

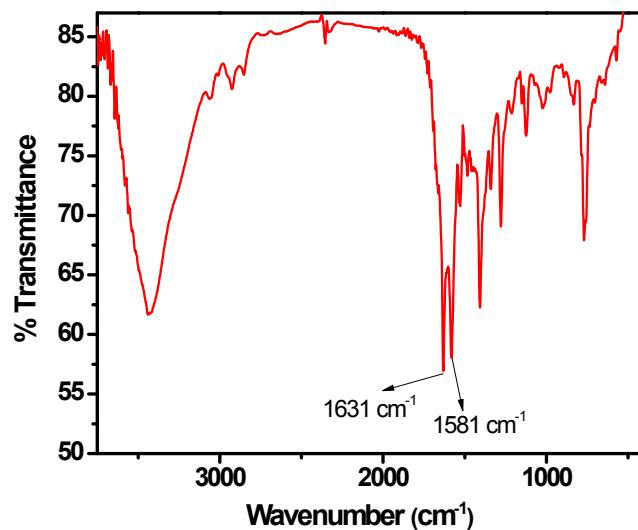
## Electronic Supplementary Information (ESI)

### A novel probe for selective colorimetric sensing of Fe(II) and Fe(III) and specific fluorometric sensing of Fe(III): DFT calculation and logic gate application

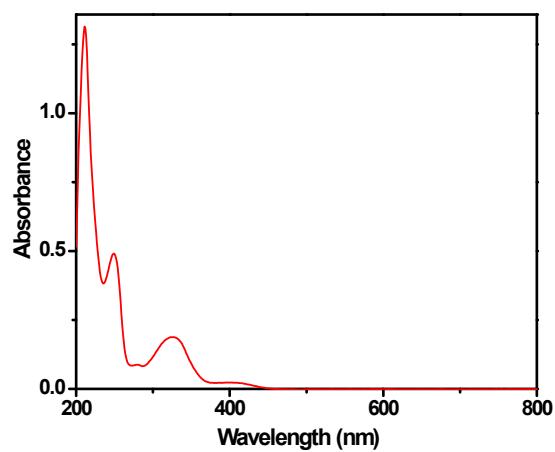
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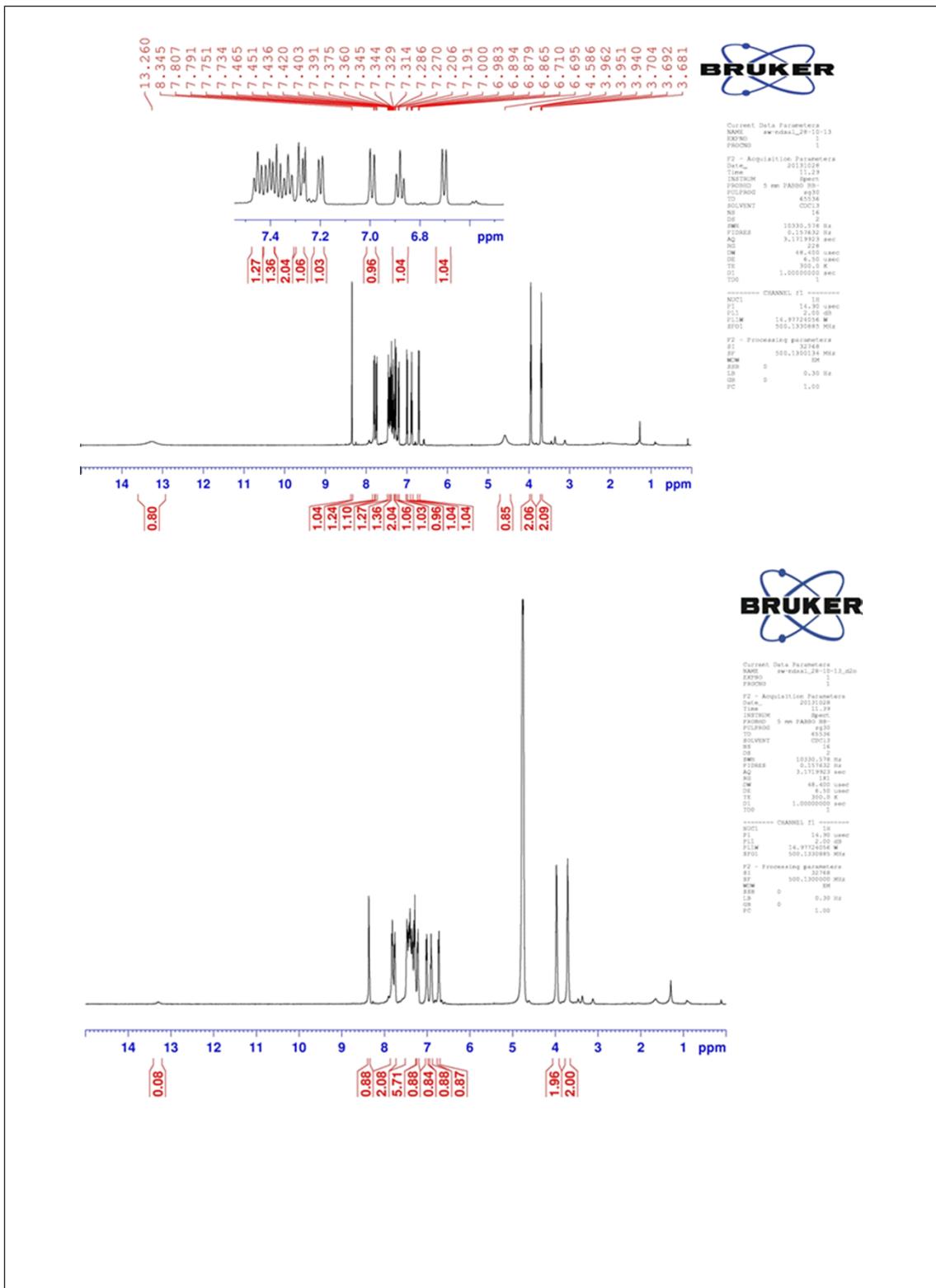
E-mail: [ghoshfcy@iitr.ernet.in](mailto:ghoshfcy@iitr.ernet.in)



**Fig. S1** IR spectra of probe 1.

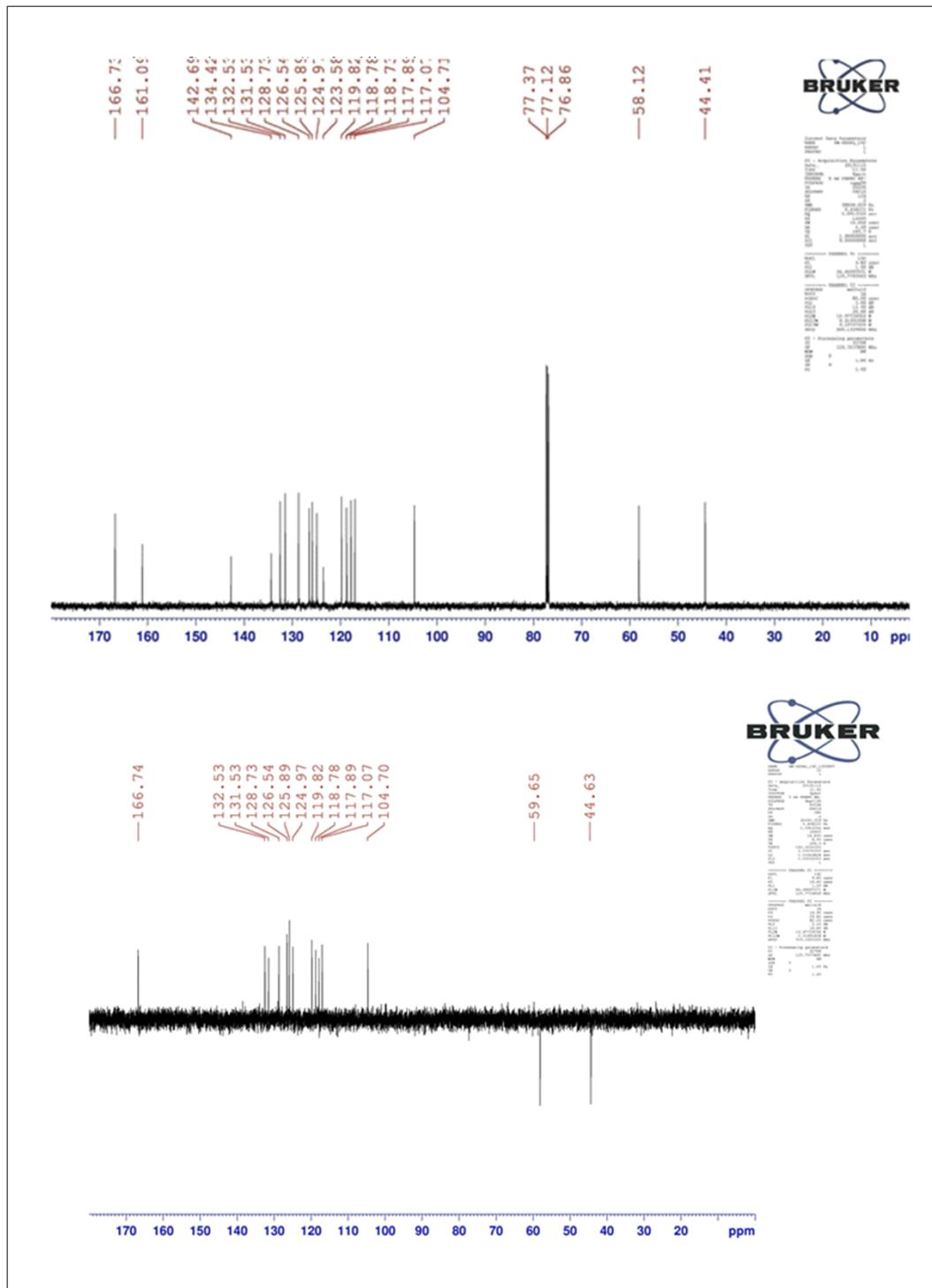


**Fig. S2** UV-vis absorption spectra of probe **1** (20  $\mu\text{M}$ ) in methanol.



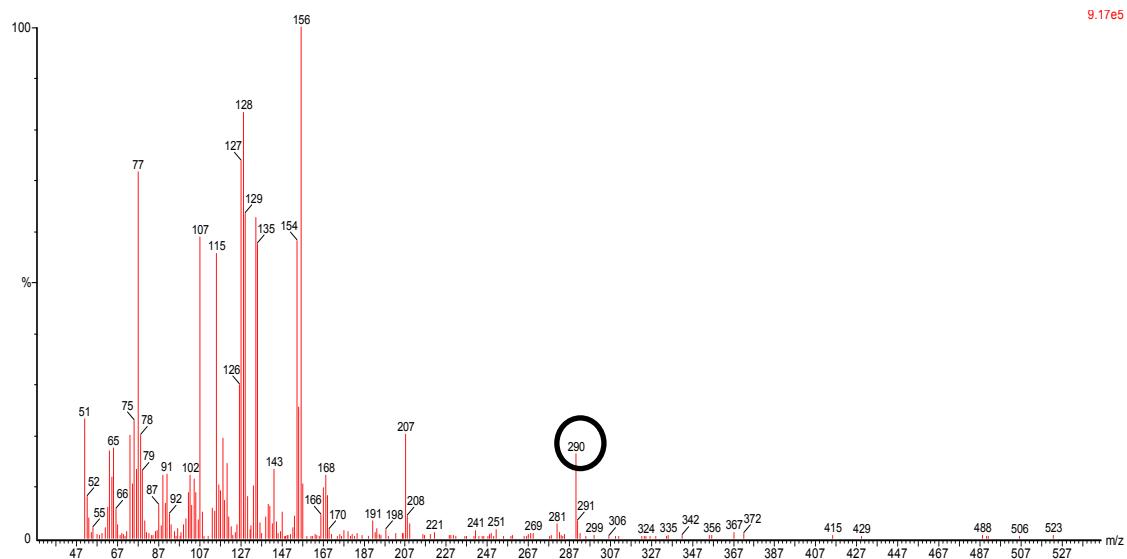
**Fig. S3 (A)**  $^1\text{H}$ -NMR spectrum of probe **1** in  $\text{CDCl}_3$ . **(B)**  $^1\text{H}$ -NMR  $\text{D}_2\text{O}$  exchange spectrum of

probe **1** in  $\text{CDCl}_3$  at room temperature.

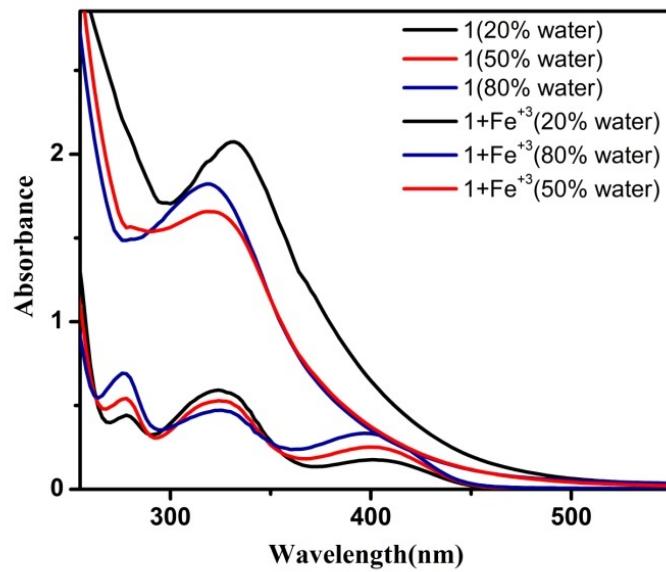


**Fig. S4** (A)  $^{13}\text{C}$ -NMR spectrum (B)  $^{13}\text{C}$ -NMR 135 dept of probe **1** in  $\text{CDCl}_3$  at room

temperature.

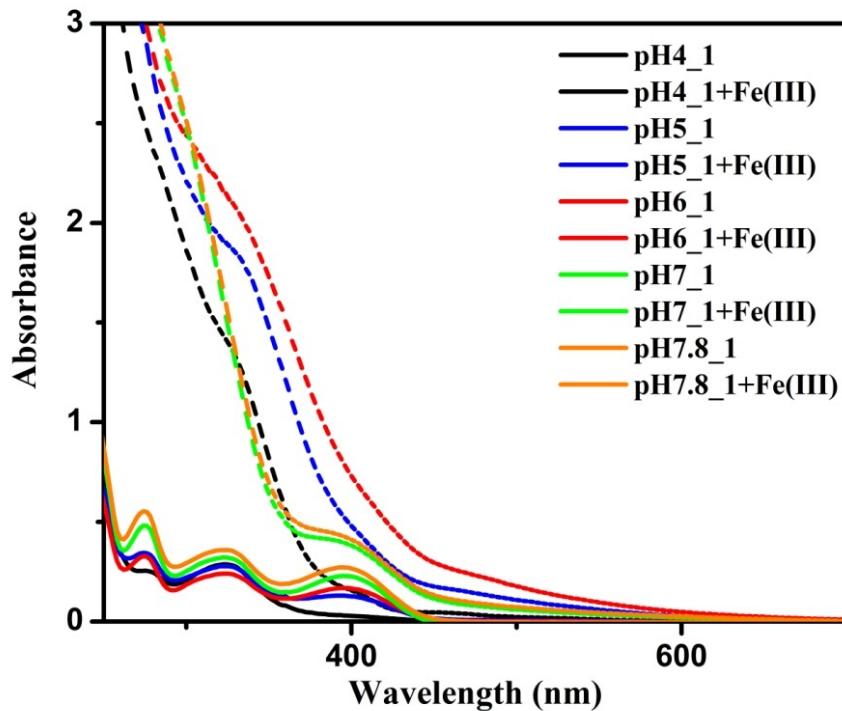


**Fig. S5** GC-MS spectrum of probe **1**.

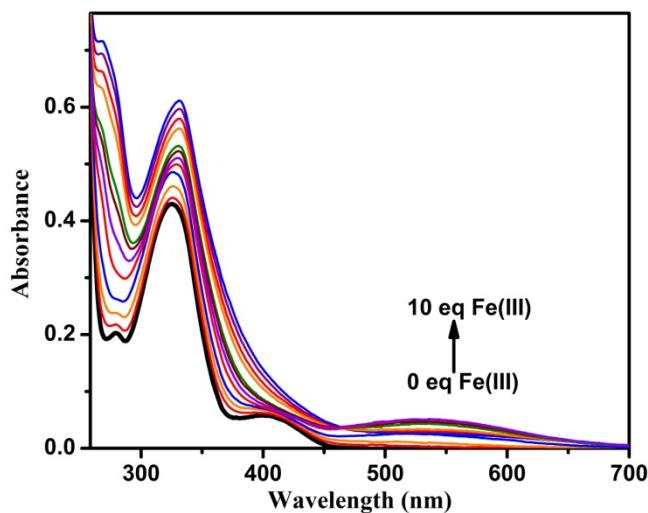


**Fig. S6** Changes observed during absorption spectral studies of **1** (20  $\mu$ M) and **1+ Fe(III)**

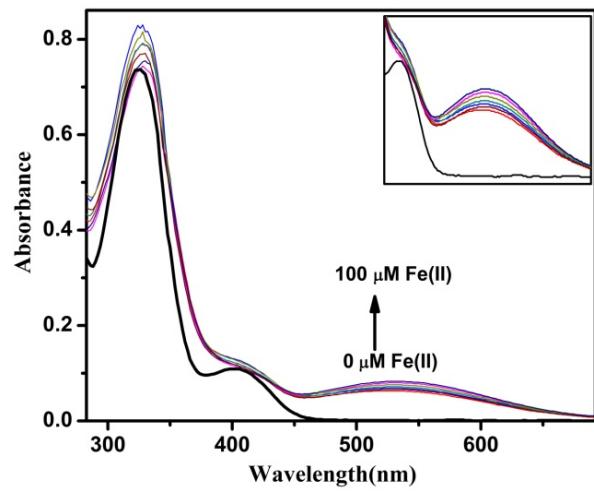
upon addition of different amounts of water in methanol.



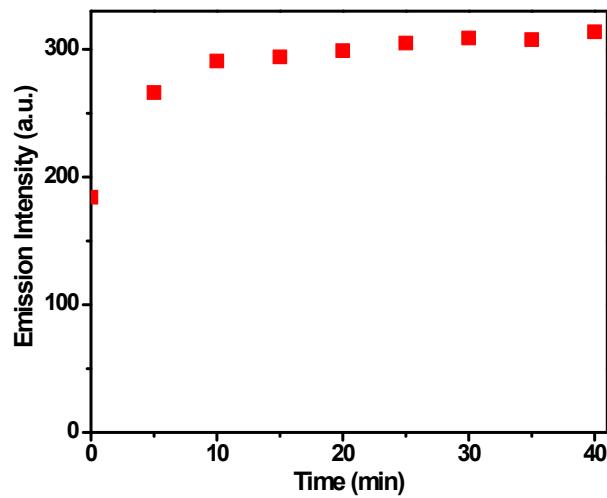
**Fig. S7** Effect of pH during absorption spectral studies of 1(20  $\mu\text{M}$ ) and 1+ Fe(III) in water –methanol(4:1).



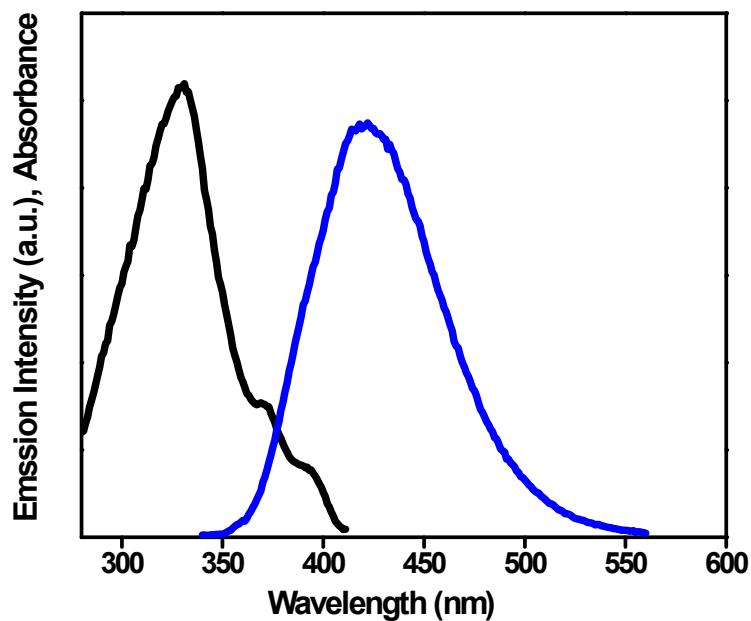
**Fig. S8** Absorption spectra of **1** (5  $\mu\text{M}$ ) upon addition of different amounts of Fe(III) in methanol.



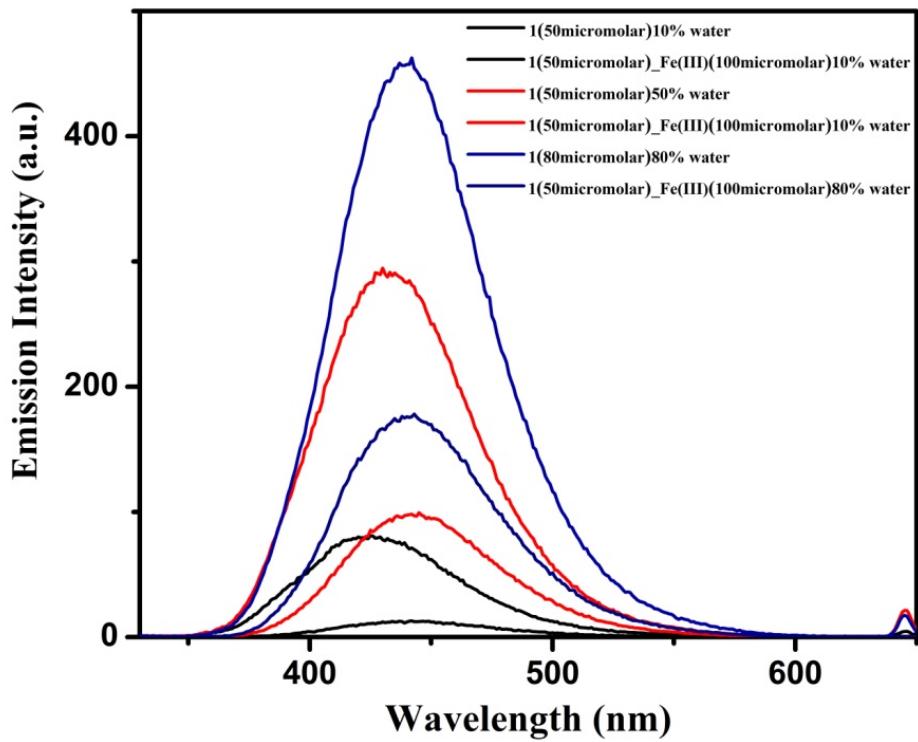
**Fig. S9** Absorption spectra of **1** ( $5\mu\text{M}$ ) upon addition of different amounts of Fe(II) in methanol.



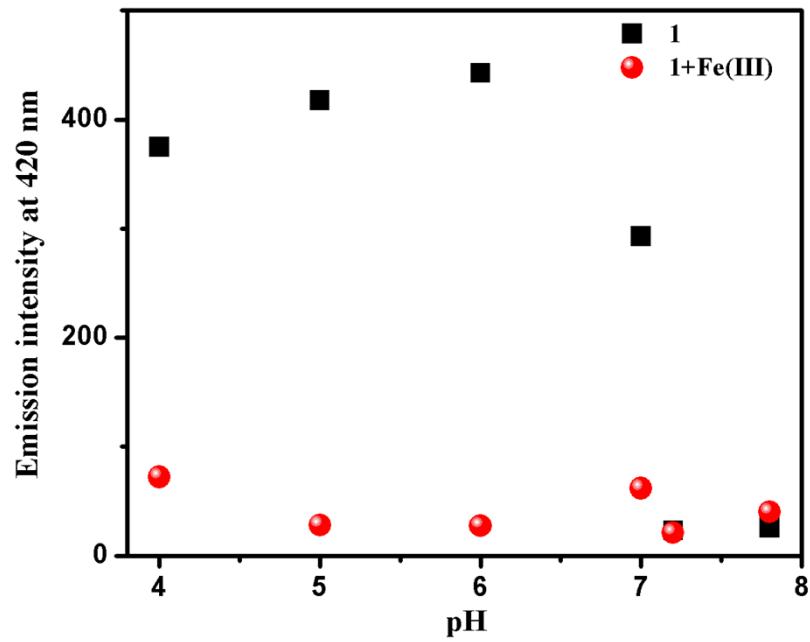
**Fig. S10** Plot of emission intensity **1** ( $5 \times 10^{-5}\text{M}$ ) from 0 min to 40 min at  $\lambda_{\text{ex}}$  320 nm.



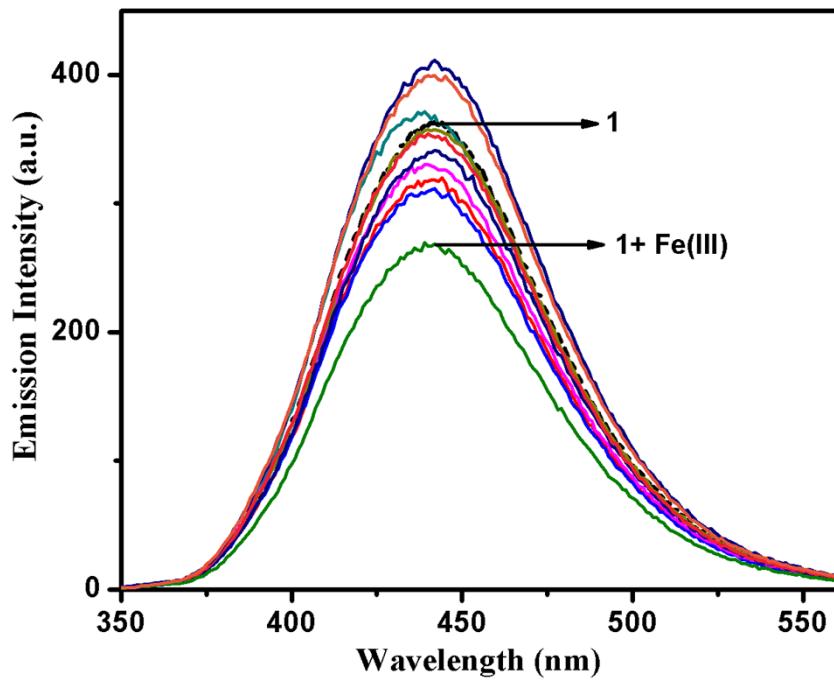
**Fig. S11** Excitation spectra and emission spectra of probe 1 ( $50 \mu\text{M}$ ) in methanol solution. Black line is the excitation spectra, and the blue line is the emission spectra. The maximum excitation and emission are at 330 nm and 420 nm, respectively.



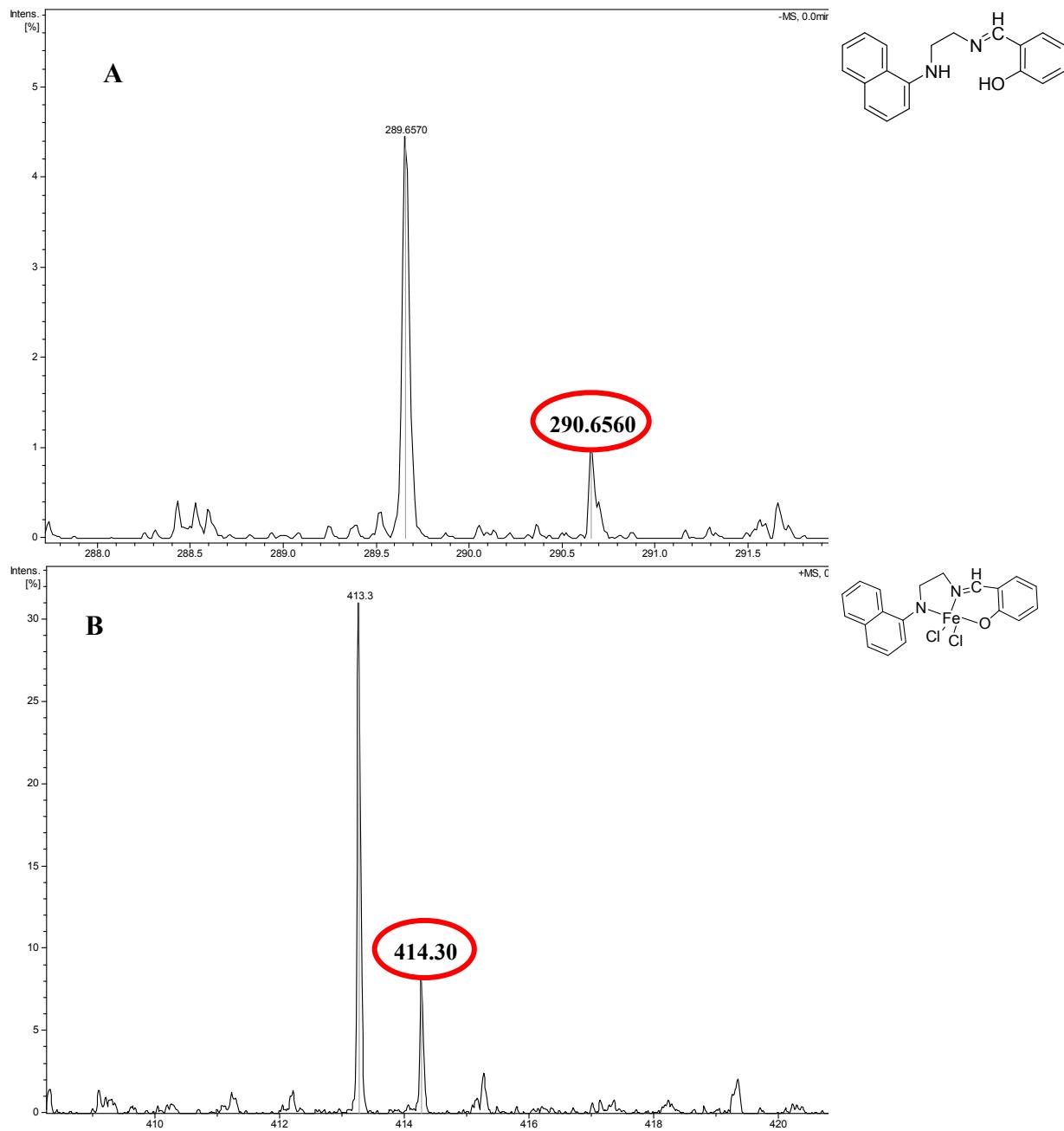
**Fig. S12** Effect of pH during absorption spectral studies of **1** (20  $\mu$ M) and **1** + Fe(III) in water-methanol(4:1).



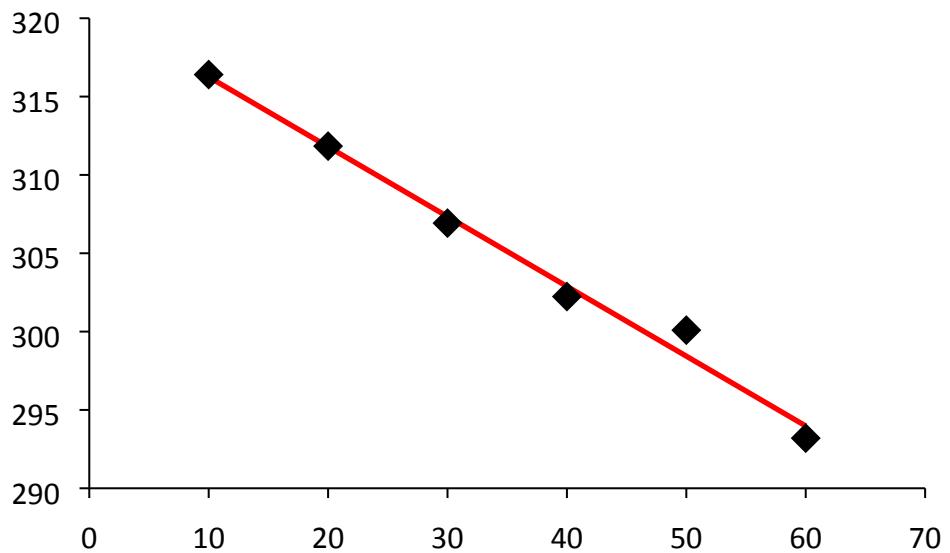
**Fig. S13** Effect of pH during emission spectral studies of **1** (20  $\mu$ M) and **1** + Fe(III) in water – methanol (4:1).



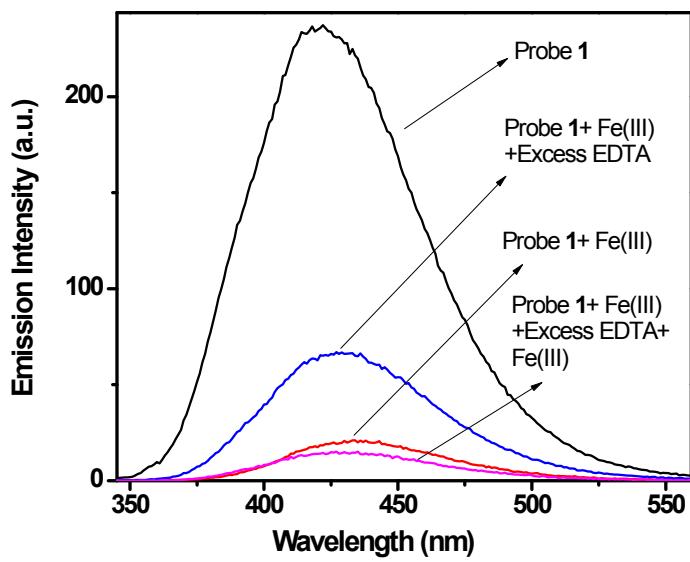
**Fig. S14** Emission spectral studies of **1** (20  $\mu$ M) and other representative metals in water – methanol (4:1).



**Fig. S15** ESI mass spectra of (A) **1** and (B) **1+Fe** (III) 10 eq in methanol-acetonitrile mixture.



**Fig. S16** Limit of detection calculated for a linear range of 10-60  $\mu\text{M}$  at wavelength 426 nm.



**Fig. S17** Reversibility experiment from fluorescence spectra of probe **1** in the presence of Fe(III) with excess of EDTA methanol.