

Electronic Supplementary Information for

How much do coulombic interactions stabilize a mesophase? Ionic pair and non-ionic binary isosteric derivatives of monocarbaborates and carboranes

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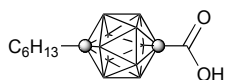
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1. Additional synthetic details

4-Undecylphenol (8[11]).¹ A solution of *n*-undecylmagnesium bromide in THF (11.6 mmol in 10 mL) was added to a stirred suspension of PdCl₂(dppf)₂ (47 mg, 0.2 mmol) and 4-bromophenol (1.00 g, 5.8 mmol) in THF (5 mL) at -78 °C. The mixture was stirred at -78 °C for 15 min, then cooling was removed. After stirring for 3h at 20 °C the mixture was quenched with 5% HCl and extracted with ether. The combined extracts were dried (Na₂SO₄), solvent was evaporated and the crude mixture was passed through a silica gel plug (CH₂Cl₂/hexane, 2:1). The product was further purified by distillation (180 °C / 0.5 mm Hg) to give 0.50 g (35% yield) of phenol **8[11]**, whose ¹H NMR was identical with that reported.¹

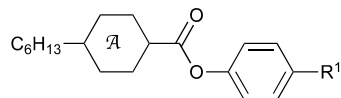
Preparation of [*closo*-1,12-C₂B₁₀H₁₀-1-COOH-12-C₆H₁₃] (**5[12]**).



p-Carborane (1.44 g, 10 mmol) was dissolved in dry THF (20 mL), cooled to -80 °C, and 1.8 M *n*-BuLi in hexane (6.0 mL, 10.8 mmol) was added dropwise. After 30 min 1-bromohexane (1.65 g, 10.0 mmol) was added and the mixture was stirred at ambient temperature for 2 hrs. The mixture was cooled to -80 °C and another portion of 1.8 M *n*-BuLi in hexane (6.0 mL, 10.8 mmol) was added dropwise. After 30 min dry CO₂ gas was passed through the mixture for 40 min. The mixture was evaporated to dryness, dissolved in 2 M KOH and washed with hexanes (2x) to remove 1,12-dihexyl-*p*-carborane. The aqueous layer was acidified and precipitation extracted with ether (3x). The extracts were dried (Na₂SO₄), ether evaporated, and the solid residue dried in vacuum. The crude acid **5[12]** was extracted into hot hexane leaving behind insoluble *p*-carborane-1,12-dicarboxylic acid (0.42 g, 18% yield). Hexane was evaporated and the residue (1.40 g, 51% yield) was purified by sublimation (130 °C / 0.3 mm Hg) followed by recrystallization from heptane giving pure acid **5[12]**: mp 126-128 °C; ¹H NMR (400 MHz, CDCl₃) δ 0.84 (t, *J* = 7.1 Hz, 3H), 1.08-1.25 (m, 8H), 1.5-3.5 (m, 10H), 1.59 (t, *J* = 7.9 Hz, 2H); {¹H} ¹¹B NMR (128 MHz, CDCl₃) δ -13.5 (5B), -12.7 (5B). Anal. Calc. for C₉H₂₄B₁₀O₂: C, 39.68; H, 8.88. Found: C, 39.61; H, 8.98%.

2. Additional DSC data

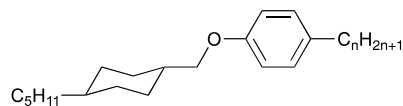
Table S1. Transition temperatures for **2**. ^a



a	R	A	Temperature (<i>enthalpy</i>)
		[10] [12] [Ph] ^b	Cr 67 (23.2) N 197 (1.4) I Cr 92 (22.7) N 206 (1.5) I Cr ₁ 84 Cr ₂ 91 (72.8) N 206 (1.8) I
b		[10]	Cr ₁ 104 (11.9) Cr ₂ 115 (15.3) N 176 (1.0) I
		[12]	Cr 145 (16.4) N 187 (1.2) I
		[Ph]	NA
c	-OCOC ₈ H ₁₇	[10]	Cr 43 (38.9) (N 37 (0.9)) I
		[12]	Cr 46 (52.8) (N 45 (0.6)) I
		[Ph]	NA

^a Cr-crystal, N-nematic, I-isotropic. Temperatures in °C and enthalpies in kJ/mol (in parentheses) obtained on heating at 5 K min⁻¹. ^b ref ²

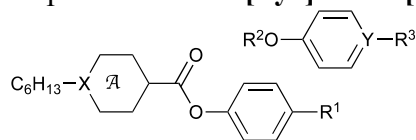
Table S2. Transition temperatures for [Ph]. ^a



	n	Temperature (<i>enthalpy</i>)
	3	Cr 32 (Sm 10) I ^b
	5	Cr 35 (Sm 31) I ^b
	7	Cr 34 Sm 38 I ^b
c	11	Cr 32 (26.4) B 42 (12.7) I

^a Cr-crystal, Sm-smectic, B - soft crystalline (SmB_{cryst}), I-isotropic. Temperatures in °C and enthalpies in kJ/mol (in parentheses) obtained on heating at 5 K min⁻¹. ^b Ref. ³

Table S3. Transition temperatures for **1-[Pyr]** and **2-[Ph]**. ^a



	X, Y	Temperature (<i>enthalpy</i>)
1[10]c-[Pyr]c	N ⁺ , B ⁻	Cr 126 (12.6) E 163 (11.8) SmA 207 (15.2) I
2[10]c-[Ph]c	C, C	Cr 27 (56.1) (N 26 (2.0)) I
1[12]c-[Pyr]c	N ⁺ , B ⁻	Cr ₁ 85 (18.6) X 148 (5.8) E 200 (12.3) SmA 210 (14.5) I
2[12]c-[Ph]c	C, C	Cr 30 (50.2) (N 29 (2.2)) I
2[10]c-[Pyr]a^b	N ⁺ , B ⁻	Cr ₁ 64 (9.2) Cr ₂ 101 (3.4) X 107 (15.1) I
2[12]c-[Pyr]a^c	N ⁺ , B ⁻	Cr 117 (27.4) I
1[12]c-[Pyr]b	N ⁺ , B ⁻	Cr ₁ 77 (33.9) Cr ₂ 132 (8.4) E 172 (24.8) I

^a Cr-crystal, N-nematic, I-isotropic. Temperatures in °C and enthalpies in kJ/mol (in parentheses) obtained on heating at 5 K min⁻¹. ^b ref ⁴. ^c ref ⁵.

3. Powder XRD measurements

X-ray diffraction experiments were performed with Bruker D8 GADDS (Cu K α radiation, Göbel mirror, point collimator, Vantec 2000 area detector) equipped with a modified Linkam heating stage and with Bruker D8 Discover system (Cu K α radiation, Göbel mirror, scintillation counter, Anton Parr DCS350 heating stage). Samples were prepared in a form of a thin film or a droplet on heated surface. XRD data was analyzed using program TOPAS 3 (Bruker). The asymmetric wide angle signal in diffractograms was fitted with two functions type PV (pseudo-Voight).

2D patterns, integrated intensities and layer spacing as function of temperature, $d(T)$, are shown in Figures S1 – S8 for individual materials.

XRD data was analyzed using program TOPAS 3 (Bruker). The asymmetric wide angle signal in diffractograms was fitted with two functions type PV (pseudo-Voight).

Ion pair 1[10]c-[Pyr]c

Table S4. Positions of diffraction signals in the 2D patterns obtained for 1[10]c-[Pyr]c.

<i>SmA phase, 175 °C</i>			<i>E phase, 150 °C</i>		
$d_{\text{exp}}(\text{\AA})$	(hkl)	$d_{\text{calc}}(\text{\AA})$	$d_{\text{exp}}(\text{\AA})$	(hkl)	$d_{\text{calc}}(\text{\AA})$
27.63	(001)		29.84	(001)	
13.84	(002)	13.82	14.92	(002)	14.92
9.23	(003)	9.21	9.95	(003)	9.95
			8.99	(100)	
			8.61	(101)	8.61
5.06	diffused		7.70	(102)	7.70
4.59	diffused		6.12	(010)	
			6.00	(011)	6.00
			5.06	(110)	5.06
			4.99	(111)	4.99
			4.50	(200)	4.50
			4.45	(201)	4.45
			unit cell dimensions:		
			$a = 8.99$; $b = 6.12$; $c = 29.84$		

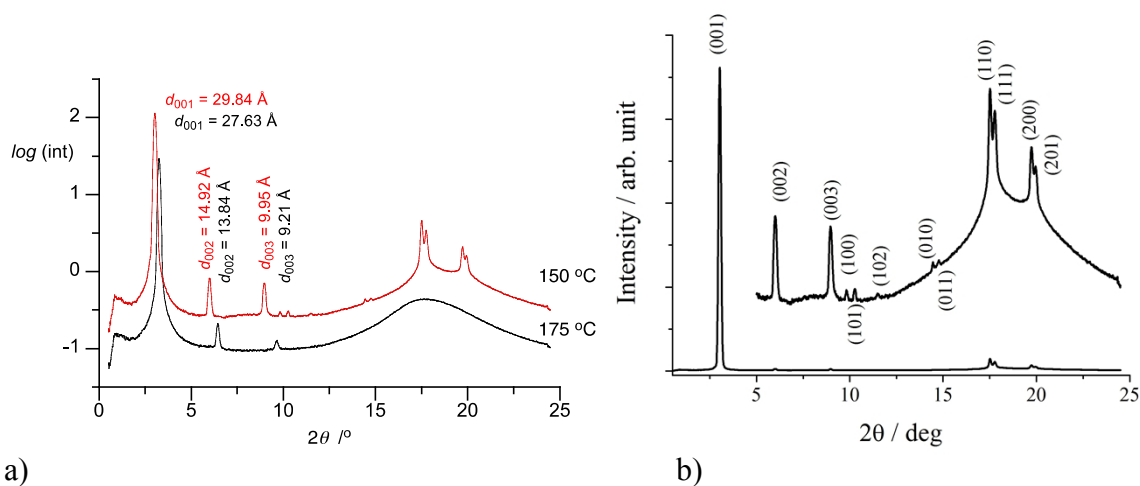


Figure S1. a) A comparison of integrated XRD patterns for SmA (175 °C, black) and E (150 °C, red) phases of 1[10]c-[Pyr]c, b) integrated XRD pattern in E phase (150 °C) of 1[10]c-[Pyr]c with full indexing for assumed orthorhombic lattice with parameters $a = 8.99 \text{ \AA}$, $b = 6.12 \text{ \AA}$ and $c = 29.84 \text{ \AA}$.

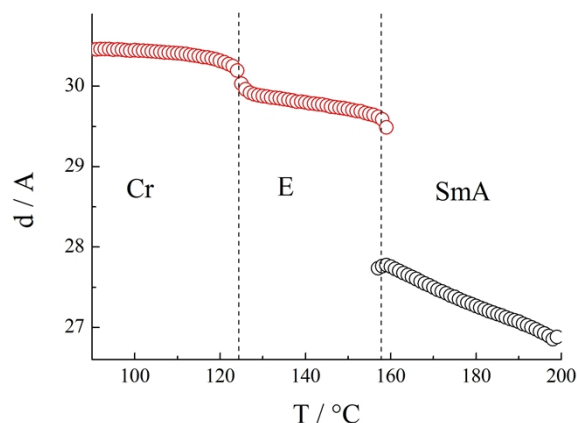


Figure S2. Layer spacing calculated from signal (001) for **1[10]c-[Pyr]c** as a function of temperature obtained on heating. Thermal expansion coefficients: $\kappa_A = -24.7 \pm 0.2 \times 10^{-3} \text{ Å K}^{-1}$ in SmA phase (temperature range 180–160 °C) and $\kappa_E = -8.1 \pm 0.1 \times 10^{-3} \text{ Å K}^{-1}$ in E phase (temperature range 130–150 °C).

Ion pair 1[12]c-[Pyr]c

Table S5. Positions of diffraction signals in the 2D patterns obtained for **1[12]c-[Pyr]c**.

<i>SmA phase, 205 °C</i>			<i>E phase, 180 °C</i>		
$d_{\text{exp}}(\text{Å})$	(hkl)	$d_{\text{calc}}(\text{Å})$	$d_{\text{exp}}(\text{Å})$	(hkl)	$d_{\text{calc}}(\text{Å})$
26.64	(001)		28.71	(002)	
13.36	(002)	13.32	14.33	(004)	14.35
8.89	(003)	8.88	9.55	(006)	9.57
			9.05	(101)	9.05
5.35	diffused		8.23	(103)	8.26
4.85	diffused		6.46	(011)	6.46
			5.30	(110)	5.30
			5.22	(112)	5.21
			4.58	(200)	4.58
			4.53	(202)	4.52
			unit cell dimensions:		
			$a = 9.16, b = 6.50, c = 57.42$		

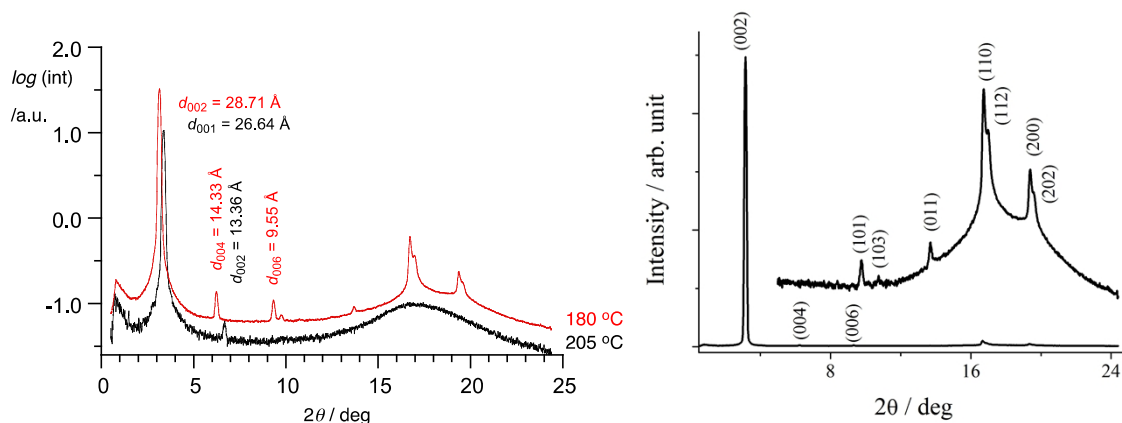


Figure S3. a) A comparison of integrated XRD patterns for **1[12]c-[Pyr]c** of a SmA (205 °C, black) and of E phase (180 °C, red), b) integrated XRD pattern in the E phase of **1[10]c-[Pyr]c** (150 °C) with full indexing for assumed orthorhombic lattice with parameters $a = 9.16 \text{ \AA}$, $b = 6.50 \text{ \AA}$ and $c = 57.42 \text{ \AA}$. For the enlarged part of the graph the 2D pattern has been integrated only in limited azimuthal angle range to avoid signal overlapping, thus (004) and (006) peaks are not visible.

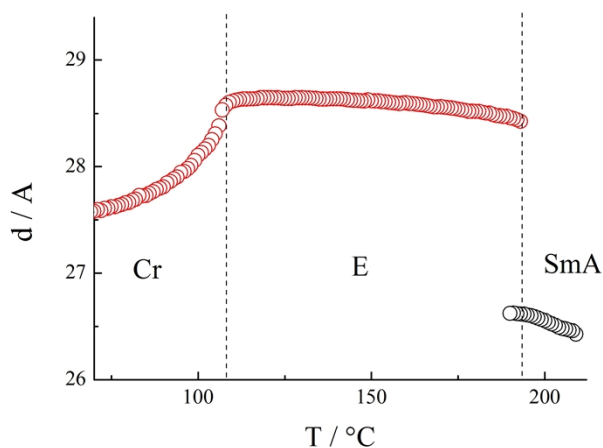


Figure S4. Layer spacing calculated from signal (001) for **1[12]c-[Pyr]c** as a function of temperature obtained on cooling. Thermal expansion coefficients: $\kappa_A = -12.1 \pm 0.3 \times 10^{-3} \text{ \AA K}^{-1}$ for SmA phase (temperature range 193–208 °C), and $\kappa_E = -2.5 \pm 0.1 \times 10^{-3} \text{ \AA K}^{-1}$ for E phase (temperature range 140–170 °C).

Ion pair 1[12]c-[Pyr]b

Table S6. Positions of diffraction signals in the 2D patterns obtained for **1[12]c-[Pyr]b**.

<i>E phase, 150 °C</i>		
$d_{\text{exp}}(\text{\AA})$	(hkl)	$d_{\text{calc}}(\text{\AA})$
25.26	(001)	
12.67	(002)	12.63
9.11	(100)	
8.56	(101)	8.57
6.44	(010)	
6.23	(011)	6.24
5.25	(110)	5.26
5.15	(111)	5.15
4.56	(200)	4.56
4.49	(201)	4.48

unit cell dimensions:
 $a = 9.11$, $b = 6.44$, $c = 25.26$

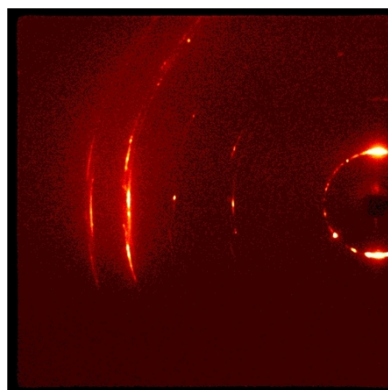
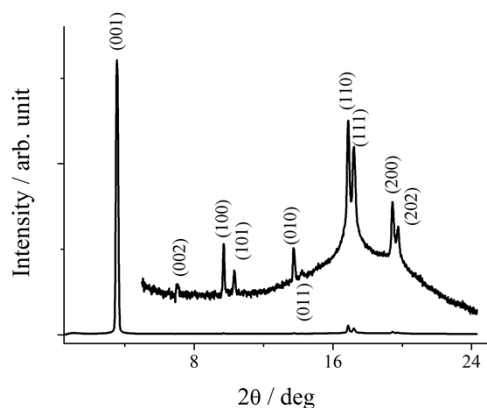


Figure S5. a) Integrated XRD pattern taken at 150 °C in E phase of **1[12]c-[Pyr]b**. Indexing done assuming orthorhombic lattice with parameters $a = 9.11 \text{ \AA}$, $b = 6.44 \text{ \AA}$ and $c = 25.62 \text{ \AA}$; b) 2D XRD pattern for **1[12]c-[Pyr]b** at 150 °C (E phase). Due to direct Iso – E phase transition well aligned sample could not be obtained.

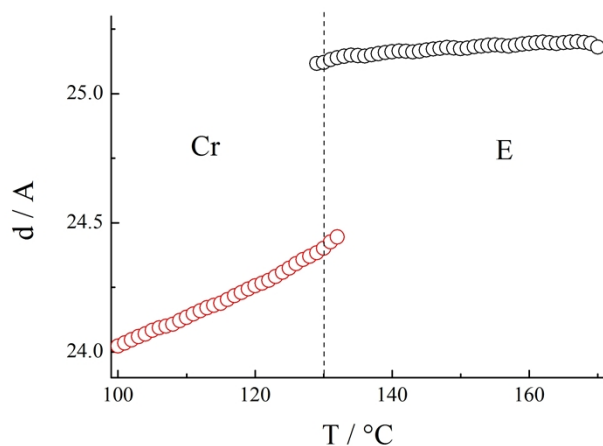


Figure S6. Layer spacing calculated from signal (001) for **1[12]c-[Pyr]b** as a function of temperature obtained on cooling. Thermal expansion coefficient: $\kappa_E = +1.9 \pm 0.1 \times 10^{-3} \text{ \AA K}^{-1}$ for E phase (temperature range 130—160 °C)

Ion pair [Pyr]c-Br

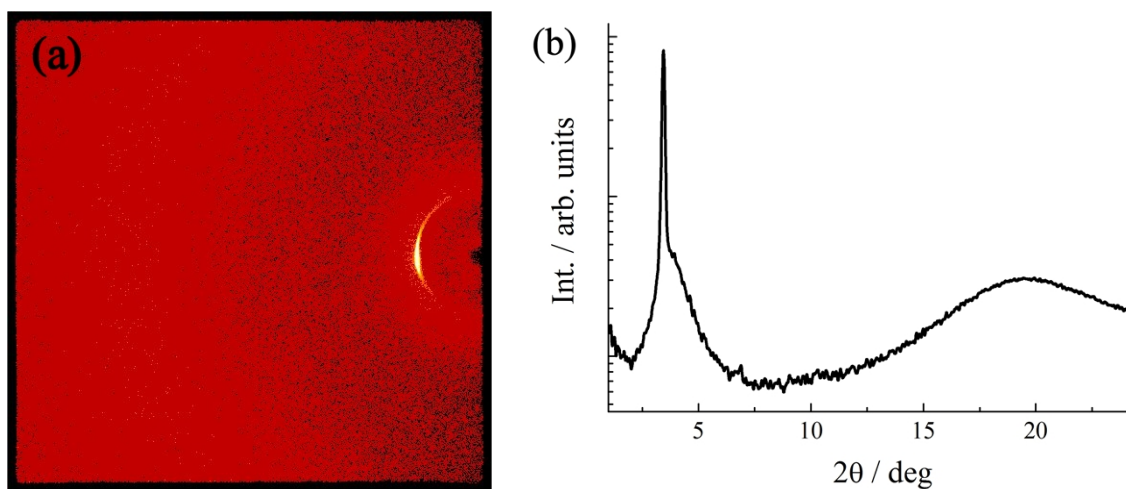


Figure S7. a) 2D XRD pattern for **[Pyr]c-Br** at 155 °C and b) intensity vs. diffraction angle obtained by integration of the 2D pattern over azimuthal angle. Note that due to low thermal stability of the material SmA phase coexists with the isotropic liquid.

[Ph]c

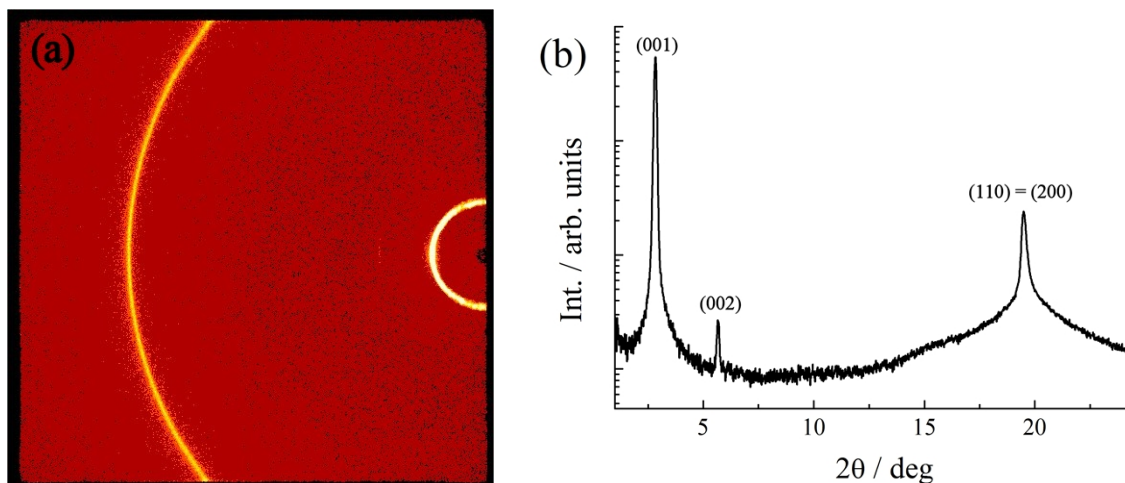


Figure S8. a) 2D XRD pattern for powder sample of **[Ph]c** in B phase (SmB_{cry}) at 29 °C and b) intensity vs. diffraction angle obtained by integration of the 2D pattern over azimuthal angle.

Table S7. Positions of diffraction signals in the 2D patterns taken for for **[Ph]c**

<i>B phase, 42 °C</i>	<i>B phase, 35 °C</i>	<i>B phase, 29 °C</i>
$d_{\text{exp}}(\text{\AA})$ (<i>hkl</i>) $d_{\text{calc}}(\text{\AA})$	$d_{\text{exp}}(\text{\AA})$ (<i>hkl</i>) $d_{\text{calc}}(\text{\AA})$	$d_{\text{exp}}(\text{\AA})$ (<i>hkl</i>) $d_{\text{calc}}(\text{\AA})$
31.14 (001)	31.16 (001)	31.19 (001)
15.56 (002) 15.07	15.55 (002) 15.08	15.57 (002) 15.09
4.57 (110)=(200)	4.56 (110)=(200)	4.55 (110)=(200)
4.53 diff	4.52 diff	4.50 diff

4. Additional molecular models

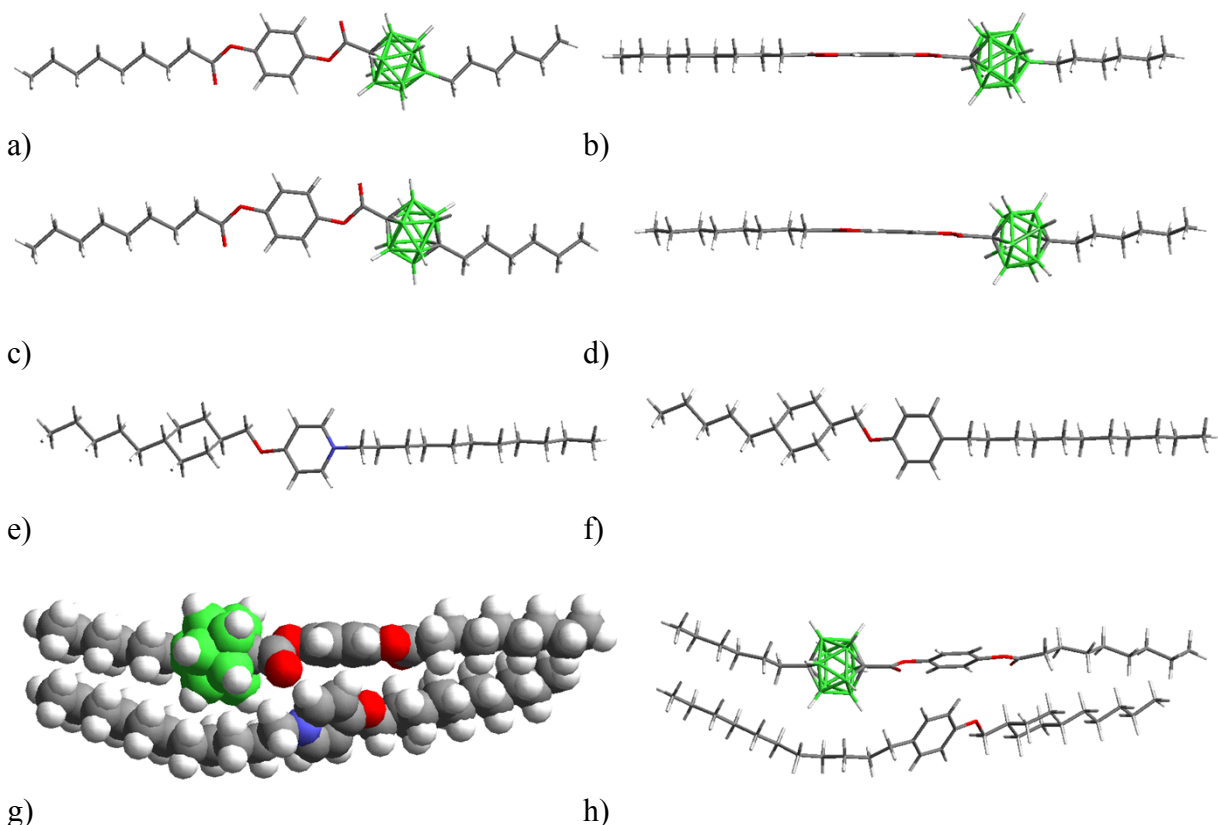


Figure S9. Equilibrium ground state geometry obtained at the M06-2x/3-21G* level of theory for a,b) **1[12]c** anion, c,d) **2[12]c**, e) **[Pyr]c** cation, f) **[Ph]c**, g) ion pair **1[12]c-[Pyr]c**, h) non-ion pair **2[12]c-[Ph]c**.

5. Archive files for M06-2x/3-21G* calculations

1[12]c

```
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2[12]c

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[Pyr]c

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94138234\C,1.2095725262,-1.089861147,0.198304579\C,0.5348227672,-0.911
5857503,-1.0271718649\C,-0.7777704114,-1.3071229925,-1.1407432996\N,-1
.4426606092,-1.8695520181,-0.1000173764\O,2.4757887881,-0.7483031948,0
.4332879423\C,-2.8964519786,-2.2199296651,-0.2463585874\C,-3.783261093
7,-0.9897080967,0.0006667471\C,-5.2709776313,-1.3688464884,-0.14523432
99\C,-6.1976637276,-0.1503679537,0.03725781\C,-7.681140411,-0.55444232
59,-0.0625146636\C,-8.6332465202,0.6516678827,0.0477137439\C,-10.11006
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51,-2.4784235\PG=C01 [X(C28H50N1O1)]\@

[Ph]c

1\1\GINC-OCTOPUS\FOpt\RM062X\3-21G*\C29H50O1\PIOTR\05-Feb-2014\0\#\P M
062x/3-21G* FOpt Geom(NoAngle, noDistance) fcheck #P freq=noraman\C5C
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2 [12]c-[Ph]c

1\1\GINC-OCTOPUS\FOpt\RM062X\3-21G*\C53H94B10O5\PIOTR\07-Feb-2014\0\\#P M062x/3-21G* FOpt Geom(NoAngle, noDistance) fcheck freq=noraman\\C6-C2B10-COO-PhOCC8 - C5-Chx-CH2OPh-C11, C1 antiparalel in vacuum\\0,1\B,-5.9118572848,-2.1942463327,-2.3806032445\B,-5.1921620031,-2.8234359755,-0.8697146885\B,-6.4677719426,-0.5209947974,-2.0772503898\B,-5.3102684155,-1.5449599982,0.3706971716\C,-6.527045096,-1.7251659525,-0.8396223692\B,-6.0985385332,-0.124489734,-0.3795332459\B,-4.3399568563,-0.1678195885,-0.1574643651\B,-3.7795024424,-1.8461666439,-0.4633893619\B,-5.059553642,0.4702790399,-1.675742466\B,-4.1484836322,-2.2472291361,-2.1730701673\B,-4.9456313525,-0.813041297,-2.9231488386\C,-3.7540300019,-0.6453466565,-1.6952013783\C,-2.4424818851,-0.0764533037,-2.1355982499\C,-0.0573349815,-0.6283396144,-2.0136254347\C,0.3846606617,0.087267941,-3.1228395189\C,1.7549400948,0.2479202423,-3.3061899214\C,2.66063

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