

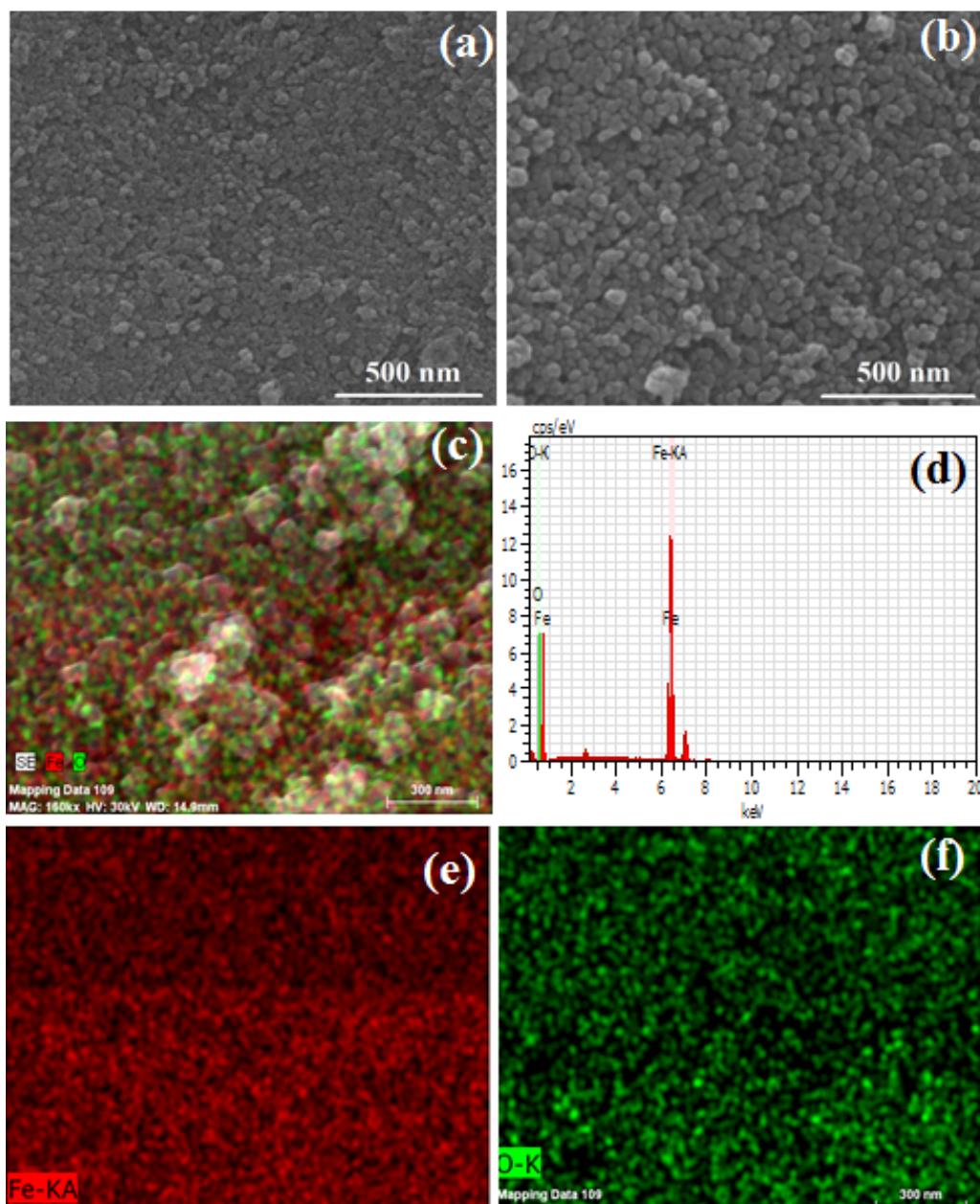
Supporting Information

Facile *in-situ* growth of Fe_3O_4 nanoparticles on hydroxyapatite nanorods for pH dependent adsorption and controlled release of proteins

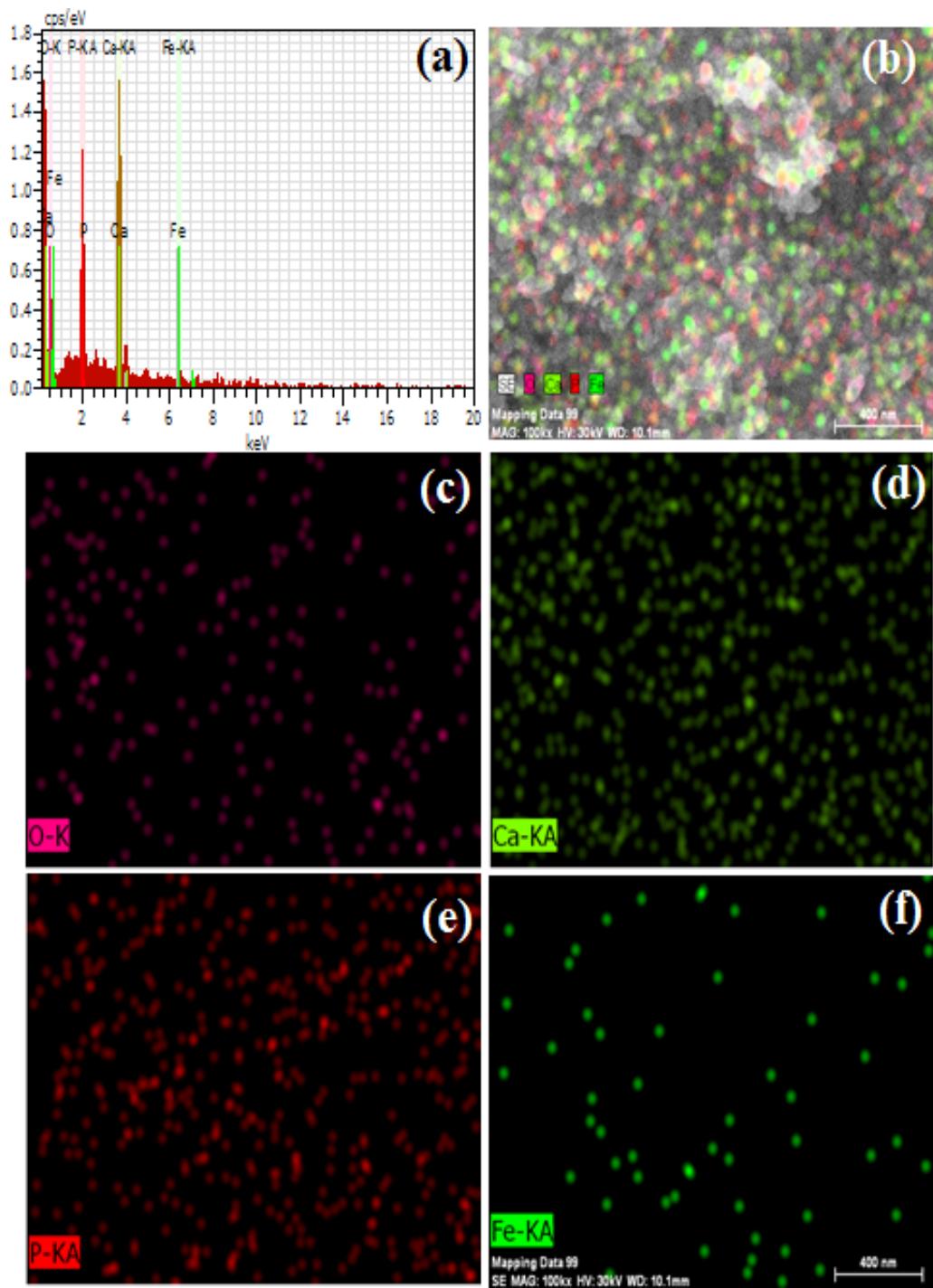
G. Bharath,^a D. Prabhu,^b D. Mangalaraj,^a C. Viswanathan,^a N. Ponpandian*^a

^aDepartment of Nanoscience and Technology, Bharathiar University, Coimbatore 641 046, India

^bInternational Advanced Research Centre for Powder Metallurgy and New Materials (ARCI), Balapur PO, Hyderabad 500 005, India



SI Fig. 1. FESEM images of two different molar concentrations of ($\text{Fe}^{2+} + \text{Fe}^{3+} = 0.5$ and 1.0 mM) Fe_3O_4 nanoparticles synthesized by coprecipitation technique. (a) 0.5 mM and (b) 1.0 mM of Fe_3O_4 . (c-f) EDX and elemental distributions of Fe and O species.



SI Fig. 2 EDS spectrum and elemental mapping of $\text{Fe}_3\text{O}_4/\text{HAp-1}$ (0.5 mM Fe_3O_4), (a) EDX spectrum and (b-f) distributions of elements (Ca, P, O and Fe) over the nanocomposites. (f) Higher Fe ions distributions on $\text{Fe}_3\text{O}_4/\text{HAp-2}$ (1.0 mM Fe_3O_4) nanocomposite.