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Supporting information

6 Nanorod-like α -Bi₂O₃ nanoparticles: a highly active photocatalyst 7 synthesized by using g-C₃N₄ as a template

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1 Sample Characterization:

2 The microstructures of the samples were observed by field emission microscope (FESEM, JEOL 6701F). Transmission electron
3 microscopy images were recorded with a field emission transmission electron microscope (2100F, JEOL Co., Japan). A surface-area
4 analyzer (BEL Sorp-mini II, BEL Japan Co., Japan) was used to measure the surface areas of samples through nitrogen adsorption and
5 desorption isotherms at 77K. The crystal structures of the samples were determined with an X-ray diffractometer (X'pert Powder,
6 PANalytical B. V., Netherlands) with Cu-K α radiation. Fourier transform infrared (FT-IR) spectra were obtained with an IRPrestige-21
7 FTIR (Shimadzu) spectrophotometer. The optical properties of the samples were measured by UV-vis diffuse reflectance spectroscopy
8 method by using UV-2600 (Shimadzu Co., Japan) spectrophotometer equipped, with BaSO₄ as the reflectance standard, and the optical
9 absorptions were converted from the reflection spectra according to Kubelka-Munk equation. Thermogravimetric-differential thermal
10 analysis (TG-DTA; Shimadzu, DTG-60H, Japan) was used to study the formation mechanism of the nanorod-like α -Bi₂O₃.

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13 Photocatalytic activity measurements:

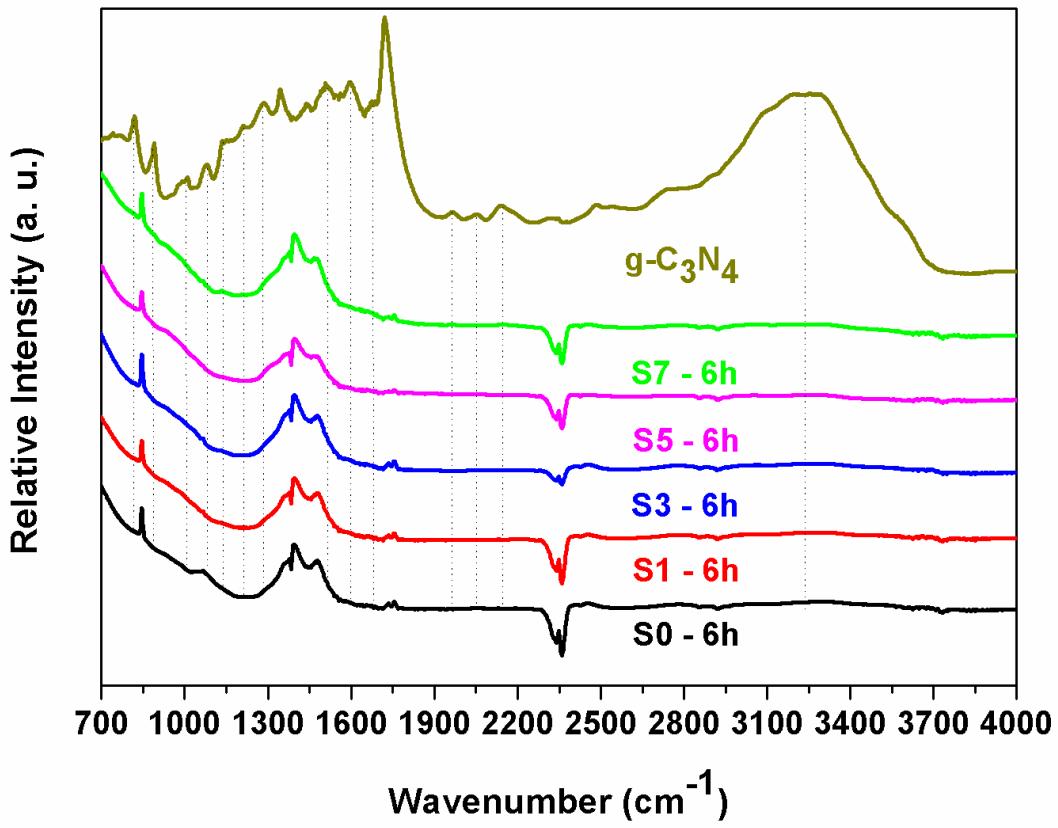
14 The photocatalytic oxidations of gaseous IPA were carried out under the visible light with the wavelengths of $420 \text{ nm} \leq \lambda \leq 800 \text{ nm}$. The
15 light source was a 300 W Xe-arc lamp (10A imported current, focused through a 50×50mm shutter window) equipped with some
16 wavelength cutoff filters and a water filter. Typically, 100 mg photocatalyst was bespread uniformly on a glass dish with an area of 9 cm².
17 A certain amount (~1400 ppm) of gaseous IPA was injected into the vessel and kept for 2 h in the dark before irradiation. During the
18 irradiation by visible light, 0.5 ml of the gas was sampled everyone 1 h intervals. The products were analyzed with a gas chromatograph
19 (GC-2014, Shimadzu, Japan) with a flame ionization detector (FID).

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2 **Figure S 1** The FT-IR spectra of all the α -Bi₂O₃ samples induced by g-C₃N₄ with 6 hours' calcinations
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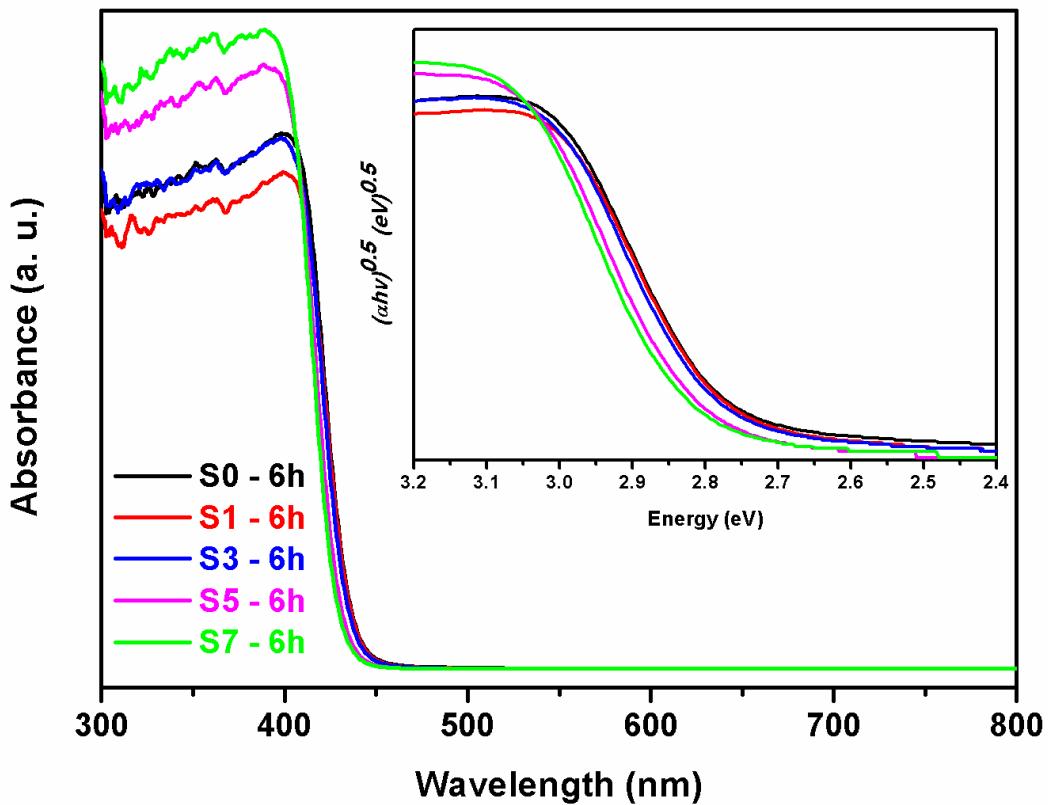
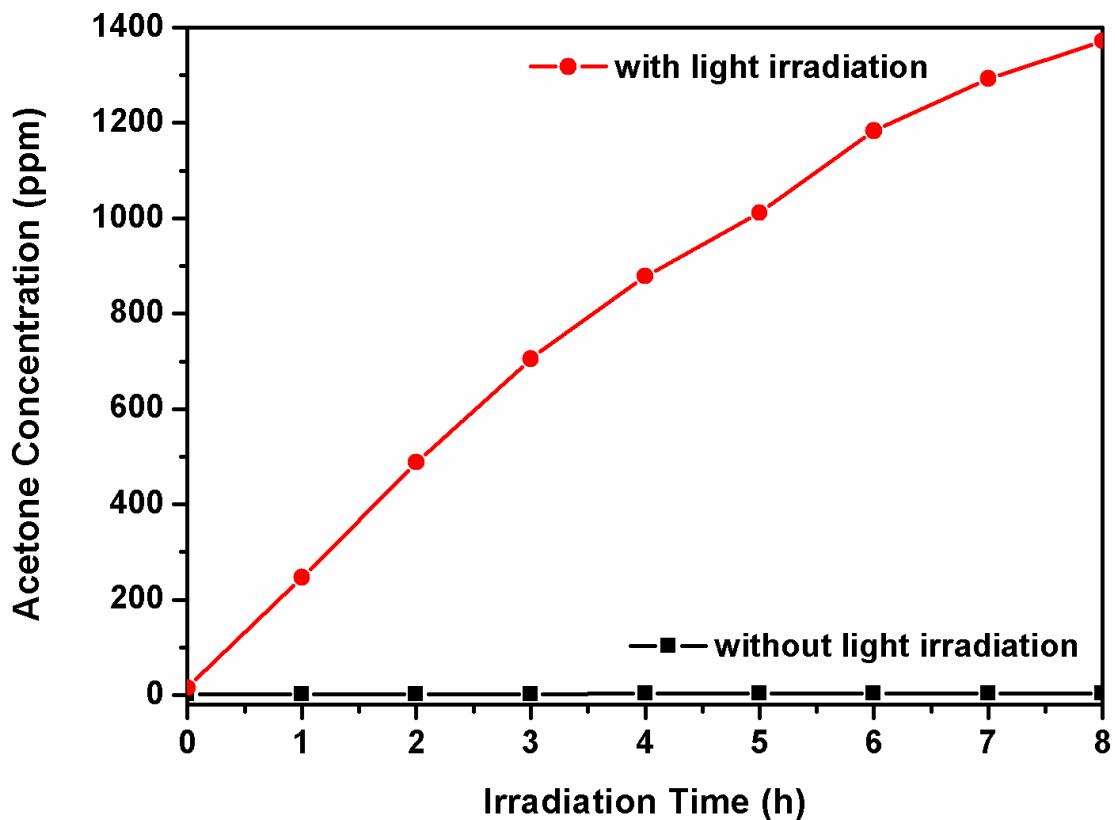


Figure S 2 The UV-vis DRS spectra of all the α -Bi₂O₃ samples induced by g-C₃N₄ with 6 hours' calcinations



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2 **Figure S 3** Photocatalytic oxidations of gaseous IPA over S7 - 6h with and without visible light irradiation.

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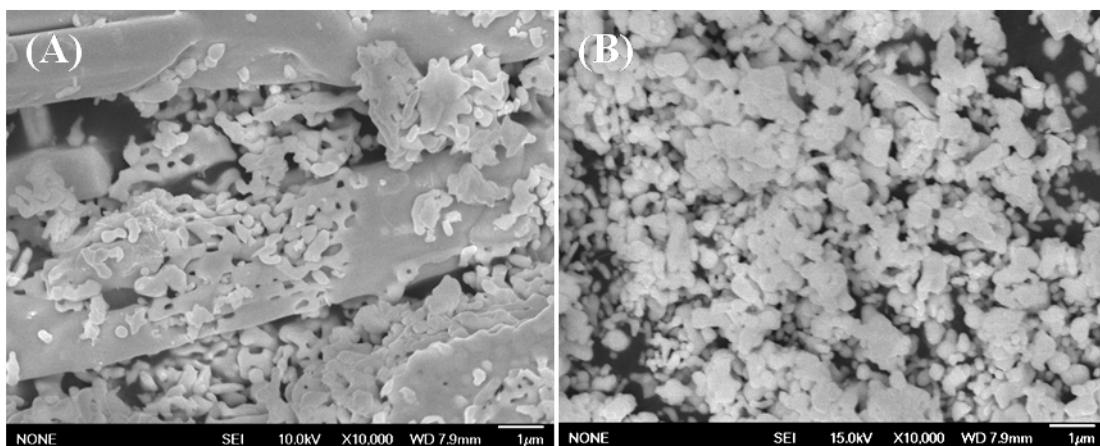
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5 **Figure S 4** The SEM spectra of the α - Bi₂O₃ by using the (NH₄)₂CO₃ [inorganics (A)] and citric acid [organics (B)] as the template to treat
6 the microrod-like α -Bi₂O₃