

## Electronic Supplemental Information

### Effect of Different Gel Electrolytes on Graphene Based Solid-State Supercapacitors

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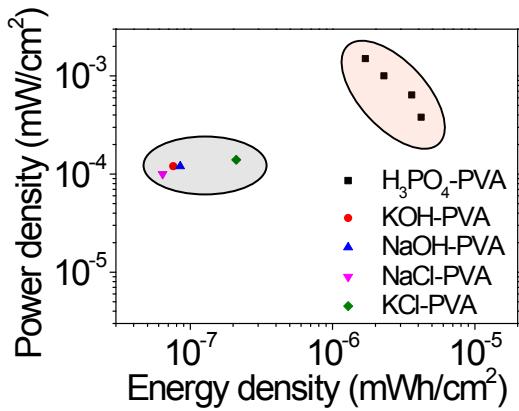
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**Table S1.** Specific capacitance, energy and power densities of devices with different gel electrolytes.

Electrolyte (0.01mol-1gPVA- 10mlH <sub>2</sub> O)	Specific capacitance (μF/cm <sup>2</sup> )	Energy density (mWh/cm <sup>2</sup> )	Power density (mW/cm <sup>2</sup> )
H <sub>3</sub> PO <sub>4</sub>	15	1.3×10 <sup>-6</sup>	1.1×10 <sup>-3</sup>
H <sub>2</sub> SO <sub>4</sub>	820	4.5×10 <sup>-4</sup>	1.6×10 <sup>-1</sup>
KOH	2.2	7.6×10 <sup>-8</sup>	1.2×10 <sup>-4</sup>
NaOH	2.4	8.5×10 <sup>-8</sup>	1.2×10 <sup>-4</sup>
NaCl	0.5	6.4×10 <sup>-8</sup>	1.0×10 <sup>-4</sup>
KCl	1.5	2.1×10 <sup>-7</sup>	1.4×10 <sup>-4</sup>



**Figure S1.** Ragone plots for the supercapacitors with the  $\text{H}_3\text{PO}_4$ -PVA, KOH-PVA, NaOH-PVA, KCl-PVA and NaCl-PVA gel electrolytes. The plots in up-right circle region is for devices using  $\text{H}_3\text{PO}_4$ -PVA electrolyte, while the left circle is for those using base and salt electrolytes.