Supporting Information

A rare dihydroxo copper(II) complex with ciprofloxacin; a combined experimental and ONIOM computational study of the interaction of the complex with DNA and BSA

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	compound	$K_{\rm vs} {\rm M}(^{-1})$	$K_{q} M(^{-1}s^{-1})$	$K_{\rm b} {\rm M}(^{-1})$	п	Ref
DNA	Ciprofloxacin	1.09×10^{4}	1.09×10^{12}	3.98×10^{4}	1.12	1
	Cu(II) complex	1.2×10^4	$1.2 imes 10^{12}$	$1.17 imes 10^4$	0.88	work this

Table. S1. The DNA binding constants and parameters derived for ciprofloxacin and the Cu(II) complex.

Table. S2. The BSA binding constants and parameters derived for ciprofloxacin and the Cu(II) complex.

	compound	$K_{\rm vs} { m M}(^{-1})$	$K_{q} M(^{-1}s^{-1})$	$K_{\rm b} {\rm M}(^{-1})$	п	Ref
BSA	Ciprofloxacin	2.33×10^{4}	2.33×10^{12}	2.17×10^{4}	0.91	2
	Cu(II) complex	$6.38 imes 10^4$	$6.38 imes 10^{12}$	$4.08 imes 10^4$	0.97	work this

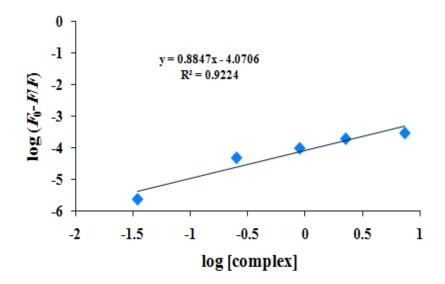
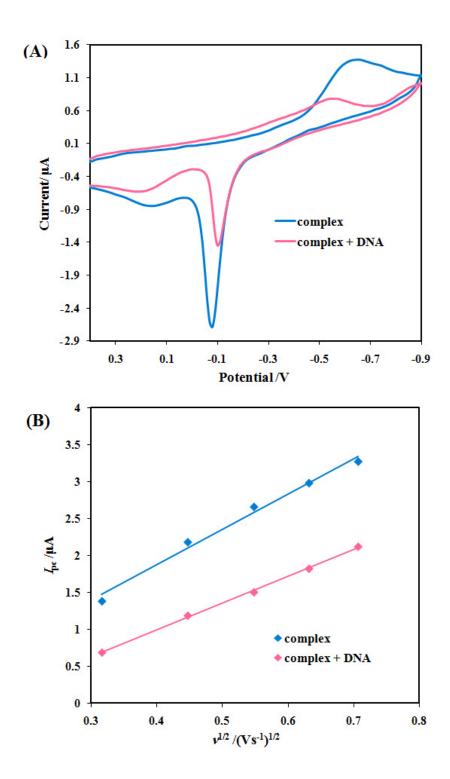


Fig. S1 Plot of $\log (F_0 - F)/F$) versus log [complex].



(A) Cyclic voltammograms Fig. **S2** 1.0 10-3 М of × of the trans- $[Cu(cip)_2(OH)_2] \cdot 2CH_3OH \cdot 6H_2O$ in the absence and presence of CT-DNA (scan rate = 0.1) V/s). (B) The plots of the cathodic peak currents of the complex in the absence and presence of CT-DNA versus the square root of the scan rates ($v^{1/2}$).

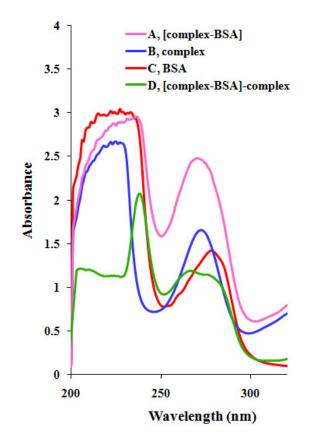


Fig. S3 Electronic absorption spectra of BSA in the presence of the Cu(II) complex. (A) (pink) The absorption spectra of BSA-complex system when the mol ratio is 1:1; (B) (blue) the absorption spectra complex only; (C) (red) the absorption spectra of BSA only; (D) (green) the difference absorption spectra between the BSA-complex and the complex at the same concentration.

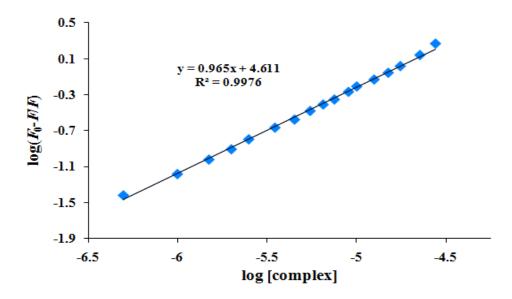


Fig. S4 Determination of the complex-BSA binding constant and the number of binding sites on BSA.

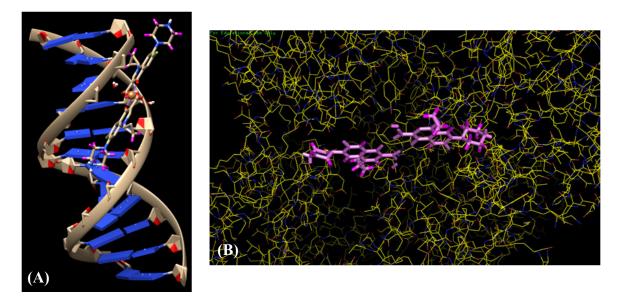


Fig. S5 Molecular docking of the complex with DNA (A) and BSA (B).

References

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- 2 Y.J. Hu, Y. Ou-Yang, Y. Zhang and Y. Liu, Protein J., 2010, 29, 234–241.