

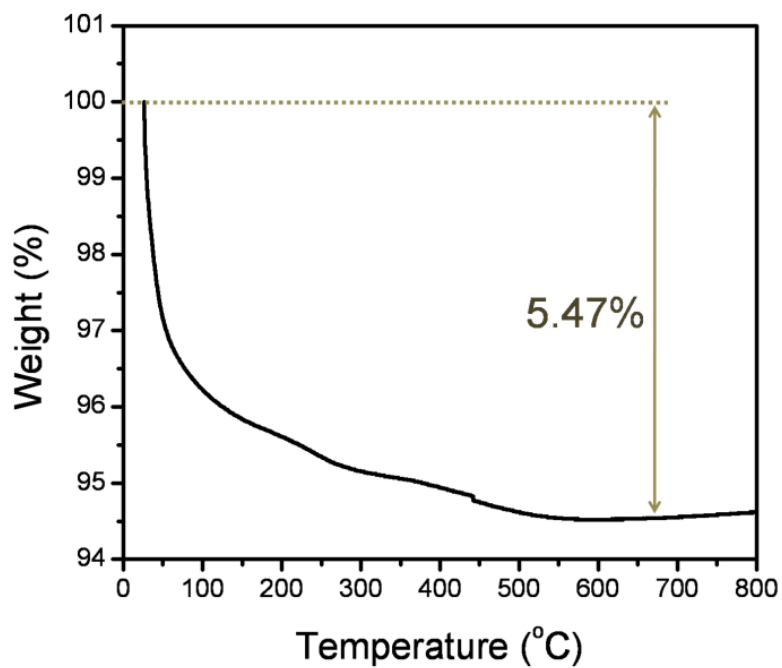
Supporting Information:

## Facile synthesis of nanostructural carbon nanotubes/iron oxide hybrids for lithium-ion battery anodes

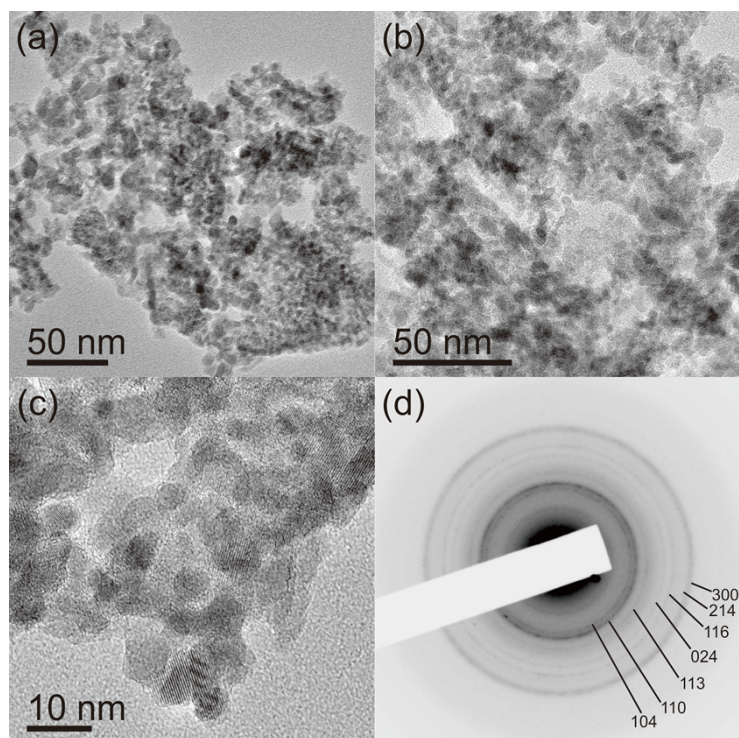
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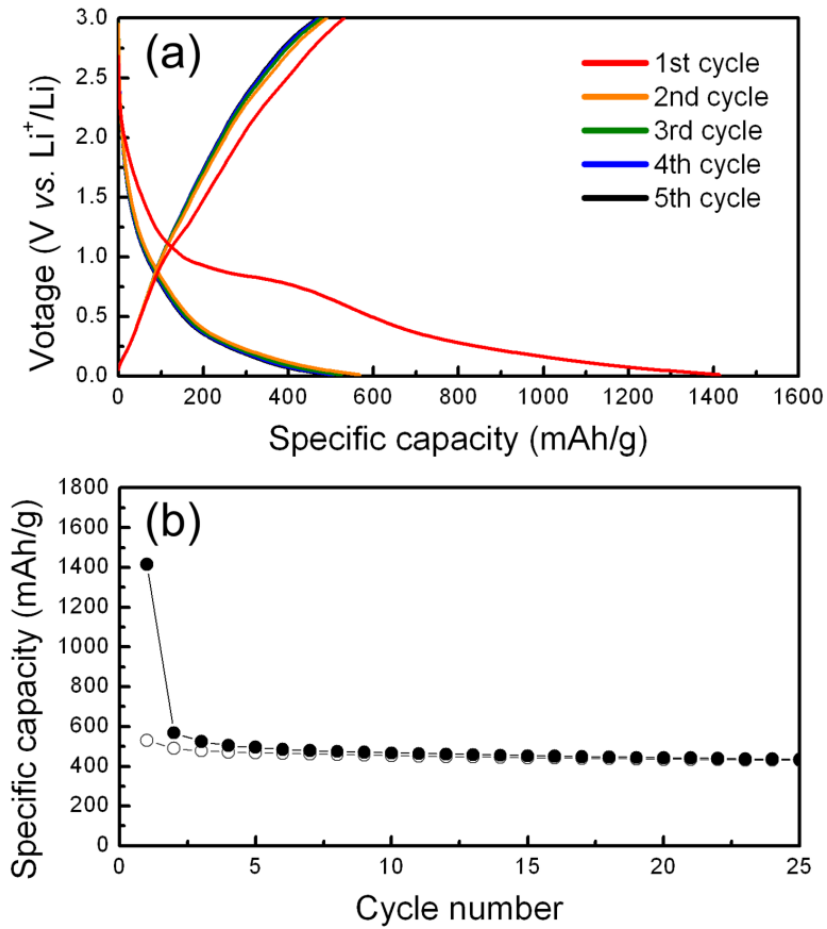
**Fig. S1** Large scale synthesis of carbon nanotubes/iron oxide hybrids (CNIOHs).



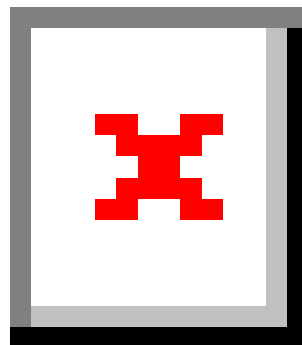
**Fig. S2** Thermogravimetric analysis of the iron (II) oxalate dehydrate precursor in the atmospheres with a heating rate of 5 °C/min.



**Fig. S3** Structural analysis for iron oxide nanoparticles formed via strong ultrasonic treatment for 0.5 h; (a and b) different magnifications TEM images; (c) HRTEM image; (d) Selected area electron diffraction.



**Fig. S4** (a) The charge-discharge voltage profiles of SWCNT for initial 5 cycles. (b) Cycle performance of SWCNT at a current density of 200 mA/g.



**Fig. S5** (a) SEM image of CHIONs-3 after 30 cycles.

**Table S1.** Electrochemical properties of Fe<sub>2</sub>O<sub>3</sub> or Fe<sub>2</sub>O<sub>3</sub>/carbon nanotube composites for lithium-ion batteries.

Samples	Electrochemical properties	Reference
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub>	662 mAh/g (100th cycle, 200 mA/g)	[34]
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub>	710 mAh/g (100th cycle, 200 mA/g)	[35]
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub>	680 mAh/g (80th cycle, 65 mA/g)	[36]
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub> /CNT	800 mAh/g (100th cycle, 500 mA/g)	[17]
Fe <sub>2</sub> O <sub>3</sub> /CNT	619 mAh/g (80th cycle, 50 mA/g)	[37]
Fe <sub>2</sub> O <sub>3</sub> /MWCNT	515 mAh/g (50th cycle, 100 mA/g)	[38]
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub> /SWCNT (CHIONs-3)	1179 mAh/g (80th cycle, 200 mA/g)	this work