

1 Influence of surfactant alkyl length in functionalizing

2 sol-gel derived microporous cobalt oxide silica

3

4 Gianni Olguin^{1,2}, Christelle Yacou¹, Simon Smart^{1,*}, João C. Diniz da Costa¹

5 ¹The University of Queensland, FIMLab - Films and Inorganic Membrane Laboratory, School of

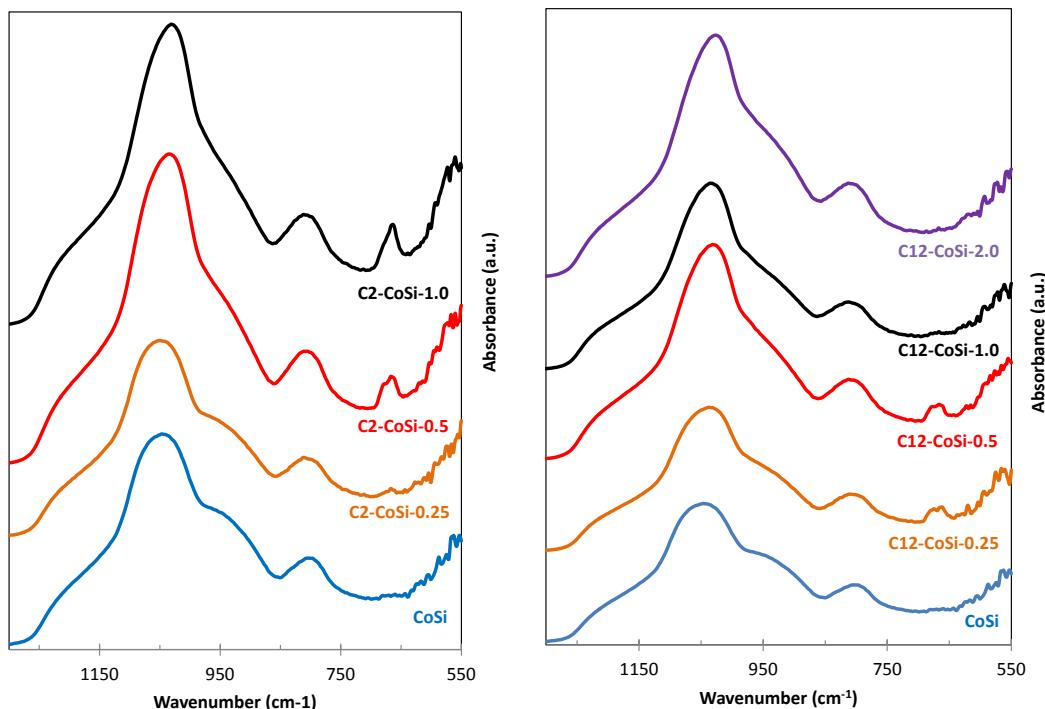
6 Chemical Engineering, Brisbane, Qld, 4072, Australia

7 ²Pontificia Universidad Católica de Valparaíso, Escuela de Ingeniería Química, Valparaíso, Chile

8 (*) corresponding author (s.smart@uq.edu.au)

9 Supplementary Information

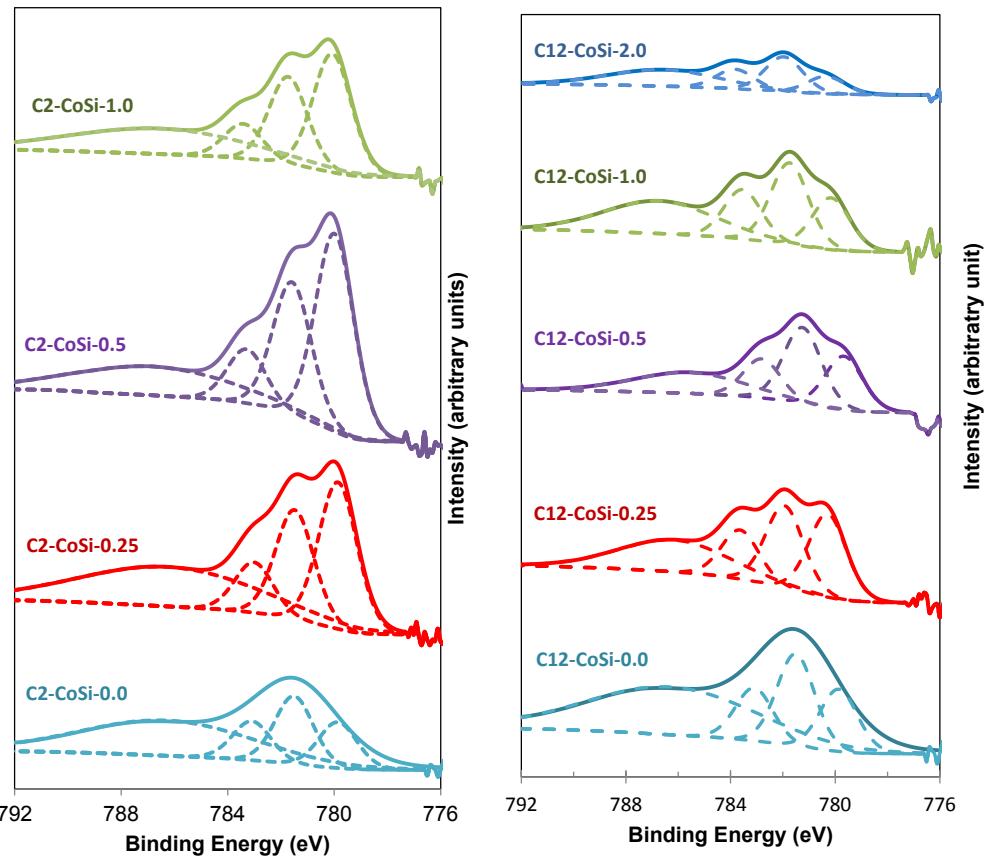
10



11

1 Figure S1: FTIR normalized spectra of calcined xerogels as a function of surfactant concentration. Left: C2-AB.
 2 Right: C12-AB. The alteration of the Co_3O_4 characteristic peak at 667 cm^{-1} ($\text{Co}^{3+}\text{-O}$ vibration) can be observed as
 3 surfactant load increases.

4



5

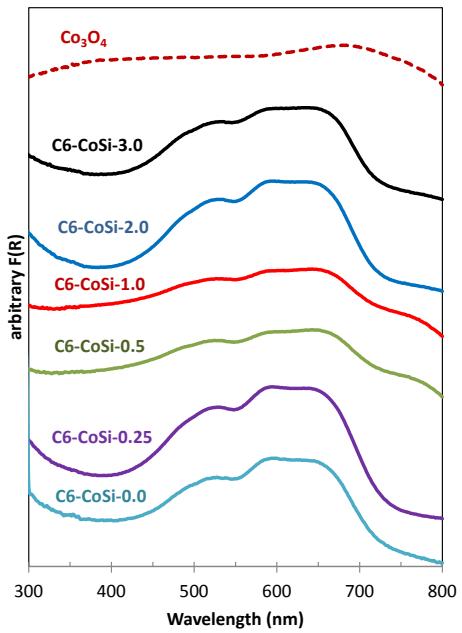
6 Figure S2: Evolution of XPS spectra near $\text{Co}2\text{p}$ region for calcined xerogels as function of surfactant
 7 concentration. Left: C2-AB. Right: C12-AB. The alteration of the characteristic low energy peak at 779.9 eV
 8 which is connected with a high valence state of cobalt, can be observed as surfactant load increases.

9

10 Table S1: XPS peak deconvolution data. Peaks position (eV), width, area and goodness of fit for each sample
 11 displayed in Figure S2.

| Sample | $\text{Co}^{+3} (2\text{p}_{3/2})$ (fwhm 1.7 eV) | | $\text{Co}^{+2} (2\text{p}_{3/2})$ (fwhm 1.7 eV) | | $\text{Co}^{+2} (2\text{p}_{3/2})$ (fwhm 1.7 eV) | | $\text{Satellite} (2\text{p}_{3/2})$ (fwhm 5-8 eV) | | Chi square STD residual |
|---------------|-----------------------------------------------------|--------|-----------------------------------------------------|--------|-----------------------------------------------------|-------|-------------------------------------------------------|--------|----------------------------|
| | Position | Area | Position | Area | Position | Area | Position | Area | |
| CoSi | 779.9 | 757.5 | 781.6 | 1126.3 | 783.2 | 644.7 | 786.4 | 2619.6 | 0.804 |
| C2-CoSi-0.25 | 779.9 | 2420.6 | 781.5 | 1808 | 783.1 | 831.2 | 786.3 | 3070 | 0.969 |
| C2-CoSi-0.5 | 779.9 | 3263.4 | 781.6 | 2181.8 | 783.3 | 891.1 | 786.8 | 2025 | 0.892 |
| C2-CoSi-1.0 | 780.0 | 1957.8 | 781.7 | 1441.8 | 783.4 | 571.3 | 786.6 | 1987 | 0.756 |
| C12-CoSi-0.25 | 780.1 | 265.4 | 781.8 | 260.6 | 783.5 | 155.8 | 785.3 | 375.1 | 0.795 |
| C12-CoSi-0.5 | 779.7 | 170.4 | 781.3 | 242.7 | 782.9 | 124.1 | 785.5 | 195.5 | 0.864 |
| C12-CoSi-1.0 | 780.1 | 632.5 | 781.7 | 996 | 783.5 | 601.8 | 786.76 | 1171.6 | 0.907 |
| C12-CoSi-2.0 | 780.1 | 55.4 | 781.8 | 110.4 | 783.5 | 62.15 | 786.4 | 40.3 | 0.796 |

12



1

2 Figure S3: DR-UV Visible spectrum as function of surfactant load for the calcined C6-AB samples and pure
3 Co_3O_4 .

4