

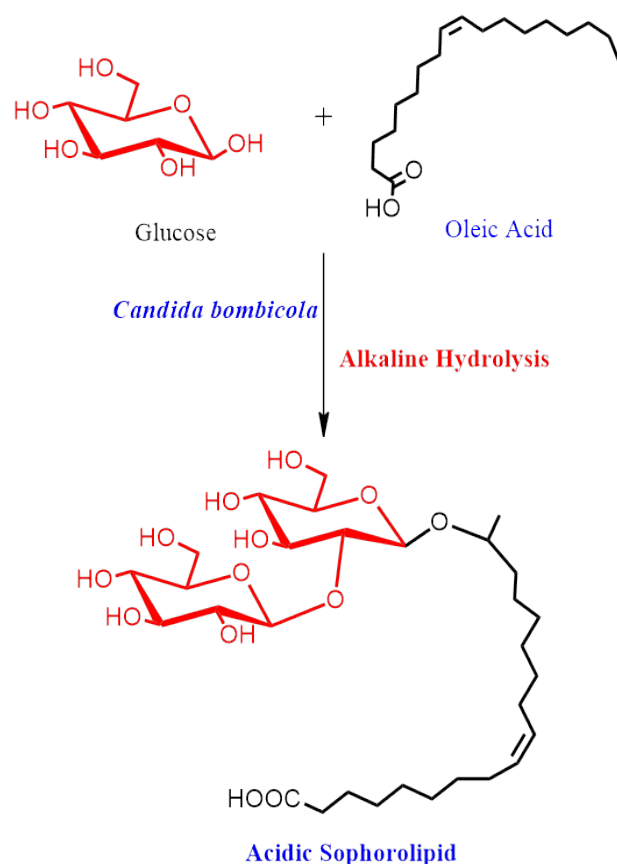
Electronic Supplementary Information (ESI)

From Micron to Nano-Curcumin by sophorolipid co-processing: Highly enhanced bioavailability, fluorescence, and anti-cancer efficacy

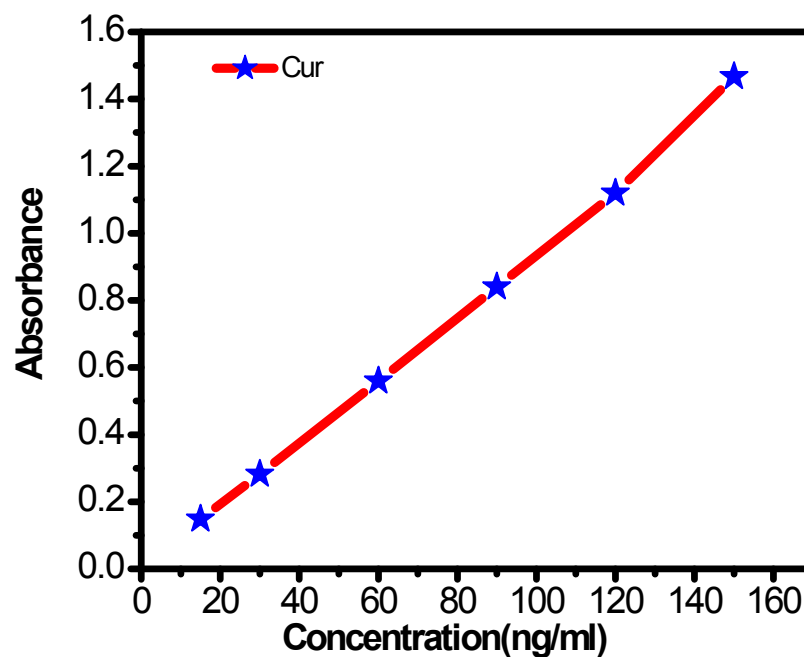
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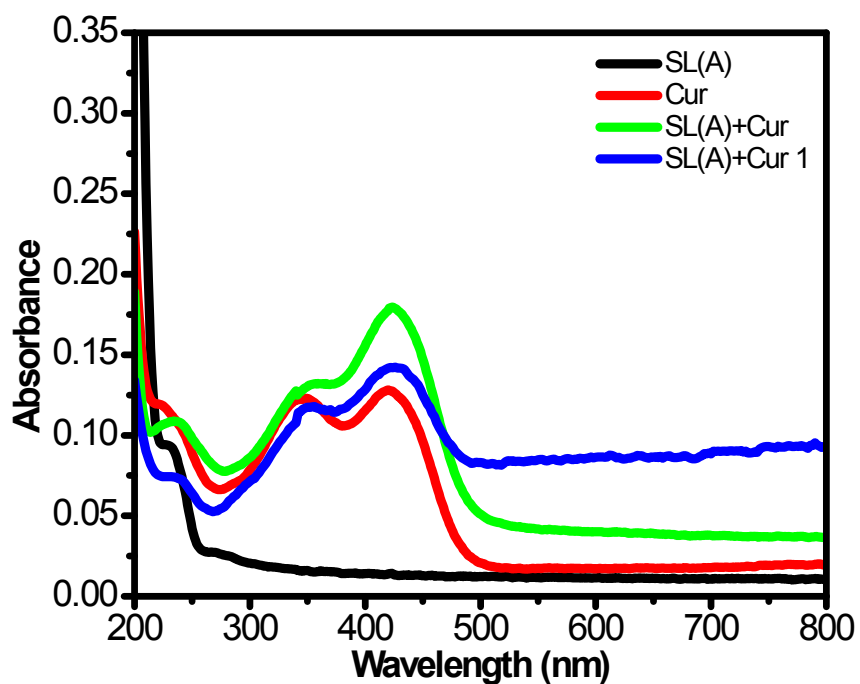
1. Sophorolipid (SL(A)) synthesis



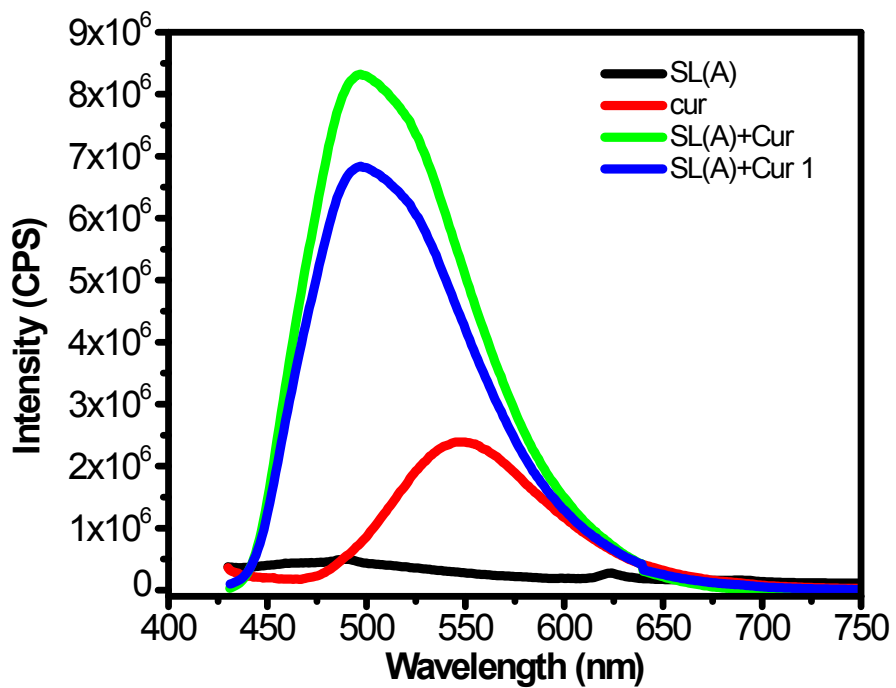
ESI Fig. S1 Basic structure of SL produced by yeast *C. bombicola* after purification process



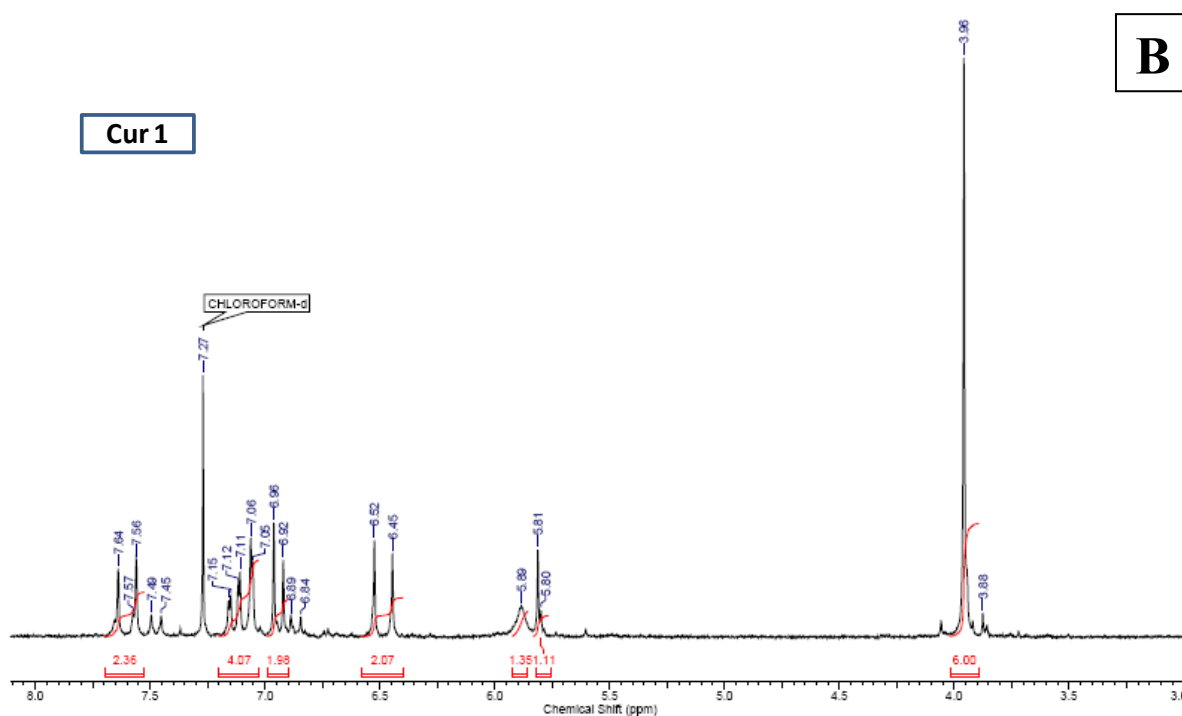
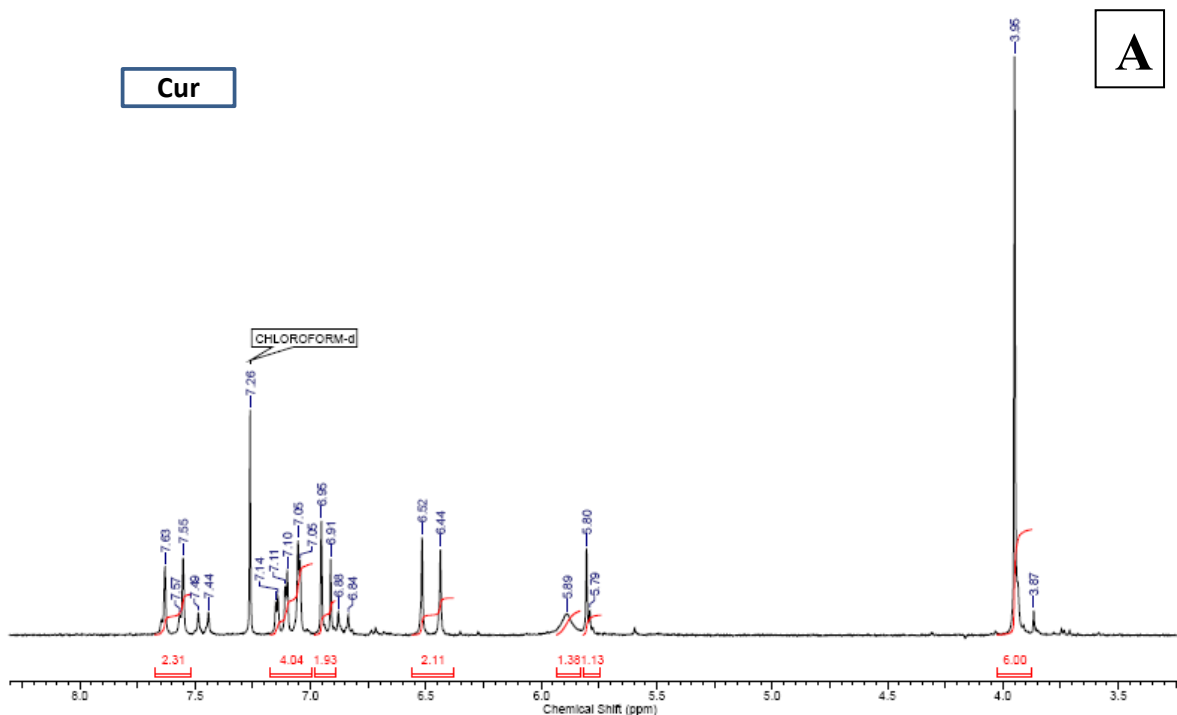
ESI Fig. S2 The curcumin concentration calculated using standard solutions with different amounts of pristine curcumin dissolved in ethanol. Curcumin was dissolved in ethanol and measured for absorbance at 420 nm using a UV-Vis spectrophotometer to determine its concentration in the SL(A)+Cur nano-complex.



ESI Fig. S3 UV-Visible spectra of SL (A), Curcumin and SL(A)+ Cur solutions. SL(A) solution shows absorbance at $\lambda=234$ nm and Curcumin solution at $\lambda=344$ and 420 nm while SL(A) + Cur showing increase absorption at one peak at 420 nm. Blue line shows the stability of nano complex after four months.



ESI Fig. S4 Photoluminescence study of SL(A). Curcumin and SL(A)+ Cur solutions. The SL(A) solution shows no PL. Curcumin solution shows PL at 550 nm while SL(A) + Cur showing very strong Photoluminescence at 500 nm; Blue line shows the stability of nano complex after four months.



ESI Fig. S5 NMR study of **(A)** Curcumin and **(B)** curcumin in SL(A)+ Cur solutions after four month stability.

Sample preparation for NMR

Chloroform was added in SL(A)+Cur aqueous solution to extract curcumin from nano-complex. The sample was rotator evaporator and vacuum dried for 60 min to remove chloroform. Curcumin crystals, as a control, were dissolved in chloroform and vacuum-dried

as above the encapsulated curcumin. The obtained curcumin powders were dissolved in deuterated chloroform for ^1H NMR study.