Supporting information for:

A Highly Selective PET-Based Chemosensor for Instant Detecting Zn^{2+}

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Figure S2. The Job's plot examined between Zn^{2+} and **L2**, indicating the 1: 1 stoichiometry, which was carried out by UV-vis spectra.

Figure S3. Fluorescence spectra response of **L2** (20 μM) upon addition of Zn^{2+} (20 equiv.) in DMSO/H_2O (8:2, v/v, containing 0.01 M HEPES, pH=7.24), (λ_{ex}=423 nm). Inset: Photograph of **L2** (20 μM) upon adding 20 equiv. of Zn^{2+}, which was taken under a UV-lamp (365 nm).

Figure S4. Plot of the intensity at 508 nm for a mixture of **L2** (20 μM) upon adding of an increasing concentration of Zn^{2+} in DMSO/H_2O (8:2, v/v, containing 0.01 M HEPES, pH=7.24),

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(λ_{ex}=423 nm).

Figure S5. ESI-MS spectra of **L2**.

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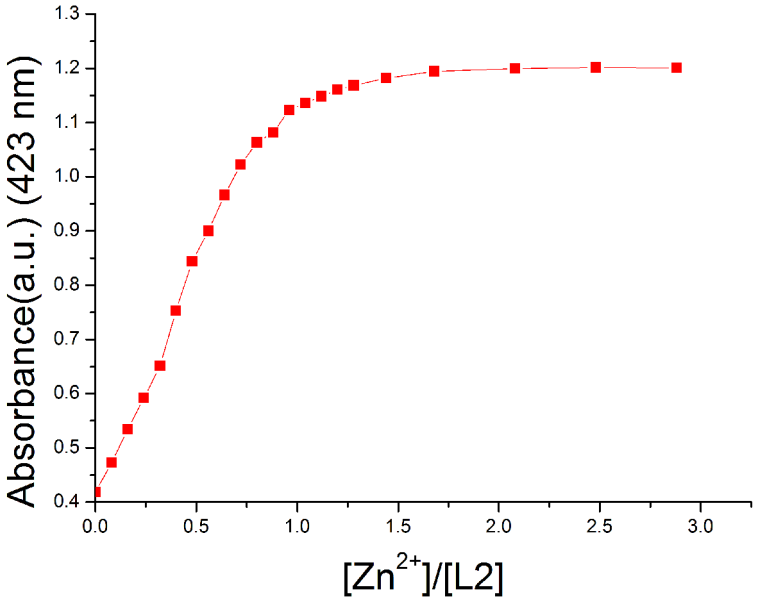
Figure S10. ^{1}H NMR spectra of **L2**.

Figure S11. ^{13}C NMR spectra of **L2**.

Figure S1



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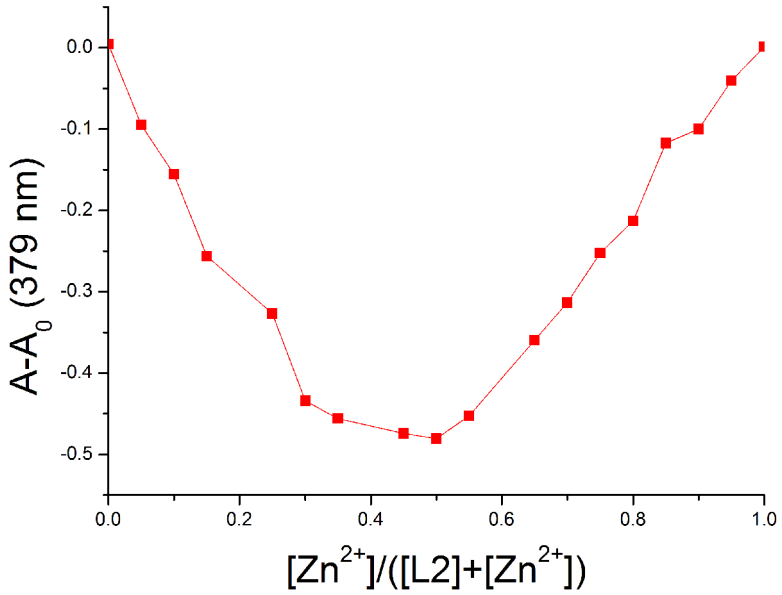

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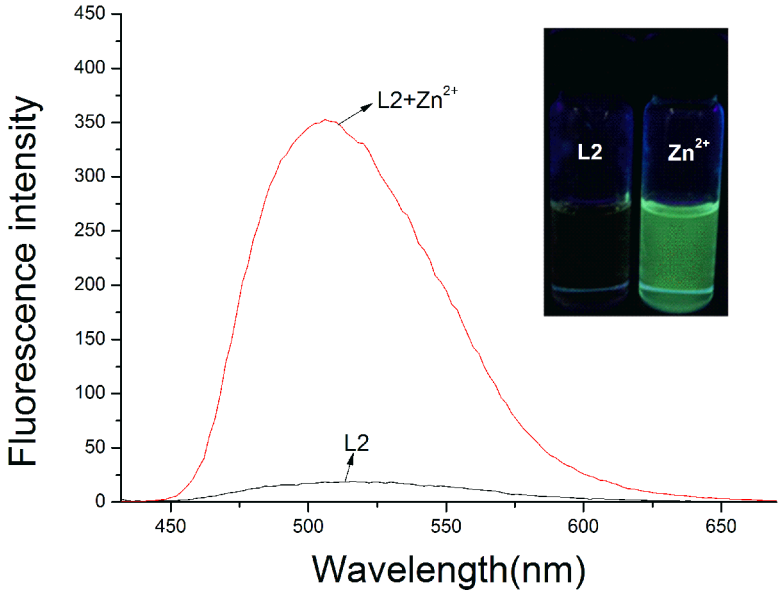

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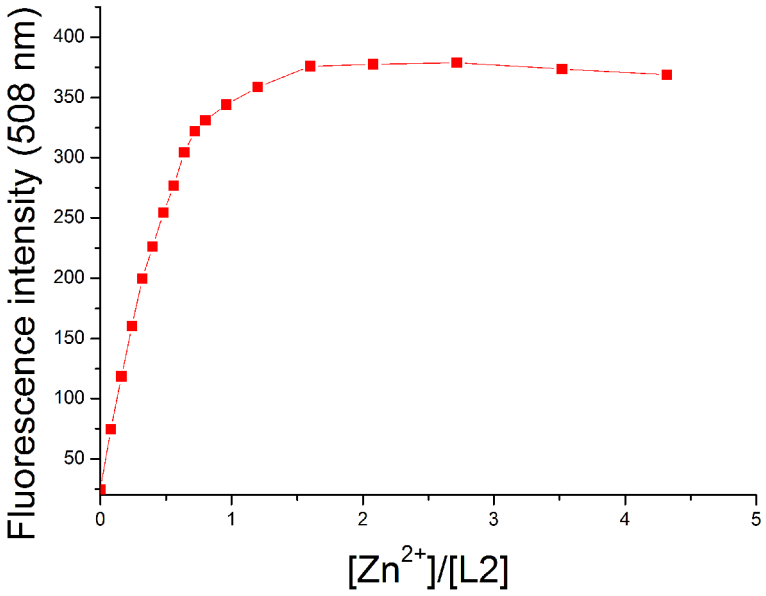

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Figure S5

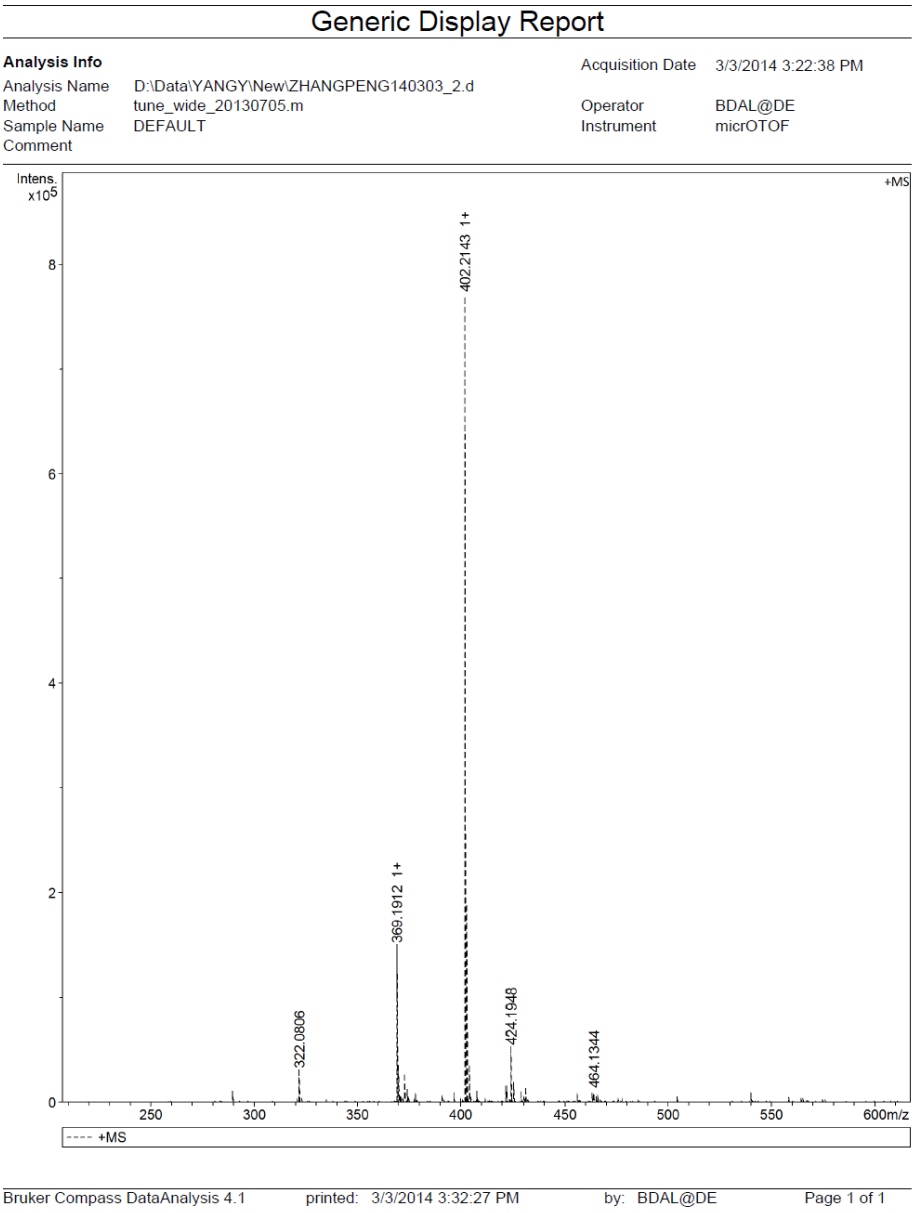

Figure S5. ESI-MS spectra of **L2**.

Figure S6

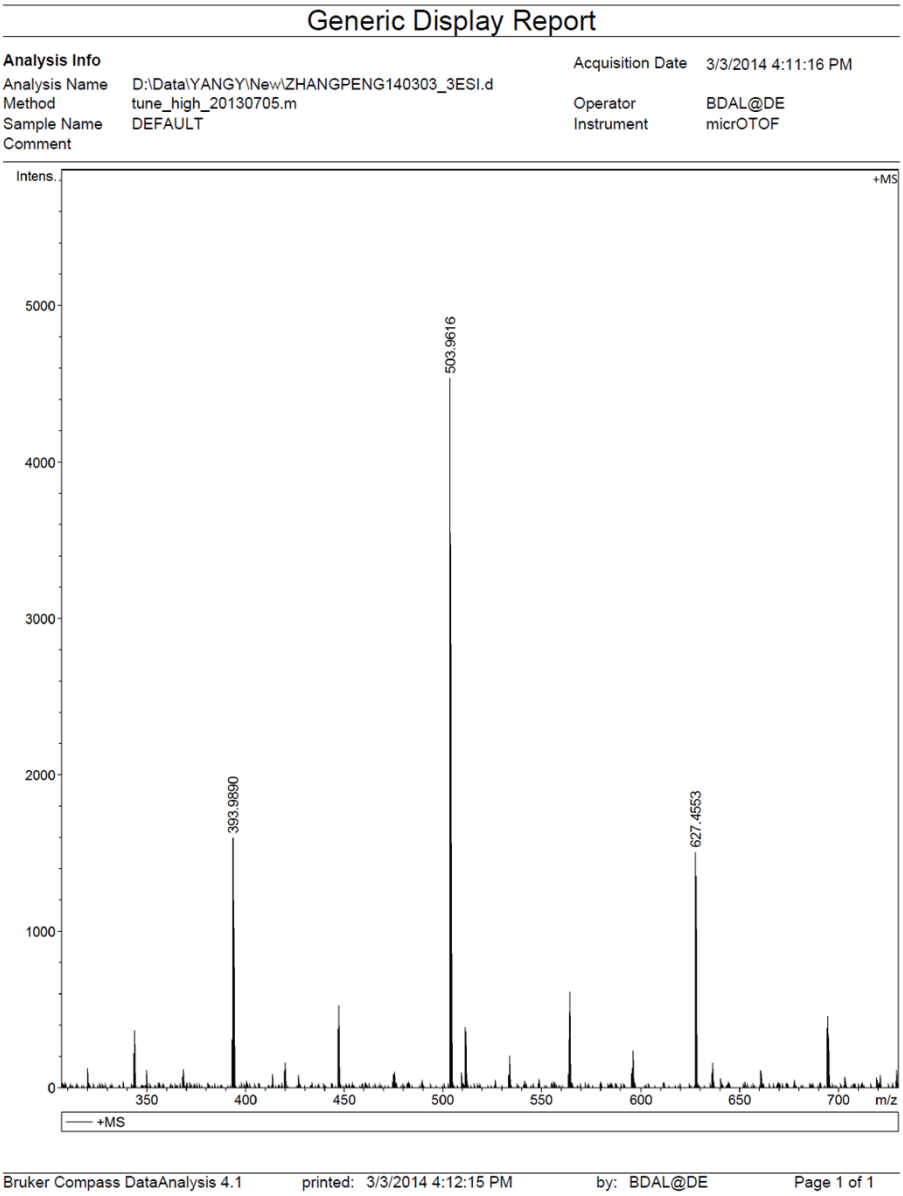

Figure S6. ESI-MS spectra of **L2**-Zn^{2+}.

Figure S7

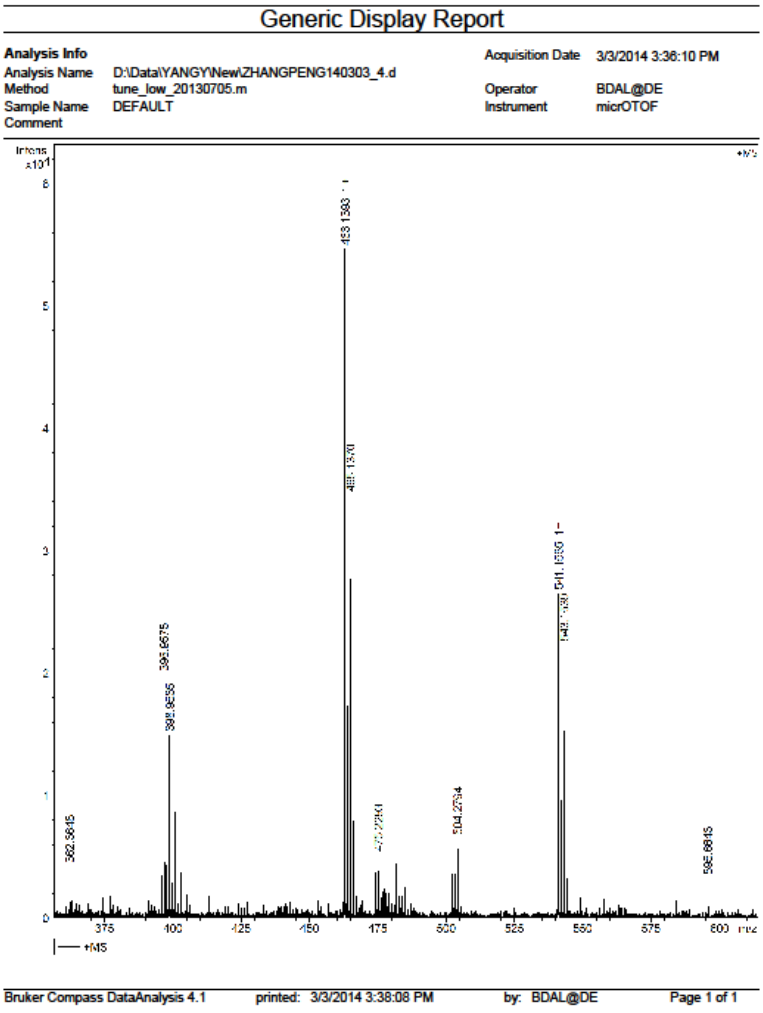

Figure S7. ESI-MS spectra of **L2**-Zn^{2+} upon addition of Cu^{2+}.

Figure S8

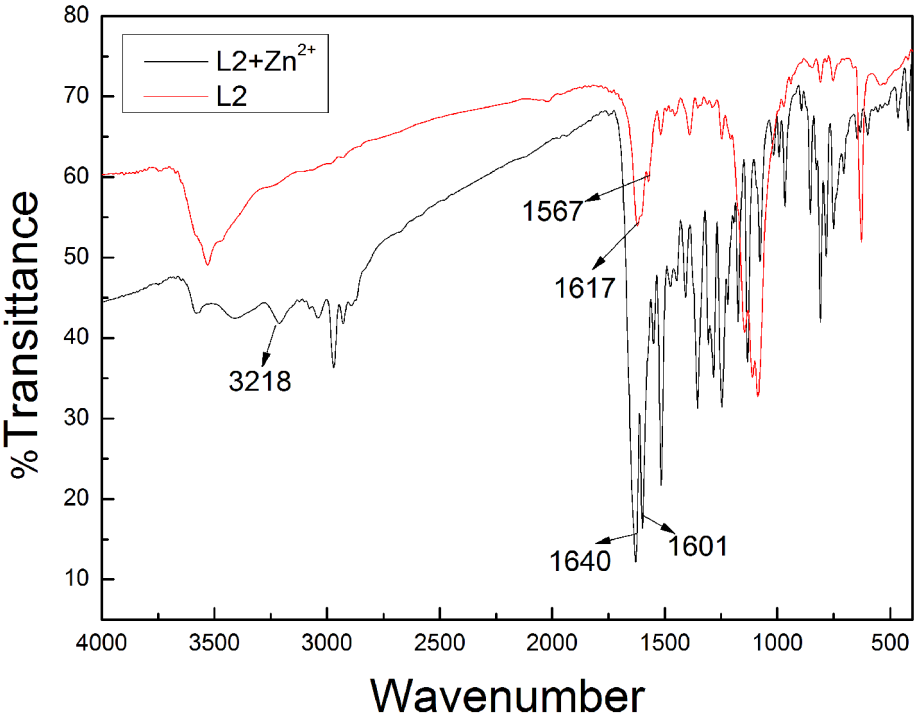

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Figure S9

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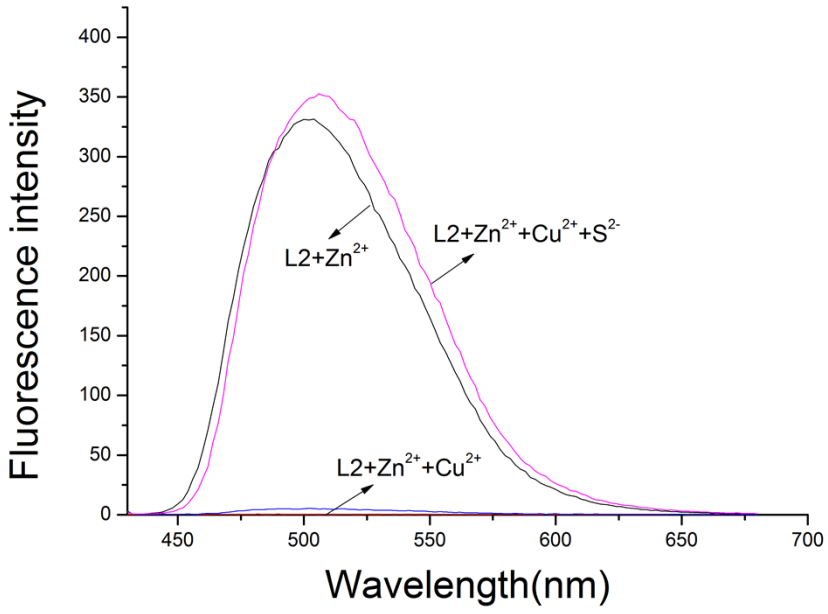

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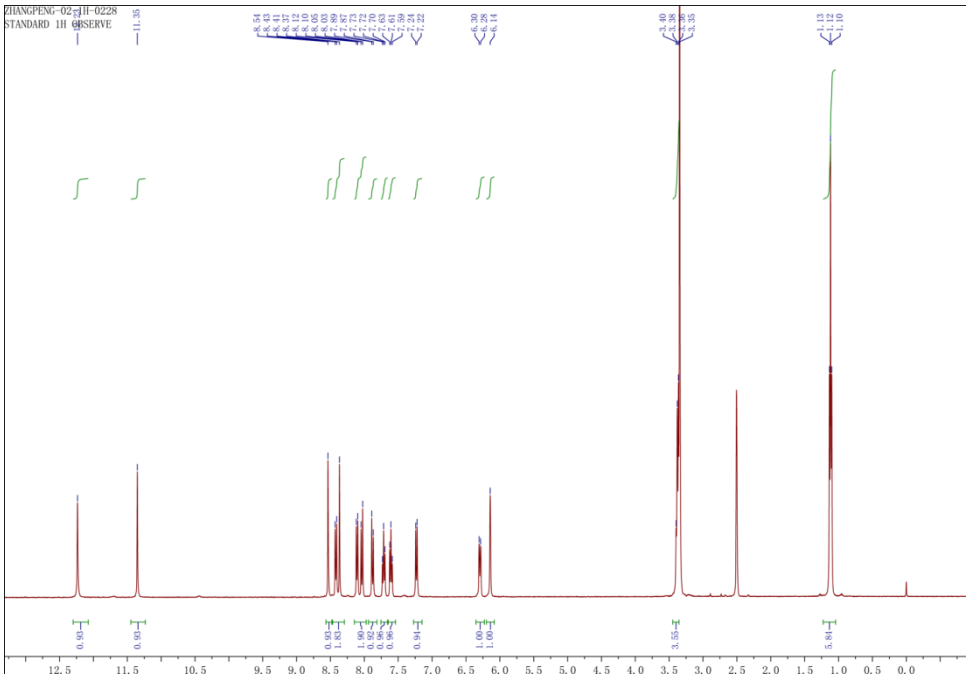
Figure S10



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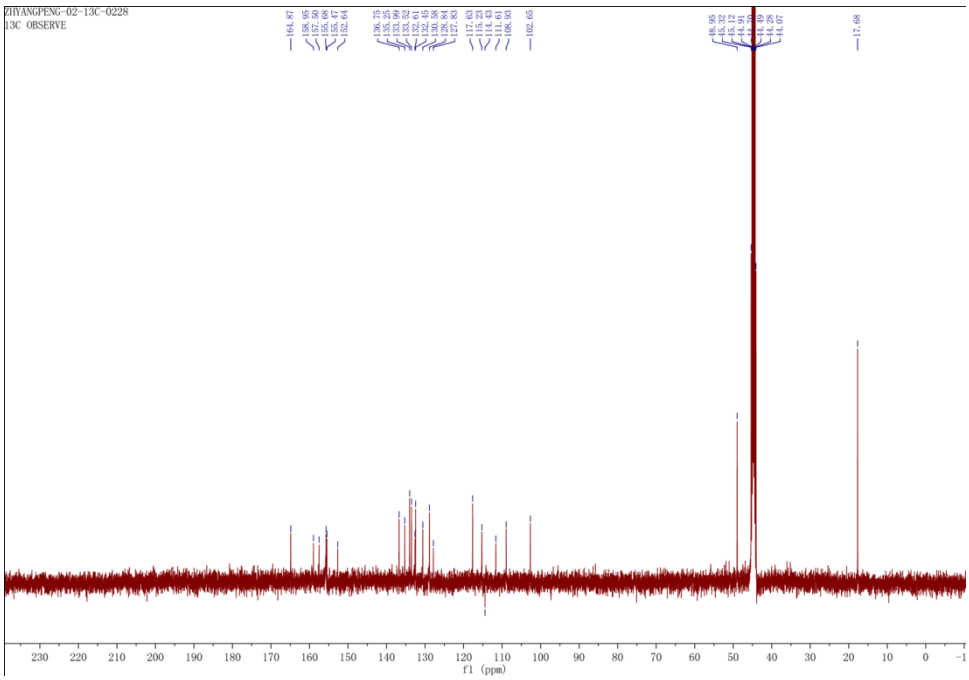

Figure S11. [13]C NMR spectra of **L2**.