Supplementary information for:

Effects of Cu addition on band gap energy, density of state effective

mass and charge transport properties in Bi₂Te₃ composites

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Figure S1. XPS Cu $2p_{3/2}$ spectrum of Cu_{0.04}Bi₂Te₃ composite.



Although the XPS spectrum was not very sharp due to the small content of Cu in the composite, the binding energy exhibited a maximum peak around 933 eV and it was quite consistent to the reported values. In literature, the binding energy of Cu has been reported to be ~ 933 eV regardless of its oxidation state (Cu, Cu₂Te, CuTe).^{S1~S3} Especially, Teeter reported that the binding energies for Cu⁰, Cu¹⁺ and Cu²⁺ were equal within about 0.05 eV.^{S1} Although the identification of the oxidation state of the doped Cu was not possible due to the similarity in their binding energies, our result clearly reveals the existence of Cu-Te bonding in the Cu_xBi₂Te₃ composite.

References

- S1. G. Teeter, Appl. Phys. Lett., 2007, 102, 034504.
- S2. E.P. Domashevskaya et al., J. Electron Spectrosc. Relat. Phenom., 2001, 114-116, 901.
- S3. J. Carmona-Rodríguez et al., J. Mater. Chem., 2011, 21, 13001.