-a and (b) one Pd atom, (c) Pd<sub>2</sub> and (d) Pd<sub>3</sub> of Pd<sub>6</sub>@ G-2O-1-a. The Fermi level is set to 0.