

SUPPLEMENTARY INFORMATION

Efficient one-pot synthesis of deoxyfructosazine and fructosazine from D-glucosamine hydrochloride under basic ionic liquid as a dual solvent-catalyst

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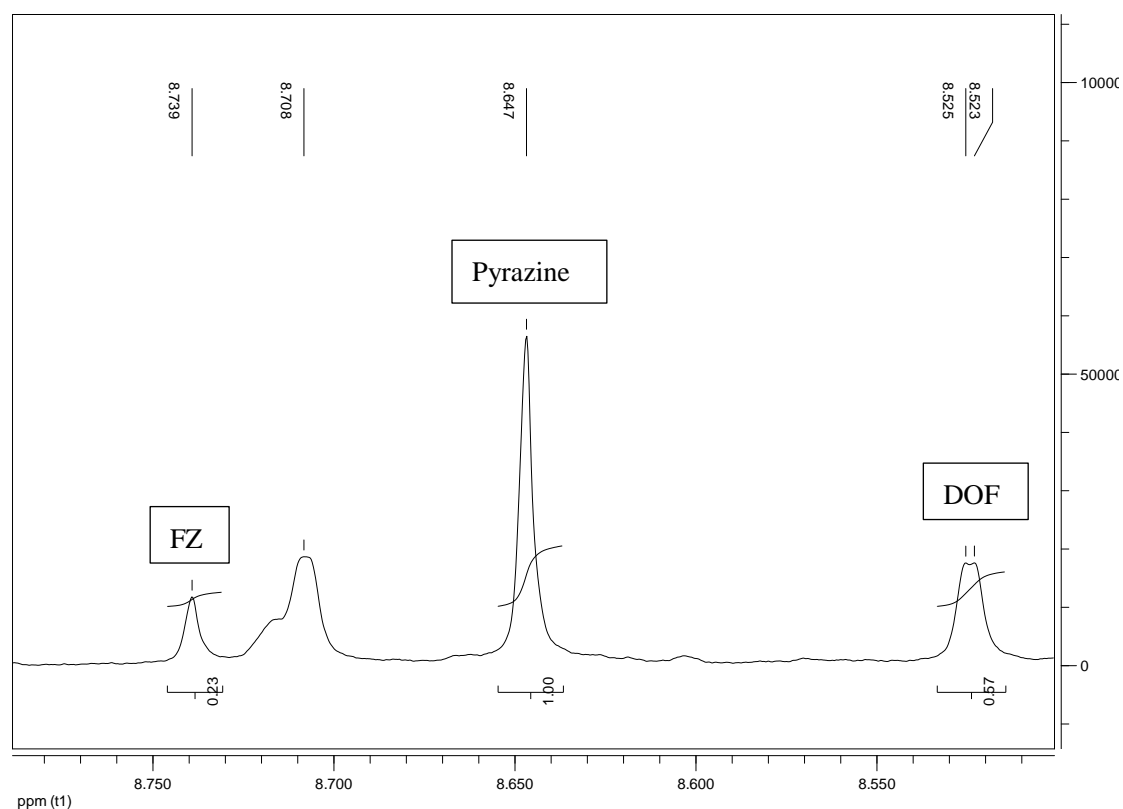


Fig. S1. A typical quantitative ^1H NMR spectrum for the measurement of DOF and FZ yields using pyrazine as internal standard compound.

The quantitative result of products is calculated as:

$$m(x) = p(std) \frac{mw(x)}{mw(std)} \frac{nH(std)}{nH(x)} \frac{m(std)}{p(x)} \frac{A(x)}{A(std)}$$

$m(x)$ and $m(std)$ are the masses (weights) in g, $MW(x)$ and $MW(std)$ are the molecular weights in g/mol, $P(x)$ and $P(std)$ are the purities, $nH(x)$ and $nH(std)$ are the number of protons generating the selected signals for integration, $A(x)$ and $A(std)$ are the areas for the selected peaks of the analyte and the internal standard.

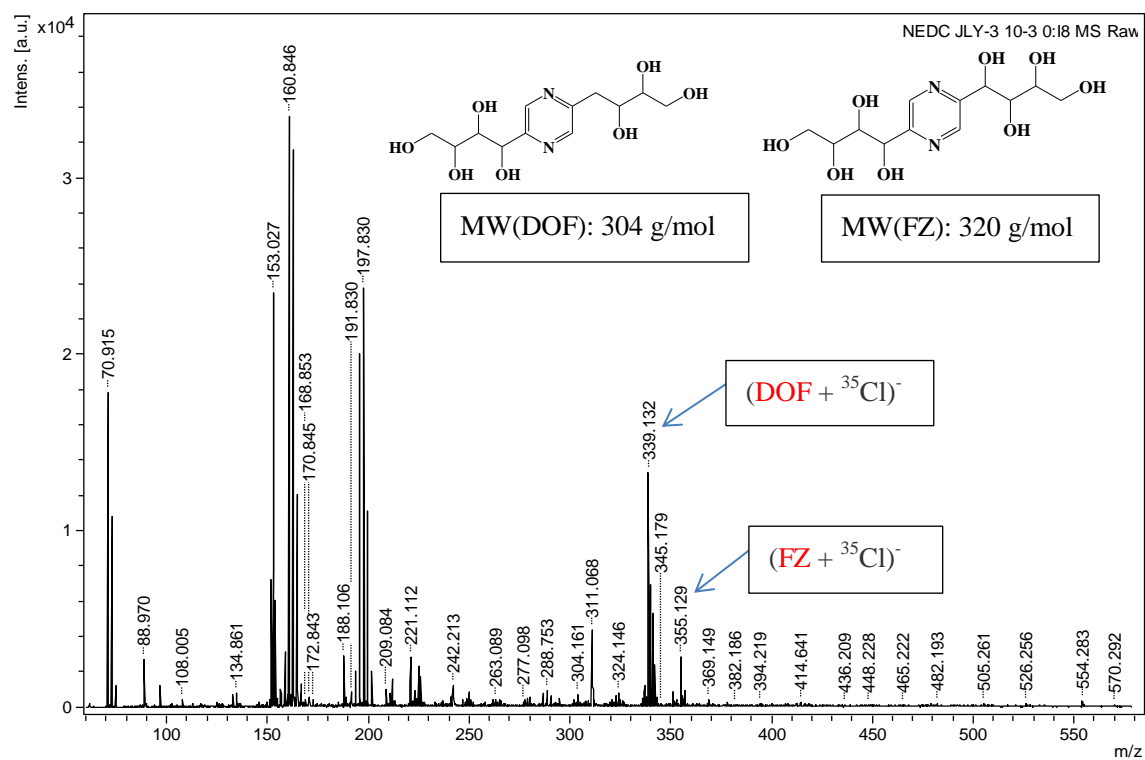


Fig. S2. Negative ion MALDI-TOF mass spectra of DOF and FZ using NEDC as a matrix.

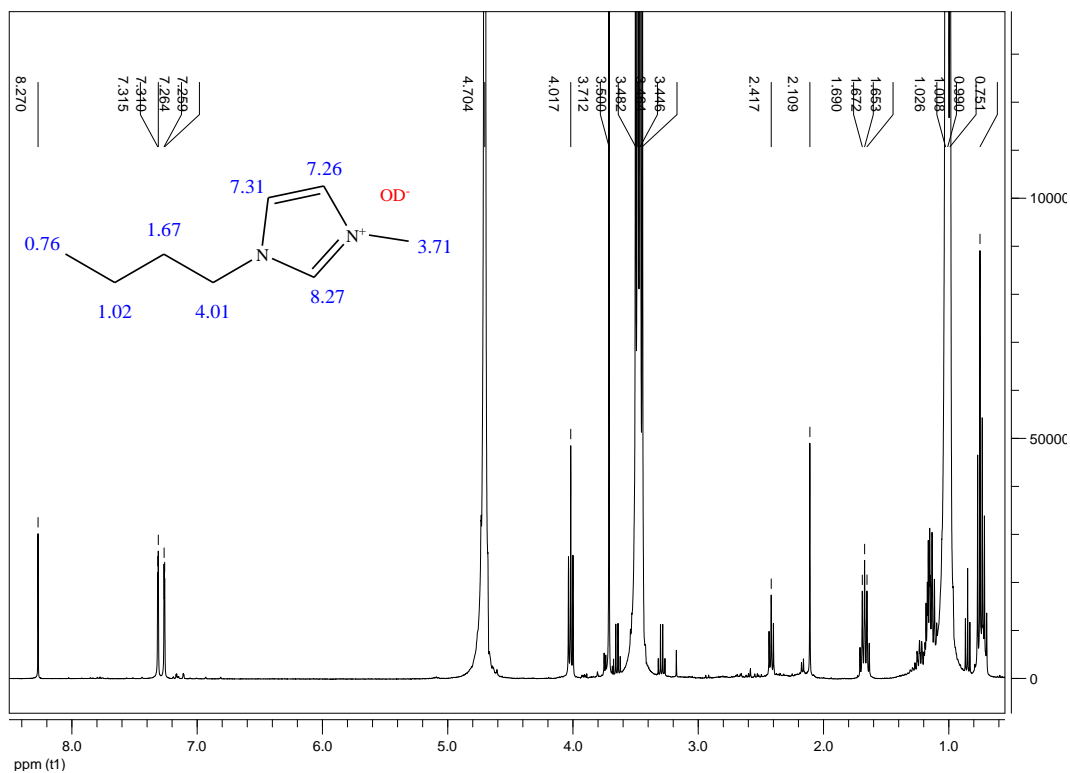
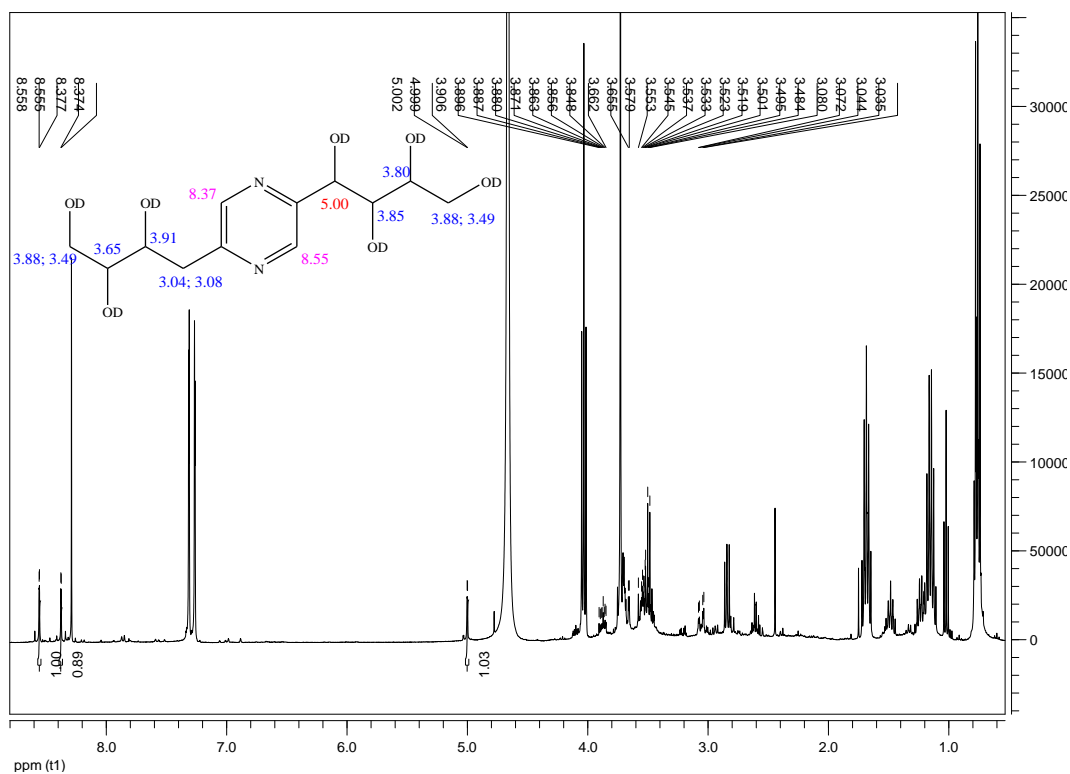


Fig. S3. ^1H NMR spectra of products and alcoholic solution of 12% alkaline [BMIM]OH (400.13MHz, D_2O , DSS). The spectra were recorded at ambient

temperature. Chemical shifts are given in ppm. DSS was used as the internal reference.

D₂O was used for the deuterium lock.

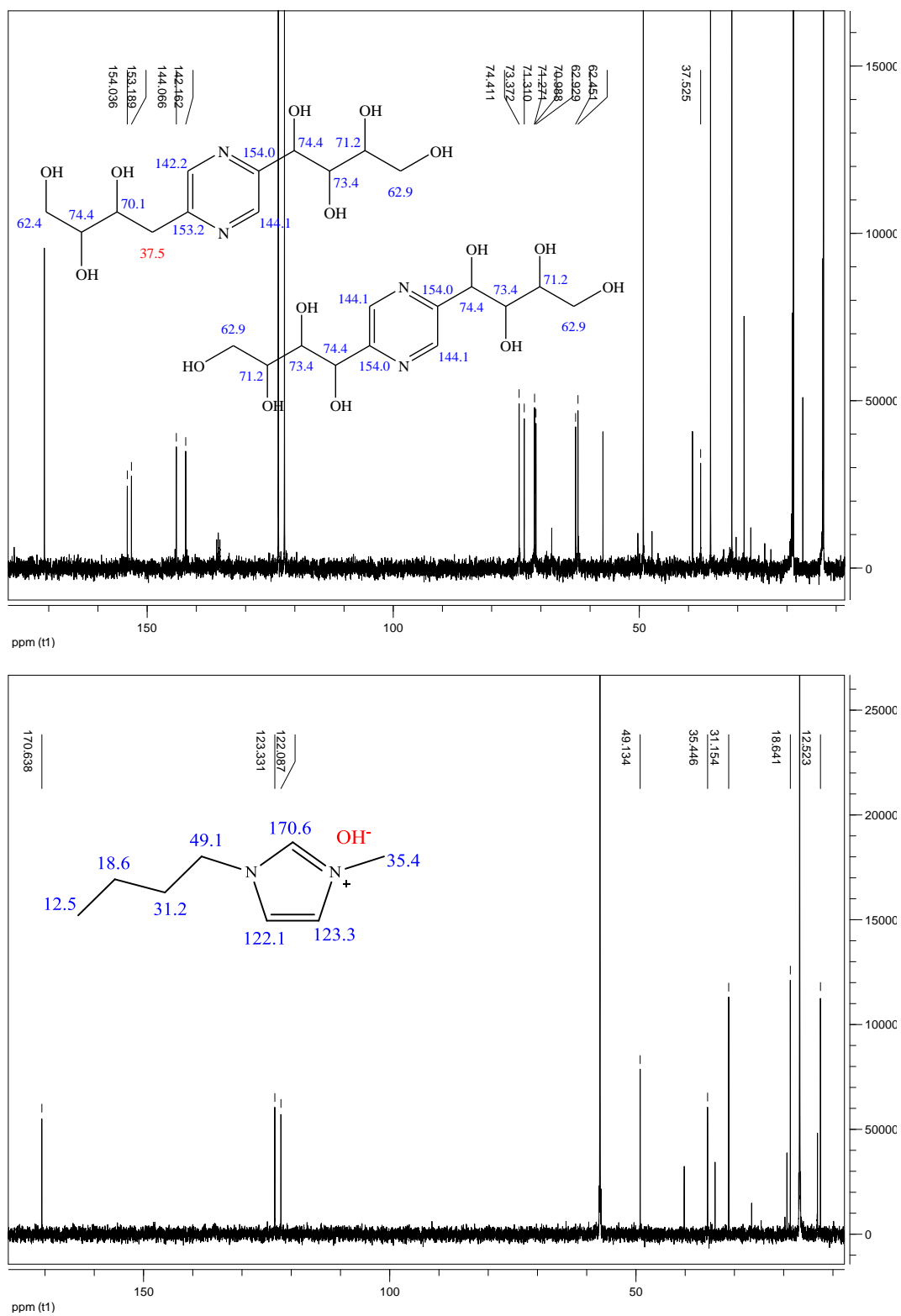
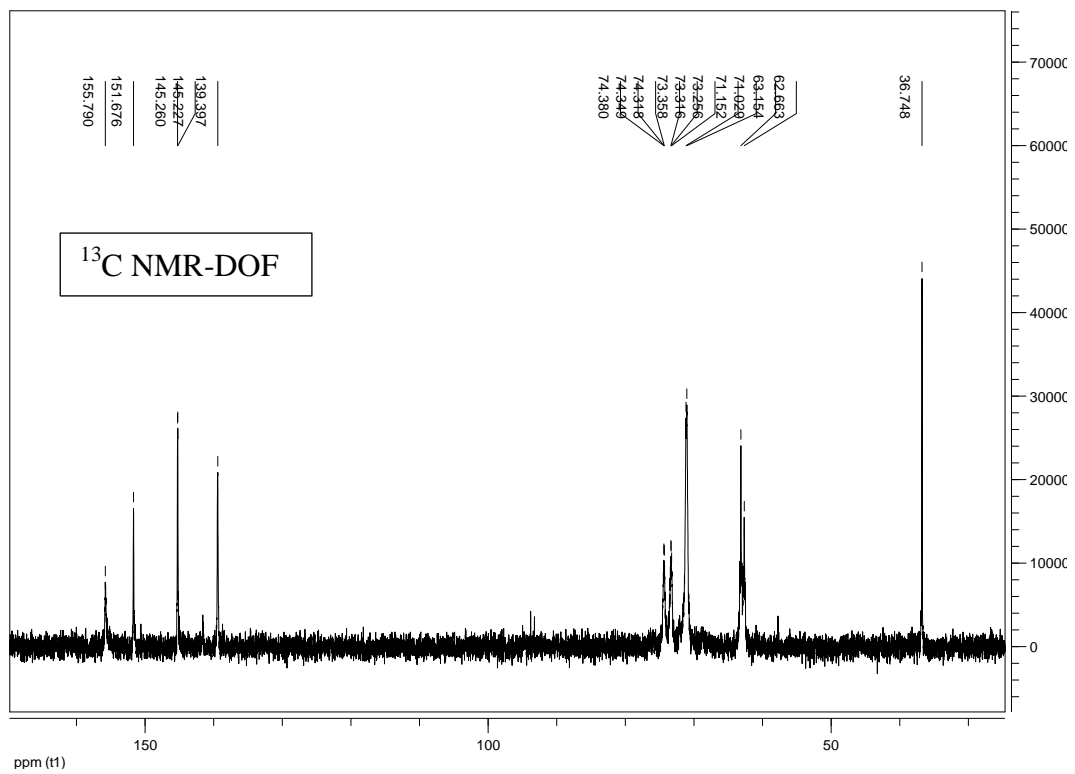
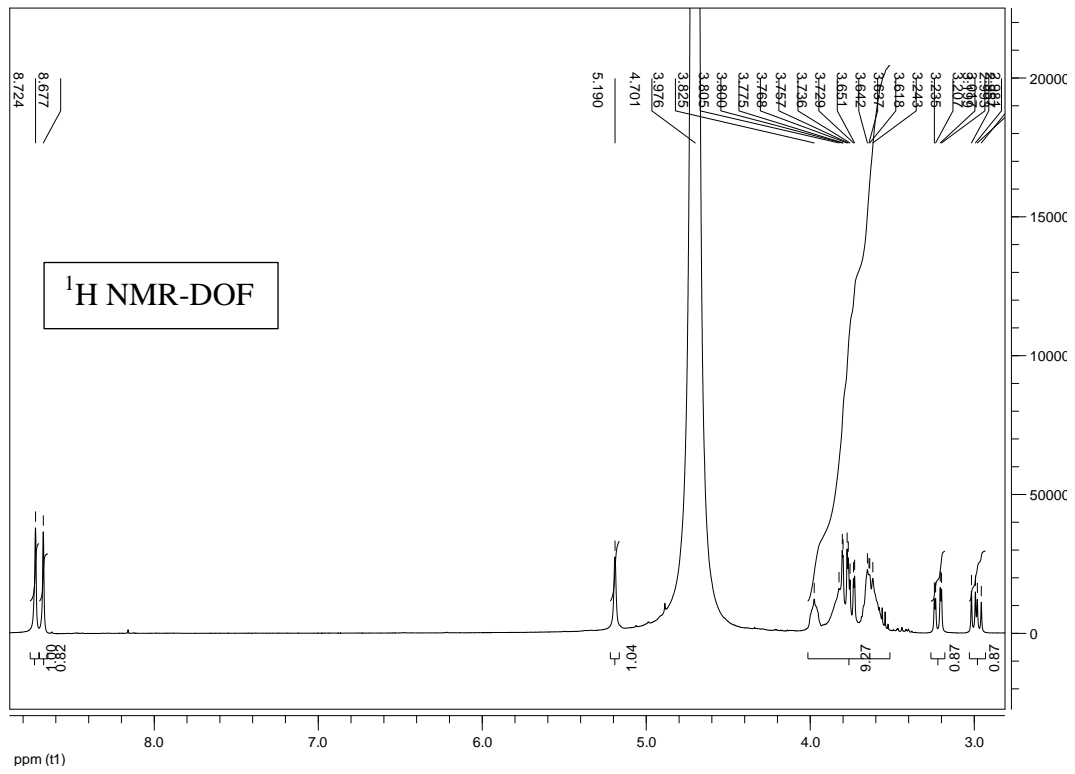


Fig. S4. ¹³C NMR spectra of products and alcoholic solution of 12% alkaline [BMIM]OH (100.61MHz, D₂O, DSS).



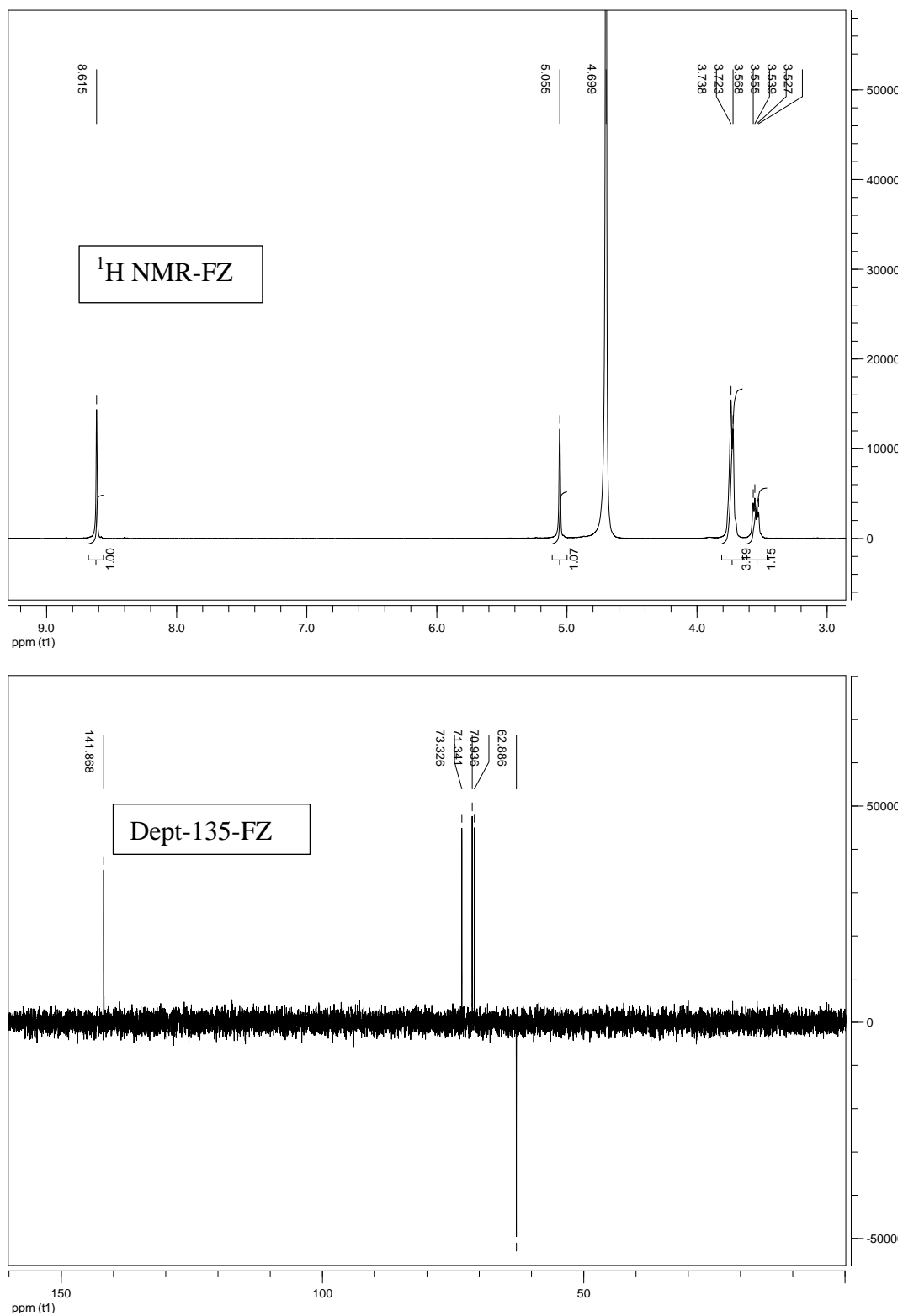


Fig. S5. ^1H NMR and ^{13}C NMR spectra of authentic sample (deoxyfructosazine, DOF and fructosazine, FZ).

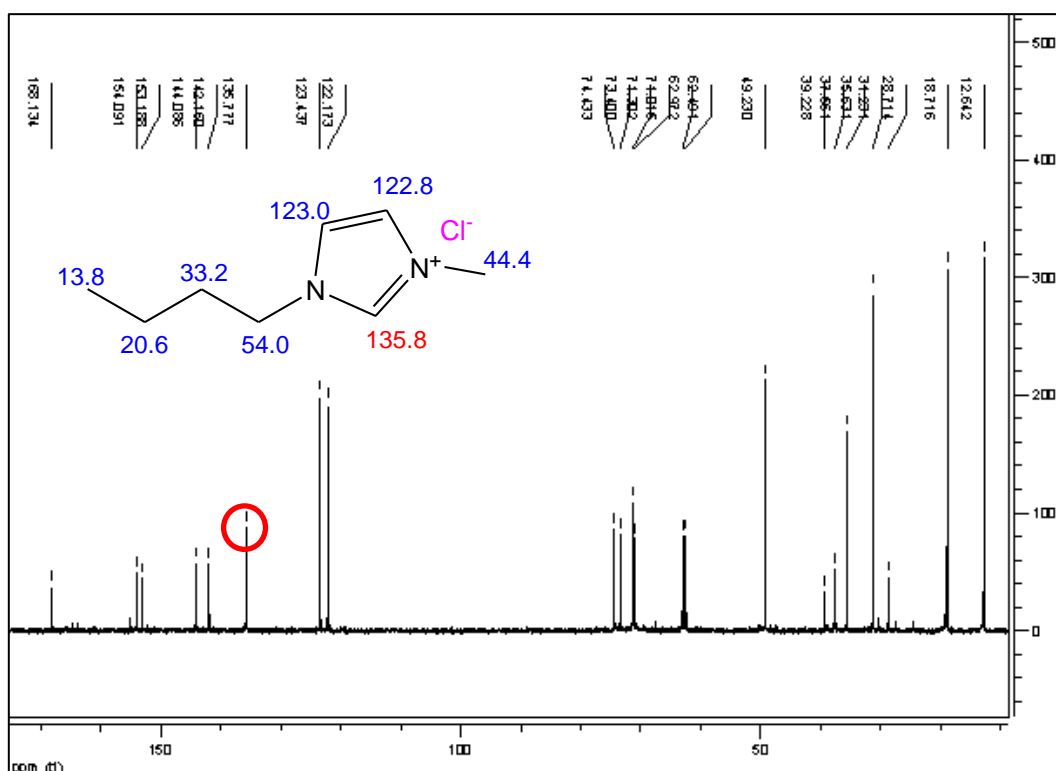
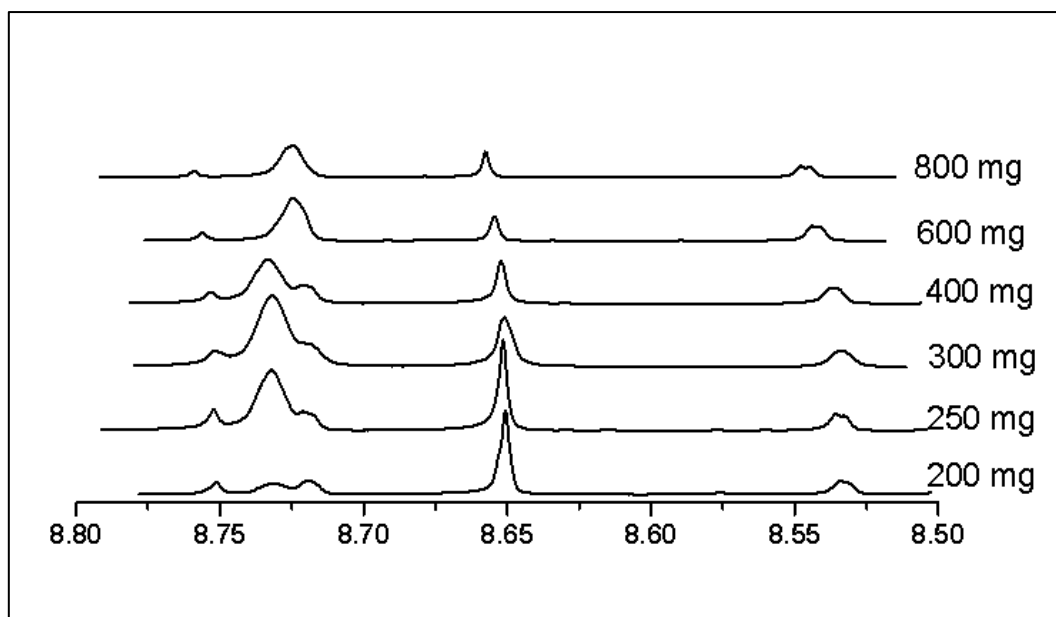


Fig. S6. ^1H NMR and ^{13}C NMR spectra of [BMIM]Cl from ion-exchange reaction between hydrochloride and the hydroxyl group of the basic ionic liquid [BMIM]OH.

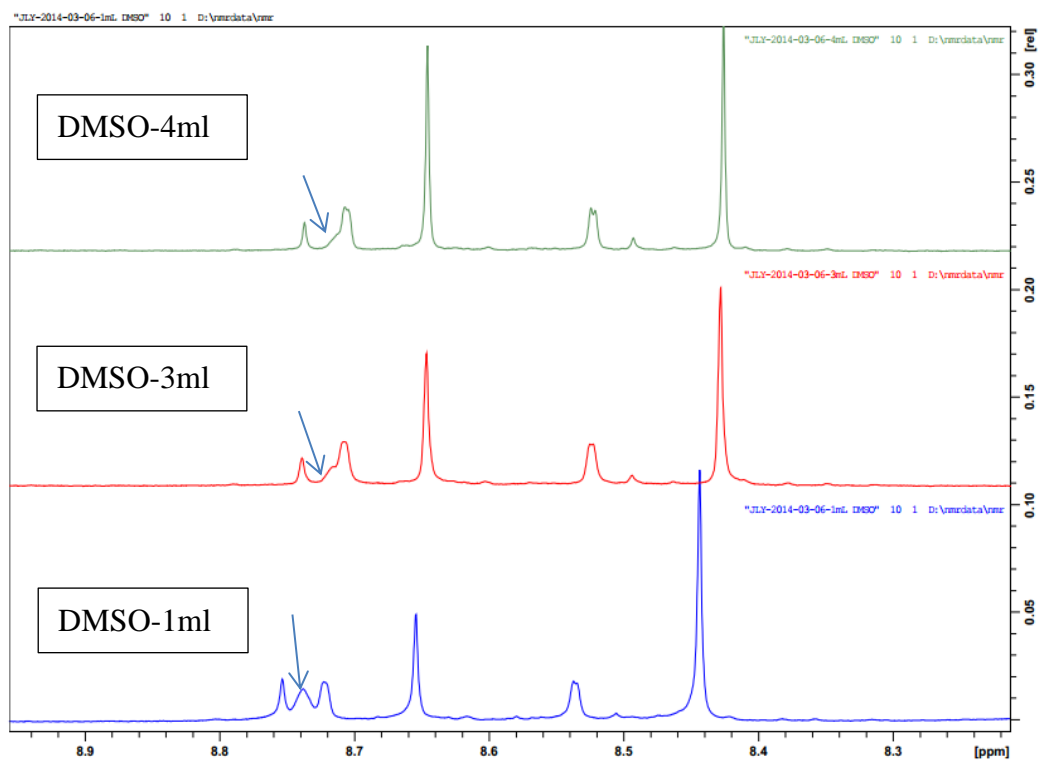


Fig. S7. ^1H NMR spectra of products with increasing amounts of DMSO in this reaction system.