## **Supporting Information for**

## Controlled RAFT Synthesis of Side-Chain Oleic Acid Containing Polymers and Their Post-Polymerization Functionalization

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Fig. S1 FT-IR spectrum of MAEO.



Fig. S2 ESI-MS spectrum of MAEO (calculated m/z for [M + Na<sup>+</sup>]: 417.308 and observed: 417.309).



Fig. S3 First-order kinetic plot for the RAFT polymerization of MAEO (A) and the corresponding  $M_n$  versus monomer conversion plot (B). Polymerization reaction was carried out at [MAEO]/[CDP]/[AIBN] = 25:1:0.2 in THF at 60 °C.

Solvent	MAEO	PMAEO
Water	-	-
Acetone	+	+
Chloroform	+	+
DCM	+	+
Methanol	+	-
Ethanol	+	-
DMF	+	-
DMSO	+	-
THF	+	+
Pet ether	+	+
Diethyl ether	+	+
Ethyl acetate	+	+
Hexanes	+	+
Toluene	+	+
Acetonitrile	+	-

Table S1 Solubility of MAEO and PMAEO at room temperature in different solvents.

The symbols (+) and (-) indicate soluble and insoluble, respectively.



**Fig. S4** MALDI-TOF spectrum of PMAEO ( $M_{n,GPC}$  = 3000 g/mol).



**Fig. S5** <sup>1</sup>H NMR spectra of PMAEO-*b*-PMMA (A) and PMAEO-*b*-PMMA in epoxide form (B). The \* indicates CDCl<sub>3</sub>.



Fig. S6 <sup>1</sup>H NMR spectra of PMAEO-*b*-PPEGMA.



**Fig. S7.** Alkene bond conversion versus ratios of [butanethiol]/[MAEO repeating unit] plot for PMAEO reacting with butanethiol at a constant [MAEO repeating unit]/[AIBN] = 1:1 in THF at 60 °C (reaction time = 12 h). Conversion values were calculated by <sup>1</sup>H NMR spectroscopy based on the consumption of the alkene bonds.



**Fig. S8** <sup>1</sup>H NMR spectra of products from thiol-ene reaction of PMAEO with (A) 3mercaptopropanoic acid, (B) butanethiol and (C) dodecanethiol. The inset corresponds to -COO*H* peak.



**Scheme S1** Post-polymerization modifications of the double bonds in PMAEO *via* epoxidation reaction at [MAEO repeating unit]/[mCPBA] = 2:1 ratio.



**Fig. S9** <sup>1</sup>H NMR spectrum of epoxide of PMAEO obtained from the reaction at [MAEO repeating unit]/[mCPBA] = 2:1 ratio.