Visible light-driven decomposition of gaseous benzene on robust Sn²⁺-doped anatase TiO₂ nanoparticles

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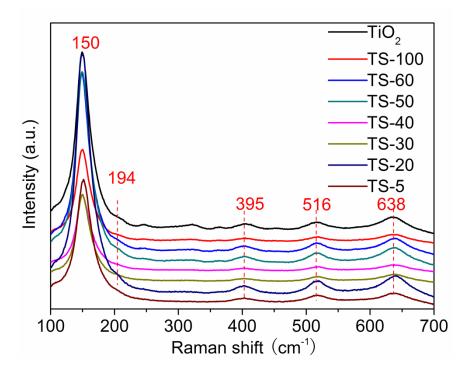


Figure S1 The Raman spectrum of the obtained TS-x samples and pure TiO₂.

To verify the structure of Sn-incorporated TiO₂, the TS-x samples were further characterized by the Raman technology, as shown in Figure S1. There are one strong Raman peak at 149.7cm⁻¹ and four weak peaks at 194, 395, 516 and 638 cm⁻¹ are assigned to Eg, Eg, B1g, A1g, and Eg, respectively, which are consistent with anatase structure TiO₂. [1, 2] From the Raman spectra of the TS-x samples, the main Raman peaks at 113cm⁻¹ and 211cm⁻¹ attributed to SnO crystalline form can't be detected, which indicates there don't exist a separate SnO crystalline phase in the Sn-doped TiO₂ system. [3] The result is in good agreement with the result from the XRD pattern of the TS-x samples.

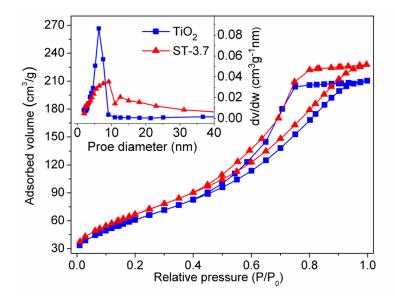


Figure S2 Nitrogen adsorption-desorption isotherms of the pure TiO₂ and TS-40 samples. The Insert shows the corresponding Barret-Joyner-Halenda (BJH) pore size distributions.

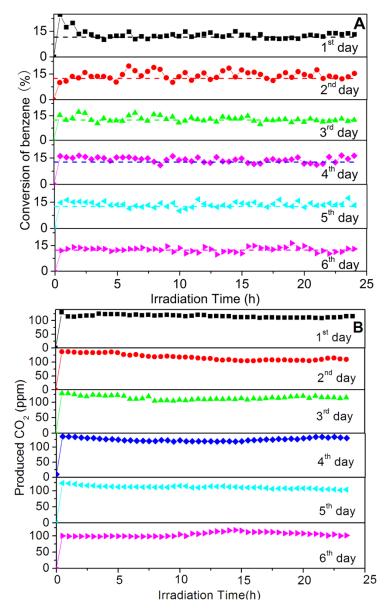
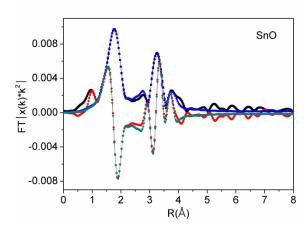
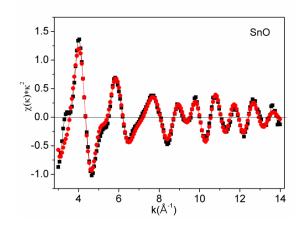
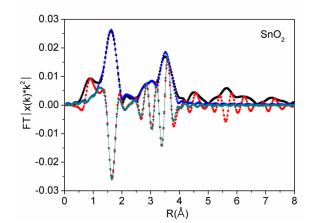
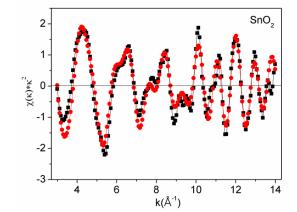


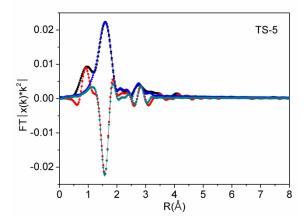
Figure S3 Cycle runs for the photooxidation of benzene on TS-20 sample: (A) the conversion of benzene; (B) The yield of produced CO_2 .

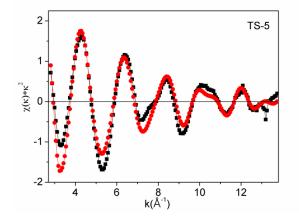












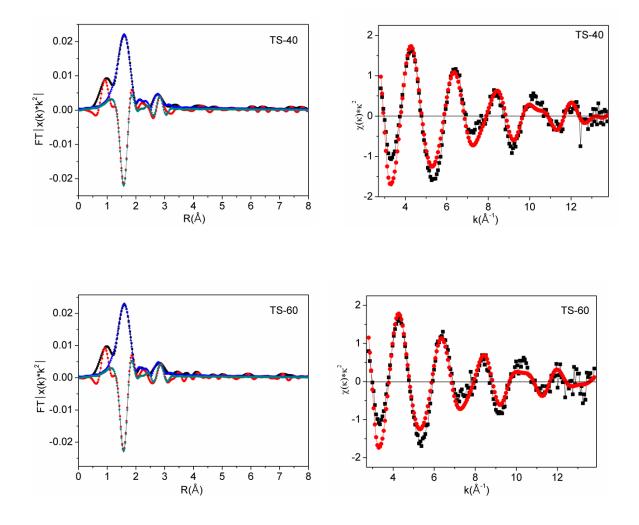


Figure S4 Sn K-edge EXAFS fit results for the SnO, SnO_2 , TS-5, TS-40 and TS-60 samples.

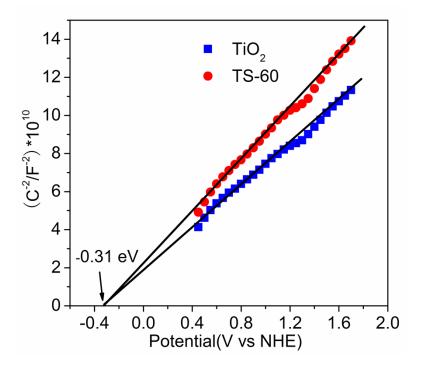


Figure S5 Mott-Schottky plots of the pure TiO_2 and TS-60 material.

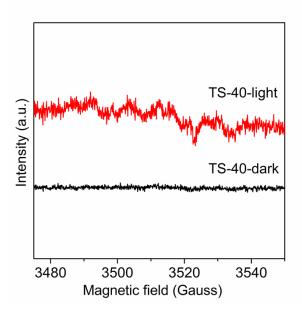


Figure S6 ESR signal of the DMPO-[•]O₂ spin adducts for TS-40 sample without irradiation and with visible light irradiation.

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