

Preparation and characterization of flame retardant polyurethane foams containing phosphorus-nitrogen-functionalized lignin

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Supporting Information

List of Figures and Tables

Fig. S1 Mass spectra of gas products during the thermal degradation of PU

Fig. S2 Mass spectra of gas products during the thermal degradation of PU-LPMC₁₅

Table S1 Identification of gas products during the degradation of PU.

Table S2 Identification of gas products during the degradation of PU-LPMC₁₅.

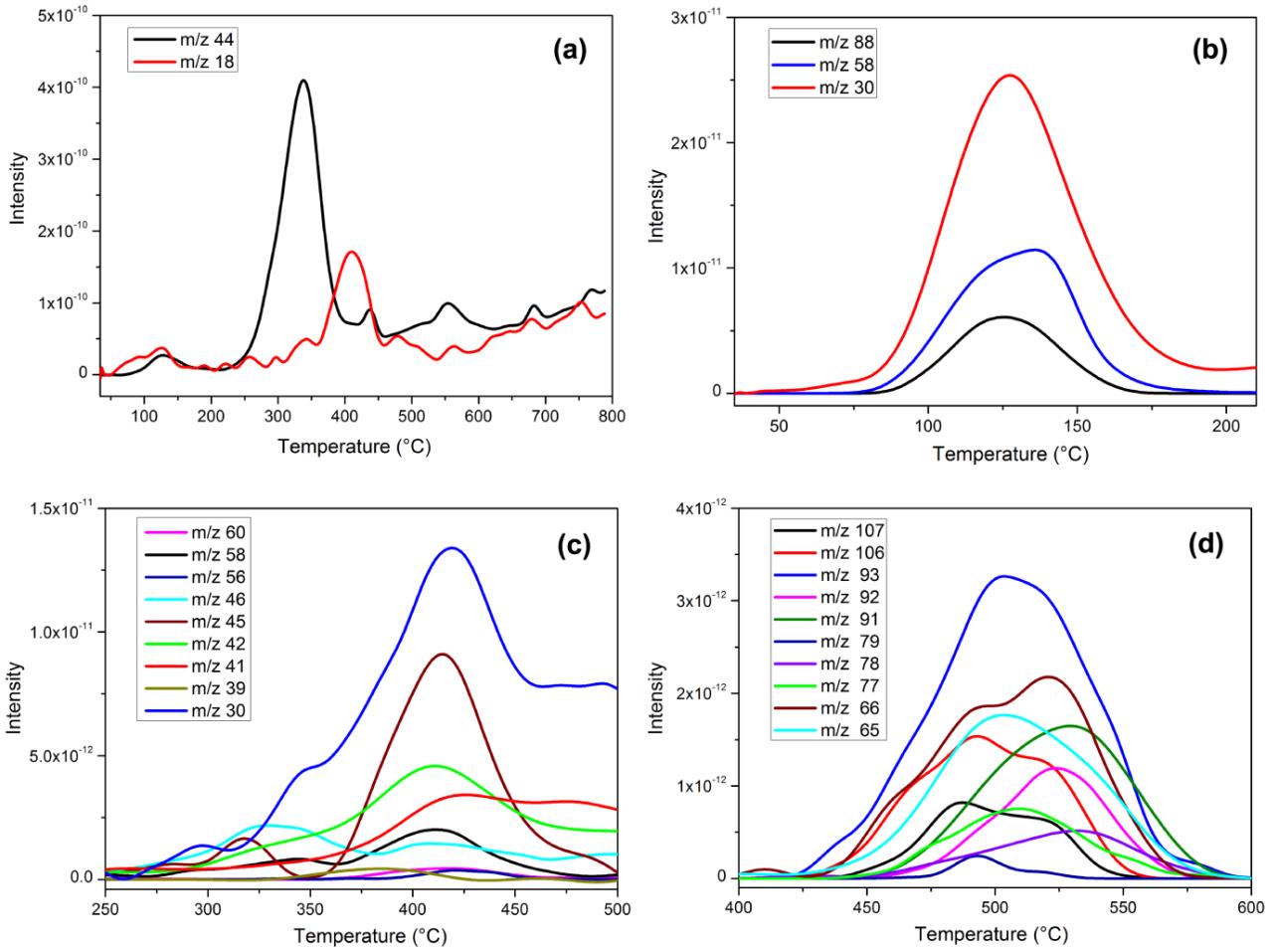


Fig. S1 Mass spectra of gas products during the thermal degradation of PU

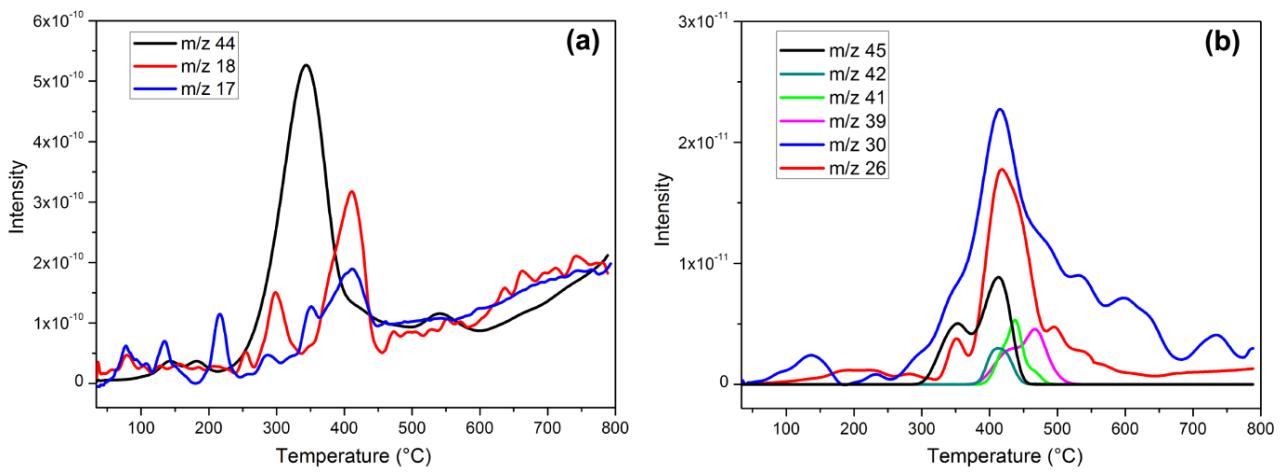


Fig. S2 Mass spectra of gas products during the thermal degradation of PU-LPMC₁₅

Table S1 Identification of gas products during the degradation of PU.

Temperature (°C)	Products	^a FTIR (cm ⁻¹)	Mass (m/z)
126.5		2968, 2864, 1456, 1369 (C-H in CH ₂) 1261, 1126 (C-O-C) 881 (cyclic C-O-C)	88, 58, 43, 30 (Fig. S1b)
290-450	H ₂ O	3500-4000	18 (Fig. S1a)
	CO ₂	2312, 2359, 670	44 (Fig. S1a)
	CH ₃ -CH ₃	2955, 2868 (C-H in CH ₃)	30 (Fig. S1c)
		2955, 2868 (C-H in CH ₂ or CH ₃) 1589 (C=C) 1126 (C-O-C) 881 (cyclic C-O-C)	58 (Fig. S1c)
		2955, 2868 (C-H in CH ₂ or CH ₃) 1126 (C-O-C)	60, 46, 45 (Fig. S1c)
		2955, 2868 (C-H in CH ₂ or CH ₃) 1589 (C=C)	56, 42, 41, 39 (Fig. S1c)
450-570		2935 (C-H in CH ₃), 1622(N-H) 1516 (aromatic ring) 744 (N-H)	107, 106, 79, 77, 65 (Fig. S1d)
		1622(N-H) 1516 (aromatic ring) 744 (N-H)	93, 66, 65 (Fig. S1d)
		2955, 2868 (C-H in CH ₂ or CH ₃) 1516 (aromatic ring)	106, 91 (Fig. S1d)
		2955, 2868 (C-H in CH ₂ or CH ₃) 1516 (aromatic ring)	92, 91, 65 (Fig. S1d)
		1516 (aromatic ring)	78, 77 (Fig. S1d)

^a The absorption peaks of FTIR are shown in Fig. 5 in the paper.

Table S2 Identification of gas products during the degradation of PU-LPMC₁₅.

Temperature (°C)	Products	^a FTIR (cm ⁻¹)	Mass (m/z)
126.5		2966, 2864 (C-H in CH ₂) 1263, 1126 (C-O-C) 881 (cyclic C-O-C)	
290-570	H ₂ O	3500-4000	18 (Fig. S2a)
	CO ₂	2312, 2359, 670	44 (Fig. S2a)
	NH ₃		17 (Fig. S2a)
	CH ₃ -CH ₃	2935, 2881 (C-H in CH ₃)	30 (Fig. S2b)
	CH≡CH		26 (Fig. S2b)
	~NH ₂ or ~ 	2935, 2881 (C-H in CH ₂ or CH ₃) 1622 (N-H)	45, 30 (Fig. S2b)

^a The absorption peaks of FTIR are shown in Fig. 6 in the paper.

References

- (1) L. Jiao, H. Xiao, Q. Wang and J. Sun, *Polym. Degrad. Stab.* 2013, **98**, 2687-2696.
- (2) X. Chen, L. Huo, C. Jiao and S. Li, *J. Anal. Appl. Pyrol.* 2013, **100**, 186-191.
- (3) K. Pielichowski and D. Slotwinska, *Thermochim. Acta* 2004, **410**, 79-86.