

Supporting Information

A NbO Type Microporous Metal-Organic Frameworks Constructed from Naphthalene Derived Ligand for CH₄ and C₂H₂ storage at Room Temperature

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Table S1. Crystallographic Data Collection and Refinement Results for **ZJU-7**

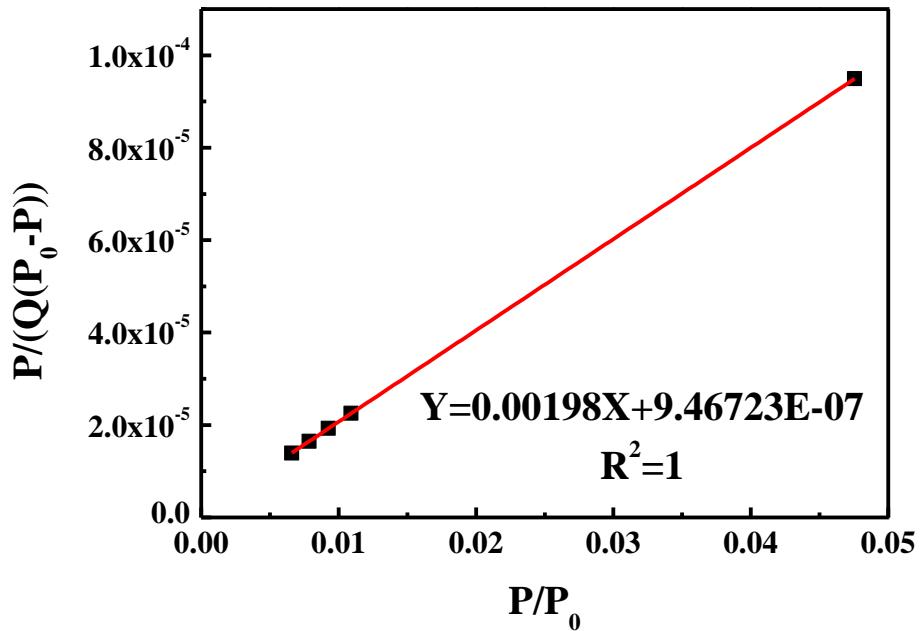
ZJU-7	
Overall chemical formula	C ₃₆ H ₄₄ Cu ₂ N ₄ O ₁₆ [Cu ₂ (C ₂₆ H ₁₂ O ₈)(H ₂ O) ₂](DMF) ₂ (MeCN) ₂ (H ₂ O) ₄
Overall formula weight	915.83
Framework chemical formula	C ₂₆ H ₁₆ Cu ₂ O ₁₀
Framework formula weight	613.48
Temperature(K)	293(2)
Wavelength(Å)	1.54178
Crystal system	Trigonal
Space group	R-3m
<i>a</i> (Å)	18.6843(7)
<i>b</i> (Å)	18.6843(7)
<i>c</i> (Å)	38.6052(12)
<i>V</i> (Å ³)	11671.6(7)
<i>Z</i>	9
Density (calculated g/cm ³)	1.173
Absorbance coefficient(mm ⁻¹)	0.848
No. of reflections measured	39425
No. of independent reflections	2626
<i>F</i> (000)	4266
Crystal size(mm ³)	0.46×0.43×0.26
R(int)	0.0489
Goodness of fit on <i>F</i> ₂	1.057
R1,wR2[<i>I</i> >2σ(<i>I</i>)] ^a	0.0489, 0.0446
R1,wR2(all data) ^a	0.1380, 0.0586
Largest difference peak and hole(e/Å ³)	0.897,-0.381

$$^a\mathbf{R1} = \sum(|\mathbf{F}_o| - |\mathbf{F}_c|) / \sum|\mathbf{F}_o| ; \mathbf{wR2} = \left[\frac{\sum w(|\mathbf{F}_o| - |\mathbf{F}_c|)^2}{\sum w\mathbf{F}_o^2} \right]^{1/2}$$

Table S2. C₂H₂ adsorption on various porous MOFs at room temperature and atmospheric pressure.

Material	S_{BET} [m ² g ⁻¹]	V_p^a [cm ³ g ⁻¹]	D_c^b [g cm ⁻³]	C ₂ H ₂ uptaked [cm ³ g ⁻¹] ([cm ³ cm ⁻¹])	Ref.
ZJU-7a	2198	0.8945	0.750	180 (135)	This work
ZJU-5a	2823	1.074	0.679	193 (131)	S1
MOF-505	1139	0.67	0.767	148 (137)	S2
PCN-16	2273	1.06	0.724	176 (126)	S3
NOTT-101	2805	1.080	0.6838	184 (126)	S4
CoMOF-74	1018	-	1.169	197 (230)	S5
MnMOF-74	695	-	1.085	168 (182)	S5
MgMOF-74	927	-	0.909	184 (167)	S5
ZnMOF-74	747	-	1.231	122 (150)	S5
FeMOF-74	1350	-	1.126	156 (176)	S6
HKUST-1	1401	0.76	0.879	201 (177)	S1

^a Pore volume. ^b Crystal density calculated from the single-crystal structure without molecules and terminal ligands. ^c



$$S_{\text{BET}} = 1 / (0.00198 + 9.46723E-07) / 22414 \times 6.023 \times 10^{23} \times 0.162 \times 10^{-18} = 2198 \text{ m}^2/\text{g}$$

Figure S1. The BET surface area of **ZJU-7a** obtained from N₂ adsorption isotherm at 77 K

References

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