

Supporting Information

Metal-free *n*-Et₄NBr-catalyzed radical cyclization of disulfides and alkynes leading to benzothiophenes under mild conditions

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General experimental procedures

All reagents and solvents were obtained from commercial suppliers and used without further purification. Mass analyses and HRMS were obtained on a Finnigan-LCQDECA mass spectrometer. Flash chromatography was performed on silica gel (200 ~ 300 mesh). ^1H and ^{13}C NMR data were recorded at 400 and 100 MHz on a BRUKER 400 spectrometer. Chemical shifts (δ) are expressed in parts per million (ppm) coupling constants (J) are in Hz. Proton and carbon magnetic resonance spectra (^1H NMR and ^{13}C NMR) were recorded using tetramethylsilane (TMS) in the solvent of CDCl_3 as the internal standard (^1H NMR: TMS at 0.00 ppm, CDCl_3 at 7.28 ppm; ^{13}C NMR: CDCl_3 at 77.0 ppm).

Kinetic Isotopic Effect (KIE) Studies:

1a-d₅ were synthesized deuterium substrates according the literature procedure.

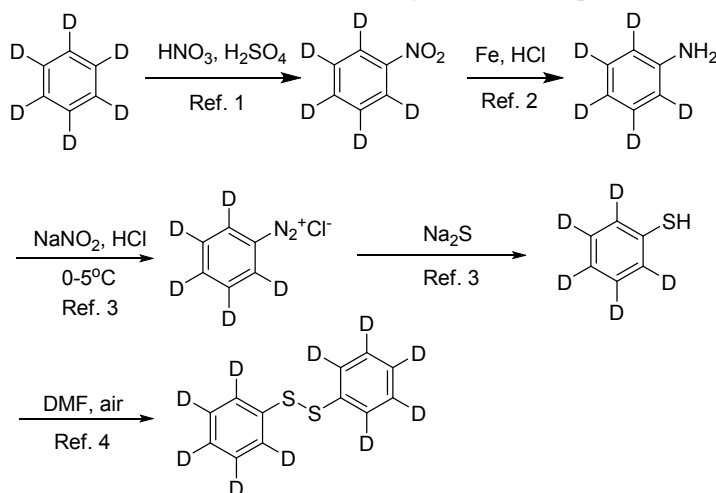


Figure S1 General synthesis route for preparation of **1a-d₅**

A 25 mL Schlenk tube equipped with a magnetic stirring bar was charged with TEAB (10.5 mg), potassium persulfate (1.1 mmol), 1,2-diphenyldisulfide disulfides (**1a**) (0.25 mmol), **1a-d₅** (0.25 mmol) and diethyl but-2-ynedioate (**2b**) (1.7 mmol). The tube was evacuated twice and backfilled with nitrogen, and DCE (1.5 mL) was added to the tube under nitrogen atmosphere. The tube was sealed with a balloon and then the mixture was allowed to stir under nitrogen atmosphere at 90 °C for 24 h. After completion of the reaction, the resulting solution was cooled to room temperature, and the solvent was removed with the aid of a rotary evaporator. The residue was purified

by column chromatography on silica gel using petroleum ether/ethyl acetate as eluent to provide the desired product, the product was analyzed by ^1H -NMR (400 MHz) (Figure S2). The result was summarized in equation S1:

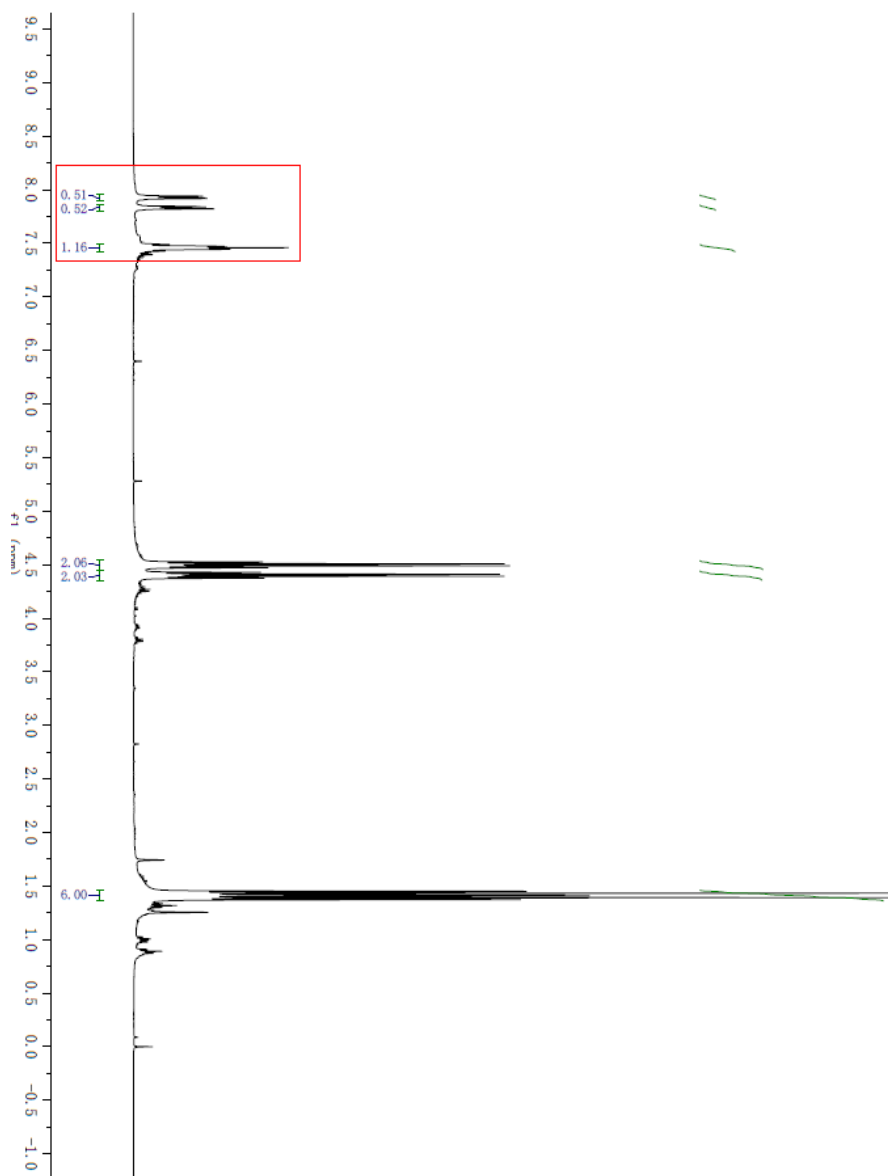
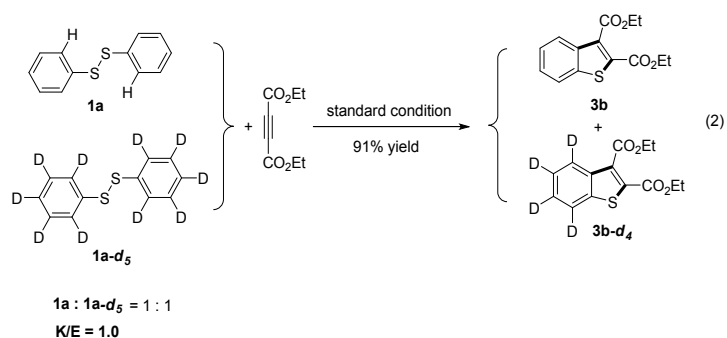


Figure S2 The ^1H -NMR spectrum of 3b and 3b-d₄

General procedure for synthesis of substituted benzothiophenes:

A 25 mL Schlenk tube equipped with a magnetic stirring bar was charged with TEAB (10.5 mg), potassium persulfate (1.1 mmol), substituted various disulfides (**1**) (0.5 mmol) and alkynes (**2**) (1.7 mmol). The tube was evacuated twice and backfilled with nitrogen, and DCE (1.5 mL) was added to the tube under nitrogen atmosphere. The tube was sealed with a balloon and then the mixture was allowed to stir under nitrogen atmosphere at 90 °C for 24 h. After completion of the reaction, the resulting solution was cooled to room temperature, and the solvent was removed with the aid of a rotary evaporator. The residue was purified by column chromatography on silica gel using petroleum ether/ethyl acetate as eluent to provide the desired product (**3**).

References

- [1] M. P. Mehn, S. D. Brown, D. M. Jenkins, J. C. Peters, L. Que, *Inorg. Chem.*, 2006, 7417-7427
- [2] R.S. Downing, P.J. Kunkeler, H.V. Bekkum, *Catalysis Today.*, 1997, **37**, 121-136.
- [3] A. Khazaei, M. Kazem-Rostami, A. R. Moosavi-Zare, M. Bayat, S. Saednia. *Synlett.*, 2012, **23**, 1893-1896.
- [4] J. L. G. Ruano, A. Parra, J. Alemán, *Green Chem.*, 2008, **10**, 706-711.

Crystal preparation and X-ray diffraction analysis of compound 3b

Crystal preparation of compound 3b.

Compound **3b** (20 mg) was dissolved in 5 mL of diethyl ether, and it was crystallized to give crystal as colorless prisms after the solvent was slowly volatilized in 2 days at room temperature (~ 25 °C).

X-Ray diffraction analysis of compound 3b.

Table 1. Crystal data and structure refinement for **a**.

Identification code	a
Empirical formula	C ₁₄ H ₁₄ O ₄ S
Formula weight	278.31
Temperature	295(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P21/n
Unit cell dimensions	a = 10.7820(18) Å alpha = 90 deg. b = 10.5128(17) Å beta = 106.290(2) deg. c = 12.334(2) Å gamma = 90 deg.
Volume	1341.9(4) Å ³
Z, Calculated density	4, 1.378 Mg/m ³
Absorption coefficient	0.248 mm ⁻¹
F(000)	584
Crystal size	0.36 x 0.28 x 0.20 mm
Theta range for data collection	2.22 to 25.99 deg.
Limiting indices	-13<=h<=13, -12<=k<=12, -15<=l<=14
Reflections collected / unique	10113 / 2639 [R(int) = 0.0256]
Completeness to theta = 25.99	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9521 and 0.9160
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	2639 / 0 / 173

Goodness-of-fit on F^2	1.048
Final R indices [$I > 2\sigma(I)$]	$R1 = 0.0442$, $wR2 = 0.1241$
R indices (all data)	$R1 = 0.0504$, $wR2 = 0.1302$
Extinction coefficient	0.015(2)
Largest diff. peak and hole	0.542 and -0.243 e. $\text{\AA}^{-3}$

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for a. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

x	y	z	$U(\text{eq})$	
C(1)	1194(2)	10568(2)	1340(2)	42(1)
C(2)	1380(2)	11881(2)	1256(2)	51(1)
C(3)	645(2)	12691(2)	1684(2)	56(1)
C(4)	-273(2)	12236(2)	2194(2)	55(1)
C(5)	-460(2)	10955(2)	2281(2)	48(1)
C(6)	277(2)	10099(2)	1851(1)	39(1)
C(7)	248(2)	8735(2)	1841(1)	40(1)
C(8)	-661(2)	7995(2)	2319(2)	46(1)
C(9)	-825(3)	6765(2)	3887(2)	67(1)
C(10)	-823(3)	5428(2)	3511(2)	77(1)
C(11)	1098(2)	8221(2)	1338(2)	40(1)
C(12)	1229(2)	6844(2)	1161(2)	43(1)
C(13)	2247(2)	5235(2)	410(2)	54(1)
C(14)	3190(3)	5118(2)	-245(2)	75(1)
O(1)	-1781(2)	7870(2)	1850(2)	72(1)
O(2)	-68(2)	7567(1)	3343(1)	55(1)
O(3)	576(1)	6059(1)	1448(1)	58(1)
O(4)	2117(1)	6587(1)	639(1)	51(1)

S(1)	1983(1)	9355(1)	858(1)	47(1)
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Table 3. Bond lengths [Å] and angles [deg] for a.

C(1)-C(2)	1.403(3)
C(1)-C(6)	1.403(2)
C(1)-S(1)	1.7284(18)
C(2)-C(3)	1.367(3)
C(2)-H(2)	0.9300
C(3)-C(4)	1.398(3)
C(3)-H(3)	0.9300
C(4)-C(5)	1.370(3)
C(4)-H(4)	0.9300
C(5)-C(6)	1.399(3)
C(5)-H(5)	0.9300
C(6)-C(7)	1.434(3)
C(7)-C(11)	1.356(2)
C(7)-C(8)	1.495(2)
C(8)-O(1)	1.191(2)
C(8)-O(2)	1.324(2)
C(9)-O(2)	1.461(2)
C(9)-C(10)	1.480(3)
C(9)-H(9A)	0.9700
C(9)-H(9B)	0.9700
C(10)-H(10A)	0.9600
C(10)-H(10B)	0.9600
C(10)-H(10C)	0.9600
C(11)-C(12)	1.476(3)
C(11)-S(1)	1.7316(18)
C(12)-O(3)	1.200(2)

C(12)-O(4)	1.323(2)
C(13)-O(4)	1.463(2)
C(13)-C(14)	1.470(3)
C(13)-H(13A)	0.9700
C(13)-H(13B)	0.9700
C(14)-H(14A)	0.9600
C(14)-H(14B)	0.9600
C(14)-H(14C)	0.9600
C(2)-C(1)-C(6)	120.82(18)
C(2)-C(1)-S(1)	127.30(15)
C(6)-C(1)-S(1)	111.87(13)
C(3)-C(2)-C(1)	118.34(19)
C(3)-C(2)-H(2)	120.8
C(1)-C(2)-H(2)	120.8
C(2)-C(3)-C(4)	121.38(19)
C(2)-C(3)-H(3)	119.3
C(4)-C(3)-H(3)	119.3
C(5)-C(4)-C(3)	120.6(2)
C(5)-C(4)-H(4)	119.7
C(3)-C(4)-H(4)	119.7
C(4)-C(5)-C(6)	119.47(19)
C(4)-C(5)-H(5)	120.3
C(6)-C(5)-H(5)	120.3
C(5)-C(6)-C(1)	119.36(17)
C(5)-C(6)-C(7)	129.31(16)
C(1)-C(6)-C(7)	111.33(15)
C(11)-C(7)-C(6)	112.74(15)
C(11)-C(7)-C(8)	125.11(16)
C(6)-C(7)-C(8)	122.13(15)
O(1)-C(8)-O(2)	125.35(18)

O(1)-C(8)-C(7)	123.60(18)
O(2)-C(8)-C(7)	110.96(16)
O(2)-C(9)-C(10)	110.7(2)
O(2)-C(9)-H(9A)	109.5
C(10)-C(9)-H(9A)	109.5
O(2)-C(9)-H(9B)	109.5
C(10)-C(9)-H(9B)	109.5
H(9A)-C(9)-H(9B)	108.1
C(9)-C(10)-H(10A)	109.5
C(9)-C(10)-H(10B)	109.5
H(10A)-C(10)-H(10B)	109.5
C(9)-C(10)-H(10C)	109.5
H(10A)-C(10)-H(10C)	109.5
H(10B)-C(10)-H(10C)	109.5
C(7)-C(11)-C(12)	124.26(16)
C(7)-C(11)-S(1)	112.99(13)
C(12)-C(11)-S(1)	122.69(13)
O(3)-C(12)-O(4)	124.52(18)
O(3)-C(12)-C(11)	122.88(17)
O(4)-C(12)-C(11)	112.59(16)
O(4)-C(13)-C(14)	107.83(18)
O(4)-C(13)-H(13A)	110.1
C(14)-C(13)-H(13A)	110.1
O(4)-C(13)-H(13B)	110.1
C(14)-C(13)-H(13B)	110.1
H(13A)-C(13)-H(13B)	108.5
C(13)-C(14)-H(14A)	109.5
C(13)-C(14)-H(14B)	109.5
H(14A)-C(14)-H(14B)	109.5
C(13)-C(14)-H(14C)	109.5

H(14A)-C(14)-H(14C)	109.5
H(14B)-C(14)-H(14C)	109.5
C(8)-O(2)-C(9)	116.80(17)
C(12)-O(4)-C(13)	114.48(15)
C(1)-S(1)-C(11)	91.06(9)

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for a. The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$$

U11	U22	U33	U23	U13	U12	
C(1)	44(1)	38(1)	42(1)	2(1)	10(1)	-1(1)
C(2)	57(1)	39(1)	57(1)	6(1)	16(1)	-9(1)
C(3)	68(1)	34(1)	64(1)	3(1)	13(1)	-5(1)
C(4)	65(1)	38(1)	62(1)	-3(1)	19(1)	5(1)
C(5)	54(1)	42(1)	52(1)	1(1)	21(1)	1(1)
C(6)	41(1)	36(1)	39(1)	2(1)	10(1)	-1(1)
C(7)	41(1)	36(1)	43(1)	1(1)	12(1)	-1(1)
C(8)	49(1)	32(1)	64(1)	0(1)	24(1)	1(1)
C(9)	93(2)	49(1)	74(1)	12(1)	46(1)	-6(1)
C(10)	87(2)	55(1)	96(2)	11(1)	41(2)	-3(1)
C(11)	40(1)	35(1)	45(1)	1(1)	12(1)	-2(1)
C(12)	43(1)	40(1)	47(1)	-3(1)	13(1)	1(1)
C(13)	59(1)	40(1)	67(1)	-6(1)	25(1)	1(1)
C(14)	88(2)	55(1)	98(2)	-10(1)	53(2)	5(1)
O(1)	45(1)	73(1)	99(1)	21(1)	20(1)	-6(1)
O(2)	67(1)	48(1)	54(1)	7(1)	24(1)	-9(1)
O(3)	66(1)	40(1)	80(1)	-6(1)	37(1)	-7(1)

O(4)	53(1)	40(1)	66(1)	-6(1)	26(1)	1(1)
S(1)	47(1)	43(1)	55(1)	2(1)	23(1)	-2(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for a.

	x	y	z	U(eq)
H(2)	1989	12191	918	61
H(3)	758	13564	1635	68
H(4)	-762	12808	2478	66
H(5)	-1071	10657	2622	58
H(9A)	-1707	7076	3701	81
H(9B)	-465	6808	4700	81
H(10A)	-1326	4917	3875	115
H(10B)	49	5116	3706	115
H(10C)	-1189	5383	2707	115
H(13A)	1419	4890	-15	65
H(13B)	2543	4768	1114	65
H(14A)	3291	4238	-408	113
H(14B)	4006	5457	186	113
H(14C)	2887	5583	-939	113

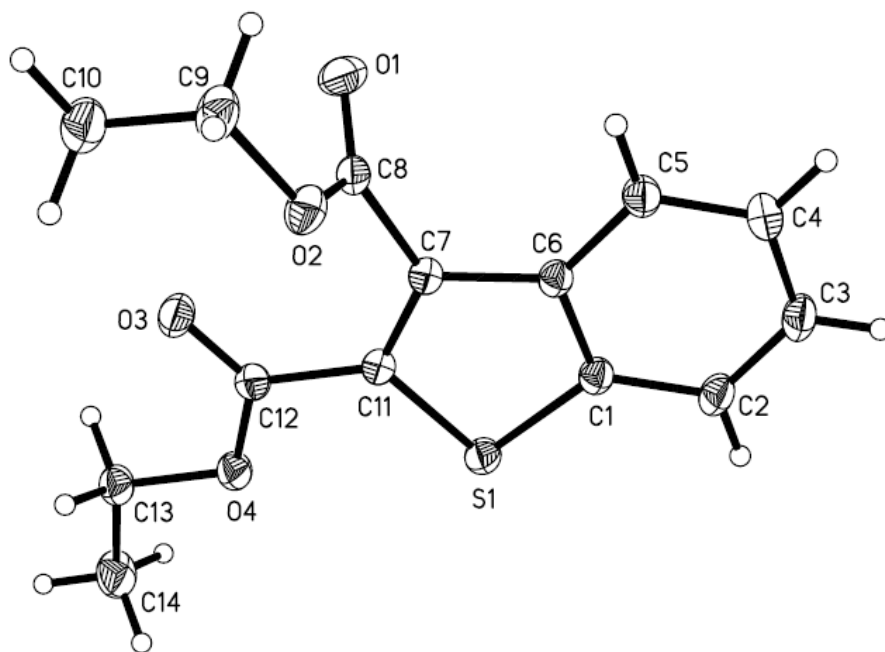


Figure 1. ORTEP drawing of C₁₄H₁₄O₄S with 50% probability ellipsoids, showing the atomic numbering scheme.

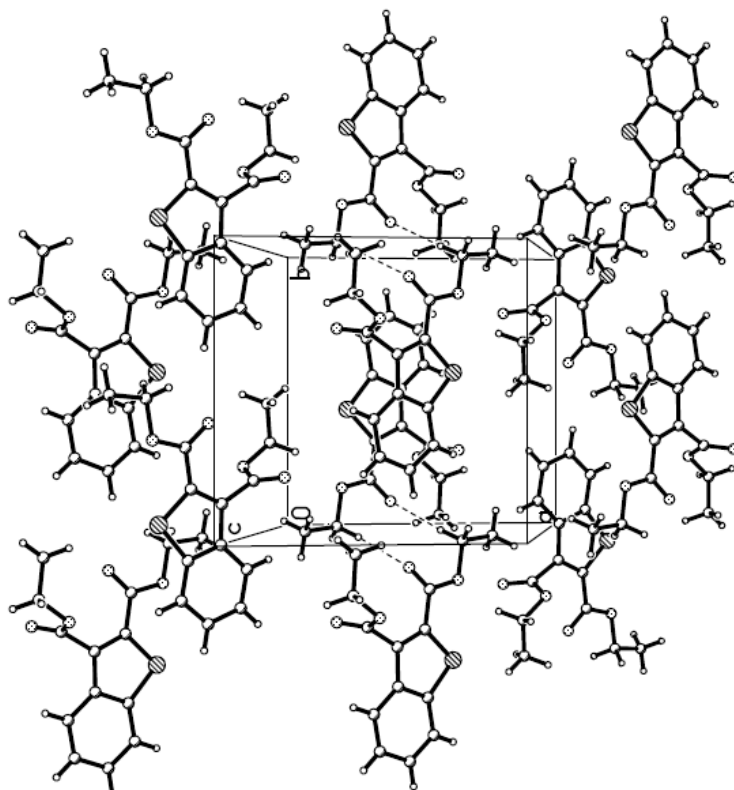
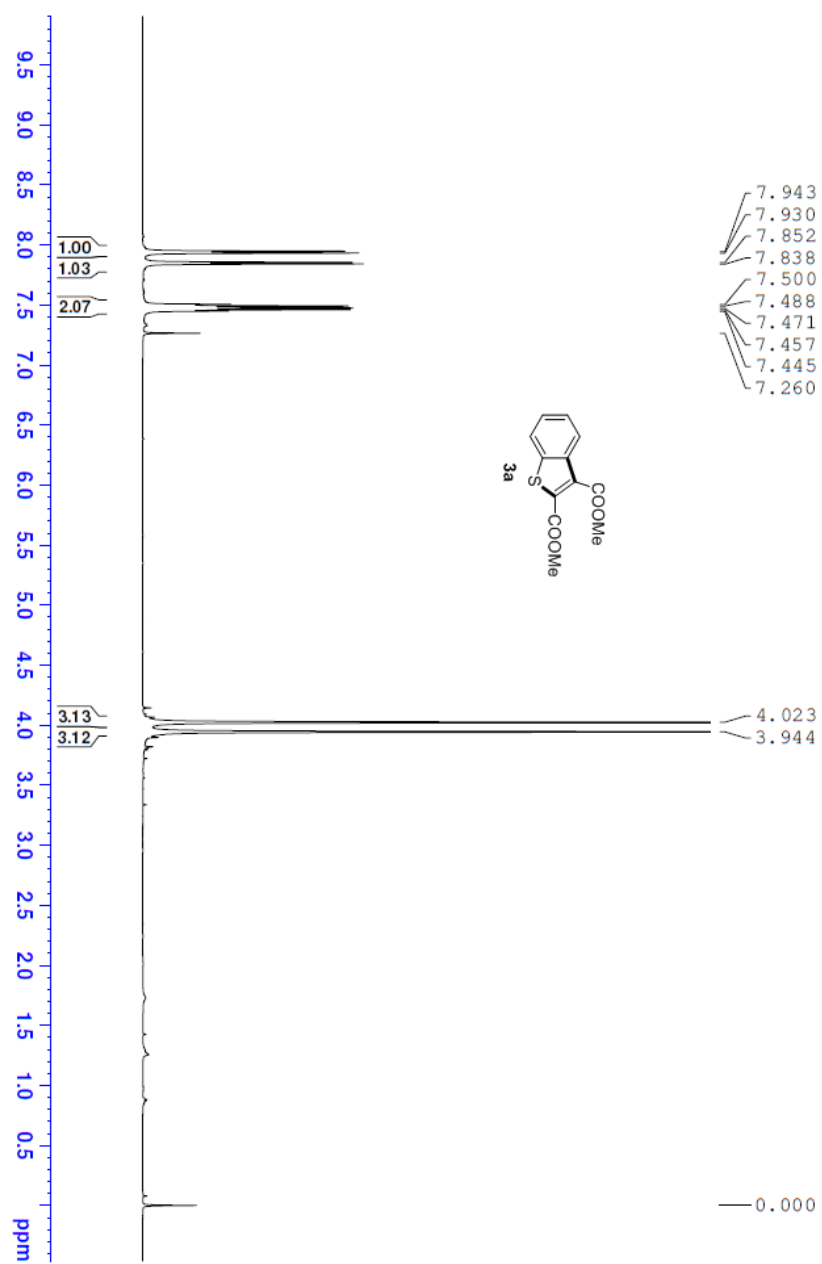
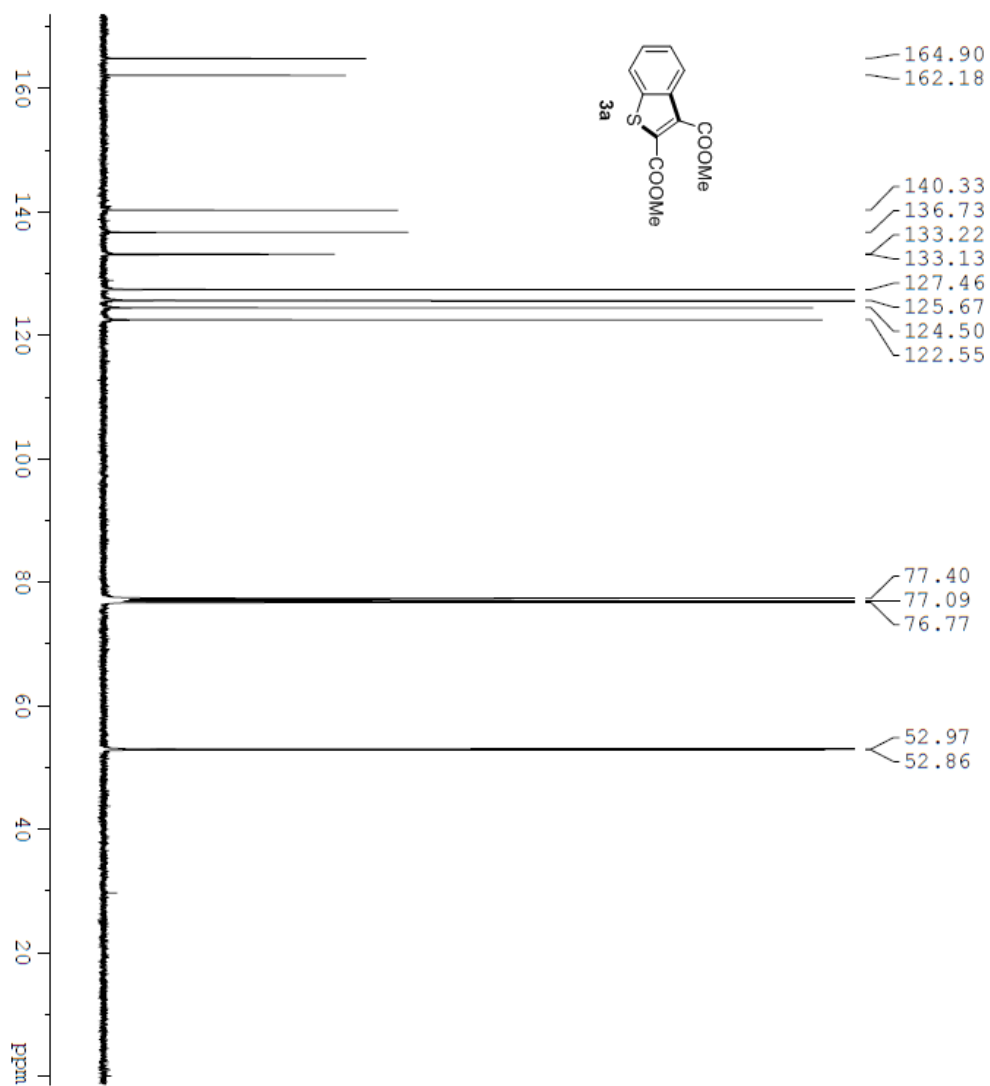
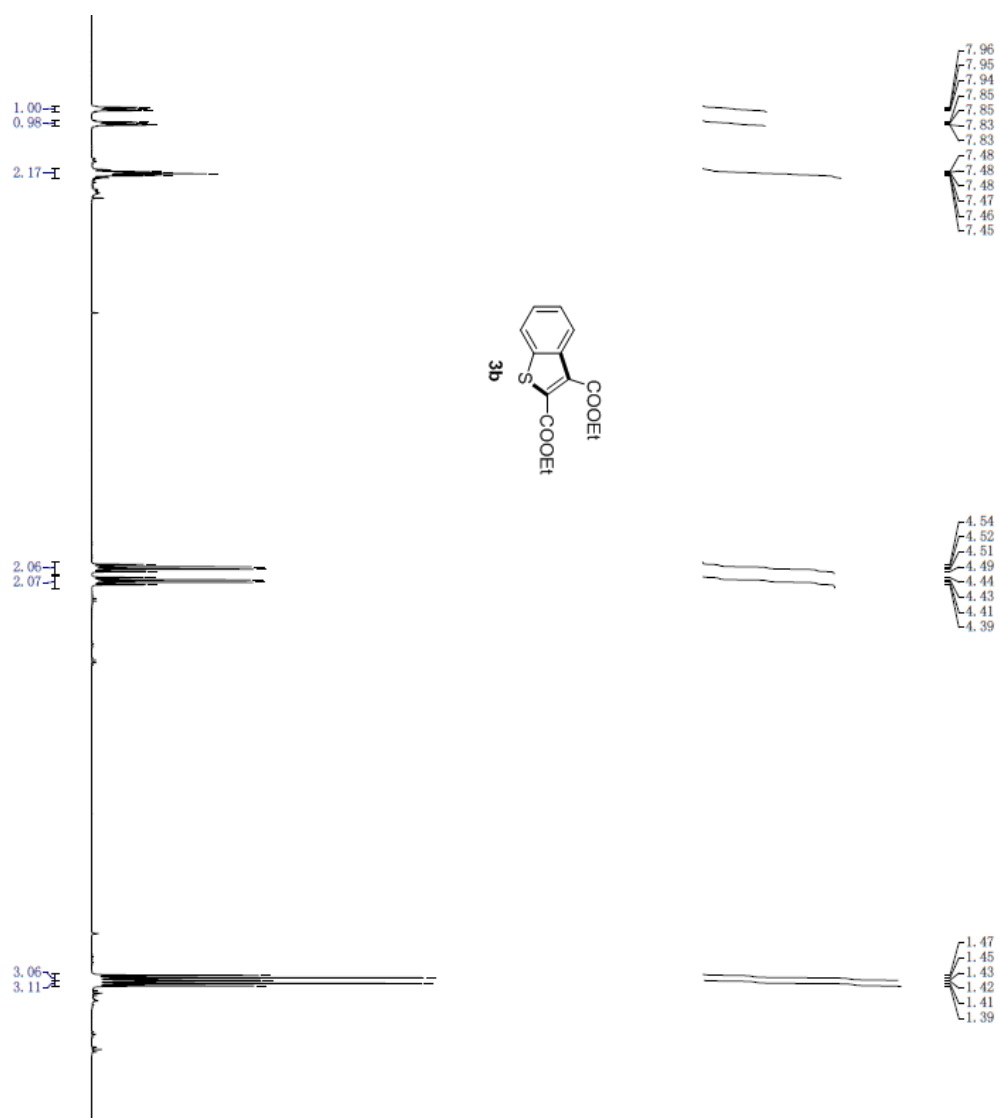
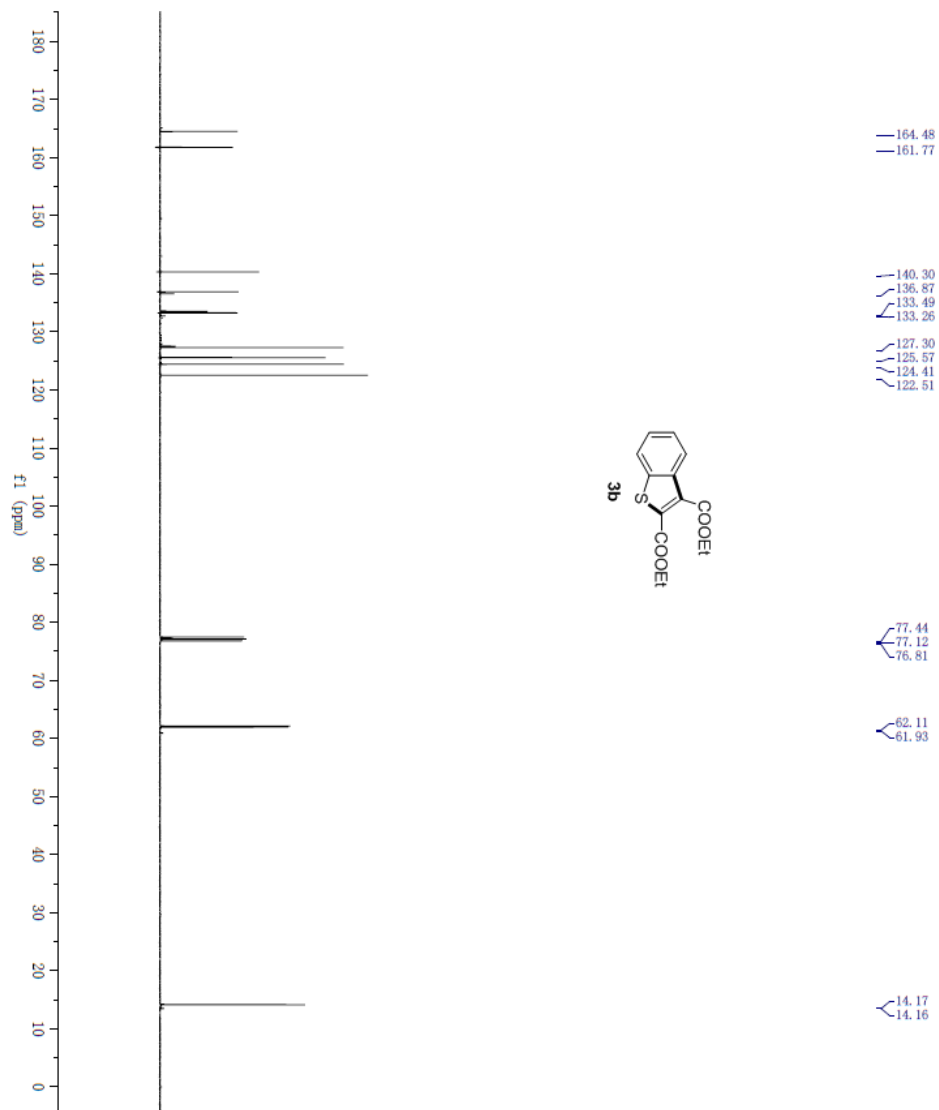


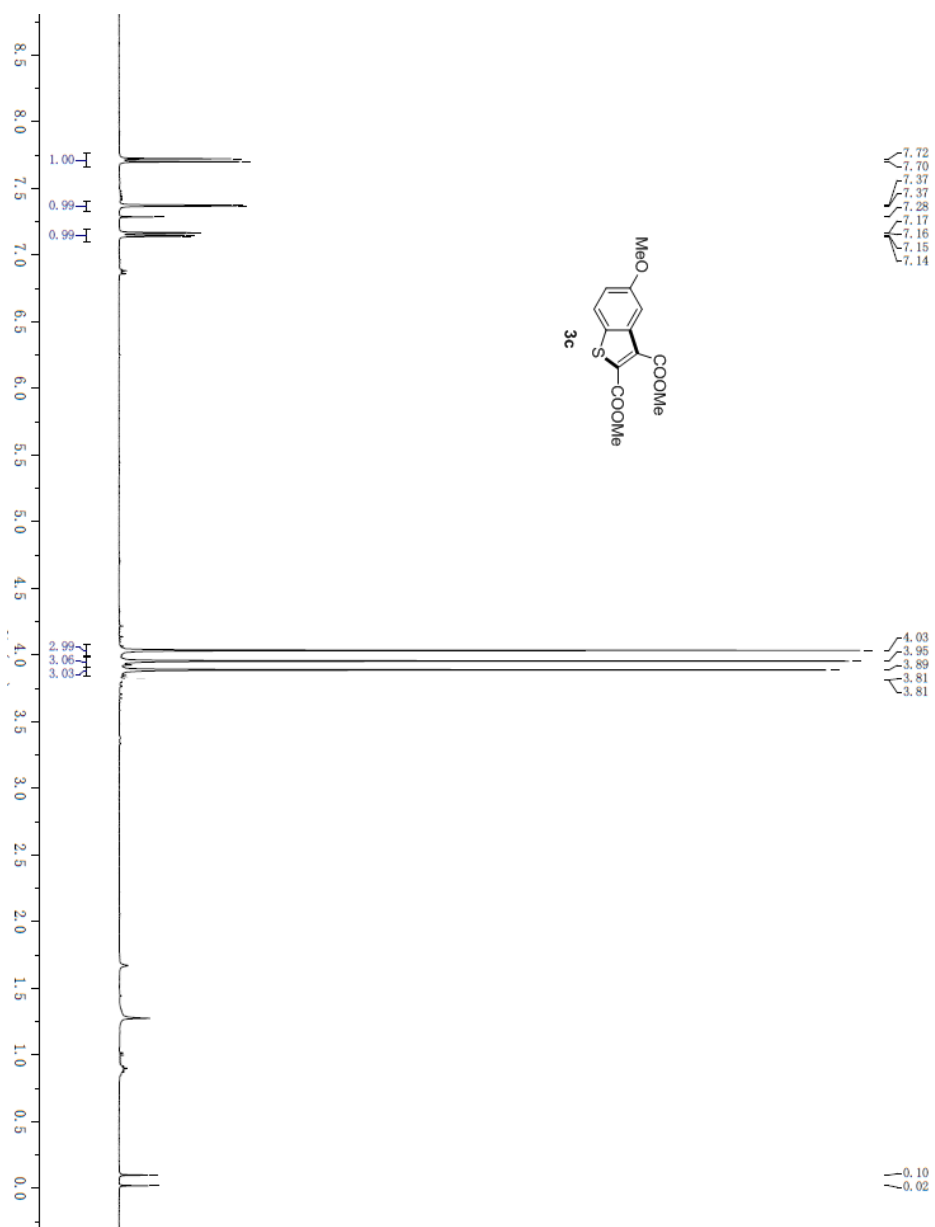
Figure 2. A packing view along the *a* direction

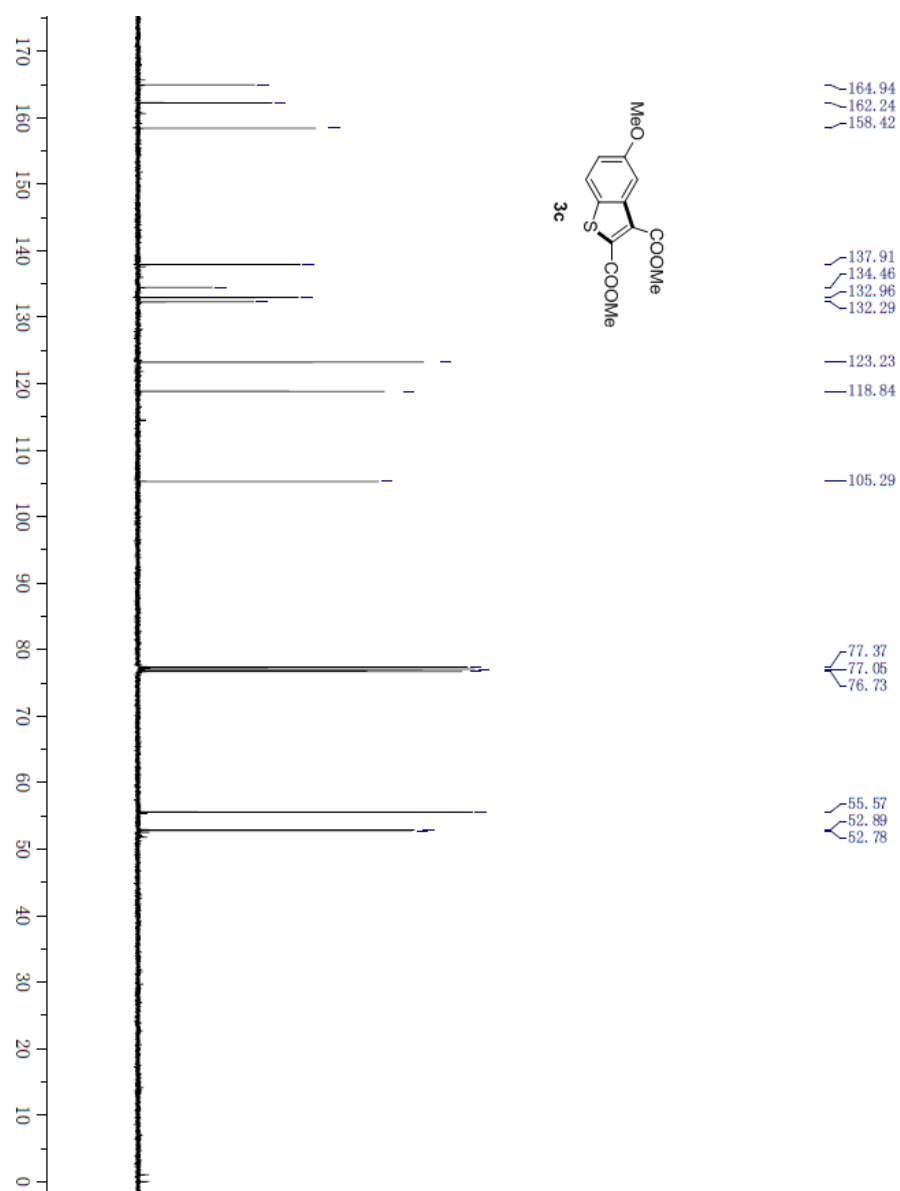


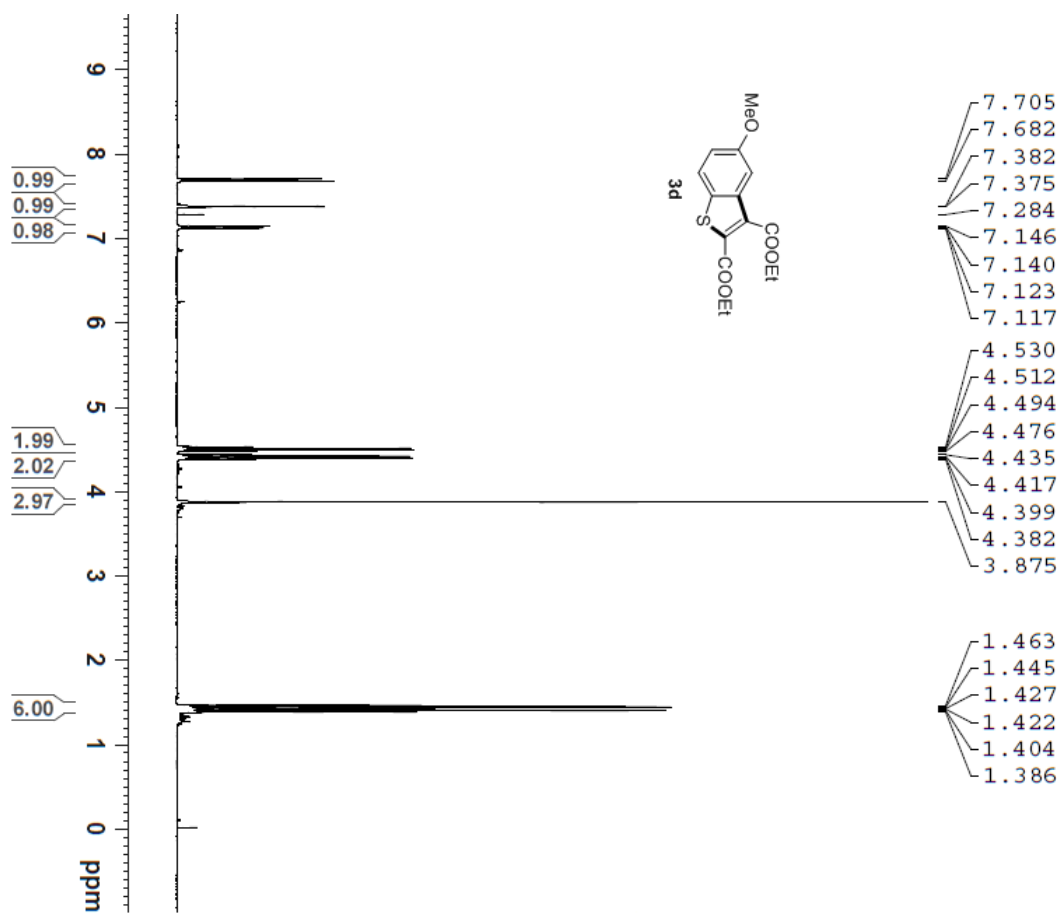


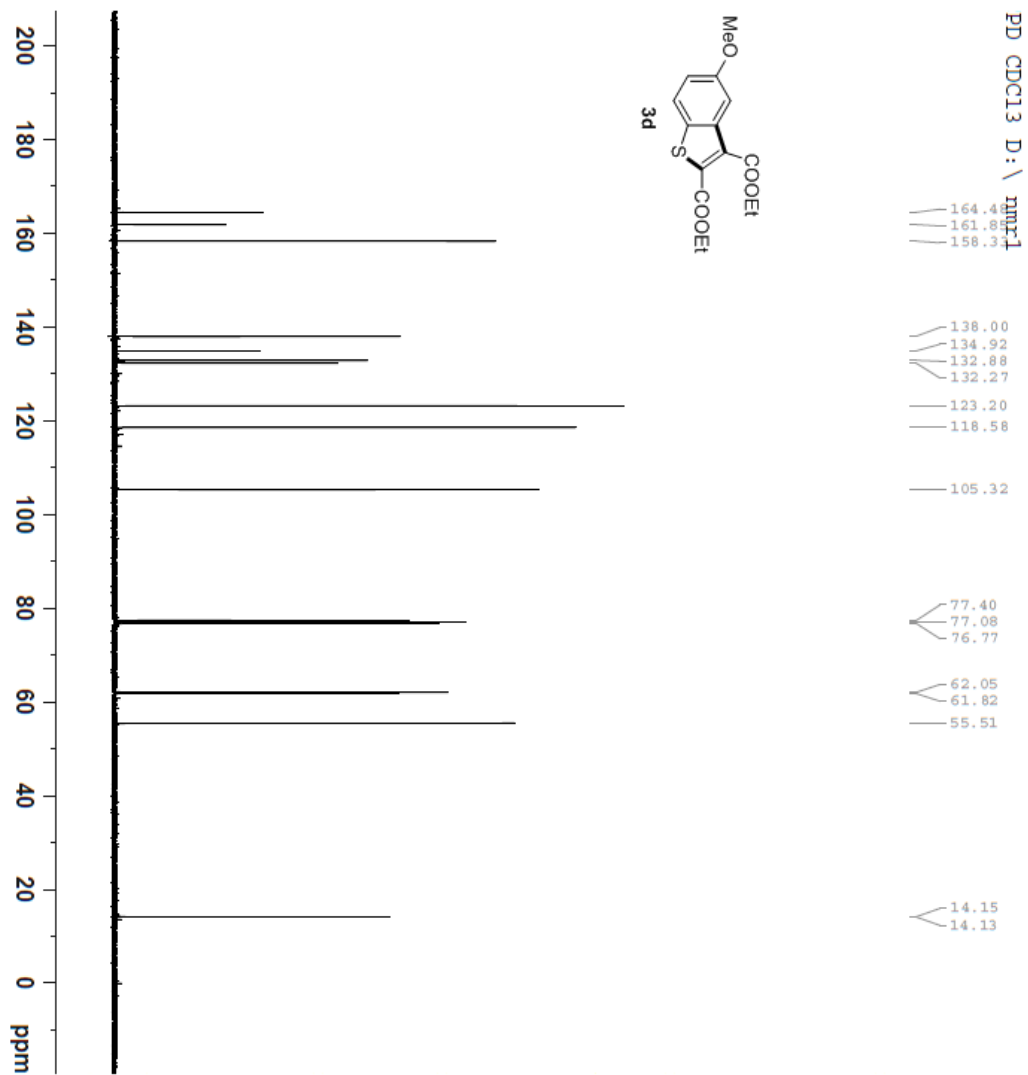


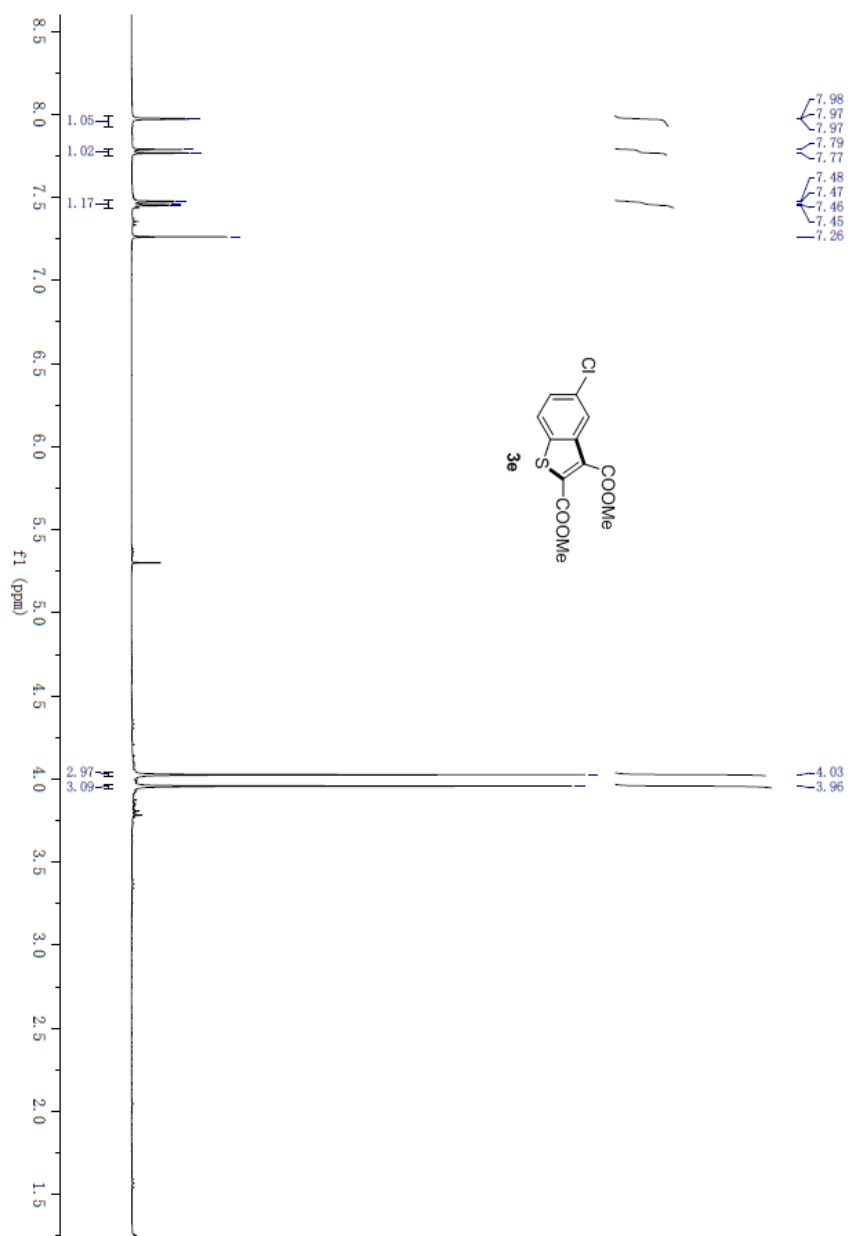


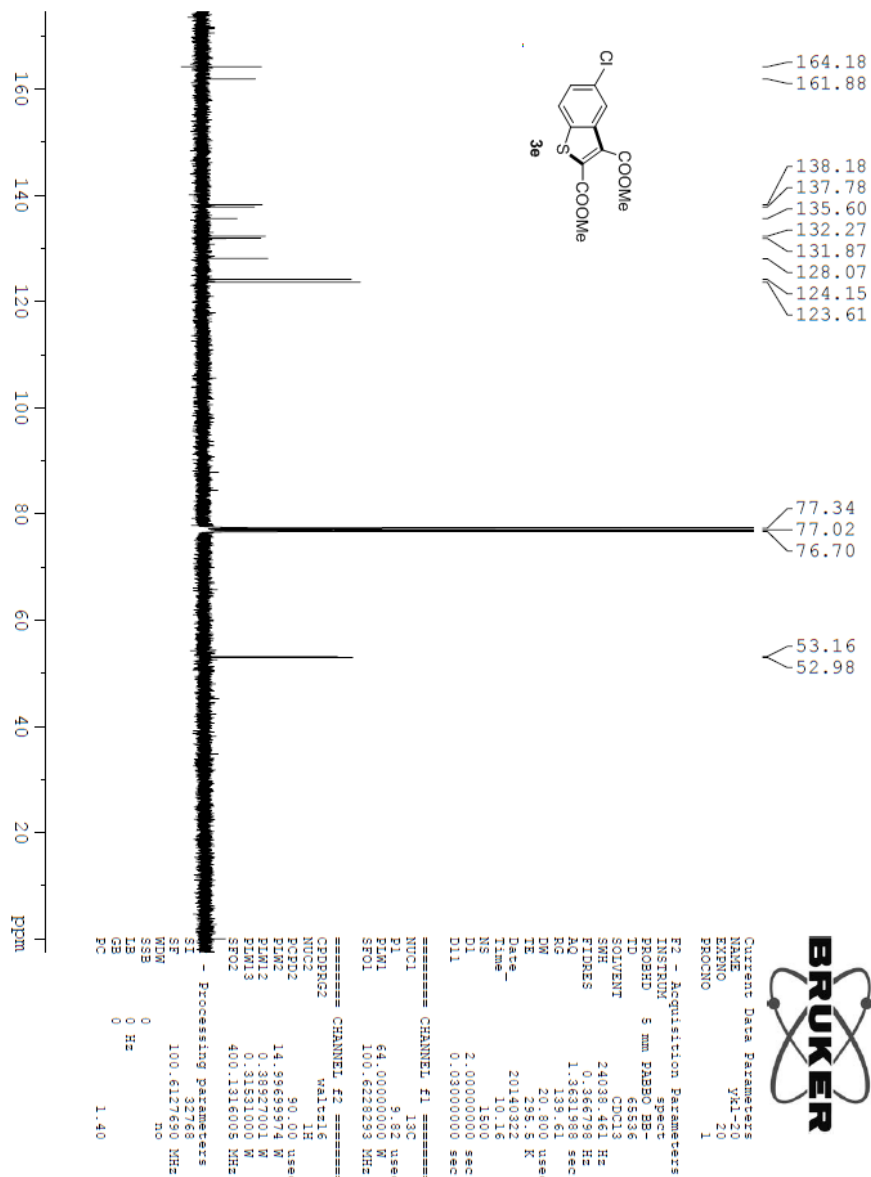


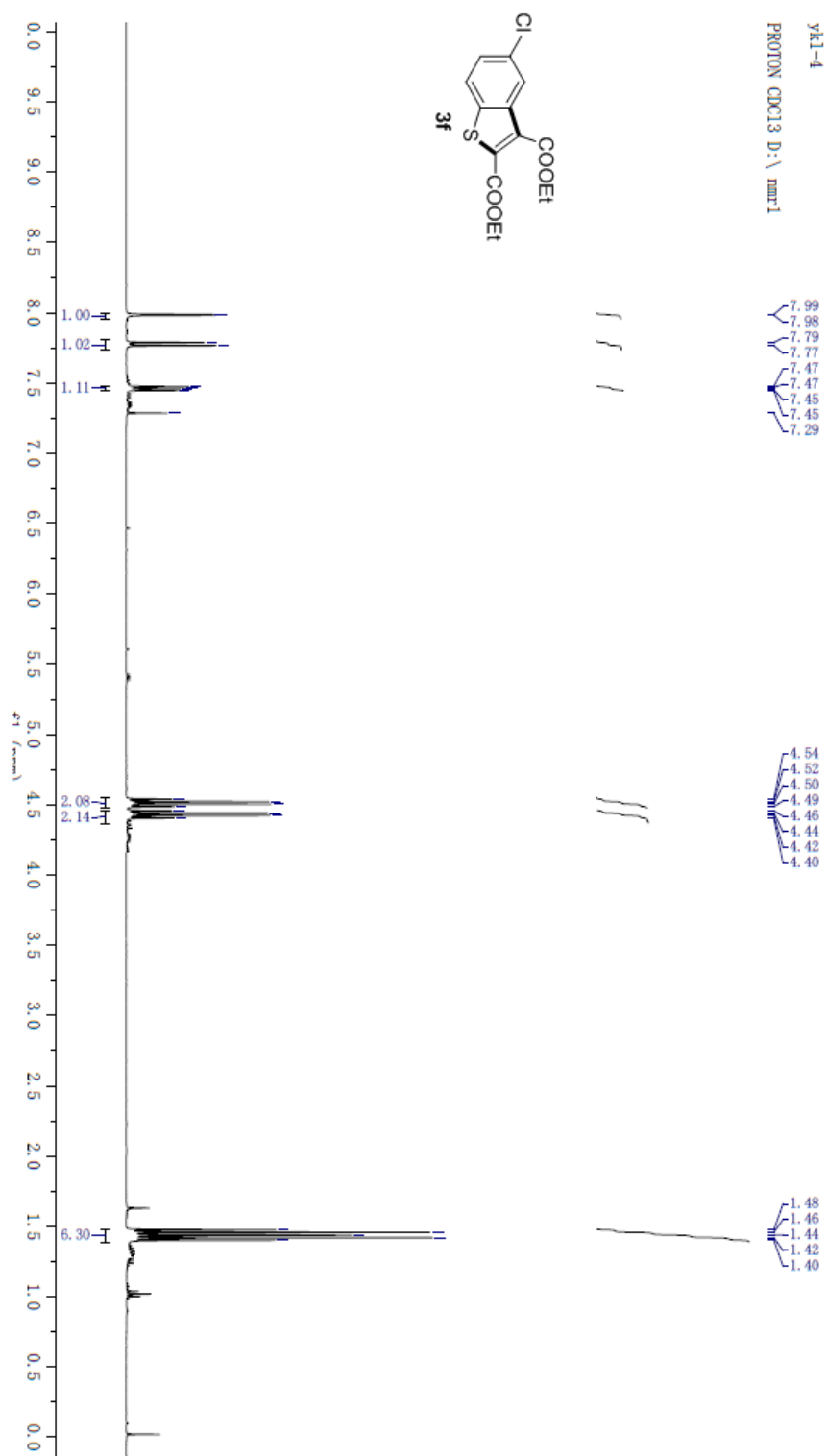


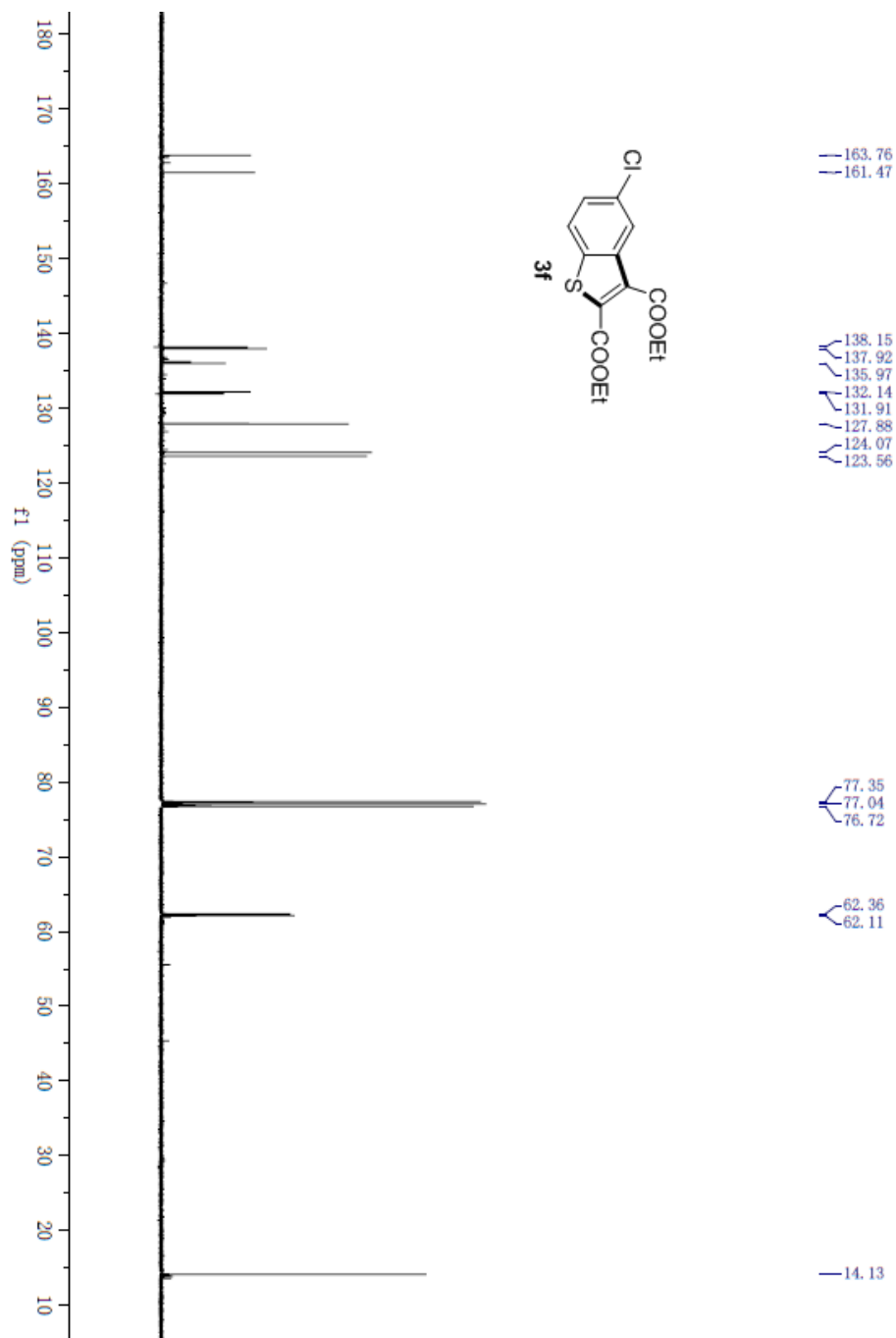


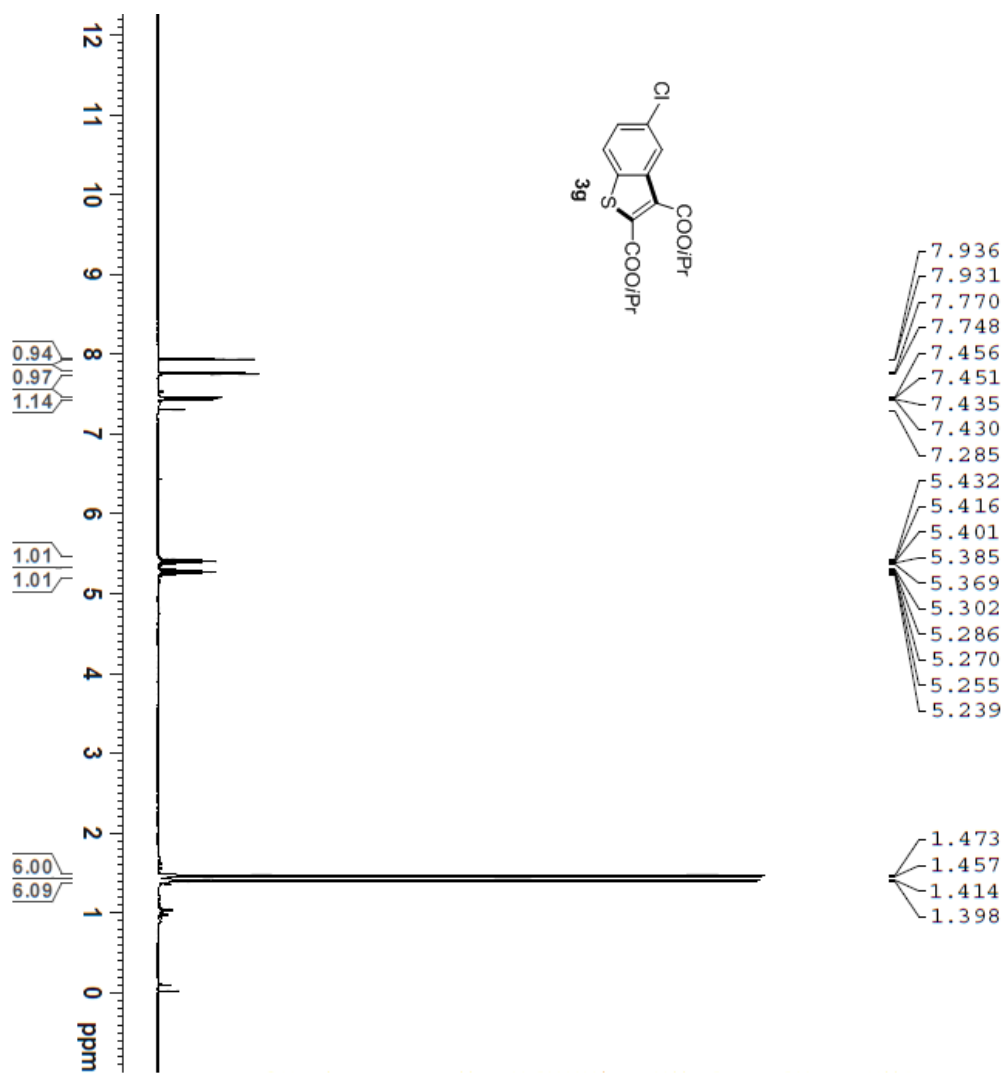


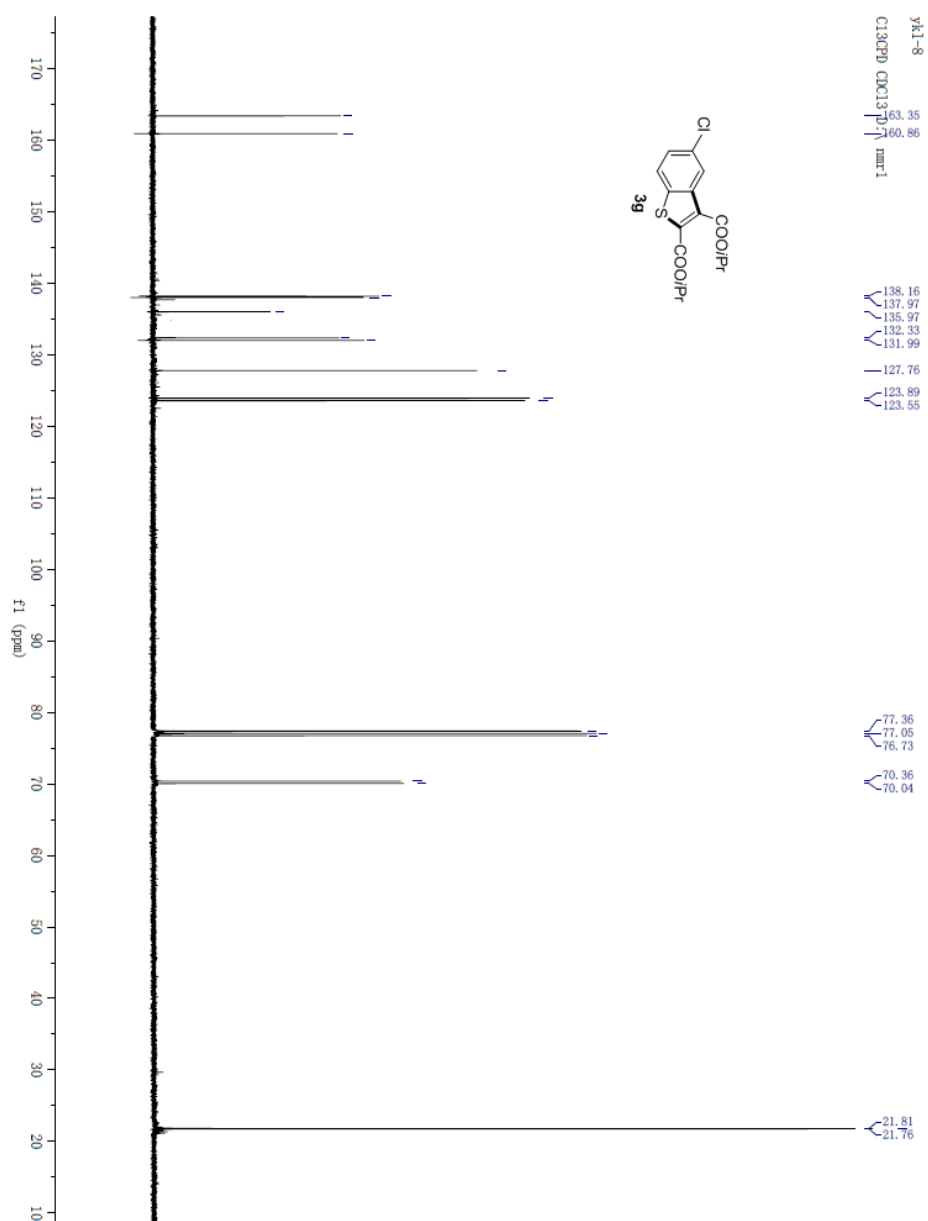


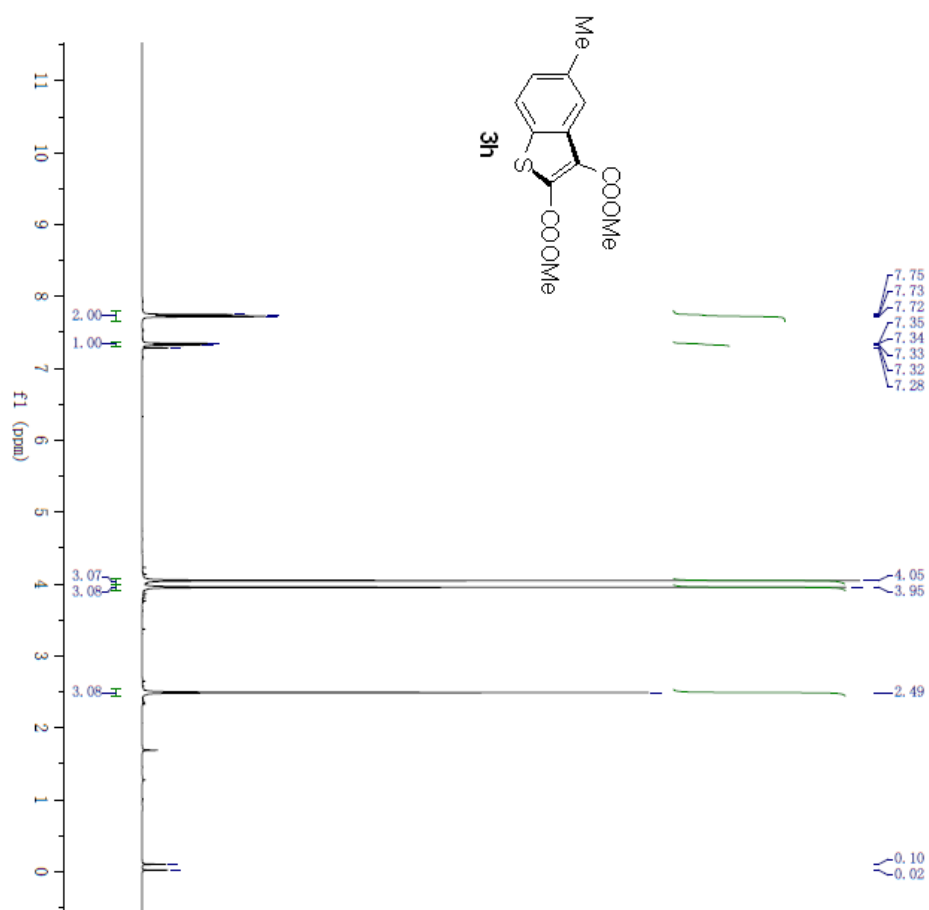


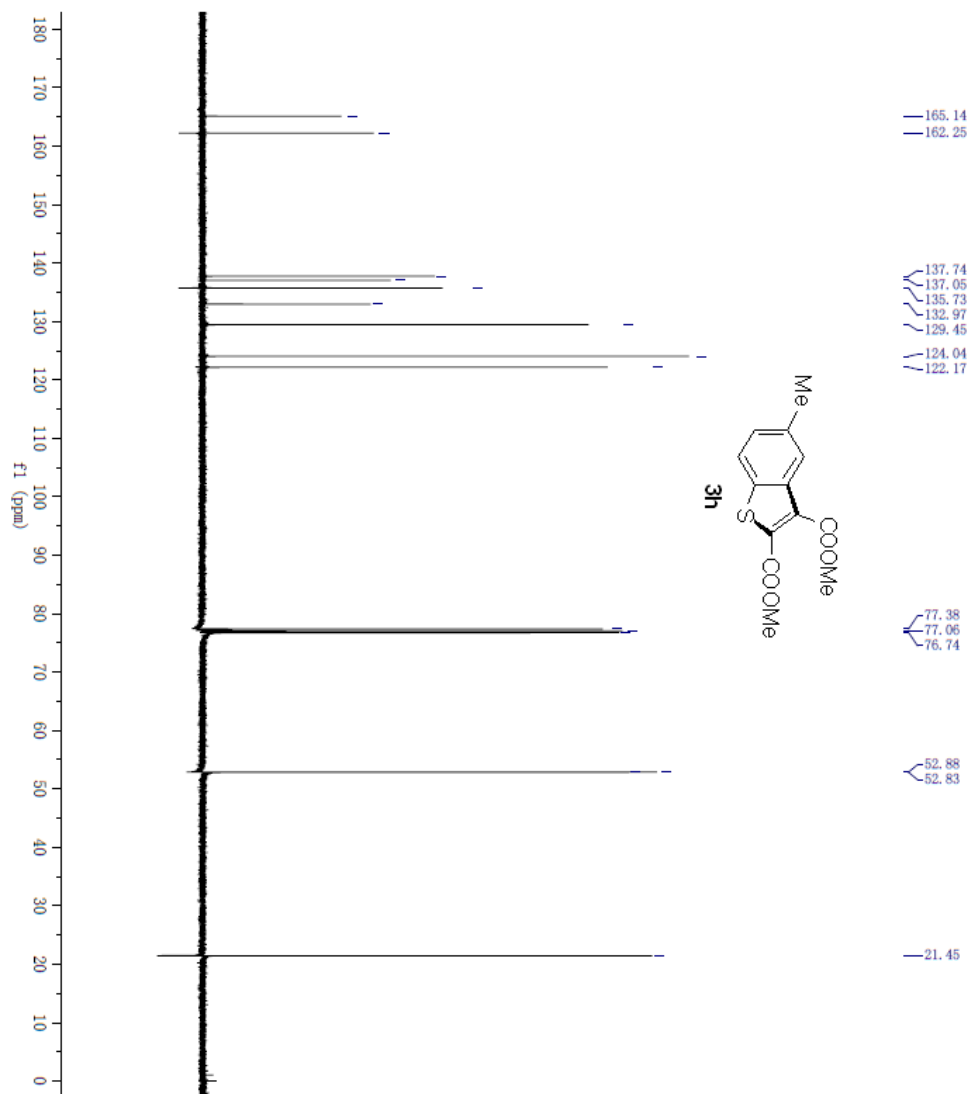










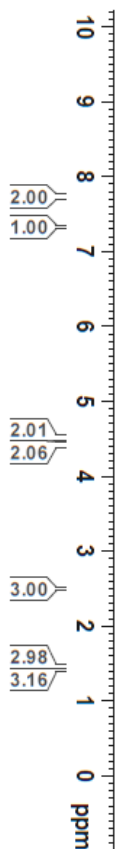
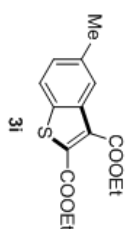


nmr1

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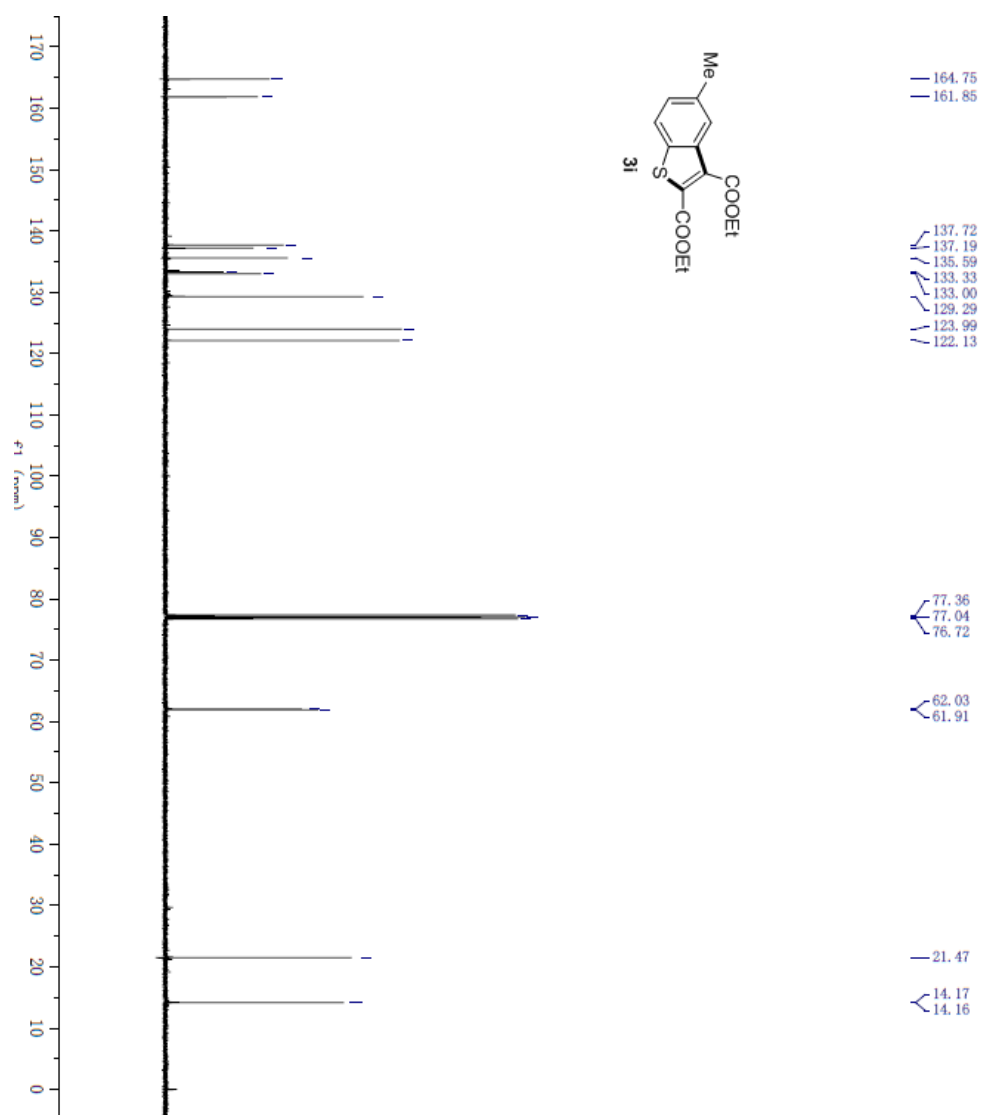
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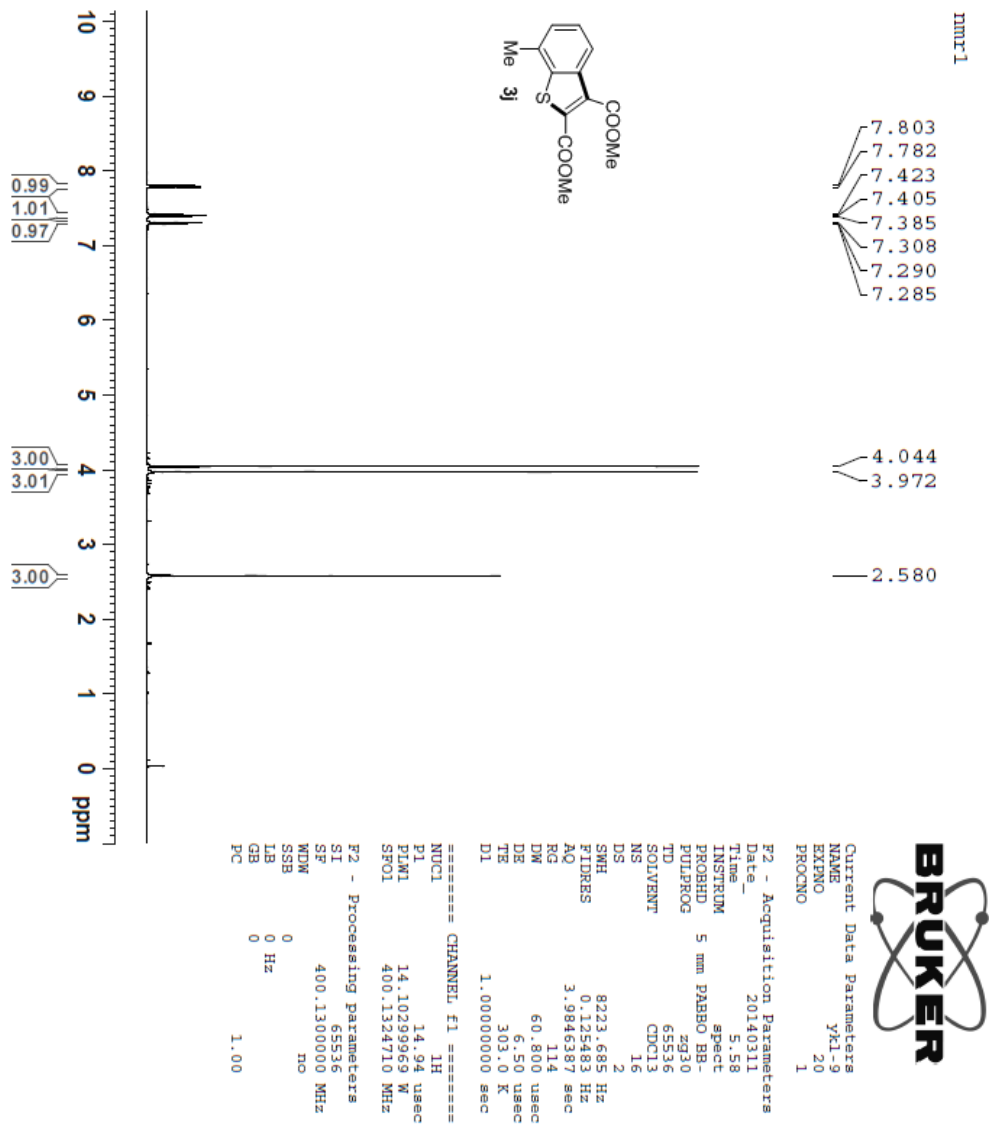


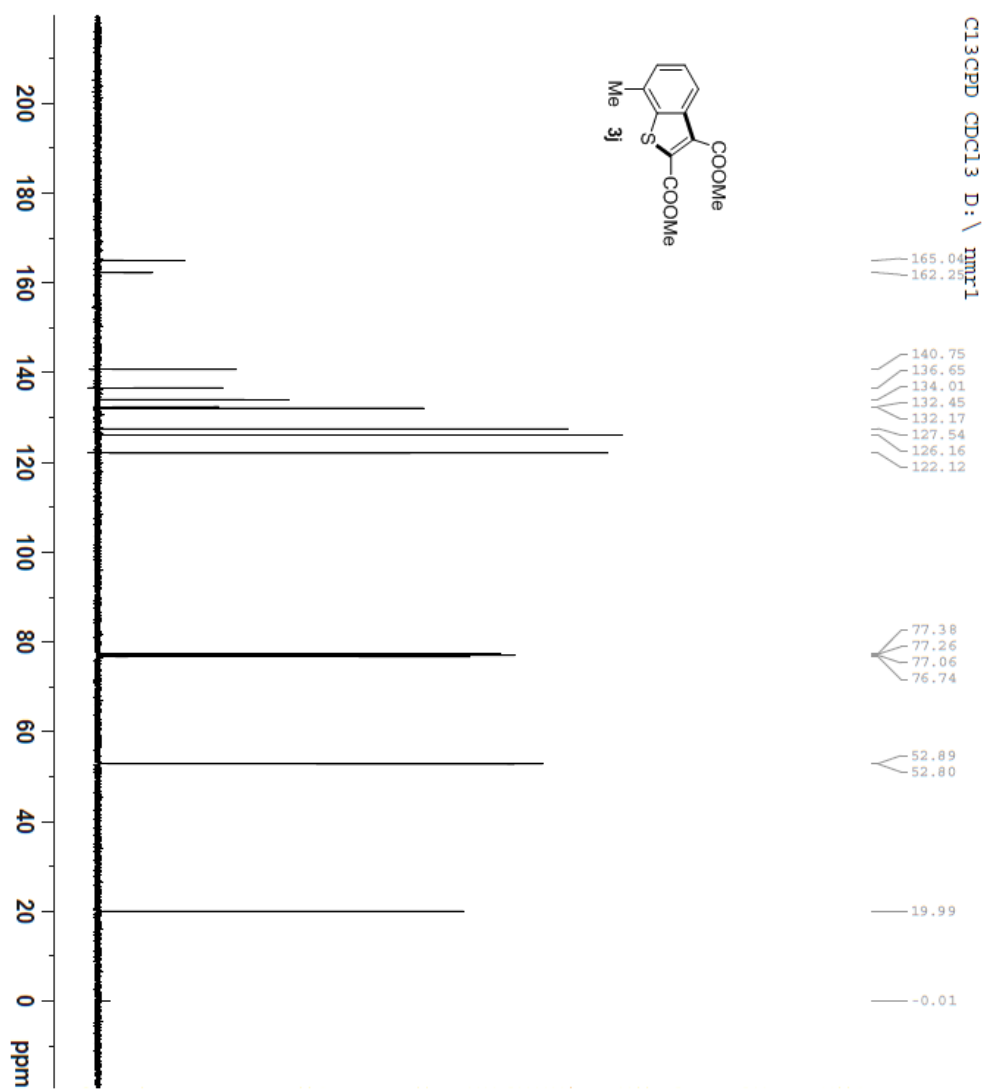
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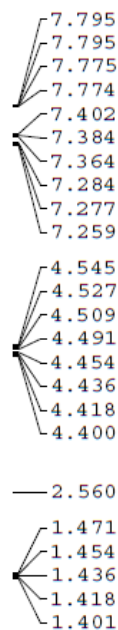
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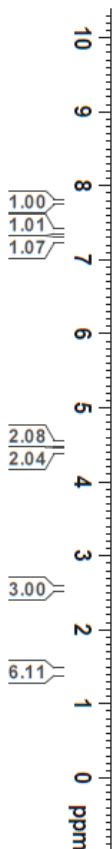
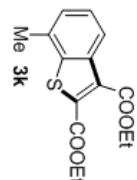
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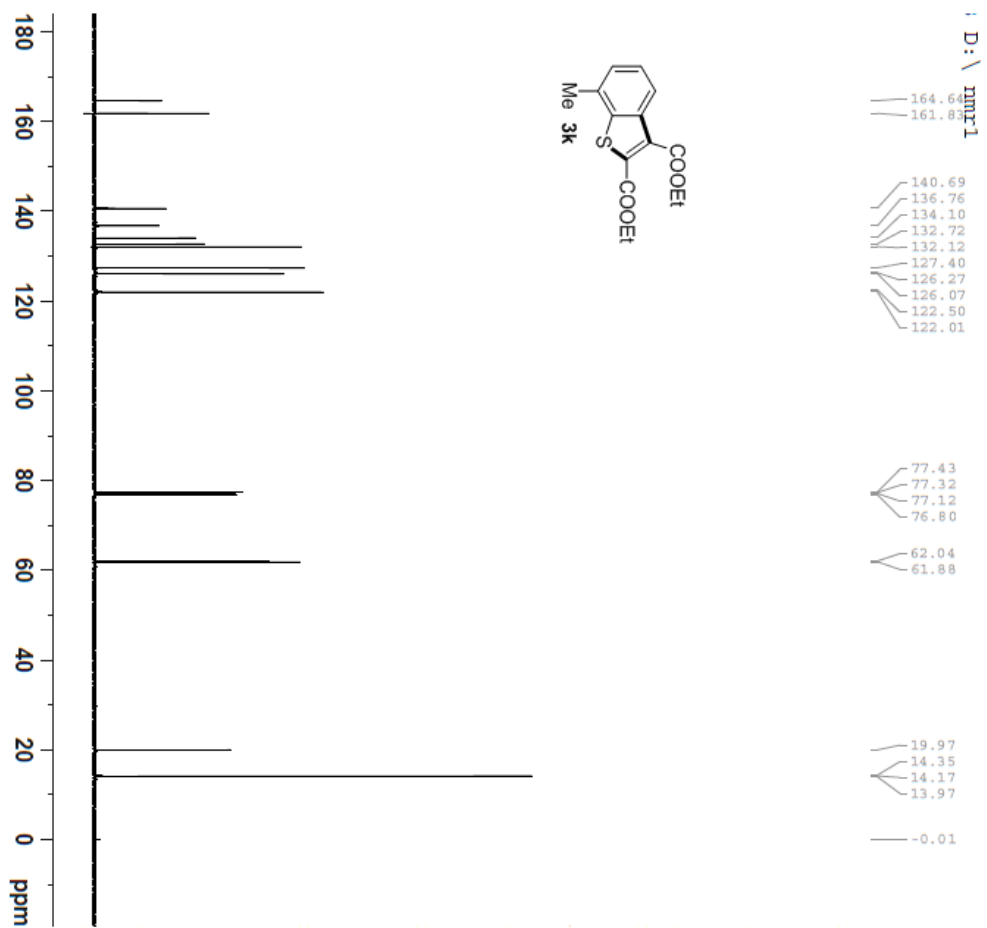


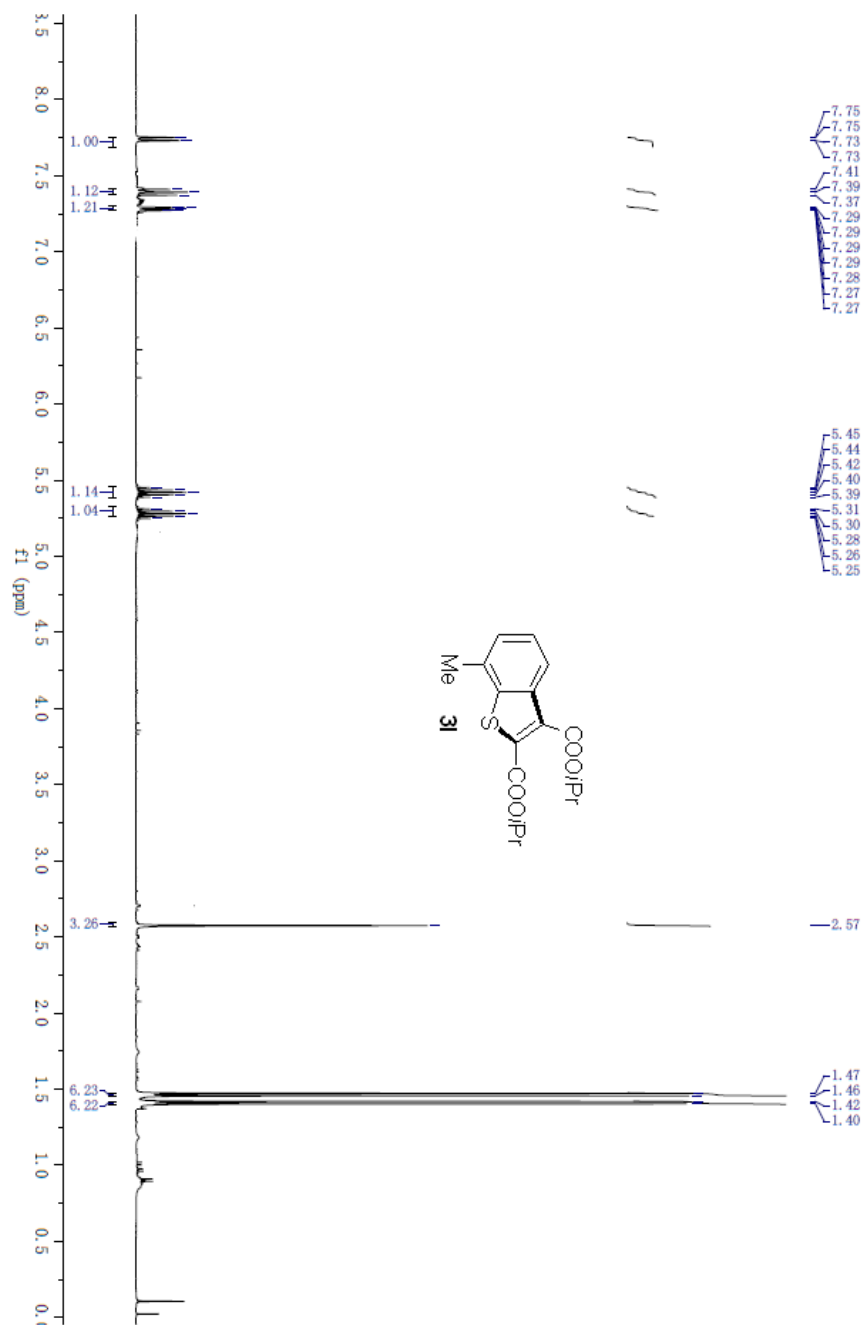
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EXPNO 22
PROCNO 1

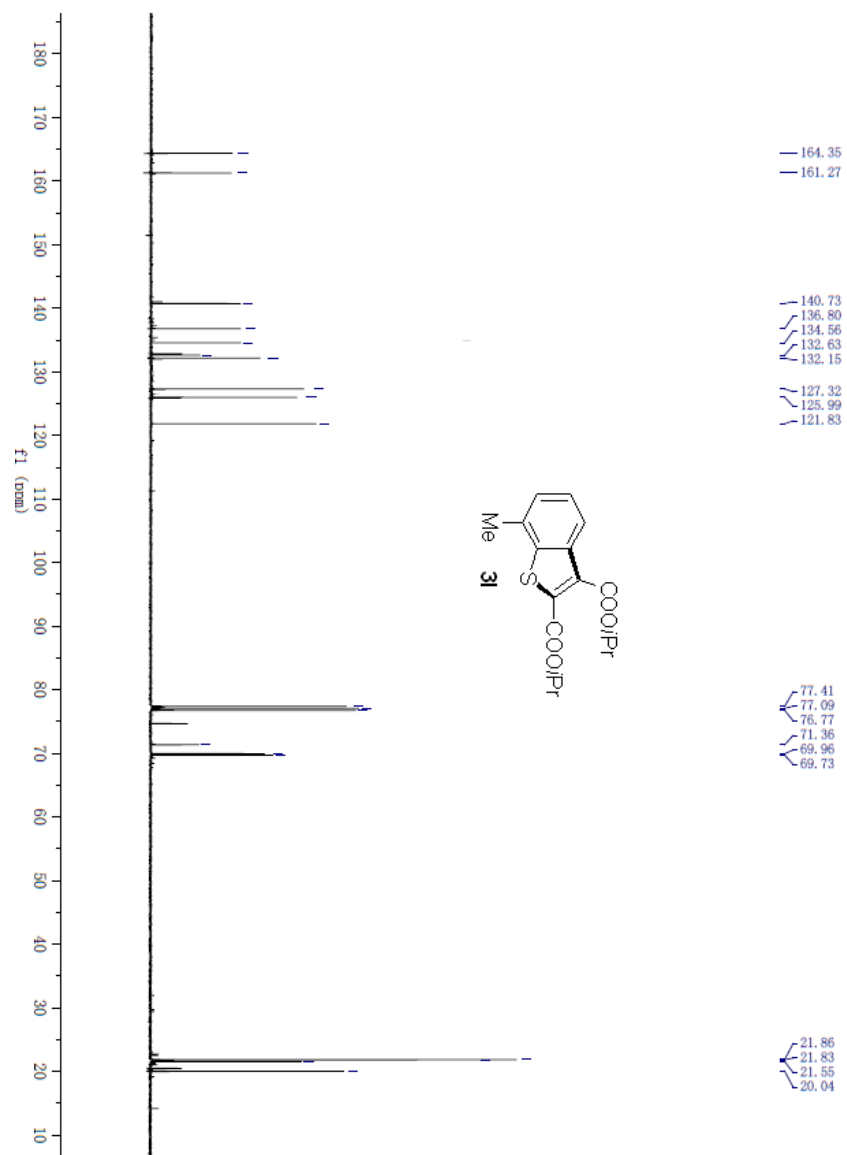
F2 - Acquisition Parameters
Date_ 20140311
Time 7.50
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 16
DS 2
SWH 8223.685 Hz
FIDRES 0.125483 Hz
AQ 3.9846387 sec
RG 57
DW 60.800 usec
DE 6.50 usec
TE 303.0 K
D1 1.00000000 sec

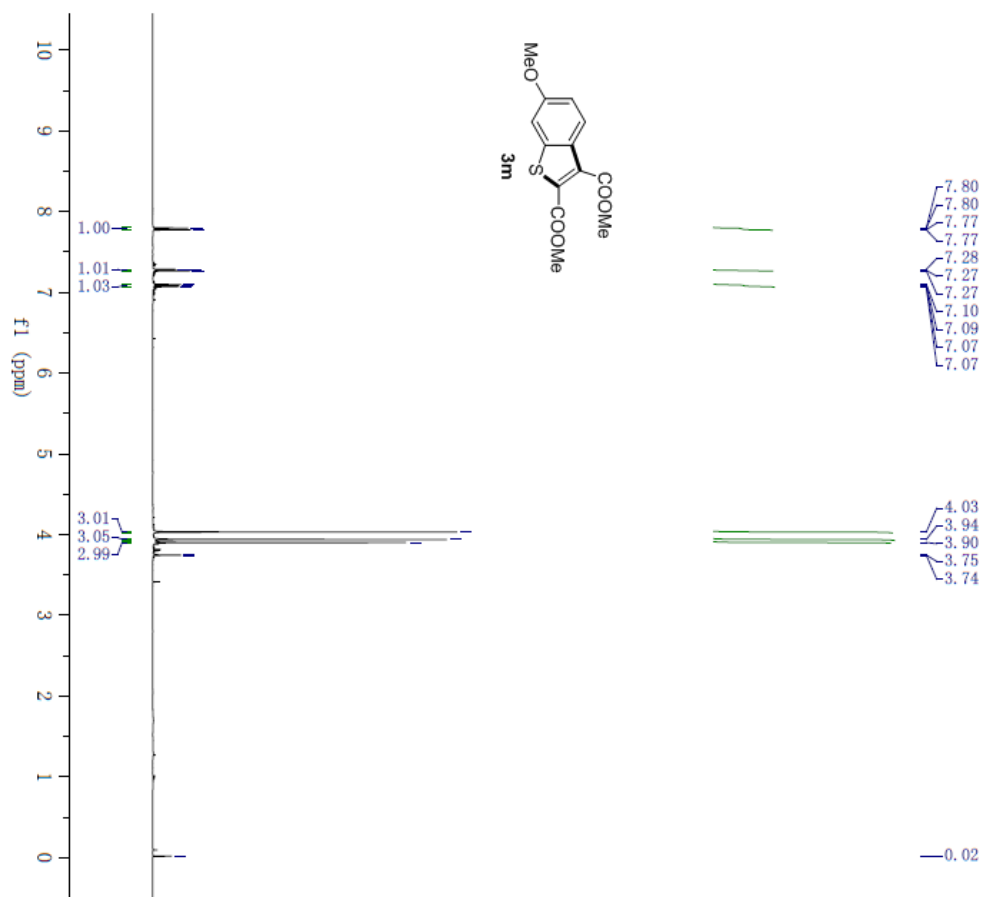
===== CHANNEL f1 =====
NUC1 1H
P1 14.94 usec
PLMT 14.1029369 W
SFO1 400.1324710 MHz
F2 - Processing parameters
SI 65536
SF 400.1300000 MHz
WDW no
SSB 0
LB 0 Hz
GB 0
PC 1.00

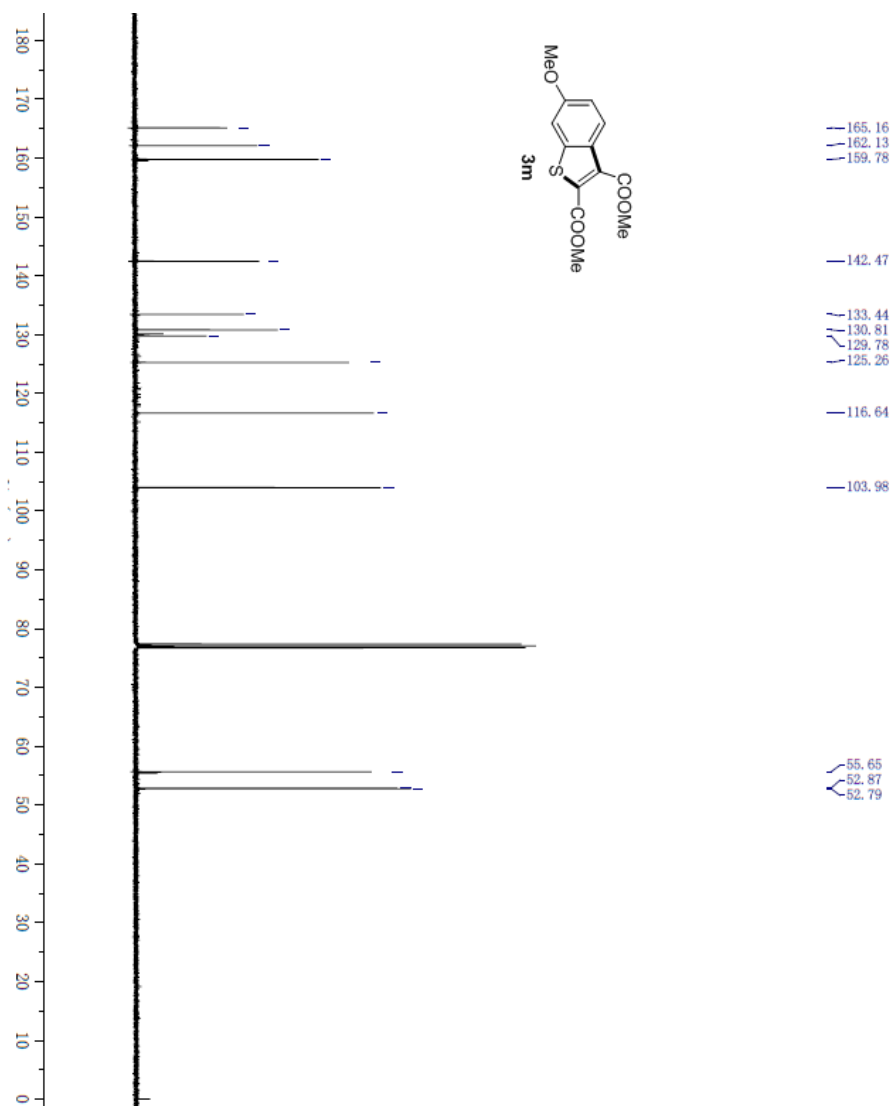


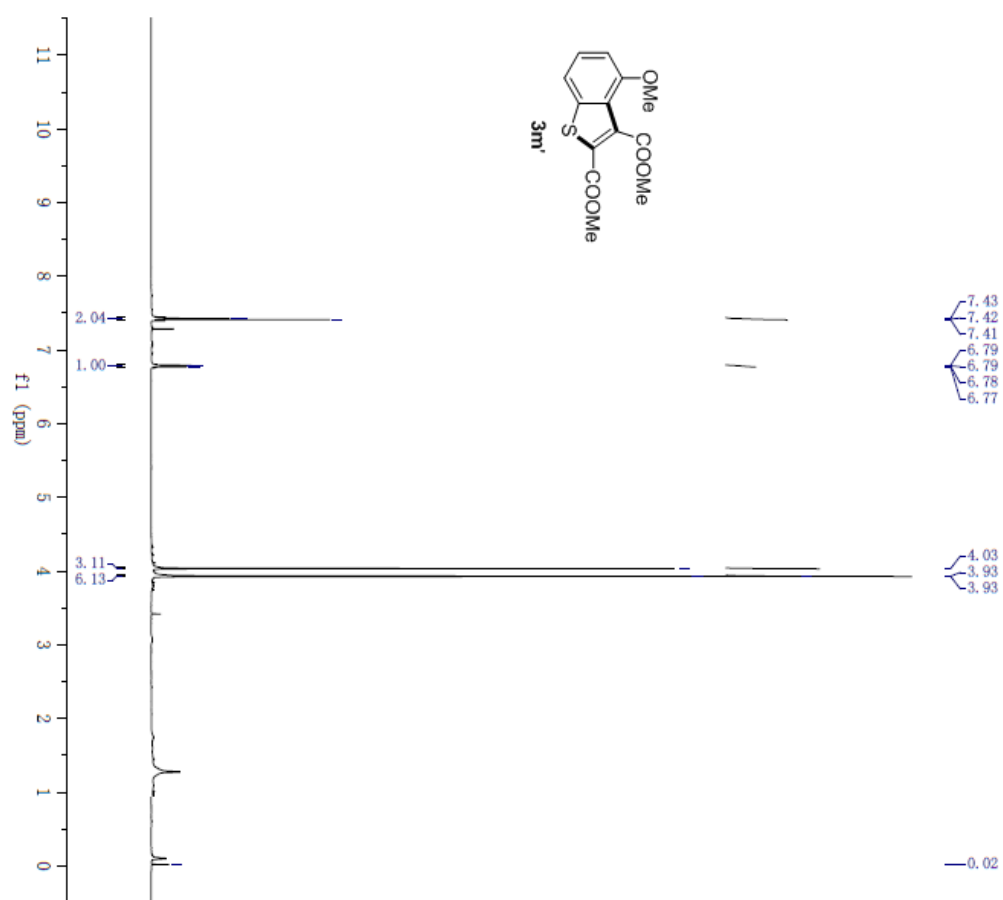


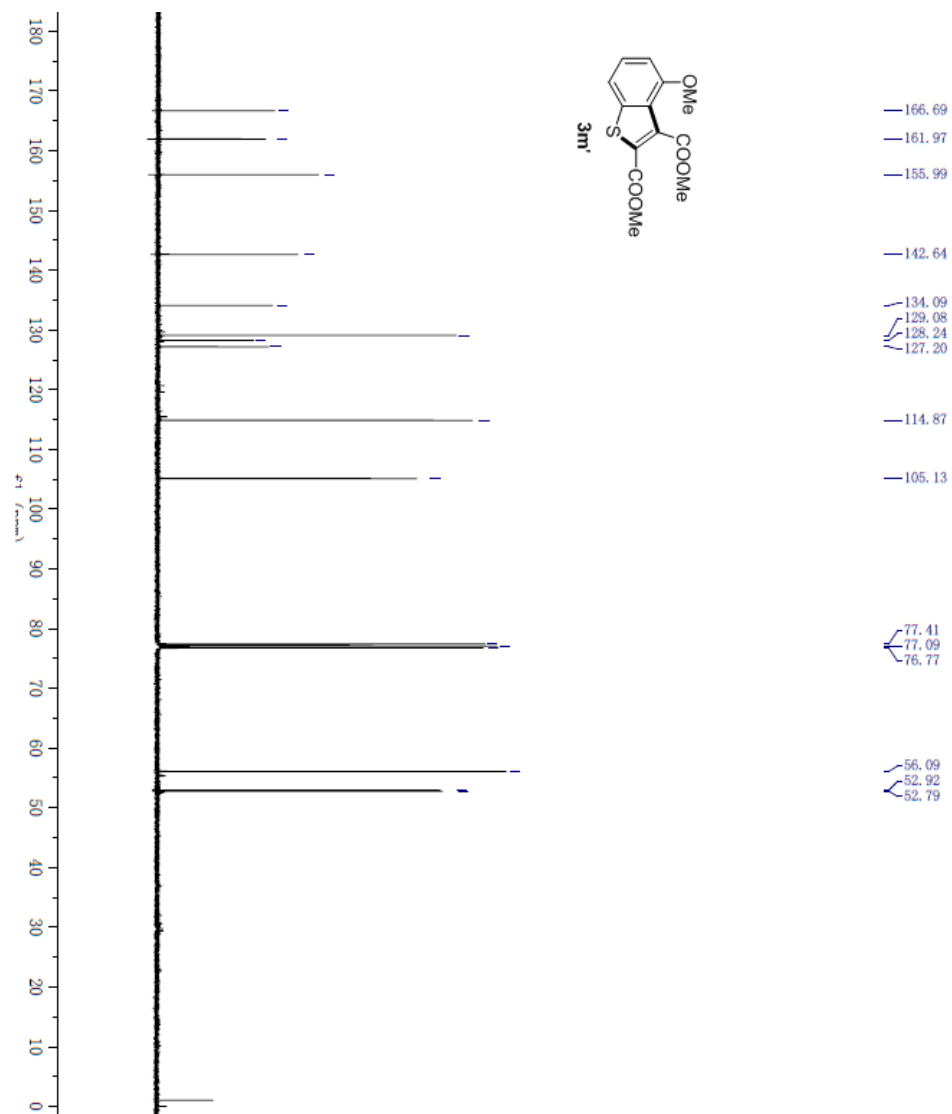












nmr1

7.801
7.779
7.285
7.274
7.268
7.098
7.092
7.076
7.070
4.532
4.514
4.497
4.479
4.430
4.412
4.394
4.377
3.904
1.468
1.450
1.433
1.424
1.406
1.388
1.280



Current Data Parameters
NAME ykl-15a
EXPNO 10
PROCNO 1

F2 - Acquisition Parameters

Date_ 20140311
Time 4.53
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 16
DS 2
SWH 8223.685 Hz
FIDRES 0.125483 Hz
AQ 3.9846387 sec
RG 128
RG 60.800 usec
DE 6.50 usec
TE 303.0 K
D1 1.00000000 sec

===== CHANNEL f1 =====
NUC1 1H
P1 14.94 usec
PLW1 14.1029969 W
SFO1 400.1324710 MHz

F2 - Processing Parameters
SI 65536
SF 400.1300000 MHz
WDW no
SSB 0
LB 0 Hz
GB 0
PC 1.00

